

**SEDIMENT CHARACTERISTICS AROUND THE KENYON  
ISLAND GROUP, GREAT SACANDAGA LAKE (NY): ECONOMIC  
POTENTIAL OF DREDGING AND LAND RECLAMATION**

A thesis presented to the Faculty  
of the University at Albany, State University of New York  
in partial fulfillment of the requirements  
for the degree of  
Master of Science  
College of Arts and Sciences  
Department of Earth and Atmospheric Sciences

**Arthur M. Ambrosino**  
2001

University at Albany, State University of New York

College of Arts and Sciences

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The thesis for the master's degree submitted by

**Arthur M. Ambrosino**

under the title

**SEDIMENT CHARACTERISTICS AROUND THE KENYON  
ISLAND GROUP, GREAT SACANDAGA LAKE (NY): ECONOMIC  
POTENTIAL OF DREDGING AND LAND RECLAMATION**

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## ABSTRACT

The Kenyon Island Group lies within the Great Sacandaga Lake's largest and most unnavigable shallow water shoal. The shoal measures approximately  $5.0 \text{ mi}^2$ . The Kenyon Islands, including Mead and Deer Islands, occupy approximately  $\frac{3}{4} \text{ mi}^2$  during the months of annual high lake level. During the late shallow water season of 1998, a NE to SW diagonal transect of nine core samples were collected by a Geoprobe coring device, with recoveries between 8' and 16' deep sections. In this study these cores were used to identify the stratigraphy, classify the sediment grain sizes, evaluate the abundance of economic minerals, and calculate the most cost effective and environmentally sound method of deepening the lakebed for navigation and recreational purposes and reconfiguring the new real property created.

Here I present evidence for the existence of a previously unidentified large moraine field, of approximately  $12 \text{ mi}^2$ , located between two different end moraines of the Pleistocene Epoch. Today, about  $\frac{1}{3}$  of that moraine field surrounds the Kenyon Island Group, within the shorelines of the Great Sacandaga Lake. The south and east end moraine is visible in an arcuate path which includes the two major Kenyon Islands and Deer Island. The south and east end moraine is approximately  $\frac{3}{4}$  mile wide and is confined by the valley walls of the Sacandaga Basin. This end moraine must have been at least 300 feet high during the Pleistocene Epoch. The north and west moraine was deposited before the south and east end moraine and is also visible in exposures around the lake. This forms moraine rock fields in arcuate bands, which are confined by the valley walls of the Sacandaga Basin. The north and west end moraine was originally

½ miles wide and must have stood at least 120 feet high.

The moraine field contains all the classic landmarks including kames, kettle lakes, eskers, drumlins, flutes and fossil streambeds. The moraine field sediments that exist are no more than 45 feet thick beneath the lakebed and 45 to 120 feet thick outside the shorelines and above bedrock. The bedrock may be Cambrian (570-510 Ma), including Little Falls Dolomite, Theresa Dolomite and/or Potsdam Sandstone.

This thesis proposes a remedy involving dredging and earth-moving heavy equipment to permanently deepen part of the lakebed of the Great Sacandaga Lake that is currently of little use because the area is typically too shallow for regular lake navigation. If there are minerals within the sediments of the study area that are of current economic importance and have values significant enough to pay for a lakebed deepening, the geochemistry, mineralogy and mineral chemistry will reveal them. These minerals can then be identified and evaluated as the mechanism that will support the tremendous costs associated with such a major deepening effort. The long term outcome of a Great Sacandaga Lake deepening, beside improving the navigational, recreational, and fisheries of the lake will create room for an additional 200 billion gallons of water, which could be priceless in the next frontier.

The moraine field sediment profile in the study area may be unique to the whole of the Great Sacandaga Lake. The question of dredging the lakebed versus simpler earth-moving heavy equipment to reconfigure the lakebed, anywhere else in the Great Sacandaga Lake, will most certainly require further research and could yield different results.

## **Acknowledgements**

I would particularly like to thank Dr. John I. Garver for all the effort he extended this project. He gave of himself and the splendid Union College facilities, far more than would have been expected and his insight and counsel was greatly appreciated.

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# CHAPTER 1

## INTRODUCTION

### *Purpose of the study*

During the last several decades, there has been an increasing dispute between the downstream “interests” and the “recreational users” of the water of the Great Sacandaga Lake in upstate New York. The downstream “interests” include power generation dams, Hudson River water levels, ocean water/salt front concerns at Poughkeepsie and flood control at Albany, among others.

Large annual water-level variations to accommodate the downstream interests cause large areas of very shallow water in the Great Sacandaga Lake (GSL), especially in midsummer, which is a situation the landowners and boating public find unacceptable. In an effort to reach accommodations to all, an Upper Hudson-Sacandaga Comprehensive Settlement Negotiation Team is presently studying water level remedies (See Appendix A-Permits and History). They have proposed raising the GSL high water mark from 768' to 770' over a twenty-year time span. Any remedy they agree upon will be a compromise between upstream and downstream interests.

This thesis proposes a remedy involving dredging and earth-moving heavy equipment to permanently deepen part of the lakebed of the Great Sacandaga Lake that is currently of little use because the area is typically too shallow for regular lake navigation. To accomplish this task it is necessary to identify the stratigraphy of the lakebed sediments at various points within the lake, classify the sediment grain sizes,

evaluate the potential economic mineral wealth, and determine the most cost effective and environmentally sound method of deepening the lakebed and reconfiguring existing Islands and shorelines.

The Kenyon Island Group, perhaps the largest and most unnavigable shallow water shoal in the Great Sacandaga Lake, is the ideal location for investigating such a project. The Kenyon Island Group contains the former and now submerged community of Osborne Bridge. To address the above issues a transect of core samples were collected by Geoprobe in the Kenyon Island Group using gravel roads which were serviceable during the late shallow water season of 1998. Nine Geoprobe 1 3/4" diameter cores were collected, between 8' and 16' in length.

### ***Location of study area***

The Kenyon Island Group is located in the north central portion of the Great Sacandaga Lake, in the town of Northampton, in northeast Fulton County, New York State (Figure 1.1). The Kenyon Island Group is just over two miles south of Northville, New York. Some portions of the Great Sacandaga Lake are located in Northwest Saratoga County, but are outside this project review. The whole of the lake lies within the perimeter ("Blue Line") of the Adirondack State Park (Figures 1.2).

The 1908 New York State Water Supply Commission Maps, numbers 20 and 21, were digitized, merged and enhanced to show the GPS and coring sites (Figure 1.3). These maps depict the area before the Conklingville Dam was built in 1929, which subsequently flooded the area creating the Sacandaga Reservoir.



**Figure 1.1 The Kenyon Island Group, Great Sacandaga Lake Navigation Map, Jimapco, 1991.**

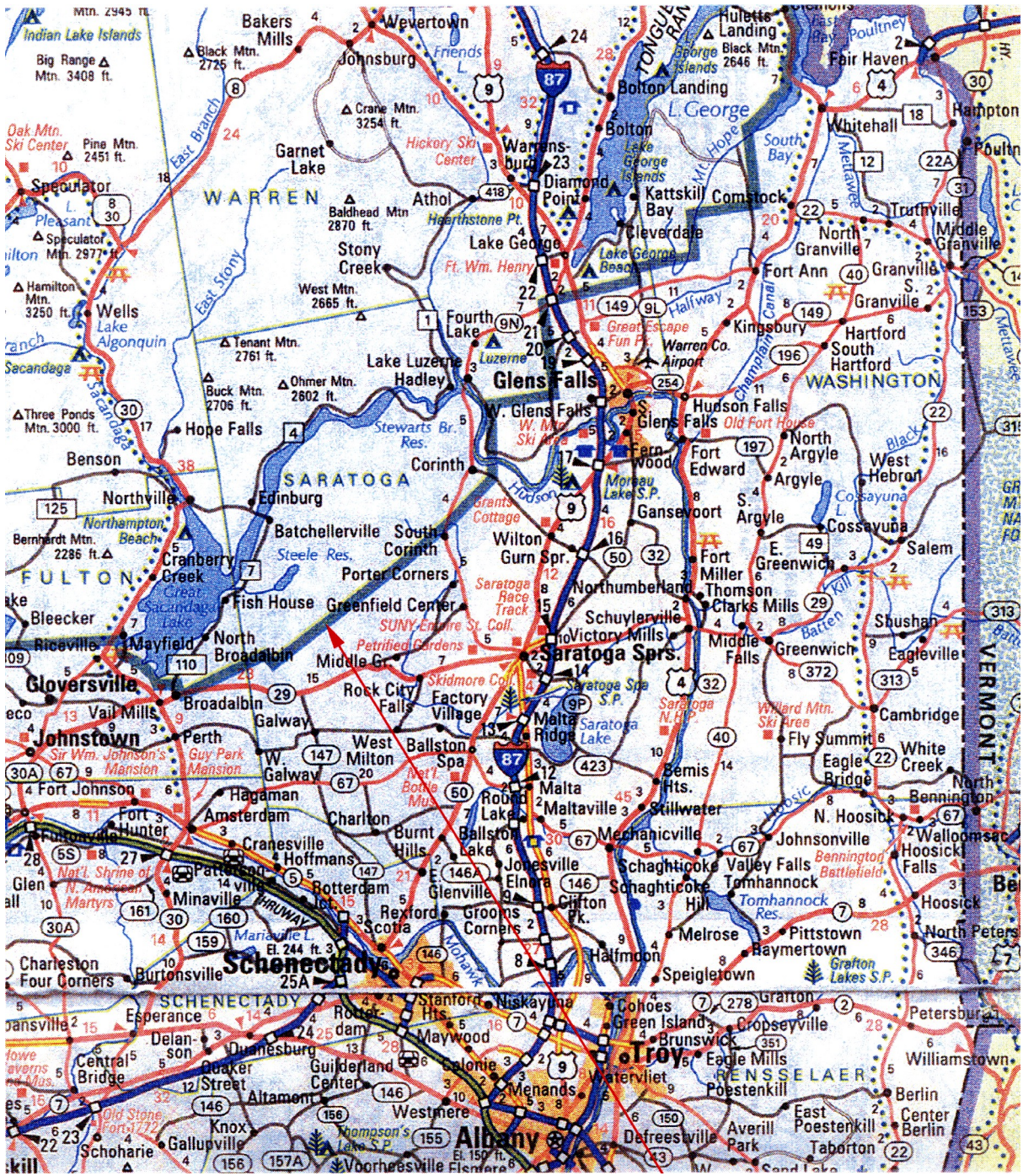


Figure 1.2 The Great Sacandaga Lake lies within the “Blue Line” of the Adirondack State Park, Rand McNally, 1999.

NEW YORK STATE WATER COMMISSION MAPS (20-21) 1908

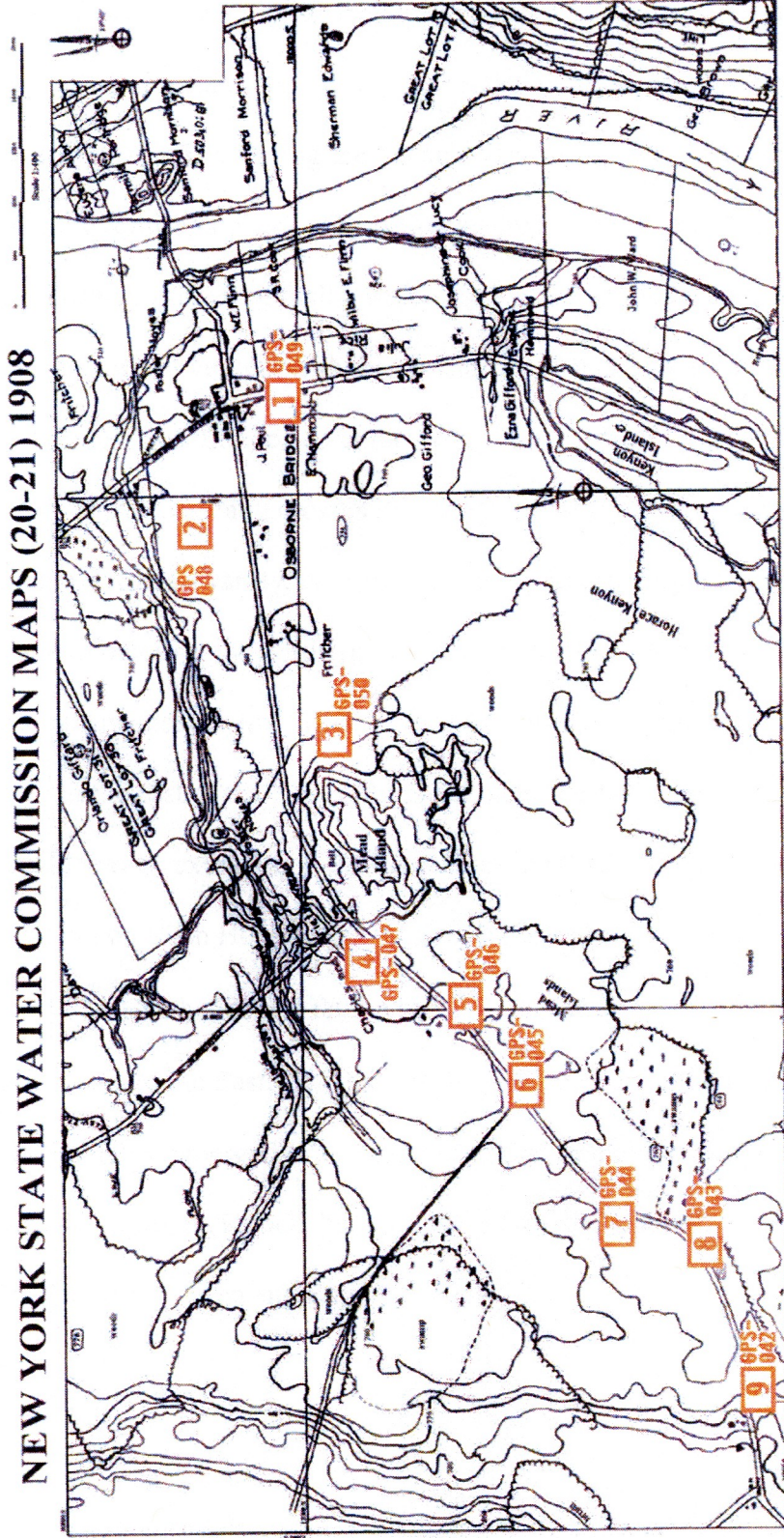


Figure 1.3 The 1908 New York State Water Commission Maps, Numbers 20 and 21, were Digitized, Merged and Enhanced to show the GPS and Coring Sites in red.

## ***Regional geology***

The geologic structure of the Sacandaga Basin lies on a great sunken block between two Precambrian crystalline blocks of the Mohawk Valley fault series. In the lower Paleozoic Era, during the Taconic Orogeny (~ 450 MA), the region was faulted due to flexure (downwarping). The bedrock across the Mohawk Valley was cut by ten or more northeast trending faults, which have greatly influenced modern topography. The faults are almost vertical with the downdropped block to the east (Figure 1.4). One exception is the fault located in the Sacandaga basin study area (Arnou, 1951).

The Noses Fault crosses Fulton County southwest of Johnstown (New York), trending northeastwardly, crossing the county line northwest of Northville. The upthrown block, at the northern most end, is on the west side of the fault. This side forms the southern-most crystalline rock escarpment of the Adirondack Mountains. Less than 6.5 *mi* to the east of the Noses, the Batchellerville fault forms another escarpment of crystalline rock extending from Batchellerville (in Saratoga County) southward past Northampton (Fish House), and then southeastward.

The Batchellerville fault also has a displacement of about 1500' but the upthrown block side is to the East. The Sacandaga Reservoir and all the nearby communities, including the submerged town of Osborne Bridge and the Kenyon Islands lie on the plain of this downdropped block, between these two Precambrian crystalline masses (Figure 1.5), (Fisher, Isachsen and Rickard, 1995).



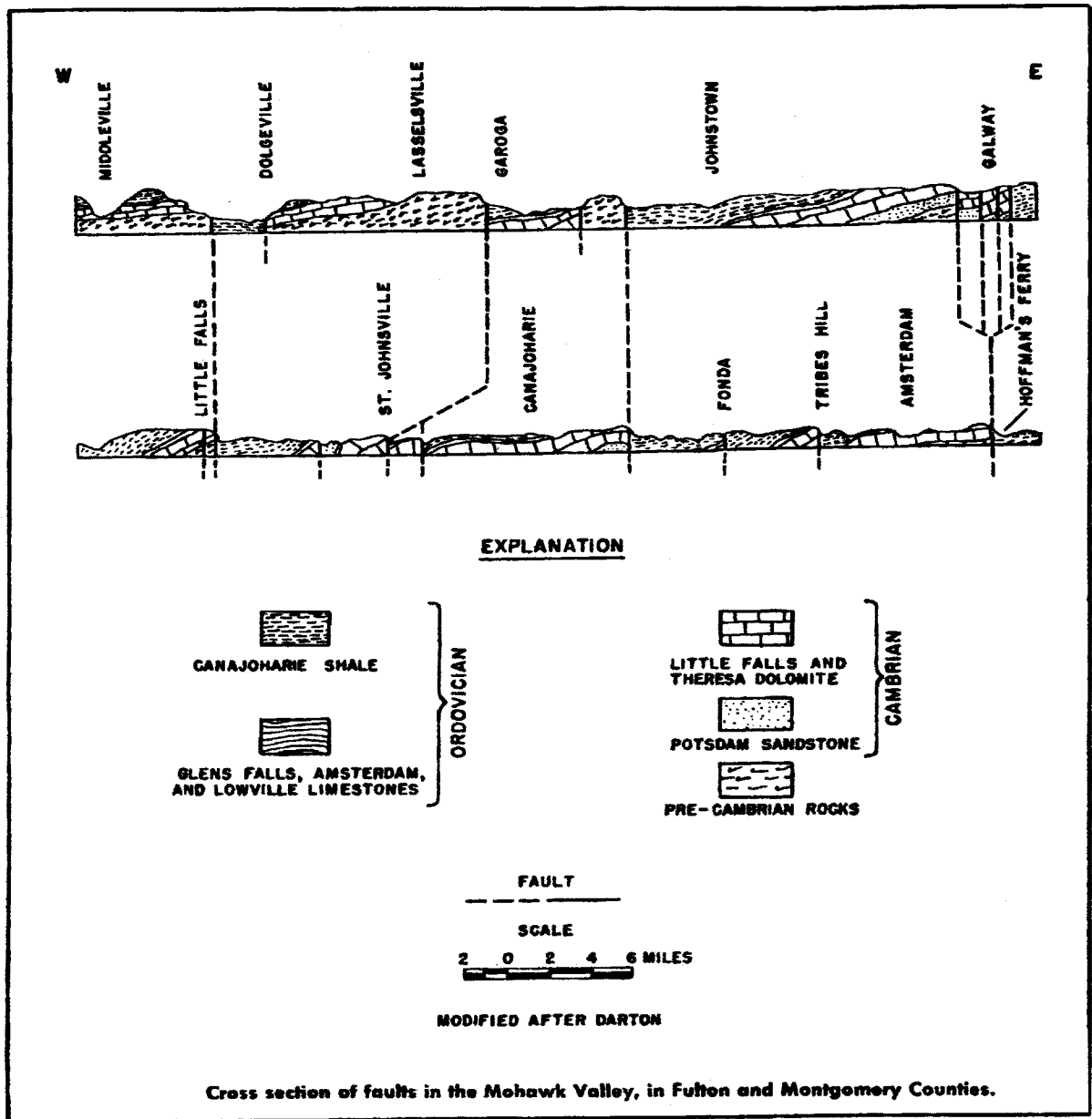
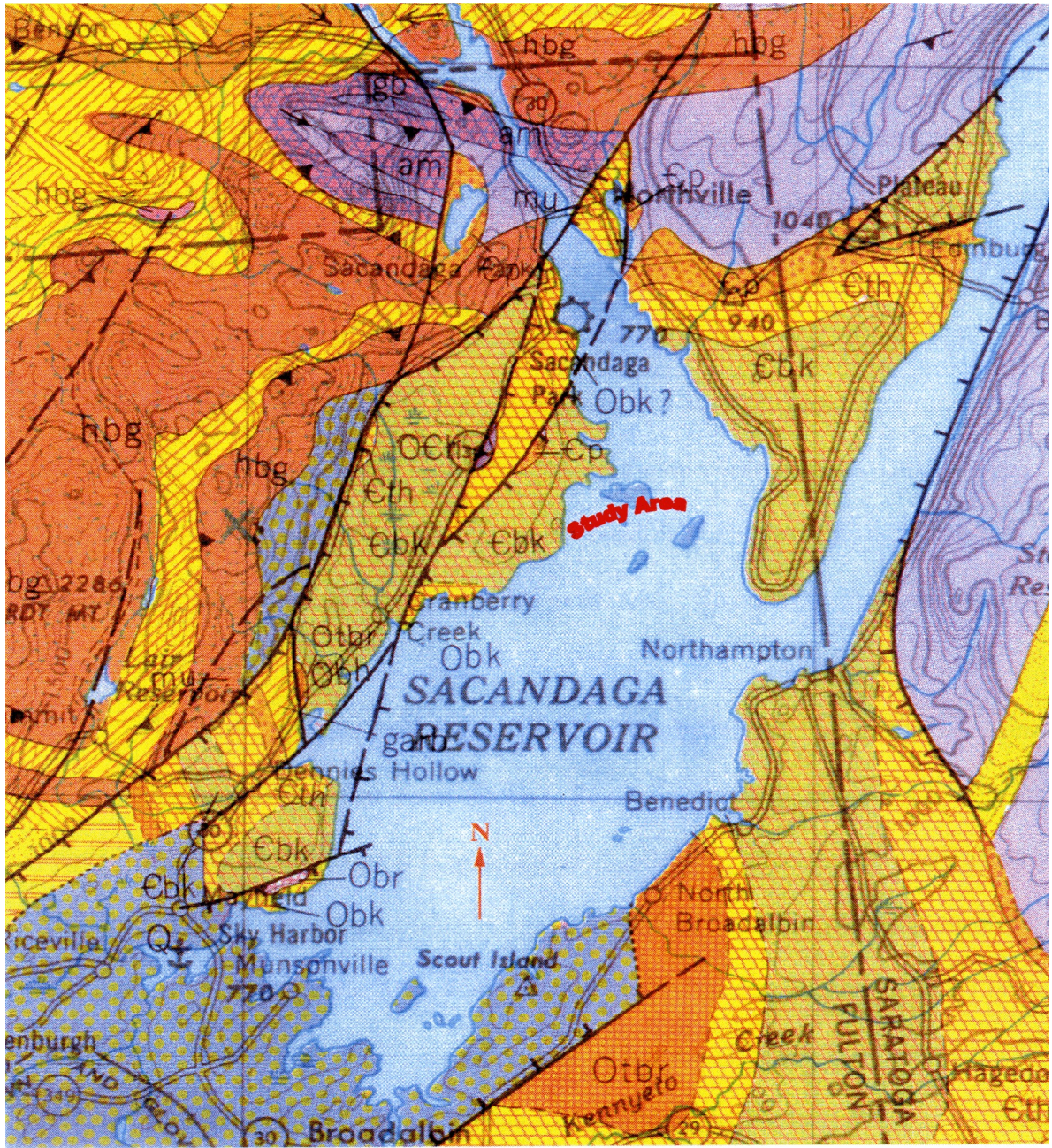


Figure 1.4 The Ground Water Resources of Fulton County, New York, U.S. Geological Survey, Bulletin GW - 24, Arnow, 1951.

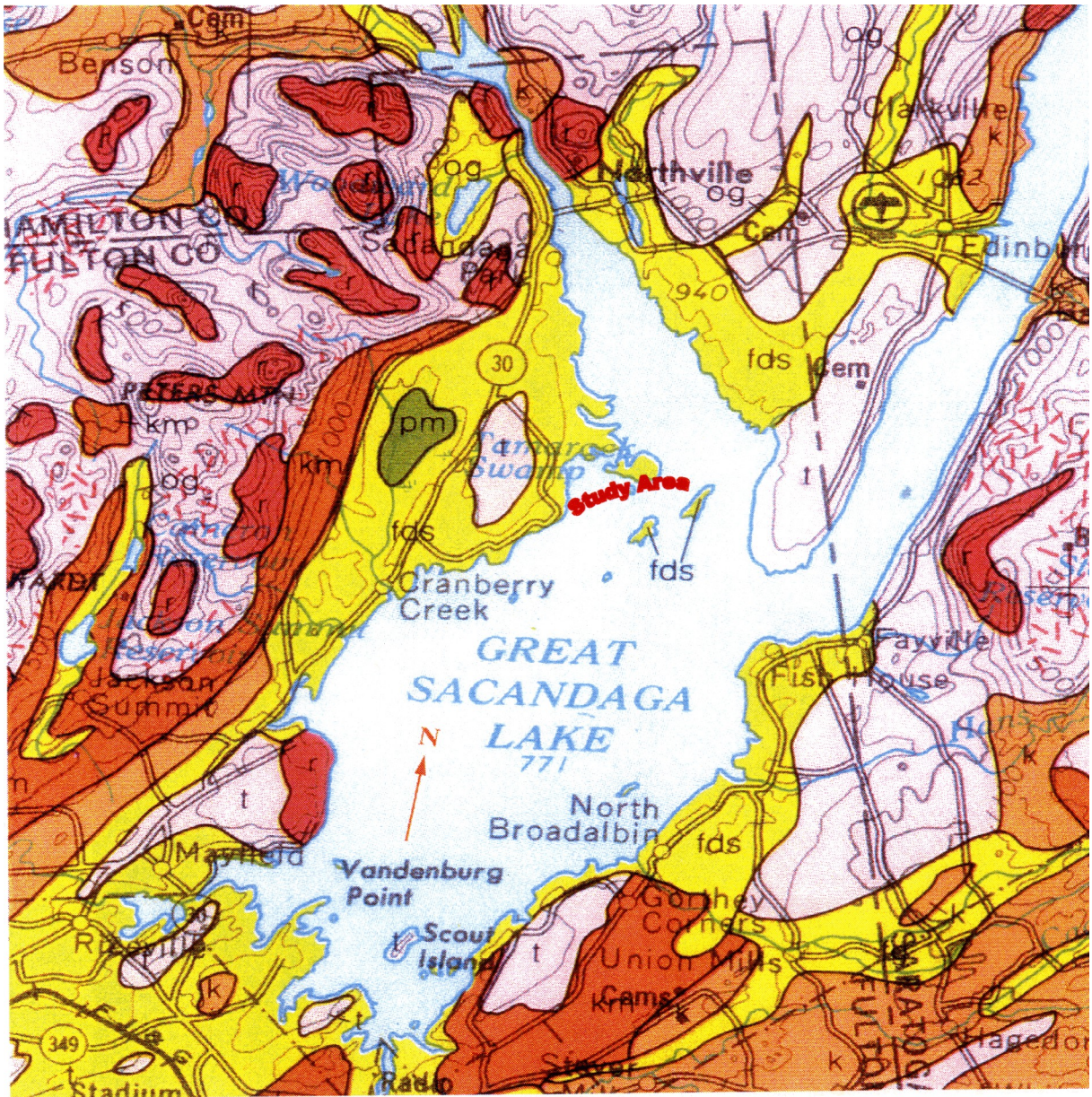


**Figure 1.5**  
**Bedrock Geologic Map of New York State, Hudson-Mohawk**  
**Sheet #15, New York State Museum and Science Service, 1995.**

Cbk	Cbk--Upper Cambrian Beekmantown Group, Tribes Hill Limestone.
Cp	Cp--Cambrian Potsdam Sandstone.
Cth	Cth--Cambrian Theresa Galway Formation Dolostone, Sandstone & Shale.
Obk	Obk--Ordovician Beekmantown Group Tribes Hill, Limestone, Dolostone.
Q	Q--Quaternary-Glacial and Alluvial deposits.

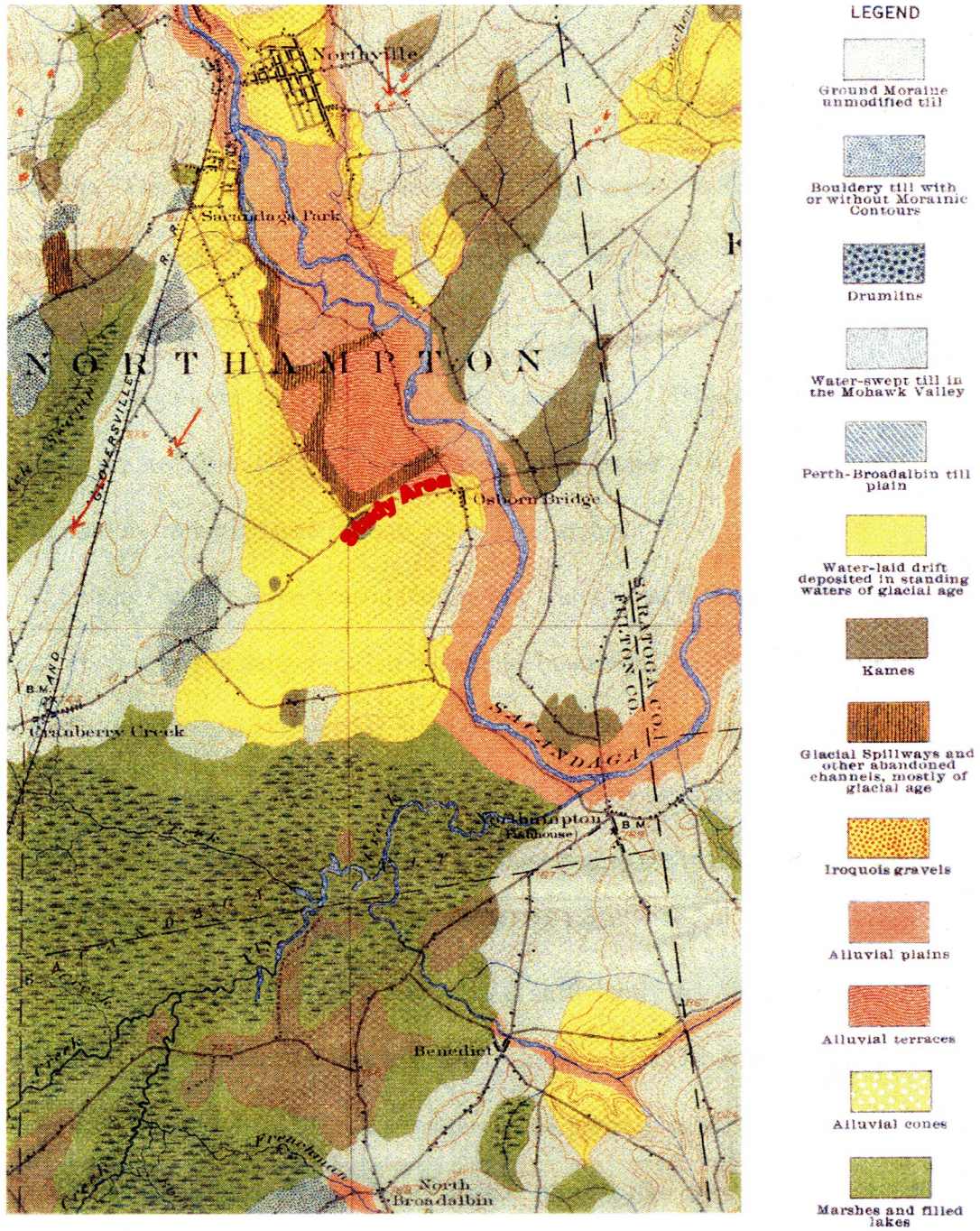
The bedrock of the Sacandaga Basin range from Precambrian crystalline rocks on the east and west escarpments to Cambrian and Ordovician beneath the overburden of the down-dropped block, within which the Kenyon Island Group resides. Perhaps the best characterization of the bedrock comes from the water well logs obtained by the 1951 U.S. Geological Survey of Fulton County Water Resources (Arnow, 1951). The wells that exist within a 2 *mi* circumference suggest that bedrock is from 10-45' beneath the overburden at the Kenyon Island site and no more than 30-120' deep closer to the valley walls of the Sacandaga Basin. Since 1951, legislation has been passed such that water well drilling companies are not required to have bedrock tailings examined and only depth and Health Department requirements exist today. No outcrop exposures are available in the study site area.

The 1987 Surficial Geology Map of New York by Donald H. Cadwell indicates that the whole of the Kenyon Island Group is fluvial deltaic sand. Some glacial tills form boundaries existing above the fluvial deltaic sand and north of Northville where outwash sands and gravels exist (Figure 1.6). In 1911 The University of the State of New York, State Museum, Topographic Map, Bulletin #153, by Miller, describes the Kenyon Islands landscape as Quaternary glacial and postglacial sediments. In 1929 The University of the State of New York, State Museum, Glacial/Geographic Map, Bulletin #280, by Brigham, describes the sediments of the Kenyon Island Group and Osborne Bridge as water laid drift (Figure 1.7).



**Figure 1.6**  
**Surficial Geologic Map of New York State, Hudson-Mohawk**  
**Sheet #40, New York State Museum and Science Service, 1987.**

- t--till
- fds--fluvial deltaic sand



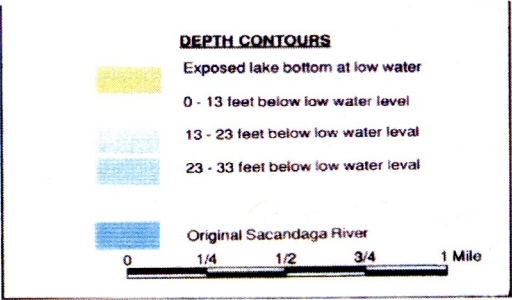
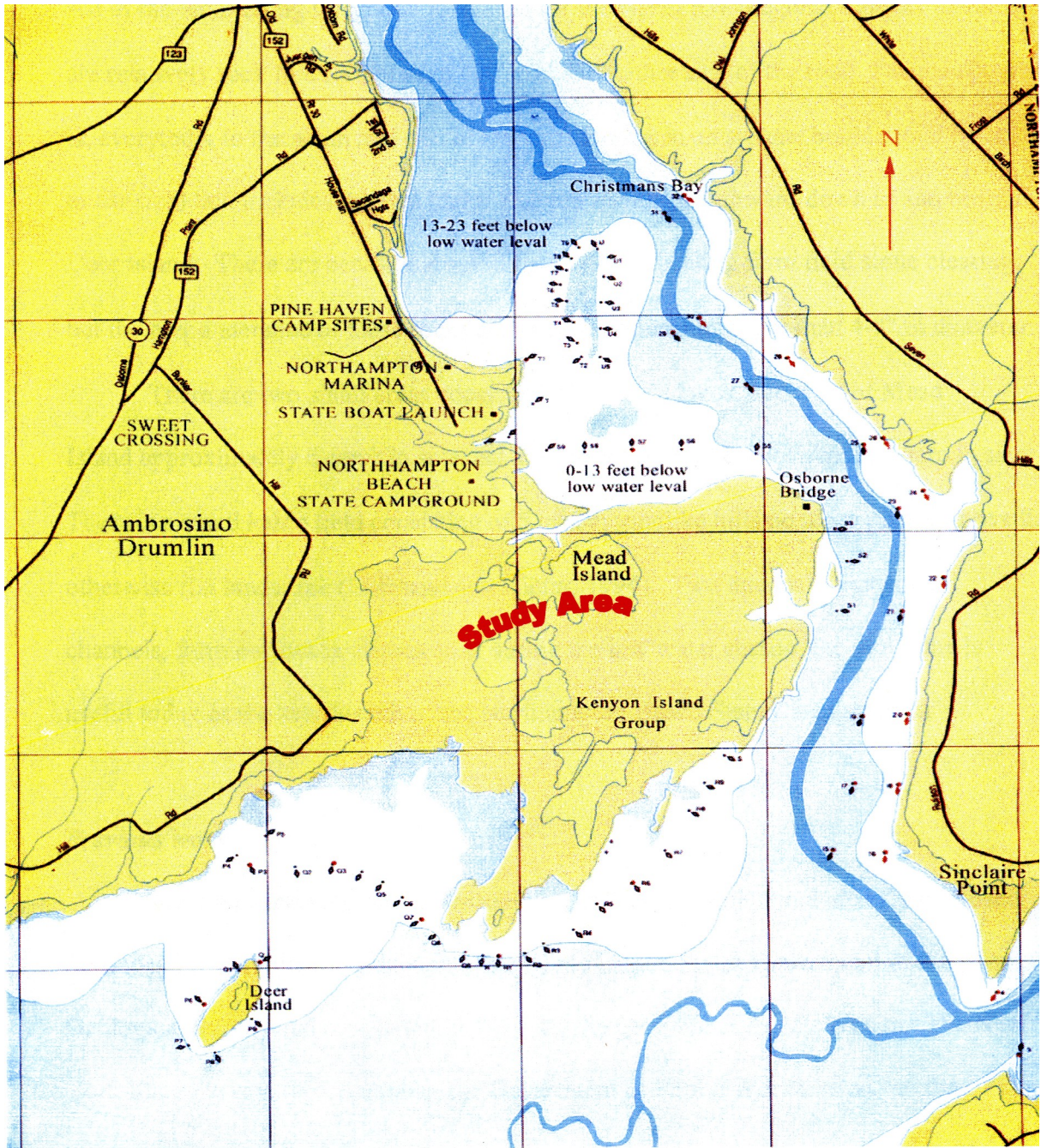
**Figure 1.7 The University of the State of New York, State Museum, Glacial/Geographic Map, Bulletin No. 280, with Cropped legend, by Brigham, 1929.**

### *Setting*

The spillway at the Conklingville Dam, which was built in 1929 to create the Sacandaga Reservoir is 771' above sea level. At high water the Kenyon Island Group including Mead and Deer Islands have shorelines at about 768'. The lake level begins to drop in (months). Water is discharged from the reservoir at an average rate of 1 1/2" per day (~1 billion gallons) and by the low water season over 150 billion gallons of water have been drained from the reservoir. It takes approximately 6 months for the transformation from high water to low water level status.

During the low-water season the shallow water shoal that surrounds the Kenyon Island Group becomes dangerous to recreational boaters on the lake. The situation becomes so serious that almost nine miles of shallow warning buoys are strung around the greater shoal, warning boaters not to enter. This buoy line extends from Deer Island to beyond Northampton Marina. Boaters are relegated to the marked Sacandaga River Channel for much of the boating season in order to by-pass the shallows (Figure 1.8).

The exposed lake bottom at low water is sufficiently dry during the summer season to travel across the old farm roads of the old town of Osborne Bridge. Dozens of remnant foundations where the old farmhouses and barns once stood are visible. Except for the occasional sandbar from anomalous lake currents like the one at core site #8 and ice-rafted sediments from just west of core site #2 to beyond core site #1, there is almost no appreciable new sedimentation across the whole shoal since the Conklingville dam flooded the area in 1929.



**Figure 1.8**  
**The Great Sacandaga Lake**  
**Navigation Map, Jimapco, 1991.**

All of the lands along the gravel road used for the NE to SW diagonal transect of cores, are relatively rock free for between 400' and 800' either side of the road. The exception is, everything to the south and east of Osborne Bridge in an arcuate boulder belt (rocks up to car size) that includes the two major Kenyon Islands, southward down to and beyond Deer Island. There are occasional small rock piles indicating farm field stone clearing, but the preponderance of these rocks are small boulders no bigger than 4-8" in diameter.

There are two small areas consistent with kettle lakes, one west of Mead Island approximately 6 acres in size and another to its southeast of about an acre in size. There is a small kame field consisting of 5 or 6 moundlike hills no more than 4' above the otherwise flat landscape (Lillesand and Kiefer, 1994). Two unusual relict stream channels, form backwater depressions during the low water season, the north one is useful today as the boat launch of the Northampton Beach State Campgrounds.

### ***Previous work***

The first surveying of the Broadalbin Quadrangle, which included the proposed Sacandaga Reservoir, was begun in 1899 by the United States Department of the Interior Geological Survey and completed in 1911, by the geographer, H. M. Wilson. The New York State Survey, represented by the Department of Public Works, surveyed the triangulations and W. H. Lovell surveyed the topography. Because construction of the Sacandaga Reservoir was anticipated, the proposed lake was stippled in and the shoreline was ambiguous. No attempt was made to depict the geology and neither the Noses fault nor the Batchellerville fault were delineated.



The first surveying and mapping of the Sacandaga terrain was undertaken by decree of the New York State Legislature in 1902. It was performed by John R. Freeman, C. E. and was finished in 1908. This study was undertaken for the purpose of creating a reservoir and only focused on the topography of the region and setting the proposed high water mark (“Flow Line”) and the dam site at Conklingville, New York. The projects legacy is the substantial body of work known as the New York State Water Commission Maps of 1908.

Simultaneous with the mapping of the Broadalbin Quadrangle by the U. S. Department of the Interior Geological Survey, The University of the State of New York, State Museum, began the first geological survey of the Quadrangle. The work was done by W. H. Miller and completed by 1911 as Bulletin #153 (Miller, 1911). Miller also stippled in the proposed Sacandaga Reservoir layered over the still viable communities within the anticipated “Flow Line.” Miller correctly identified the Noses fault to the west of the Sacandaga Basin but failed to identify the Batchellerville fault to the east. His description of the surficial geology was also very broad, calling it Quaternary glacial and postglacial sediments. Later on in his career when he had completed his geological survey of the Gloversville Quadrangle in 1922-23, Miller speculated that the Sacandaga River once drained to the Mohawk River rather than turning northward near Osborne Bridge to drain into the Hudson River, as it does today.

In 1929, The University of the State of New York, State Museum used the same Broadalbin Quadrangle and added the glacial and geographic features. The work was done by A. P. Brigham and presented in Bulletin #280 (Brigham, 1929). Brigham characterized the Kenyon Island Group and Osborne Bridge surficial geology as “water

laid drift” deposited in standing waters of glacial age. Brigham suggested that the preglacial Sacandaga River followed the present axis of the Sacandaga Reservoir and then flowed in a southwesterly direction, paralleling the Noses fault line, reaching the Mohawk River just west of Fonda, New York. Geophysical exploration in 1949 by the United States Geological Survey in cooperation with the State of New York, Department of Conservation, Water Power and Control Commission indicate the west edge of a buried escarpment and could possibly confirm Brigham’s postulate (Arnow, 1951).

The United States Geological Survey in cooperation with the State of New York, Department of Conservation, Water Power and Control Commission work, done in 1951 by Theodore Arnow, called Bulletin #GW-24 (Arnow, 1951), is the most complete report on the geology of the region. The study began in 1946 as part of a statewide survey of the groundwater resources of the State of New York. It is called “The Groundwater Resources of Fulton County, New York”. The study examined the records of 219 water wells in Fulton County, did chemical analysis on water samples, incorporated The University of the State of New York, State Museum geological work, and did the geophysical resistivity and seismic profiles of the overburden east of Gloversville, New York. It did not, however, examine the surficial geology of the County or the Sacandaga Basin, except in cooperation with local water well drilling companies whose well logs and tailings were used during the study period. It recognized in a generic way, Brigham, Miller, and others’ work in identifying the Quaternary deposits that mantle the region, but limited its scope to the sources, quantity and quality of the groundwater of Fulton County (Arnow, 1951). This report, however, did correctly interpret the former work of Darton, (1897). Darton (1897) had suggested multiple faulting north of the Hoffman’s Ferry

fault, east of Amsterdam. However, Darton did not identify the Batchellerville fault. It was later identified by the U. S. Geological Survey, Groundwater Branch workers who conducted the Groundwater Resources of Fulton County study (Arnold et al, 1951). This report and summarized by the work of Darton (1897), Miller (1911) and Wilson (1911), identifies the occurrence of the great sunken block between the Noses and the Batchellerville faults as being due to a regional uplift.

The Geologic Map of New York by The University of the State of New York, State Museum, compiled by (Fisher et al, 1970) and the Surficial Geologic Map of New York compiled by Caldwell and Dineen (1987) are both composites of previous work and aerial photography and should be considered as good references only. Neither is completely definitive nor intended to be.

The General Soils Report of Fulton County, New York (Davis et al,1971) was very important in arriving at the geologic perspectives this thesis espouses. The United States Department of Agriculture Soil Conservation Service prepared this report in cooperation with the Fulton County Soil and Water Conservation District. Most of the work was done using aerial photography and United States Geological Survey maps as working base maps. The report was prepared as a general planning guide, based on definitive soil characteristics, which affect the degree and type of limitations that can be expected for community development, recreational, or farmland use.

A soil association map was compiled with 50 soil-type associations across Fulton County. Each of the 50 soil characteristics were given soil classifications including their suitability for such things as agricultural use, campsites, roadbeds, athletic fields, and septic drainage fields, etc. (see Table 1.1, soil-type associations involving the study area).

GENERAL SOILS REPORT – SOIL ASSOCIATION AND CHARACTERISTICS								
Number	Soil Association	Soil Suitability and Limitations					Sand and Gravel	Drainage Class
		Field Crops	Camp Sites	Streets, Roadbeds	Athletic Fields	Septic Fields		
8	Charlton-Sutton Stony	Slight	Slight	Moderate	Moderate and Stony	Moderate	Not Suitable	Well & Moderate
11	Charlton, Stony and Extremely Stony	Severe Stoniness	Severe Stoniness	Severe Stoniness	Severe Stoniness	Severe Stoniness	Not Suitable	Well
12	Sutton-Leicester Stony	Slight	Moderate	Moderate	Moderate and Stony	Moderate, Wetness	Not Suitable	Moderate & Poorly
14	Leicester Stony	Moderate	Severe & Wetness	Severe & Wetness	Severe & Wetness	Severe Wetness	Not Suitable	Poorly
17	Charlton-Sutton Extremely Stony	Severe Stoniness	Severe Stoniness	Severe Stoniness	Severe Stoniness	Severe Stoniness	Not Suitable	Well & Moderate
32	Sudbury	Moderate	Moderate	Moderate	Moderate	Moderate	Good/Fair	Moderate

**Table 1.1 Soil Type Associations in Study Area from 1971 General Soils Report of Fulton County, New York.**

Although the work is interpretive and little fieldwork was actually done (Davis et al, 1971), patterns do exist that are quite revealing (Figure 1.9). The work was done in the spring of 1971 during high water level status of the Great Sacandaga Reservoir. Doing the work in the spring distorted the Kenyon Island Group soil-type perspectives and Davis's field workers did not expend the time boating out to the islands. Thus, this report overlooked the forested terrain and the lands adjacent to the Kenyon Islands that form the focus of this thesis.

Finally, the Hudson River-Black River Regulating District (the new District created in 1959 by the New York State Legislature when it combined the Hudson River and Black River Regulating Districts) commissioned a study (Gabrielson, 1982). The purpose of the study was to determine the storage elevation curve of the reservoir to estimate how much sedimentation from stream tributaries and shoreline erosion was affecting the lake capacity. Their conclusion was that after 53 years of operations, the holding capacity of the reservoir had diminished by 1/3 of a billion cubic feet or about 2.6 billion gallons.

### ***Format of this study***

In this thesis, I try to present a logical argument for a deepening project in the Great Sacandaga Lake by looking at the subsurface soil stratigraphy using Geoprobe coring in Chapter 2. Core descriptions, access, permits, maps and GPS coordinates, methods and the questions raised are presented.

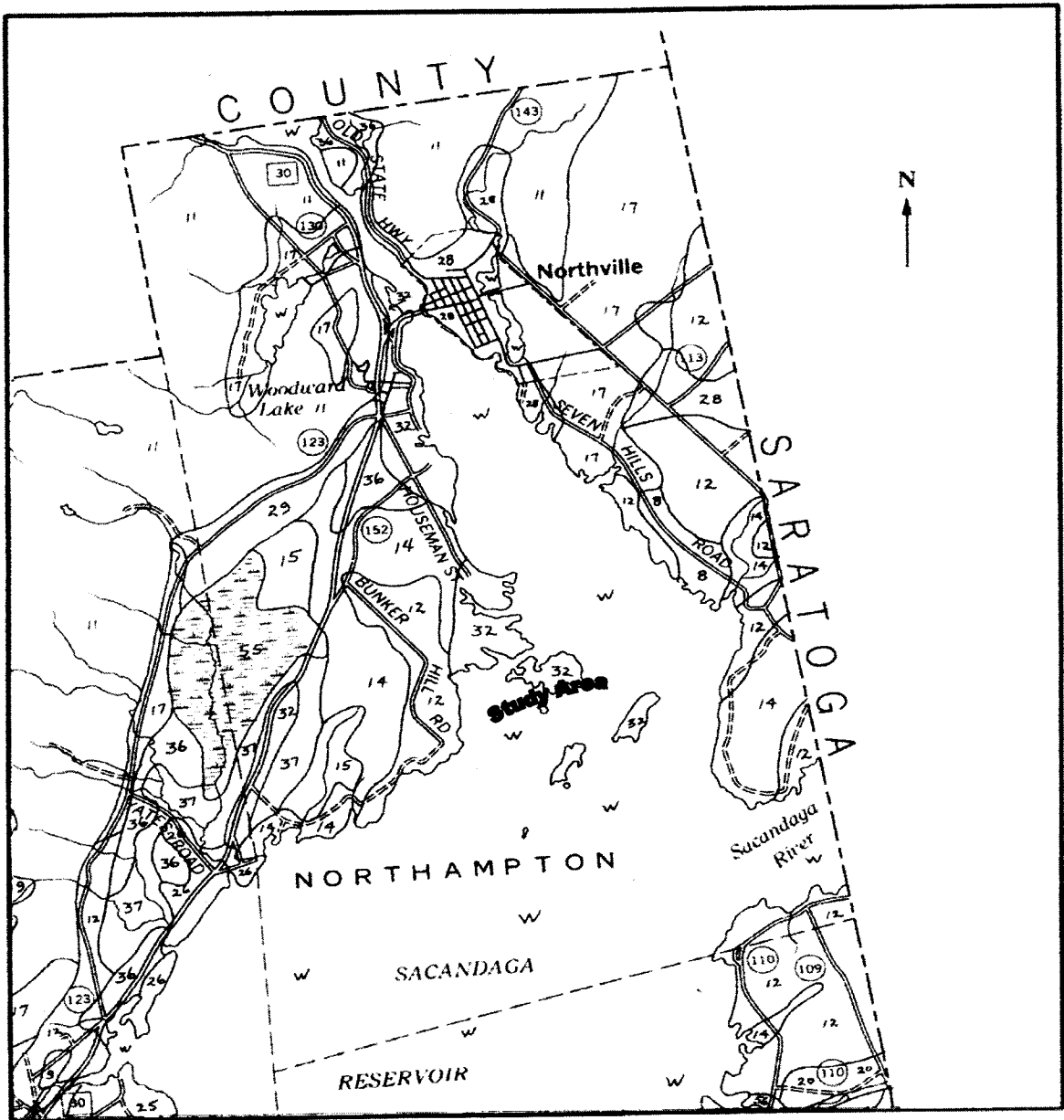


Figure 1.9 Soil Type Associations in Study Area from the General Soils Report of Fulton County, New York, Arnow et al, 1971.

In Chapter 3, a grain size analysis is conducted to try to assist in the interpretation of the cores. I have also examined the clay-sized fractions with X-Ray Diffraction and evaluate using *Ostracodes* to examine lake history. Idrisi software was used on Landsat 4, 5 imagery and isodata clustering during the late shallow water seasons of 1984 and 1988 to evaluate the bathymetry and land cover of the study area.

Some recent and historical accounts of the potential presence of heavy minerals in the sediments or bedrock around and in the reservoir are taken into account in Chapter 4. A trace metal analysis performed on different soil horizons in the geoprobe cores was used to determine which horizon minerals, if any, have value. A total heavy mineral mineralogy on stratigraphically averaged sediments from the cores was also performed. I also try to determine if the heavy minerals were entrained in glacial debris transported from many miles away, or rather, if they are beneath the overburden in the bedrock. The potential mineral resource of the heavy minerals will also be discussed.

In Chapter 5, I take a look at the issues involved in Kenyon Island reconfiguration. They include access and ownership of the lakebed sediments, the planning and geography of the reconfiguration possibilities, the costs of the undertakings, the real property values, and the timetables of the wisest approach to the deepening effort. In accordance with the new field and laboratory evidence, the potential economic impact a deepening project could have is presented and some possible future work and the conclusions are discussed.

## CHAPTER 2

### GEOPROBE CORING

#### *Geoprobe Coring*

Nine sediment cores were collected with a Geoprobe™ in a transect across the Kenyon Island Group. The goal for each core site was to obtain continuous soil/sediment core samples, but this is sometimes not attainable especially if the cutting shoe hits gravel, cobbles or large rocks.

The Geoprobe used for this coring was a model #5400 operated by Environmental Maintenance Technology (Schenectady, New York). The nine Geoprobe core samples were collected along a NE to SW diagonal transect, across the shallow water shoal, which includes the Kenyon Island Group and Mead Island. Access was accomplished during the low-water season of 1998, by driving along the gravel roadbeds of the former village roads of Osborne Bridge, which is submerged during the high water season. Everywhere along the transect of core holes, the ground was generally flat, but the ability to adjust the plumb of the core holes so precisely, made the operations very smooth and effortless. Minimal disturbance to the ground is a great advantage and the probing rate is fast between sections (from 5' to 25' per minute depending on the soil/sediment type). Each probe rod section is 48" long with a drive head at the top and a cutting shoe on the bottom. The outside diameter of the stainless steel probe rod is 2.0125" and the outside diameter of the cutting shoe is 2.2" with an inside diameter of exactly 1.8125". A 44" CAB (Cellulose Acetate Butyrate) polymer liner, exactly 1 3/4" OD fits snugly inside. The liners are approximately 1/32" thick and have a small flare on the down side to fit

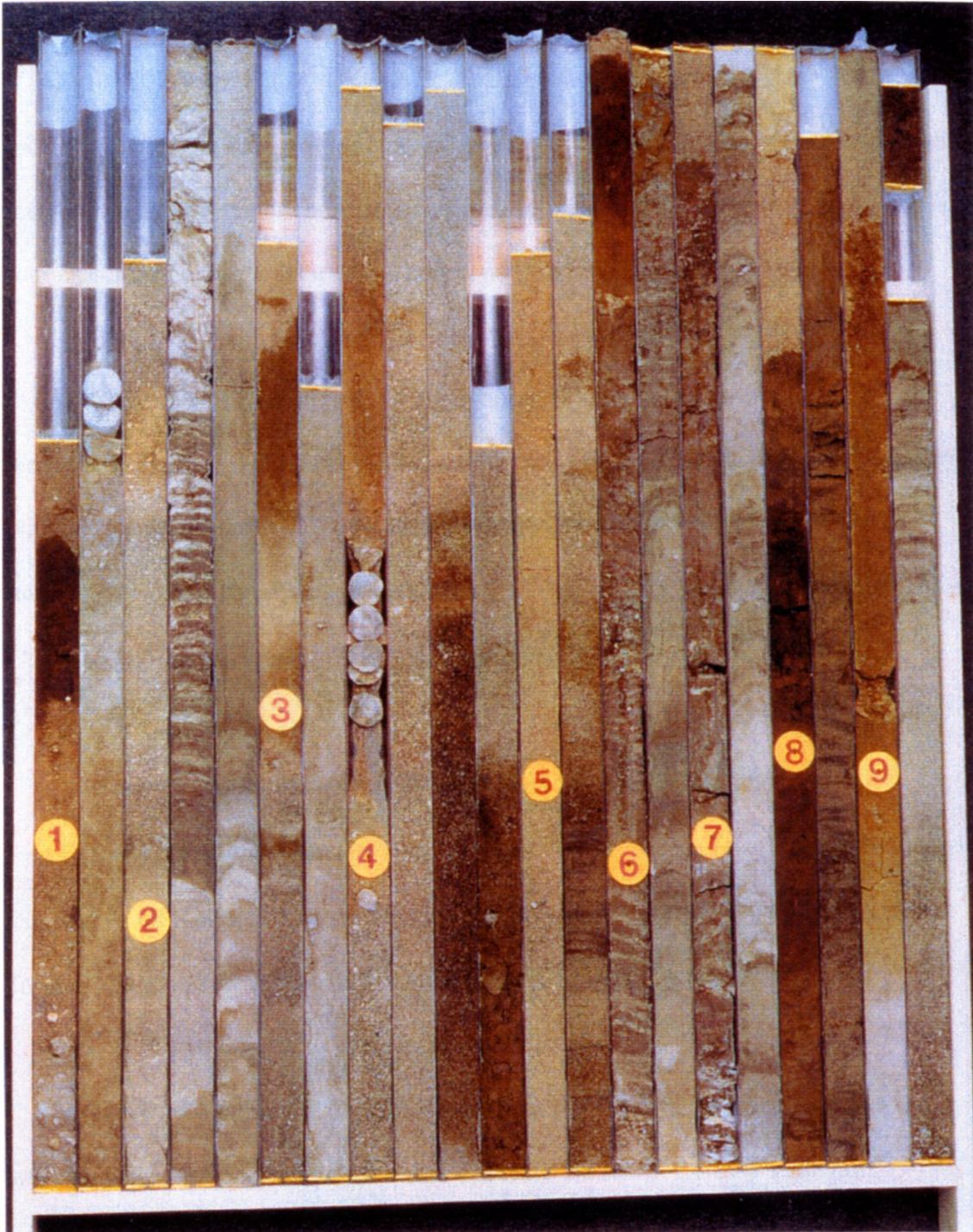


over the cutting shoe on the inside of the probe casing. These liners are clear and offer a good view of newly pushed cores, at least until the soil/sediment moisture obscures the view. For clarity the 44" liners are assigned an added 4", weighted over the sediment horizons to account for the 4' penetrations of each probe section. The first 4' section is pushed and removed. Each successive core is pushed in 4' more with the aid of 1.25" solid steel probe extensions. Extensions are added and removed one at a time, so deep cores are more and more time consuming.

Environmental Maintenance Technology's Geoprobe was powerful enough to push aside small rocks, but this never became a significant problem in any of the core holes. Our purpose was to soil/sediment probe as many holes as possible into the subsurface of the shallow water shoal known as the Kenyon Island Group, in one workday. The coring was performed on November 13, 1998, and we recovered a total of nine cores (See figure 1.3).

### *Core descriptions*

In cores #1 and #4 we encountered small rocks in gravel horizons, which were about 2" in diameter and these rocks blocked the recovery of complete cores until the cutting shoe rounded them enough to accept them (Figure 2.1). Figure 2.1 should be referred to whenever detailed descriptions of the core horizons are being discussed. None of the cores encountered any rocks of enough consequence to force us to relocate the truck for a second push. For a more complete discussion of the core horizons (See Appendix B – Core Descriptions).



**Figure 2.1 Geoprobe Core Horizons.**

**Core #1** was located just *SSW* of the Four Corners of old downtown Osborne Bridge. Its file number is 049 and its GPS (Global Positioning Satellite) coordinates are *N* (43°-10'-801") and *W* (74°-08'-998"). The elevation was approximately 766' above sea level. The site was also just north or on the edge of an end moraine, which was revealed further into the investigation of the study area (see page 53-55). We were probably on the former lands of either J. Paul or E. Hammond (see Figure 1.3).

**Core #2** was located just *NW* of Osborne Bridge on the former lands of Foster Noyes (see Figure 1.3). It was located on the banks of the Sacandaga River Floodplain, which is pervasive north to the Northville Bridge, even at low water status. Its file number is 048 and its GPS coordinates are *N* (43°-10'-865") and *W* (74°-09'-809"). The elevation was approximately 765' above sea level.

**Core #3** was located just *E* of Mead Island on the former lands of L. D. Fritcher (see Figure 1.3). It was at an elevation of about 765' above sea level. Its file number is 050 and its GPS coordinates are *N* (43°-10'-821") and *W* (74°-09'-791"). The site was formerly farmed and a few small stone piles existed adjacent to Mead Island.

**Core #4** was located at the *NW* corner of Mead Island adjacent to a backwater inlet that faces the Sacandaga River floodplain to the east. Core #4's file number is 047 and its GPS coordinates are *N* (43°-10'-715") and *W* (74°-10'-252"). Its elevation was precisely at the high water mark of 768' above sea level. The site was on the former lands of Charles and Arietta Mead (see Figure 1.3).

**Core #5** was located *WSW* of Mead Island on the edge of the largest eroded micro-island of a hummock field in the local greater Mead Islands group. The micro-island is usually submerged until late summer. The site was picked just on the eastern edge of the old Osborne Bridge Road and was 765' above sea level. Its file number is 046 and its GPS coordinates are *N* (43°-10'-527") and *W* (74°-10'-230"). It is unclear by the 1908 New York State Water Commission Maps who the former owner was but it could have been Horace Kenyon. In a straight line, the site was about 900' from core site #4 (see Figure 1.3).

**Core #6** was located about 800' further *SW* along the old Osborne Bridge gravel road from site #5 (see Figure 1.3). This site is approximately 760' above sea level and it is unclear who the former owner was, but was likely to be Horace Kenyon. The site was on the East Side of the road and appears to be a former dirt road heading northwest to an anonymous unimproved road connected to the Bunker Hill road. Its file number is 045 and its GPS coordinates are *N* (43°-10'-431") and *W* (74°-10'-421").

**Core #7** was located at a bend in the old Osborne Bridge Road, *SW* of core #6 by some 1200' (see Figure 1.3). It's not perfectly clear from the maps but this property could have been formerly the same lands of Ed Gifford. This site was also at the 760' elevation above sea level and on the West Side of the road. Its file number was 044 and its GPS coordinates were *N* (43°-10'-332") and *W* (74°-10'-487").

**Core #8** was sited approximately 800' further *SW* along the old Osborne Bridge Road. It is on the East Side of the road at 764' above sea level and was probably the former lands of Ammon Blowers (see Figure 1.3). Its file number is 043 and its GPS coordinates are *N* (43°-10'-262") and *W* (74°-10'-603").

**Core #9** is located at the high water mark (768') on the shoreline of the old Green's Beach Campsites. It would be just south of the old Osborne Bridge Road and on the former lands of Ammon Blowers (see Figure 1.3). Its file number is 042 and its GPS coordinates are *N* (43°-10'-182") and *W* (74°-10'-798"). Core #9 was approximately 1200' further *SW* along the old Osborne Bridge Road.

### ***Core Analysis***

After collecting the cores the ends were taped tight with duct tape to prevent the escape of sediments and moisture. All of the cores were labeled at retrieval with a file number and designated which ends were tops and bottoms. They were then transported to the Earth and Atmospheric Sciences Sediments Laboratory at the University at Albany.

Special racks were built to hold the cores and restrain them one by one for splitting each core liner lengthwise, proximally. The University machine shop built to my specifications a brass-cutting implement to which we attached two roofing knife blades. These were meant to slice the core liners only, at the centers 180° apart along the lengths. Hand handles were fabricated and 50# monofilament fishing line was attached to the handles and used to cut the sediments proximally through the length of the liners. A steel length weight was placed over the already split cores and a flat plastic shim was

pulled down through the center of the liners, separating the top and bottom liner and sediments. Once that was done, the top split could be inverted and laid cut side up. The University Graphics Photographer took pictures of the still moist core halves (see Figure 2.1). The remaining halves were covered with clear plastic strips, (44" x 1¾") and archived in the Sediment Lab at the University of Albany.

The actual core liner lengths, which are 44", were adjusted to reflect a 48" push Geoprobe depth. Each core was given a file number, which referred to the core site no matter what the depth of the pushed core was. The sediment horizons were designated by their status regarding depth. For example, a distinct horizon in core site #1, file number 049, at the bottom of the second section from 78" to 96" deep would hereafter be referred to as 0497896. An interesting horizon at core site #4, file number 047, at the bottom of the fourth section from 176" to 192" deep would hereafter be referred to as 047176192, etc.

Each distinct horizon, were split and weighed. Portions were used for analysis, of trace metals, particle grain size analysis, pH and conductivity. Read about the purpose and results of the particle grain size study in Chapter 3 and trace metals in Chapter 4. Additionally, plugs were taken at intervals in the varved banding of the fine sand and clay horizons in sites #6 and #7 for identification of the clay constituents and to evaluate the need for a more detailed study of the *Ostracode* fauna.

## CHAPTER 3

### PARTICLE GRAIN SIZE ANALYSIS

#### *Purpose*

The core descriptions in Chapter 2 were based on observation, texture, color and facies changes, which reflect changes in the depositional environment. Facies changes at the top 16" of core site #8 show the sandbar made by currents during the last 70 years. The top 32" of modern (<70 years) ice rafted clasts and postglacial glaciofluvial gravels at core site #2 are Recent, due to the annual deltaic flooding at this part of the Sacandaga River as it opens-up to the newly created reservoir. Alignment of the core profiles contiguously by elevation with removal of all the blank spaces failed to allow clear correlation of sediment horizons from one site to another. Similarities can be seen between cores #6 and #7 or between cores #8 and #9, but no correlation exists between cores #5 and #6 and little exists between cores #7 and #8, even though the distances between probes was a mere 800' in each case.

The depositional environment given the nearby Sacandaga River and/or the observations of the authors of previous work in the study area ie; glaciofluvial (Brigham, 1929), fluvial (Miller, 1909) and floodplain or fluvial deltaic (Cadwell, 1987) did not explain the deposition of the study area sediments. If the deposition of the core section horizons were either glaciofluvial or fluvial, we would expect to see sediment types with several components each related to processes of different flow regimes , characteristic of a polymodal distribution. Historical accounts of the Sacandaga River, before the reservoir was created, indicate numerous floodings and drownings during lumber log

jams in the Sacandaga Basin. Even the 1908 Water Commission Maps indicate flood events at 766 feet above sea level, though the Sacandaga River elevation in the study area before the reservoir averaged 740 feet above sea level. River velocities, size and density of clasts and particle shapes would all be variable aspects of transported and entrained sediments, which should be expected to impart some degree of hydrodynamic sorting (Prothero and Schwab, 1996).

If the deposition were that of a floodplain, we would expect to see great variations over both small vertical and small horizontal distances, and the importance of the fine component, which characterizes the lowland suspended load type of rivers. Any fine suspended loads outside of the main channel of the river would have had to come about by the river overtopping its banks (Brown, 1985). Fine suspended loads would appear as layers in the sediment horizons and appear as abundances in core horizon splits.

Finally, if the deposition of the core sediment horizons were those of a lacustrine or lake environment, one would expect the preponderance of the sediments to be clays and silts, less than (0.0625mm/62.5 $\mu$ m) in size and showing sequences of laminae deposited in still water with annual layering reflecting lacustrine conditions (Loizeau et al., 1994). Important clues would be clay, terrestrial organic matter and freshwater crustaceans such as *Ostracodes*. The laminated layers in the clay horizons of cores #2, #6 and #7 looked to be good candidates for such conditions.

Over the last decade or so, several automatic rapid sediment analyzers based on light diffraction have been developed which are ideal for the measurement of the particle grain size of natural sediments, especially fine-grained sediments characteristic of fluvial/glaciofluvial and floodplain or lacustrine sediments. The geology department at



Union College in Schenectady, New York, has a Coulter LS-100 with a laser light source, from Coulter Electronics Ltd (USA) in their core analysis lab. Data from this instrument was used to characterize the particle grain sizes to assist in the interpretation of the depositional environments. Grain-size distributions would be presented in two ways: graphs of grain-size distributions will be plotted and compared; and statistical measures such as mean, median, mode and sorting are arithmetically calculated. In this chapter I will also identify and analyze the clay sized fraction with X-Ray Diffraction. Although an *Ostracode* study of the varved fine sand and clay layers of the core horizons at sites #2, #6 and #7 may not be appropriate for a chapter on grain-size, I did the work and hoped to learn whether (Brigham, 1929) was right to have postulated what he called a former “Glacial Lake Sacandaga” in the basin. The results of the *Ostracode* study will determine if a coherent pollen profile study needs to be undertaken. No bulk density or pore water analysis will be undertaken in the fine sand and clay-sized fraction to estimate the physical or chemical weathering effects on the sediments. All of the horizons examined either for particle grain size or economic worth (see Chapter 4) will be repeats of the most interesting and colorful horizons. References to a pH and Conductivity analysis, in this Chapter, on the core horizons interesting enough to further our overall goal of evaluating the potential economic worth of the sediments for a deepening project, will be discussed in Chapter 4.

### ***Methods and Principles***

Core horizons thought most interesting, as described in Chapter 2, page 48, were split and weighed, with approximately 30 dry grams used for particle grain size analysis

and pH and Conductivity measurements. Each 30-gram horizon split was dried in a low temperature oven at 50° C (maximum temperature, so as not to lose any volatiles that may have existed). Core splitting revealed that there were no root systems to be removed. Absence of root systems was not assumed to indicate organics were not present, but rather an indication that no aquatic plants or terrestrial plants existed at any of the core sites. Approximately 20 grams were split-out of the 30 gram charge and set aside for the pH and conductivity measurements. Of the roughly 10 grams left, enough of a portion from each core horizon was sieved-out (<2mm) and weighed to about 2 grams (g) and placed in a pan. In each core horizon exactly 0.50g of sieved sample was weighed to the nearest 0.01g and placed in a pre-labeled and weighed centrifuge tube. Labeling recorded the core horizon sample numbers and weights and was used for duplicate records. All of the remaining core horizon (<2mm) fractions were tested with weak sulfuric acid for secondary carbonates, but none whatsoever were detected. Therefore, no carbonate extractions were necessary.

Despite the lack of visible roots, the removal of organic matter is essential, especially because it binds mineral particles together; particularly clay sized particles, and hinders dispersion. For removal of organics, 3 milliliters (ml) of concentrated (30%) H<sub>2</sub>O<sub>2</sub> (Hydrogen Peroxide) was added to each centrifuge tube and sample. Each tube and sample was placed under a fume hood and gloves and glasses were used with a calibrated pipette to add the H<sub>2</sub>O<sub>2</sub>. Each sample was stirred with a glass rod and the rod was rinsed with deionized (DI) water (rinsed back into the centrifuge tube so that no material was lost). Organic matter is oxidized by H<sub>2</sub>O<sub>2</sub> and an initial reaction or frothing indicates that organic matter is present. No organic matter was evident in any sample except for very

small amounts (as indicated by the extent of frothing) in a few of the surface sediment horizons. Still, to be absolutely sure that all of the organics would be destroyed, each centrifuge tube and sample was placed upright in beakers about half filled with water on a hotplate set to 50° C (none of the tubes were submerged more than 75% in H<sub>2</sub>O). When the small amount of frothing in any of the tubes was complete enough to ensure that no further loss of fluid in the tubes was occurring, each tube was filled evenly (to about 1½" deep) with DI water and left overnight on the hotplate in the fume hood to thoroughly digest any remaining organics. The next morning the fluid levels in the tubes were topped off with DI water (to balance the tubes for centrifuging). Caps were placed on each set of centrifuge tubes and samples and they were centrifuged at 5000-7000 RPM for 3 minutes. Each tube supernatant was decanted carefully so that no sediment escaped. The centrifuge tubes were filled to ¾ level and the caps replaced. The tubes and samples were timed and vigorously hand shaken for 30 seconds and re-centrifuged for another 3 minutes.

The removal of biogenic silica is necessary for all lake sediment samples and surface soil developed horizons. Physiological actions of organisms, both because of the study area's history of farming and the lake microorganisms, effect the cycling of the chemical constituents of the core sediment minerals. To do this, 10ml of 1 Molar NaOH (Sodium Hydroxide) was added to the centrifuge tubes and samples. Each tube was left upright in beakers without caps in water and placed back on the hotplate, set at 50° C for at least 4 hours. The samples were not allowed to dry out or cool down during this period. The centrifuge tubes were capped, placed in racks and then in a reciprocating shaker for 30 minutes. After the required time, the centrifuge tubes were put back in the

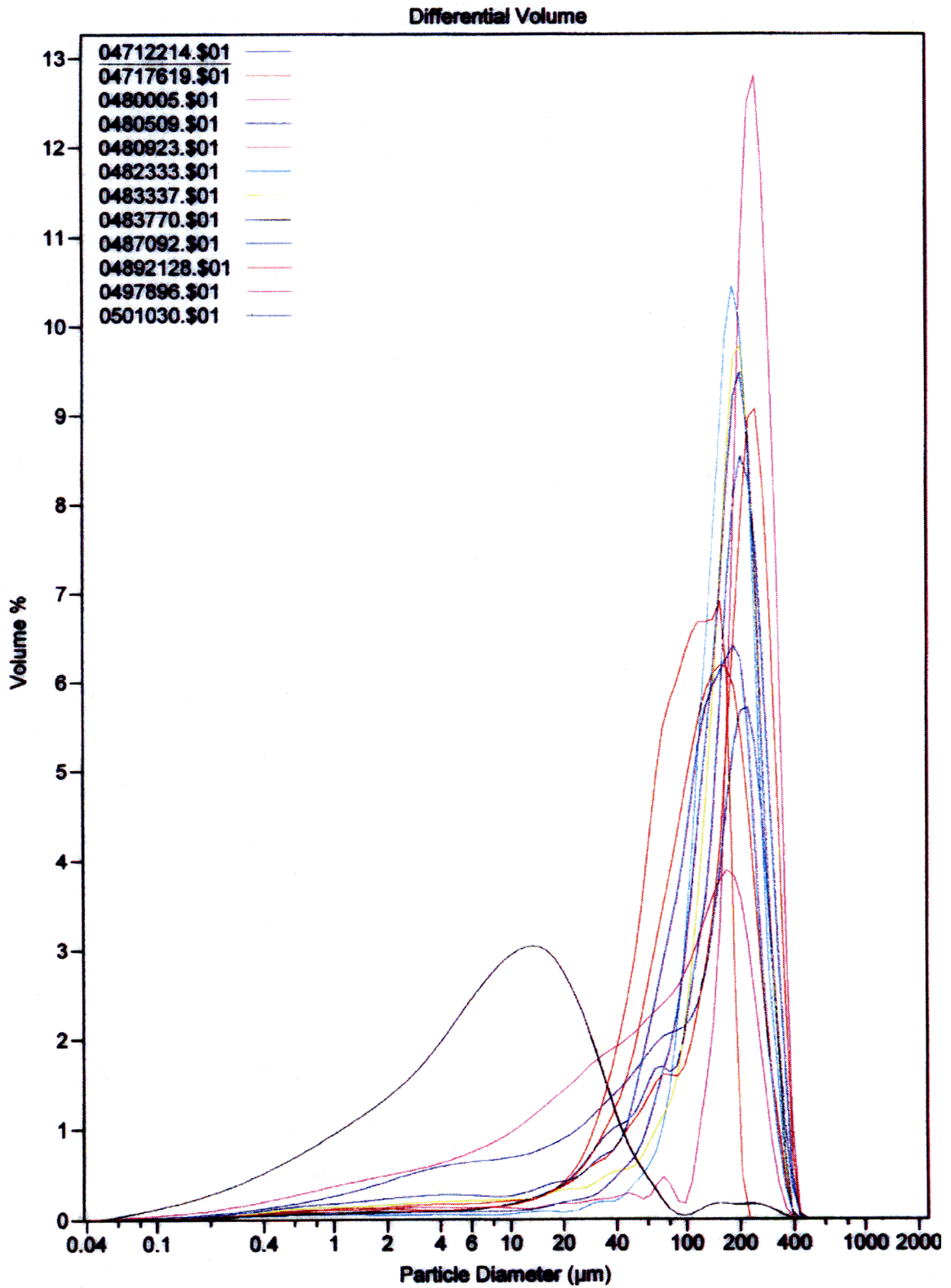
water-filled beakers on the hotplate for another 30 minutes. Each tube was again put in groups for centrifuging at 5000-7000 RPM for another 3 minutes. Each tube supernatant was carefully decanted so as to avoid losing any sample material. Virtually no reaction occurred on any sample during the removal of the biogenic silica, so the tubes were filled evenly to  $\frac{3}{4}$  full with DI water, timed and hand shaken vigorously for 30 seconds apiece and re-centrifuged one more time. Each sample was carefully decanted again, so that no sediment was lost and the tubes were filled to  $\frac{3}{4}$  full.

A chemical dispersant made of non-ionic surfactant (0.3 Molar), sodium hexametaphosphate (5 Molar) and sodium carbonate (1 Molar) was added to each centrifuge tube and sample and stirred with a glass rod. The glass rods were carefully rinsed with DI water back into the tubes to prevent sample contamination and material loss. The samples were then capped and stood upright in racks and placed on the reciprocating shaker table for two hours, after which they were left overnight to await particle size analysis with the Coulter laser diffraction machine. The dispersant is supposed to accomplish three tasks: (a) adding sodium ions to increase the exchangeable sodium and cause a repulsion between particles, (b) adding hexametaphosphate which is adsorbed on to positive electrical charges on the sesquioxides and clays, so preventing attraction to negatively charged clay and (c) adding carbonate to raise the pH of the solution and so remove positive charges (Loizeau et al., 1994).

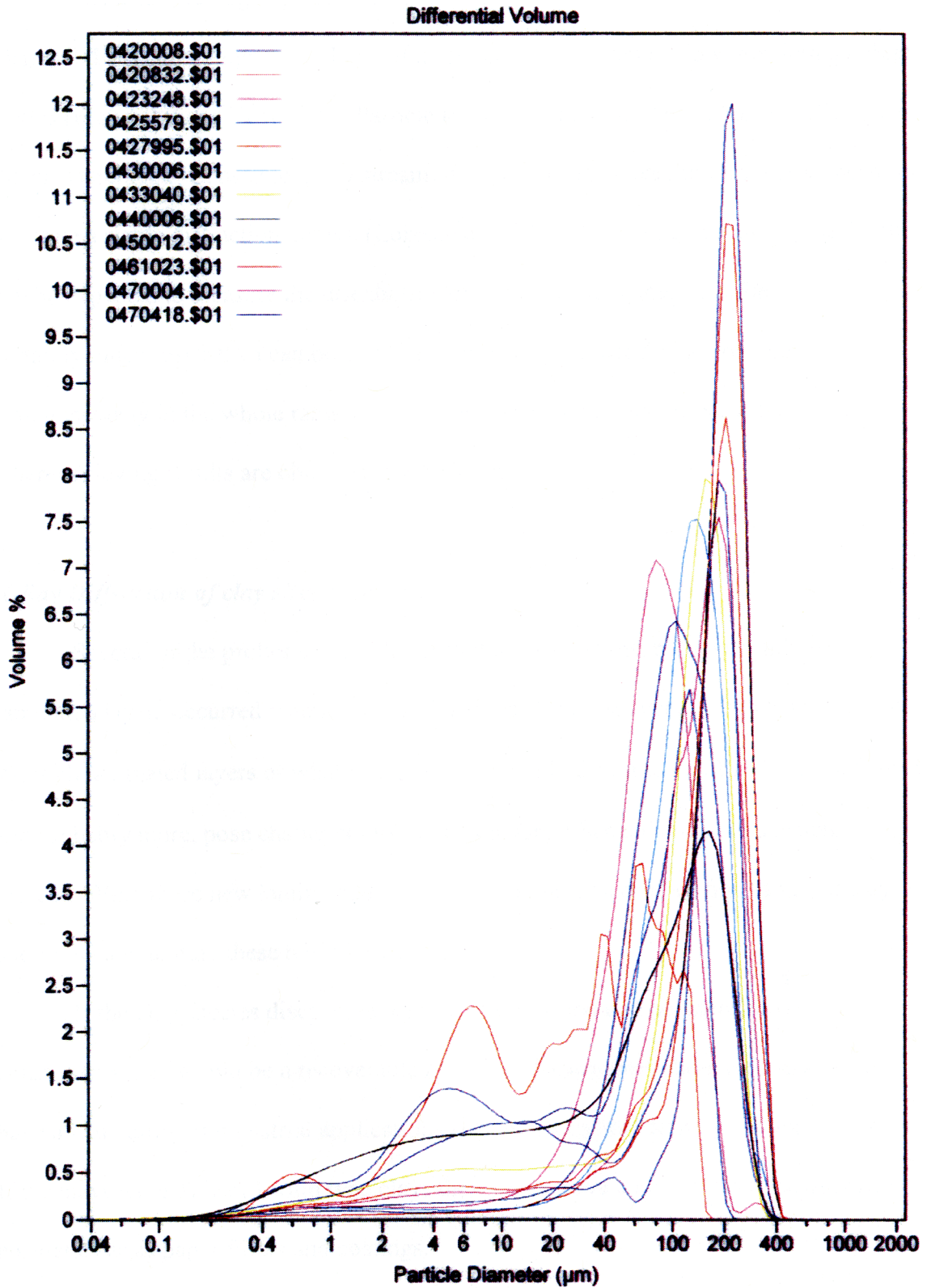
Using the Coulter Laser (LS) Particle Size Analyzer is essentially automatic. Each centrifuge tube and sample is placed upright in a beaker of water, atop an ultrasonic vibrating machine for 5 minutes prior to run time. Vibrating ensures that dissolved air from stirring and shaking is degassed because bubbles do not interfere with the particle

size distributions. The filename and sample information is entered, the sample added after the water column is flushed clear with filtered water. After each sample the Coulter automatically saves the particle size distribution as a log- $\mu\text{m}$  Cumulative Curve that can be grouped as advised (Figures 3.1a and 3.1b). The Volume Statistics generated can be seen in Appendix B. The exact same protocol, ultrasonic degassing, data entry, sample addition and chamber purging follow each sample run.

The principle of the Coulter Laser Grain size analyzer is based on the interaction between light and particles of sediment. Diffraction is the preponderant effect: particles of a given size scatter light through a given angle, the angle increasing with decreasing particle size. A laser beam of monochromatic light ( $\lambda = 750\text{nm}$ ) is collimated to form a 13 mm diameter beam that passes through the sample cell containing an upwards-moving suspension. After the samples are ultrasonically dispersed, the centrifuge tube and its sample contents are introduced into the fluid module that generates a high-speed suspension stream inside the sample cell. The stream is strong enough to move particles up to  $900\mu\text{m}$  in diameter with a density of up to  $4\text{g cm}^{-3}$ . Size results are supplied in terms of equivalent spherical cross-sectional diameter and the Coulter resolves sizes into 72 classes from 0.4 to  $900\mu\text{m}$  with an exact logarithmic progression. The Coulter Laser Particle Size Analyzer does not take into account particles calculated to be smaller than  $0.4\mu\text{m}$ , such as ultra fine clays and silts. The efficiency of their detection is proportional to the actual clay content. For ultra fine sediments containing less than 10% of clay, the efficiency is about 40-45% whereas for clay content greater than about 80% it reaches 100%.



**Figure 3.1a** Coulter LS Particle Size Analyzer, Sediment/Core Lab,  
Geology Department, Union College.



**Figure 3.1b** Coulter LS Particle Size Analyzer, Sediment/Core Lab,  
Geology Department, Union College.

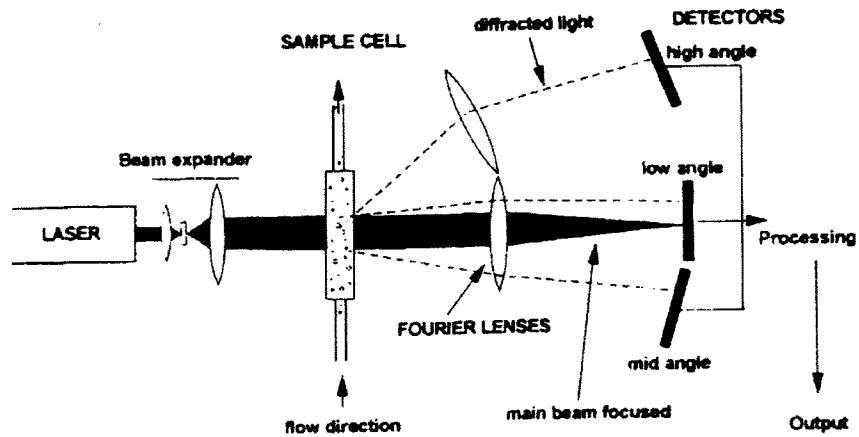
The diffracted light is focused by two Fourier lenses, which direct light of identical diffraction angle toward one of three sets of detectors for low, medium and high angles (Figure 3.2 and Table 3.1). Particle location in the sample cell has no influence on the measurement. Consequently, a stream of particles can be passed through the beam to generate a stable diffraction pattern (Loizeau et al., 1994). From all accounts, the Coulter can be expected to measure the distribution of sediment sizes equally or better, and certainly much faster than can be measured by sieving. Laser diffraction computes simultaneously in the whole range of sizes, expressed in volume (or surface) units, whereas sieving results are obtained in weight units.

### ***X-Ray Diffraction of clay sized fractions***

Several of the probes showed laminated layer horizons, but apparently the laminated layers occurred mainly in sand fractions. The discovery of two cores, #6 and #7, with laminated layers of what appeared to be solid <50 $\mu$ m to depths of at least 8' deep and probably more, pose challenges to a potential Great Sacandaga Lake deepening project. Were these new laminated layers solid clay or a mixture of fine sand and clay and to what extent do these horizons exist?

If the clay species discovered were 100% pure and had no superfluous mineral constituents they would be a recoverable resource. Expandable clays such as smectite have a wide array of industrial applications such as catalytical activity and ion exchange. If the clays were Talc or kaolin type clays they would have applications in powders and cosmetics or as paper fillers and coatings.





**Figure 3.2 Schematic diagram showing the arrangement of the main Components of the Coulter LS-100, after Loizeau, 1994.**

**Table 3.1 Main characteristics of the laser diffraction grain size analyzer Coulter LS-100, after J-I Loizequ et al., 1994.**

Laser type	solid, 4mW
Wave length of light	$\lambda=750\text{nm}$
Number of detectors	127
Size range	0.4-900 $\mu\text{m}$
Number of size classes	72
Fluid Module capacity	1-2 litres
Weight of sediment needed	0.05-2 g

If the clay has no intrinsic value, it still could have some very practical purpose, as a constituent in topsoil production for the reconstructed island or islands. Several of the smaller outlying Kenyon Islands and Mead Islands are barren with few, if any, trees. The larger islands have trees but most are stunted and are littered with dead and fallen-trees. Good topsoil, depending on the source referenced, has 15-20% clay, 40-45% sand, 35-40% sesquioxides and 5-10% humus. The right proportions necessary to build a topsoil that does not need constant nutrient additions and will support trees and grasses would have to be determined by a soil scientist, but the shallow water shoal sediments (those sediments to be dredged) should have a sufficient contingent of each in the sediment mix. The ideal island that might be reconfigured in a deepening project will be discussed in Chapter 5, but would involve approximately 3-4 million cubic yards (yd<sup>3</sup>) of topsoil (12" deep). Clays of little intrinsic value (between 500,000-800,000 yd<sup>3</sup>) could be utilized in topsoil mixtures during island reconstruction.

The extent to which the fine sand and clay horizons exist could not possibly be determined without a much more extensive probe grid work. That may not be necessary though, if the X-Ray Diffraction (XRD) results reveal no potential economically valuable clay fraction and/or it contained superfluous contamination that required separation. If the clays cannot be determined to be valuable and dredging them is impractical, then earth-moving heavy equipment used to accomplish a deepening project would overcome the clay dilemma wherever necessary.

Clay plugs (1/2" marble size) were taken from the upper core sections of core #6 and #7 at 36" of depth. Core #6 included a light colored varve layer and core #7

includes a dark colored varve layer. The plugs were sent to Dr. Rossman F. Giese Jr. at the State University of New York at Buffalo, for X-ray diffraction analysis.

The clay fraction is mostly chlorite (>80%) and lepidolite (20%), with the balance having 12Å (angstrom) spacing, suspected of being <4µm common silicates feldspar and quartz. Fine sand and clays accounted for approximately 12% of the samples by weight. There were no expandable clays in the clay fractions. No further X-Ray diffractions were planned because both cores #6 and #7 were so consistent throughout and there were no apparent facies changes evident.

Therefore, I conclude that the <4µm fraction has no intrinsic economic value and are essentially all feldspar and quartz rendering them useful only as a constituent of topsoil and/or fill. One important result of the X-Ray Diffraction is that the clay is from the Chlorite Group and chlorite and lepidolite might help with the interpretation of the rock provenance of the sediments. This also means that the <4µm horizons would have to have been deposited as rock flour (pulverized while being transported). Rock flour is finely comminuted, chemically unweathered material, consisting of silt and clay sized angular particles of rock forming minerals. The extent to which silt and clay sized sediments exist here in the study area provides clues as to the sediment depositional environment and perhaps denudation rates, suggesting that the transport mechanism was a glacier. If the clay fraction is lacustrine, then either an *Ostracode* study or a coherent pollen profile might help with interpreting the environment of deposition.

### *Ostracode Study*

Plugs were removed from cores #2, #6 and #7 at approximately 60", 36" and 60" deep respectively, on the premise that *ostracodes* could be preserved in the sediments. Most *Podocopina* are between 1-2mm, to ensure capture of juveniles or those of smaller stature, each plug was wet sieved at 80 $\mu$ m (0.07mm) and any sediments that passed through the sieve were discarded. The fraction that was saved, larger than 80 $\mu$ m, was washed back into a cup and a small amount of non-ionic surfactant was added. The sediments were mostly very fine sands after sieving out the smaller clay fraction, but did have clumps of clayed particles. The sediments were stirred and the surfactant aided in de-clumping the clayed particles. The cups were covered and left for several days, with occasional stirring. The sediments were dried in an oven at 50° C and examined under the microscope (x 100 magnification).

Absolutely no *Ostracodes* were discovered in any of the core horizons examined. Because of this, it was decided not to do any coherent pollen profiles, which might turn out to be of limited value in this context. The implications are telling and indicate that no surface water exposure to sunlight and photosynthesis, or eolian (wind blown) organics lent a part in living organisms becoming part of the fabric of the depositional environment of the >80 $\mu$ m (often referred to as slimes) core sediments. This could mean that no lake existed in the study area at anytime within the framework of the depths of our coring since the last glacial epoch. Or it could just mean that *Ostracodes* were not preserved or present in our limited analysis. One could argue that three sample plugs are not fully representative, but without *Ostracodes* it could be difficult to make a case that the subsurface sediments were viable to life and living organisms. Other interpretation

could be imagined but might be impractical given reasonable alluvial denudation rates in a valley such as the Sacandaga Basin. One would expect the sediments brought about by erosional or due to a fluvial nature would show some degree of rounding and these sediments do not. Therefore, my conclusion is that if a lake existed either now or since the last glacial epoch, life ought to have left a fingerprint, at least some organics would be detected at depth (see page 32 and 33 – methods) and that the total overburden is probably the result of a single glacial event in the late Pleistocene.

### *Interpretations*

In the New York State Museum Bulletin No. 280, 1909, Brigham recognized what he called strong moraine, recessional moraine, outwash terraces and kames in the Sacandaga Valley, north of Northville, New York. He even points out that the Northville Quadrangle Topographic Map adequately shows these features. He was convinced though, that the Sacandaga Vly, represented a former glacial lake he called “Glacial Lake Sacandaga” (Brigham, 1909). Perhaps the basin, covered in trees, grasses and swamps, did not lend itself to easy field work and he was only able to interpret the Sacandaga River shorelines.

The aerial photographs used by the United States Department of Agriculture Soil Conservation Service in preparing the 1971 General Soils Report of Fulton County, New York, tell another story (Davis et al, 1971). Here, the mapping clearly suggest, the existence of remnant end moraine sections forming further arcuate bands south of Northville, New York, particularly at Hampton Point, Northampton Marina and across the Kenyon Island Group.

Landsat Thematic Mapper Imagery, obtained from the Eros Data Center in Sioux Falls, South Dakota also indicate the existence of morainal ramparts (morainal ramparts as described by Jackson, 1997) throughout the Great Sacandaga Lake. I used Thematic Mapper Spectral Bands 1: (0.45 to 0.52 $\mu\text{m}$  – blue), 4: (0.76 to 0.90 $\mu\text{m}$  – reflective infrared) and 7: (2.08 to 2.35 $\mu\text{m}$  – mid-infrared) on Idrisi 32 software to run an Unsupervised Land-Cover Examination with isodata clustering classification to study the land cover around the Great Sacandaga Lake (Jensen, 1996). I also used orthophoto's and the USGS Topographic map to improve the image classifications, obtaining close to 80% accuracy. The images used were obtained on 11/08/1984 and 09/16/1988, during exceptionally low water periods for the reservoir and were in raster format. The land cover classifications are very similar to the bathymetry of the GSL Navigation Maps.

Examination of the Coulter LS Particle Size Distribution Cumulative Curves in Figures 3.2a and 3.2b from 24 different sediment horizons shows that the mean of volume percentages is about 130 $\mu\text{m}$  (0.15mm). The consistency is remarkable given the possible range of depositional environments that could be imagined, such as fluvial, glaciofluvial, floodplain or lacustrine. In fact, the particle grain sizes that most reveal the parent material are somewhere between fine sand and very fine sand. The transect of core probing covered over 2 miles in length, diagonally across more than 5 $\text{mi}^2$  of the shallow water shoal study area. Except for horizon (0483770) in core #2, which was almost entirely fine sand with some clay, little variation existed. Glaciers, which erode and deposit sediment in a very distinctive manner may be the reason the deposition appears to be subglacial.

Glacial sediment is entrained in various ways and the mechanisms that transport and deposit clasts by ice make them notoriously poorly sorted, angular and heterogeneous. Clasts of a wide variety of rocks, derived from whatever lithologies lying upstream from the deposit, float in the predominantly  $<50\mu\text{m}$  fine sand and clay “rock flour” matrix. The matrix is not the product of normal aqueous flow, but rather, deposited by a mode of mass transport in glacial ice. The sediment mix is generally characterized by a disrupted gravel framework, within which, stones are not generally in contact. The sediment mix is unstratified and notable for containing more matrix than gravel sized clasts (pebbles are less than 10%). All material incorporated on, within or near the base of glacial ice settles out slowly, varying with the seasons. The high viscosity and low velocity of moving ice permit no sorting and little abrasion, although grinding of the bedload against valley walls and floor can generate additional fine-grained sediment. Glacial loads are plastered onto the substrate as ground moraine (lodgement till) or accumulate wherever there is a balance between rates of ice flow and melting. This accumulation produces end, recessional, and terminal moraines (Prothero and Schwab, 1996).

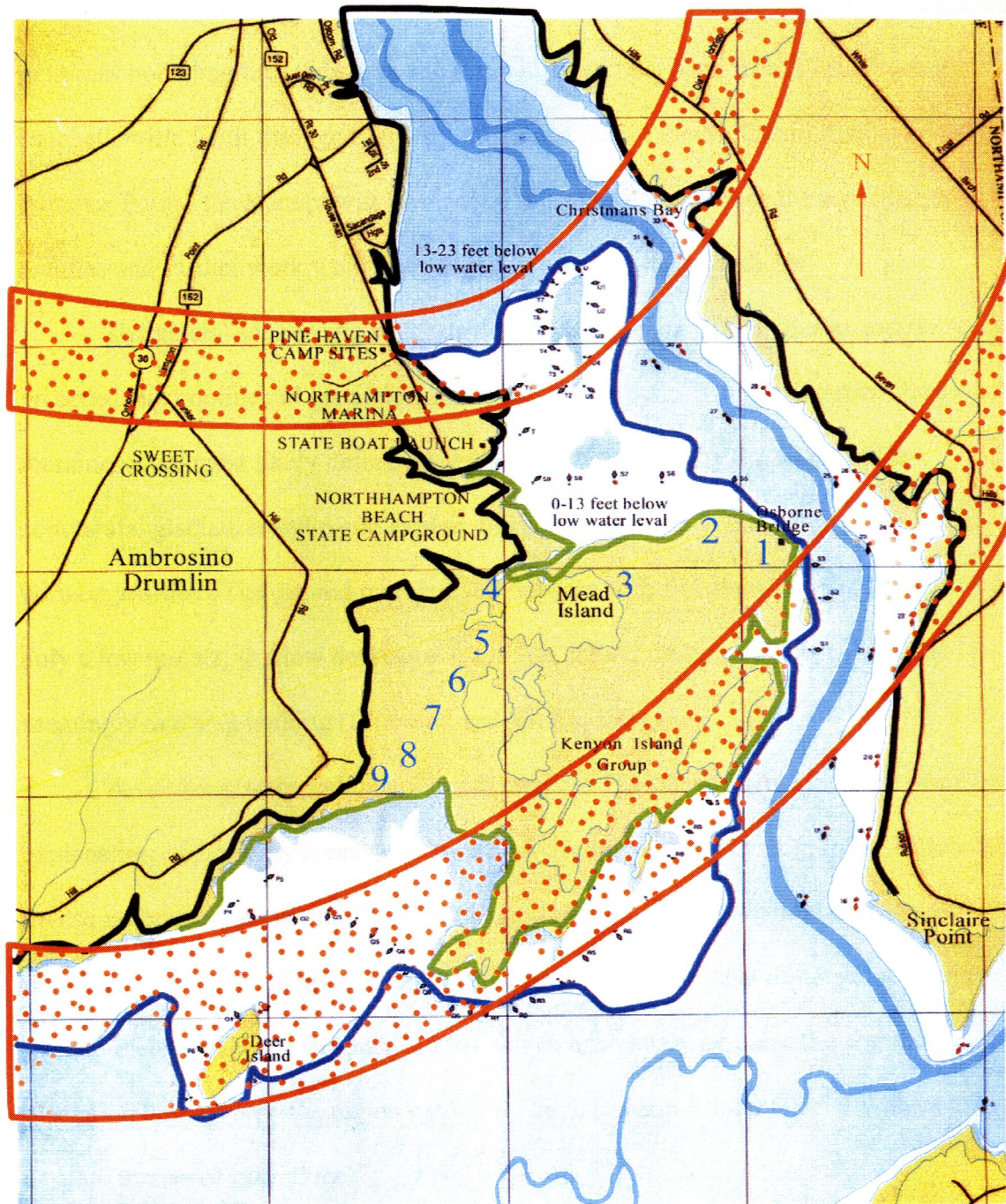
There must be supporting field evidence to identify ground moraine or deformed lodgement till including varved clay sequences (freeze and thaw cycles) and very large boulders forming lateral, end and/or terminal moraines (Prothero and Schwab, 1996). In fact, the shallow water shoal study area has all these features. The transect of cores lies almost entirely within the ground moraine field and its sediments are deformed around the flutes, kames and drumlins as stoss and lee topography. The kettle lake sediments created by the superincumbent glacial ice were deformed by the overlain lobe weight.

The study area may have been part of a sequence of outlet glacier (outlet glaciers as described by Jackson, 1997) advances called morainal ramparts. Each rampart looks as if it were an amphitheatre at the mouth of the valley, whose leading edge develops as an end moraine, its outside edges develop as lateral moraines and its central portion develops as ground moraine or deformed lodgement till (Prothero and Schwab, 1996).

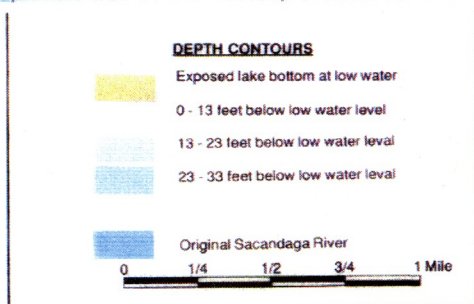
Outlet glaciers, may explain the morainal rampart topography exhibited by the study area. The skeletal end moraine to the south and east which makes an arcuate path south of Osborne Bridge across the Kenyon Islands, Deer Island, past the Ambrosino Drumlin and Cranberry Creek and to the southwest of Tamarack Swamp forms the rampart boundary of the study area to the south. The skeletal end moraine to the north and west makes a smaller arcuate path from the Johnson Road, Christmans Bay, through Northampton Marina, underneath the Ambrosino Drumlin and to the north of the Tamarack Swamp makes up the northern boundary of the study area (Figure 3.5). The greater area ( $\sim 12mi^2$ ) makes up the whole of the area confined by the two skeletal end moraines described (see USGS Topographic Map, Northville Quadrangle, 1970).

These are not all inclusive of the morainal ramparts produced by the Sacandaga Outlet Glacier during the late Pleistocene. At least one other skeletal end moraine can be described farther north of these two. This one occurs in an arcuate path across the valley from Gould Hill to just north of Hampton Point. Perhaps, based on topographic and navigation maps another existed just south of Northville across to the old Scout Island and to the north of the Mountain Road.





**Figure 3.3 Skeletal End Moraines bounding study area, Jimapco GSL Navigation Map, 1991.**



- High water mark
- 0-13' below water
- Shallow water shoal
- Skeletal End Moraine
- 1-9 Geoprobe Boreholes

A terminal moraine most likely existed as the final rampart, south of the two skeletal end moraines bounding the study area and it ran in an arcuate band from Fish House and the Batchellerville Fault through to the southernmost end of the Mountain Road, just north of Paradise Point. No attempt will be made in this report to convince the reader of these realities and further work would be needed to confirm this hypothesis.

The glacial sequencing suggests that the Sacandaga Outlet Glacier may have preceded any continental glaciation and was overlain by it. The ground moraine and end moraines were most likely deformed and probably smeared by the weight of the continental glaciation and could explain the deformation and correlation issues in the core horizon profiles. The ground moraine is consistent with flat plains having local relief of only a few meters, shallow depressions with no natural drainage outlet and gullies seemingly draining nothing (Lillesand and Kiefer, 1994).

An outlet glacier and the morainal ramparts it produced seem the only plausible explanation of the study areas depositional environment and its topography. Microscopic investigations reveal essentially no rounding of grains and they remain sharp and angular throughout. For a good general explanation of the glaciation that affected the region in the late Pleistocene (see Brigham, 1929), which adequately explains the continental glaciation that covered the region including the Sacandaga Outlet Glacier with as much as 1000 meters of glacial ice.

## CHAPTER 4

### MINERAL WEALTH

#### *Theories behind the study*

Any deepening for this part of the Great Sacandaga Lake would involve dredging or earth moving by heavy equipment. Either of these approaches to permanently deepening the lakebed would be very expensive and careful consideration of many variables, especially in the case of dredging, would need to be evaluated. Earth-moving heavy equipment would be easier and less costly, but the only way to recover these costs would be through the new real property created, or Island rebuilding and reconfigured lakefront property values. The current "State of the Art" dredging machines can pump up to 500 cubic yards of material (solids up to 50% content and move objects as large as 8" in diameter) and are the method of choice in deepening projects, where "heavies" (mineral wealth) need to be recovered. Heavies minerals have a specific gravity greater (usually  $>2.85$ ) than the surrounding sediments (a detrital mineral less than 1% in most sediments forming a minor constituent or accessory mineral). They typically include magnetite, ilmenite, zircon and noble metals and they may be recovered to offset the cost of operations. Simultaneous to extracting lake bottom sediments and separating heavies, the dredged slurries are pumped as much as 3000-4000' to be used to rebuild or reconfigure adjacent islands. Dredging machines are particularly ineffective in consolidated sediments such as clays. Dredging and pumping abilities are diminished and additional implements need to be added such as, auger heads with clay degraders or bucketwheel cutterheads to disaggregate the clay. Because clay sized particles remain suspended longer due to their shape and specific gravity, serious turbidity problems could

also be encountered if dredging was used. The topic of island rebuilding will be examined more fully in Chapter 5.

Bedrock studies and assays of the slope sediments in the feeder streams to the Great Sacandaga Lake have revealed bedrock units that contain interesting minerals (Miller, 1911). Glaciers in the Sacandaga Valley indicate that the deposition of the sediments of the study area must have come from an up-ice or upstream bedrock and overburden. There are known granite pegmatites (Miller, 1911) around the Sacandaga Valley and glacial erosion probably accounts for the many billions of tons of sediment atop the bedrock of the study area and adjacent morainal ramparts.

Dredging and separating the minerals of the study area is the only method that has the potential to generate profits beyond the altruism of the deepening cause. If the sediments have native elements, oxides, sulfides or even any of the pathfinder elements that could point to bedrock mineralization, I can then evaluate their economic importance. I have found mineral abundances and/or mineralizations using ICP-MS (Inductively Coupled Plasma Mass Spectroscopy), ICP-AES (Plasma Atomic Emission Spectroscopy) or standard "Fire Assay" procedures. The multi-element analysis using ICP-MS requires a tri-acid digestion procedure, which produces "near total", extended sensitivity digestion to selectively leach a much wider spectrum of elements. The Geoprobe core horizons that looked interesting and had the most colorful sediments, after cleaning-off small samples of their coatings with weak acid, were sent off for analysis. One hundred gram samples were split from the liners, marked and sent off, none of which contained any cleaned (no coatings washed-off) horizon sediments.

Multi-element analysis was performed on 26 Geoprobe samples and one nearby bedrock unit sample of the Potsdam Sandstone (from the Ridge Road by Carpenters Corners) as identified by nearby water well records FU 141, 0740900 NAD27 of the USGS, (see Appendix C). Facilities exist between the University at Albany and Rensselaer Polytechnical Institute to analyze the samples, but it was decided that due to the potential economic importance of the sediments, more credibility would be attached to the results from an outside independent source. The Assaying and ICP-MS multi-element analyses were performed by ALS Chemex Labs, Inc., in Mississauga, Ontario, Canada and Reno, Nevada (see Appendix D). Forty-Four (44) elements were analyzed and the results are displayed as ppb (parts per billion), ppm (parts per million) or as % (percent).

Occurrences of minerals in much higher abundances and worthy of recovery are not commonly found in glacially derived sediments. If the Sacandaga Outlet Glacier derived its sediments from bedrock that has economically important minerals, we might have deposits of these ore minerals at our disposal. Mineral deposits of economic value in sedimentary deposits (i.e. placers) are those in which minerals are sufficiently high enough to make it economically worthwhile to extract. Because minerals are so widely distributed in many common rocks, the issue of resources and reserves is dominated by costs of recovery and selling price.

In sediments, if the elements are locked-up in mineral ores with specific gravities above 1.8-1.9 and if the particles are no smaller than 20 $\mu$ m, dredging recoveries are commonly greater than 95% after washplant operations (Hartman, 1996). The sediments in my study area have a mean size above 125 $\mu$ m (>0.15mm) as confirmed by the Coulter

Laser Grain-Size analysis. This point is central to our whole study because there is an advantage to dredging unconsolidated sediments and mining them as *placer deposits*. In fact, some mineral deposits in Quaternary sand bodies are currently of greater importance than deposits of any other type. For example, titanium oxides from shoreline deposits account for virtually all-current production and 36% of all identified resources of titanium in the United States and 55% for the entire world. The current economic mineral ore rutile ( $\text{TiO}_2$ ) in placer deposits in Australia, have cutoff grades (0.1%  $\text{TiO}_2$ ) below the average crustal abundances of  $\text{TiO}_2$  (Force, 1991). Another advantage accorded to dredge mining is that it is a completely non-chemical extraction and environmentally friendly. Perhaps the greatest advantage though, is the low cost of equipment and operational costs, one-tenth of hard rock mining, making placer mining profitable even with very low-grade mineral ores (Holland and Petersen, 1995).

The trace metal analysis and assay abundances in my core horizon samples are listed in Appendix D. They represent only 44 of the more than 100 fundamental substances that consist of elements of only one kind. The absence of the remaining 66 elements, with a few exceptions, was chosen by design and meant to maximize the potential economic opportunities a specific gravity separation of the elements, such as dredge mining, would offer. Two elements that should be considered in further research include tin (Sn) and zirconium (Zr).

Heavy mineral mineralogy and assays of mineral fractions were used to identify those heavy minerals of economic importance. A thorough, microscopic study has been accorded the sediments of the study area and several elements pointed out in the trace element analysis have been specifically identified in their mineral ores. Several other

identification methods were employed in this study such as mineral associations; many minerals exist in sympathy with other minerals and sometimes only with other minerals associated with specific rock types (Chesterman, 1995). Magnetization and fluorescence studies were used to help identify or eliminate suspected mineralogy. A radiological study was used because of rare earth element enrichments discovered in the study area sediments (see Appendix E). Using a Landers, Frary and Clark Radiological Survey Meter (Geiger counter), made by the Victoreen Instrument Company of Cleveland, Ohio (air-ionization chamber) the sediments registered 0.11 R/hr (Roentgens per hour). A Roentgen is a measure of radiation intensity related to the ionization of air. The dose to soft tissue is about 10 milligray when exposure is 1 Roentgen.

Trace elements (those present in amounts less than 0.1% in general) play an important role in defining the nature of rocks and from which igneous partial melts they are derived. They are valuable indicators because the extent to which different minerals incorporate these elements varies sharply as a function of temperature, pressure or composition of the fluid from which the minerals crystallized. Of special interest are barium, rubidium, strontium, zirconium and the rare-earth elements. The incorporation of trace elements in igneous rocks also depends on the crystal structures of the minerals crystallizing from the melt. Minerals such as rubidium, strontium and the rare-earth elements (REE) tend to remain in the melt longer because their ionic radii and charges prevent them from incorporation into simpler structures (Raymond, 1995). Most of the sediments in the study area should be from igneous and metamorphic Grenville gneissic rocks of the Adirondacks. A correlation matrix was incorporated in an attempt to discover the rock provenances of these minerals.

The next important result of the multi-element analysis is the the role played in identifying or dismissing Dispersion Patterns; the patterns of distribution of chemical elements, especially trace elements, in the wall or surface rocks of an ore body or in the surface sediments surrounding it (Smee, 1983). The sediment in the study area lies atop bedrock that could contain mineralizations. A simple pH and ionic conductivity measurement on suitably prepared sediment horizon slurries may be a sufficient indicator of diffusion through the overburden. Any variations to background levels of elements in the sediment horizons over an ore zone would indicate an anomaly (Smee, 1983).

Measurements of soil pH and conductivity on study area soil solution slurries were made following methods of (Rowell, 1994). Trace element dispersion pattern anomalies, did not occur in our sediment horizon samples (see Table 4.1).



**Table 4.1**

**pH and Conductivity Measurements  
Great Sacandaga Geoprobe Boreholes**

<b>Borehole</b>	<b>Weight (grams)</b>	<b>pH*</b>	<b>Conductivity†</b>
0427995	11.5	6.25	.220
0433040	9.5	6.40	.275
0480005	12.5	6.20	.275
0470418	12.0	6.25	.275
0501030	12.0	6.35	.250
04892128	14.5	6.60	.165
0491832	11.5	6.00	.475
0423248	9.5	6.55	.160
0480509	10.0	6.05	.245
0497896	15.5	6.15 +[6.10]	.235+[.005]
0425579	10.0	6.10 +[6.20]	.135+[.015]
0482333	9.5	6.10	.140
0483770	22.0	6.40 +[6.35]	.180+[.050]
0480923	13.5	6.00	.210
0450012	12.0	6.00	.270
0420832	9.5	6.10	.180
0430006	8.0	5.95 +[5.85]	.195+[.040]
0420008	10.5	5.90	.155
0461023	10.5	5.90	.180
0440006	7.0	5.45	.430
047122140	12.5	6.10	.110
0487092	12.5	6.30	.150
0483337	11.5	6.10	.230
0470004	11.5	6.00	.190
0467896	13.5	6.10	.160

\* Hanna Instruments pH Meter (HI 9218) with Stability Indicator

† Solinst Conductivity + Temperature (°C) [measured in (µS x 1000) @ ~ 16.8 °C]

+ variations with agitation

Inductively Coupled Plasma (ICP) Spectroscopy developed in the early 1980's offers a way to carry out simultaneous analysis of large suites of elements including pathfinder elements associated with noble metals. Plasma Mass Spectroscopy (ICP-MS) offers the added advantage of introducing diverse selective extraction procedures for the detection of trace element suites. The object is to isolate and quantify that portion of the element in a sample horizon, which is relatively loosely bound, indicating prior mobility, either from depth (if mineralizations exist in the bedrock below the overburden) or if they were transported from short distances away, as is the case of our study area.

The secondary weathering environment is of fundamental importance and should be examined, because enough time has passed since the last glacial epoch in the study area. Secondary iron and manganese oxides occurring as coatings, concretions or as discrete particles in soils like those of the study area are significant because of their strong scavenging of important ore metals. Fe and Mn are pathfinder elements and are particularly important as scavengers of silver (Ag) and cobalt (Co). Although most sediment contains much greater amounts of Fe oxides than Mn oxides, the later possess greater sorption capability for trace elements. Such reactivity is due to characteristics of Mn such as: (1) it exists in several oxidation states (I, III, IV); (2) it forms non-stoichiometric oxides with different valencies; (3) its higher valence oxides exist in several crystalline (e.g. birnessite, lithiophorite, pyrolusite) or pseudocrystalline (manganite) forms; and (4) it forms co-precipitates and solid solutions with Fe oxides, owing to their similar chemical properties (Chao and Theobald, 1976).

There are various mechanisms of binding labile elements (water-extractable, organic and oxide-bound) in the secondary environment, including physical and chemical

sorption, precipitation, chelation and complexation. Phases present in the study area sediments, which are likely to scavenge free elements other than amorphous Mn and Fe oxides, are humic and fulvic acids of soil horizons. The characteristics of the sediment and soil profile depend not only on the parent rock material but also on climate, topography and biological activity, all issues addressed previously in this study. I know from this study that some of the Geoprobe cores had appreciable quantities of fine sands and chlorite clay. Adsorption of metals by soil constituents can be described as specific or non-specific. Specific adsorption implies exchange of cations and anions forming covalent bonds within ionic lattices. Metals most able to form hydroxyl (OH<sup>-</sup>) complexes such as hydrous oxides of Fe, Mn and Al are the major constituents that are responsible for this type of adsorption. A soil's cation exchange capacity (CEC) is a measure of its ability to provide sites for non-specific adsorption. The forces involved in the adsorption of ionic species on charged surfaces are electrostatic and governed by Coulumb's Law of Attraction and Repulsion (Press and Siever, 1986). CEC's of most soils range from a few to several hundred meq/100g and far exceeds anion exchange capacity as negative rather than positive charges dominate colloidal surfaces. The distribution of an element between soil solution and solid phases at equilibrium is governed by: (1) the density of surface binding sites for each component (e.g. clay mineral, hydrous oxides); (2) the binding intensity of the metal ion to each component; (3) the abundance of each component; (4) the chemical characteristics (e.g. pH, ligands) and (5) the concentration of other ions, major or trace, competing for binding sites (Hall, 1998). ALS Chemex's quantitative determinations such as by Fire Assay and Extended Sensitivity acid digestion leaches used in ICP-MS are designed to be specific to elemental concentrations in my

study area and to exploit elements from mineral ores that have economic potential, in today's markets.

Noble metal mineralizations, besides the obvious and most important native states of these elements, are associated with pathfinder elements such as tellurium (Te), antimony (Sb), arsenic (As), bismuth (Bi) and more rarely mercury (Hg). The trace multi-element analyses were designed to evaluate their potential noble metal complexations in our sediment horizons and also to evaluate the mineral associations of noble metals with lead (Pb), copper (Cu), chromium (Cr) and nickel (Ni). There are almost 75 platinum-group minerals, often derived from deep-seated slow cooling and more fluid ferromagnesian rocks such as gabbro and peridotite, both igneous-plutonic rocks. Because platinum-group minerals include palladium, iridium, osmium and rhodium, they have a wide range of specific gravities; much of the world supply of these metals is derived from clastic placer deposits. If platinum-group metals are found, we would expect to find enrichments in nickel, iron and chromium, which would be indicative of their close association and environment together as elements that crystallize out of a melt first (Cabri, 1976). Gold, because of its general unreactivity, occurs in nature in only about 10 mineral ores, principally with tellurides and pyrites or arsenopyrites, but sometimes with antimony and bismuth (Henley, 1975). Silver is complexed with about 20 mineral ores, mostly sulfides, arsenides, tellurides and sulfosalts. Finding silver would be more important as a guide to mineral associations than for its economic potential.

Rare-earth elements (REE) were known at the turn of the century as monazite sands, but these sands are principally sources of thorium (Th) and uranium (U) for use in

mantles in gas lamps and electrodes in arc lamps. The sands are found as littoral deposits and placer mined extensively on the seacoast of Brazil (Hedrick, 1995). The specific gravity of monazite is between 4.9-5.3, its color is yellow and its luster is resinous. The mineral crystals are resistant to weathering, hardness about 5 ½, and are derived from parent rocks of granite pegmatite, and to a lesser extent in carbonatites and gneiss. Monazite possesses slight and low radioactivity due to its Th and U content, which is strong enough to affect undeveloped film and can be measured by Geiger counters (Kithil, 1915). When first discovered, REE were known only in their oxide forms and because they resembled the oxides of the alkaline earths (CaO, BaO, etc.) and did not seem to form common minerals, they were labeled rare earths. REE really do not deserve their name because they are much more abundant than many other geochemically scarce metals and today have wide uses in many diverse applications. The REE are produced mainly from two minerals today, monazite (CeYPO<sub>4</sub>) and bastnaesite (CeFCO<sub>3</sub>).

Although the formulas of both minerals are written for cerium compounds, all of the REE substitute for cerium in their structures by atomic substitution (Boyle, 1982). There are 15 REE (see Table 4.2), starting with lanthanum and ending with lutetium. They differ only in the number of electrons in their inner electron shells and have very similar chemical properties. I designed our trace element analyses to look for cerium (Ce), lanthanum (La), thorium (Th), uranium (U), phosphorous (P-because of phosphates PO<sub>4</sub>) and yttrium (Y-because it behaves like REE) because I was interested in heavy mineral sands of current economic significance, with specific gravities heavy enough to dredge.

Most geochemical evaluations are considered on a population basis. That is, the larger population, like the one we designed for this study, will include multiple sub-

populations. Sub-populations include for example, the noble metals, Au, Pt and palladium (Pd) or those elements that noble metals are complexed with (pathfinders) such as As, Sb, Mn, Fe, Te, etc., the REE oxides and all of the elements that are associated with individual rock environments

**Table 4.2**

**The rare-earth elements. The relative amount of REE produced is in proportion to its geochemical abundance in the crust. (From Craig, 1988)**

Name	Chemical Symbol	Atomic Number	Geochemical Abundance (% by weight)
Yttrium*	Y	39	0.0035
Lanthanum	La	57	0.005
Cerium	Ce	58	0.0083
Praseodymium	Pr	59	0.013
Neodymium	Nd	60	0.0044
Promethium	Pm	61	0.0000
Samarium	Sm	62	0.00077
Europium	Eu	63	0.00022
Gadolinium	Gd	64	0.00063
Terbium	Tb	65	0.0001
Dysprosium	Dy	66	0.00085
Holmium	Ho	67	0.00016
Erbium	Er	68	0.00036
Thulium	Tm	69	0.000052
Ytterbium	Yb	70	0.00034
Lutetium	Lu	71	0.00008

\*Yttrium is commonly classed with the REE because its properties are so similar.

## ***Results and discission***

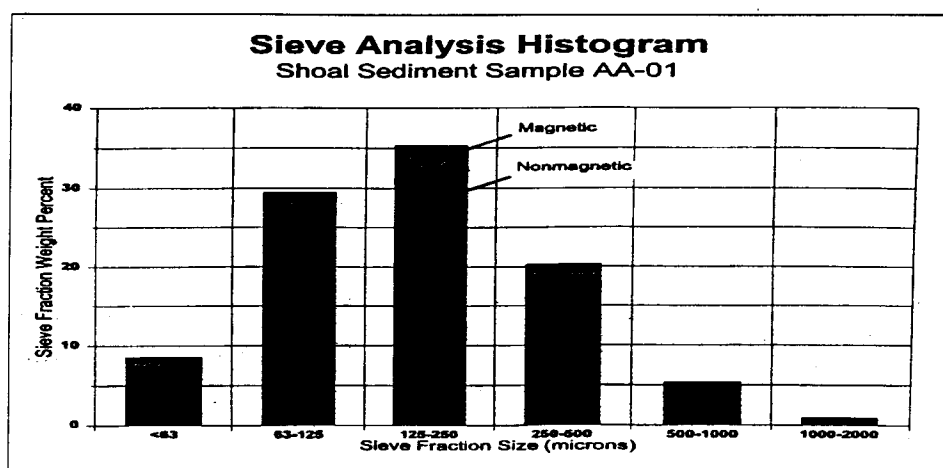
The geochemical results from the Great Sacandaga Lake study area are intriguing and promising. The fact that the sediments are related to glaciation and were locally derived means that the parent rocks deserve some further investigation. With that in mind, a hard rock sample from an adjacent outcrop, thought to be underneath the sediment overburden of the study area (USGS water well logs), was sent along with the Geoprobe core horizon samples to be analyzed by ALS Chemex. The results are listed separately in Appendix D and are not too different from the unconsolidated sediments of our study area. The rock is of course, much more ancient and metamorphosed and those elements that are more mobile like the alkalis Ca and Mg are slightly diminished. Those elements that are more labile like K and Na are slightly enriched and those elements that are more resistant or locked-up in minerals that are durable like those of the REE are about the same. Both the sediment samples of the study area and the hard rock sample would be considered weathered. Throughout this study and the results, the precise geochemical abundance backgrounds for weathered rocks in the literature is difficult to verify, but this hard rock specimen is not sufficiently enriched in any element that approaches ore grade. By incorporating otherwise “un-interesting” sample populations, though, we can get a glimpse outside of any bias the study area sediments may have introduced through secondary mobility of the elements.

The trace multi-element procedures designed for this study have given us a fair chemical analysis, but were only designed to yield the 44 elements (as in Appendix D) that would give us the strongest signals. The procedures avoid the conflicts of weaker signals, which arise using near-total tri-acid digestion. This also means that some of the

more resistant minerals and their abundances such as rutile (Ti) and REE like monazite (Ce, La, Th, and Y) may not be fully reported in this study. In further investigations, fusion and X-ray Fluorescence spectroscopy detection can be counted on for total element detection and will likely reveal somewhat higher element abundances in our sediments. Zircon (Zr) and cassiterite (Sn), also resistant minerals, are identified or indicated in the sediments and should also be determined by lithium metaborate fusion/ ICP-MS (absolute total digestion) in further investigations.

Black sands are ubiquitous in my study area sediments and every horizon sample sent-off for analysis included these sands. Rudimentary magnetization tests on these sediments reveal about 10-12% of the black sands are magnetic and less than 5% of the mineral aggregates are magnetic. Sieve analysis performed by myself and by Overburden Drilling Management Limited of Nepean, Ontario, Canada (the principal mineralogy company used for this project) are nearly identical. A histogram showing weight percent plotted against size fractions reveals the extent of the heavy mineral population of the study area sediments (see Table 4.3, below).

**Table 4.3**





A more thorough radiological survey was done at the Radiation Safety Lab at the University of California at Los Angeles and listed in Appendix G (including a decay scheme). The element isotopes of europium ( $^{152}\text{Eu}$ ,  $^{155}\text{Eu}$ ), Lead ( $^{210}\text{Pb}$ ), protactinium ( $^{234}\text{Pa}$ ) and thorium ( $^{234}\text{Th}$ ) were slightly elevated. Other lanthanide and actinide elements are also indicated. It was decided not to pursue a grams activity per sample analysis and leach test at this time because the gamma radiation appears to be no problem and the beta radiation has only slightly higher energy. No determination, without a leach test analysis will be speculated, regarding the danger of drinking from local water wells in the Sacandaga Basin. Investigations into dry beach sand recoveries of heavy mineral sands in Australia, containing monazite sands, indicate that air-borne particles have to be carefully damped to keep employees safe from radiological dangers. No such dangers are encountered by wet gravimetric dredge mining recoveries (Uranium Information Center, Ltd., Australia), such as the type we espouse in this report.

Based on a 20' averaged deepening of the study area total dredged sediment amounts to 92 million  $\text{yd}^3$  with all the calculations based on  $3.0 \times 10^{11}$  lbs of sediments (92 million  $\text{yd}^3 \times 3000 \text{ lbs}/\text{yd}^3$ ). The sediments actually weighed over 3300  $\text{lbs}/\text{yd}^3$ , but everywhere, quantities and potential worth were estimated downward and costs were estimated upward. If I was unsure of an element's mineralogy, occurrence or complexation, I opted to take the conservative approach and declare no value for purposes of this study. If we can identify mineral sand ore such as monazite and are unsure of its element mineralogy (REE all substitute for cerium in their structures by atomic substitution), then they were merged into one category because all of the minerals have high specific gravity, are considered quite valuable and are all easily separated

(Craig, 1988). Some elements of high abundance that exist as constituents of minerals were classed too light, based on their known occurrences, and others whose occurrences are not known but expected to be in minerals too light to be easily separated, were given a \$0 ore grade value until further work is done. Most elements examined, even though their ore minerals have potential economic value, are at or below average crustal abundances. Only those minerals that are clearly identified, to date, and exist in quantities appreciable enough to dredge successfully were assigned an ore grade worth. This does not mean that other dredgeable minerals are not recoverable and worth recovering.

The coatings on the sediments are identified as of this writing from the mineralogy as goethite (hydrogen iron oxide), with substantial amounts of manganese and iron. The coatings are heaviest in the heavy sand and gravelly horizons and may well be due to the freer flow of groundwater and the secondary-weathering environment.

There are no dispersion pattern anomalies in any of the sediment horizons and no enrichments of pathfinder elements either. In fact, the relative abundances of noble metals are equal to the average crustal abundances for these elements and no indication of pathfinder elements would be expected in this scenario (Tilling et al, 1973). This means that the overburden does not lie atop any noble metal mineralization and that no nearby up-ice or upstream noble metal bedrock mineralization is likely to be discovered either. This would also hold true for base metal mineralizations.

The pH and conductivity readings are as expected, uneventful as well. A more thorough, much narrower (testing every 100'-rather than every 800') gridwork of tests would need to be conducted if any anomalies were detected, but that does not seem to be

indicated in these sediments. Ionic migration is likely to electrochemically transport through as much as 100-150' of glacial sediment overburden, since the last glacial epoch, especially in areas of high groundwater content and/or clay sediment (Smee, 1983) and time and thickness should have had no effect on halos or dispersion patterns in our study area sediments.

In the secondary environment, anomalous concentrations of metals, derived from some mineralized source are taken into solution in the generally reducing and slightly acidic environment below the water table. Metal cations are incorporated into the interlayer structure of the phyllosilicates (clay minerals) by ion exchange. Phyllosilicates that have a large open structure and can offer sites to metal cations, even those with 8 and 12 fold coordination such as Cs and Sr with large radius ratios can become enriched. These clays often take on colorations (plaqueing) different than in their normal structures, with increased mass (Bradshaw et al, 1974). This does not appear to have happened to our study area clays and no different or unusual element enrichments or colorations occur in any horizons with elevated clay mineralogy. Cesium (Cs), rubidium (Rb) and lithium (Li) are often associated with lepidolite ores, which were identified in the sediment horizons. Lepidolite is a phyllosilicate mineral with a specific gravity of about 2.8-2.9. Some washplants have been designed with overflow filters to collect phyllosilicates plaqueed with noble metals, but for this writing, we chose to ignore this and assigned no ore grade value to them.

The trace multi-element procedures used on the study area sediments only pointed to element enrichments, but without total heavy mineral (THM) mineralogy done on the separable heavies of the sediments of the study area, we cannot assign any mineral

value to the results. Two companies were chosen for their special reputations in the field; Overburden Drilling Management Limited (ODM) of Nepean, Ontario, Canada as mineralogy experts and QIT-Fer et Titane (QIT) of Tracy and Sorel, Ontario, Canada as industry experts. At the University I attended to heavy mineral (heavies) separations in the same manner as these two companies, to isolate the heavies from the bulk sediment, do the sieve analysis for populations size-fractions of heavies and to identify and verify microscopically as best I could, the heavy mineral fractions that they might attain. This first step is accomplished by weighing from the archived sediment cores, collected using Geoprobe, stratigraphically averaged sediments from all the lithologies. I shipped just over 8.3 kilograms to ODM and used 1.0 kilogram for my own separations. Three 9-kilogram bags were shipped to QIT from sites around the Great Sacandaga Lake, including one from the study area, for bulk processing from an industrial perspective. Methylene iodide with a specific gravity greater than 3.2 (after re-use and dilution with acetone) was used by ODM and myself to separate the "heavies". QIT used electrostatic extraction, which separates conductors (oxides) from non-conductors (garnet, pyroxene and amphibole) followed by a magnetic fractionation at different field intensities to separate the various iron and titanium oxides, such as ilmenite, titano-magnetite and rutile, etc.

The results of these studies are all remarkably consistent, with noted differences (Appendix D). The descriptions can be seen in. Each of the three techniques measured between 8.5% and 12.5% (average 10.5%) THM in the study area sediments. My results and ODM both determined that the THM fraction resides essentially between 45 $\mu$ m and 1000 $\mu$ m, with, 1-2% of the THM residing above or below these fractions. The mean

(193µm) totals approximately 35% of the THM. The estimates of the heavy mineral populations by species of minerals from the ODM mineralogy are listed in Table 4.4, below.

**Table 4.4**

**MINERALOGY**

PERCENT	MINERAL	FORMULA
3	Augite	Ca, Na (Mg, Fe, Al) (Al, Si) <sub>2</sub> O <sub>6</sub>
3	Diopside	Ca Mg Si <sub>2</sub> O <sub>6</sub>
32	Hornblende	(Ca, Na, K) <sub>2-3</sub> (Mg, Fe <sup>2+</sup> , Fe <sup>3+</sup> , Al <sub>3</sub> ) (Si Al) <sub>8</sub>
3	Hyperthene	(Mg, Fe) Si O <sub>3</sub>
2	Zircon	Zr Si O <sub>4</sub>
2	Titanite	Ca Ti Si O <sub>4</sub>
7	Magnetite	Fe <sub>3</sub> O <sub>4</sub>
7	Ilmenite	Fe Ti O <sub>3</sub>
35	Almandine	Fe <sub>3</sub> <sup>2+</sup> Al <sub>2</sub> Si <sub>3</sub> O <sub>12</sub>
3	Apatite	Ca <sub>5</sub> (PO <sub>4</sub> ) <sub>3</sub> (F, Cl, OH, REE)
3	Lucfoxene	Ti O <sub>2</sub>

ODM and QIT used Scanning Electron Microscopy (SEM) and 200 grain-count methods to evaluate the THM. In addition to the above-mentioned minerals, QIT identified rutile (TiO<sub>2</sub> at <0.4%), monazite ((Ce, La, Y, Th)PO<sub>4</sub> at <0.3%) and zircon (ZrSiO<sub>4</sub> at ~1.0%). Potentially other minerals could be identified with continued grain-counts, but for practical purposes and given the reputations of the two companies, the results are substantive.

The most abundant mineral of economic significance is the garnet almandine, at nearly 35% of the THM; it represents more than 3.5% of the total sediments in the study area. Almandine is particularly hard (~7.5 Mohs scale) and brittle, with a specific gravity ranging from 3.56-4.32. Almandine generally occurs in Diorite of plutonic rocks (Chesterman, 1995). Examinations of the THM fractions by all the methods discussed above indicate sharp angular grains with essentially no rounding, consistent with glacial

sediments. Since the THM's are from averaged lithologies, the inference clearly indicates the transport mechanism is by entrainment in ice and not by fluvial deposition. This is a very positive circumstance for almandine because its principal use today is as an industrial abrasive. Additionally, the almandine exists in perfect size fractions and requires no crushing to be used as an industrial abrasive. In an averaged 20' deepening of the study area, 8.7 billion pounds of almandine would be recovered at 90% washplant efficiencies. No ore-grade values could be found in Standard and Poor's or Platt's Metal week, but USGS experts claim \$0.09-\$0.12 per pound Wyoming ore-grade almandine. Phone conversations with industrial mineral officials indicate \$0.05-\$0.06 per pound. Using \$0.09 per pound the mineral almandine is worth \$782,000,000 in the study area. Production costs are listed in Chapter 5.

Another of the most abundant and recoverable heavy mineral sands from the study area sediments are the oxides of titanium, principally ilmenite. Titanium minerals account for between 6.5% and as much as 20.8% according to ODM and QIT, respectively, of the THM population (expressed as  $TiO_2$ ). This represents 0.69% of the total sediments in the study area. A separate assay of the THM's, done again by ALS Chemex, verified this percentage (see Appendix F). The economically important Ti minerals in the study area THM's include; Ilmenite (average 48%  $TiO_2$ ), titanomagnetite (FeO mineral with up to 15%  $TiO_2$ ), altered ilmenite (leucocoxene, >55%  $TiO_2$ ), and rutile (>85%  $TiO_2$ ). Ilmenite deposits of magmatic origin (the anorthosite-ferrodiorite type) are currently the most important types of  $TiO_2$  mineral deposits. Ilmenite from magmatic ferrodioritic sources are known to exist just 30-50 miles north of the study area in Sanford Lake and Tahawus, N.Y. (Force, 1991). In an attempt to identify heavy mineral

provenance, the inter-element relationships were examined using a correlation matrix of elements normalized as a percent of total (see Table 4.5). Trace elements such as the noble metal assay results were eliminated. Several elements revealed significant correlations, with rare earth's (likely to exist together) and mafic elements (Ti and Fe), showing significance among themselves. Significance usually means that two elements when compared to each other have coefficients above  $\sim .60$ . The correlation matrix I generated was too large to fit in this document so I recoded the elements relevant to the THM's and while a fair fit to the regression line emerged, I was disappointed with the histogram and regression formula generated. The recoded correlation matrix ( $R^2$  -values) are presented below in Table 4.5.

**Table 4.5**

**Correlation Matrix of Selected Element Geochemistry**

	Al	Ca	Fe	Ti	Cr	La	Ce	Y	U	Th
Al	1.000	.218	-.205	-.309	-.731	.494	.555	.018	.454	.319
Ca	.218	1.000	.184	.297	-.049	-.185	-.053	.027	-.103	-.234
Fe	-.205	.184	1.000	.959	.231	.300	.370	.823	.427	.515
Ti	-.309	.297	.959	1.000	.382	.075	.160	.731	.332	.367
Cr	-.731	-.049	.231	.382	1.000	-.598	-.596	-.075	-.410	-.288
La	.494	-.185	.300	.075	-.598	1.000	.970	.662	.743	.823
Ce	.555	-.053	.370	.160	-.596	.970	1.000	.685	.779	.791
Y	.018	.027	.823	.731	-.075	.662	.685	1.000	.719	.746
U	.454	-.103	.427	.332	-.410	.743	.779	.719	1.000	.837
Th	.319	-.234	.515	.367	-.288	.823	.791	.746	.837	1.000

It is my contention that although the essential, accessory and associated minerals of ferrodioritic rocks correlate with the THM's and should be the likely source rocks for these THM's, the correlation matrix does not substantiate my conclusions. The only explanation I have for the inconsistency is that for robust calculations of correlation coefficients, the 28 samples (and 44 element variables) are too few to generate reasonable conclusions. In addition, the sediments may not be mature enough to exhibit clays scavenging mobile ions and as such, the elements do not appear to be correlating with aluminum. Nevertheless, the main purpose of this research was not to discover the rock provenances, but rather to identify minerals with enough economic potential to pay for a deepening of the Great Sacandaga Lake.

The titanium-mineral sands in our study area, principally ilmenite, at 0.69% of the total sediments of the study area equal 1.86 billion pounds at 90% washplant efficiencies. Ore-grade values from Standard and Poor's and Platt's Metal week indicate that these mineral sands are worth \$0.45 per pound. Phone conversations with industry officials indicate that prices may be as low as \$.15 per pound, until product chemistry has determined the contaminants in the ilmenite. QIT ran the ilmenite product chemistry (see Appendix D - Mineralogy). It shows the best magnetic fractions (using MicroProbe) contain 48%  $\text{TiO}_2$ , 39%  $\text{FeTi}$ , ~0.68%  $\text{Al}_2\text{O}_3$ , 0.03%  $\text{CaO}$ , 0.51%  $\text{MgO}$ , 0.02%  $\text{P}_2\text{O}_5$ , 0.47%  $\text{MnO}$ , 0.19%  $\text{SiO}_2$ , 0.03%  $\text{Cr}_2\text{O}_3$ , 0.26%  $\text{V}_2\text{O}_5$ , and 0.02%  $\text{ZrO}_2$ .

At a very conservative minimum estimate of \$0.15 per pound the ilmenite mineral sands are worth more than \$279 million dollars in the study area. Production costs are listed in Chapter 5.



Titanium metal has unique properties (density half that of steel, strength retention and atmospheric corrosion immunity) making it an ideal construction material for engines, airframes and structural purposes. Almost all Ti metal production involves tetrachlorination and reaction with either Mg or Na to form 85% pure metal and pig iron, called sponge. Ti sponge is worth from \$2-3.00 dollars per pound (Platt's, 2000). The same beginning process is used to make microcrystalline TiO<sub>2</sub> used for white pigment in paint, plastic and paper and is by far the largest single use of the metal. TiO<sub>2</sub> forms more than 20% by weight of paints and the pigment industry consumes more than 90% of all titanium minerals mined. The two communities of Gloversville and Johnstown with their under-capacity Wastewater Treatment Facilities and failing economies could benefit tremendously from further processing of titanium mineral sands. Benjamin Moore Paint Company has a new and very large facility in Johnstown and another Paint Company is planning to move to the area soon. Again, the titanium mineral sands estimated for the study area, and/or if the whole lake is deepened, do not have to be value added to, or further processed, to be worth what I anticipate to be their in-place value. It should also be pointed out here, that the whole lake may not contain heavy mineral sands such as REE or TiO<sub>2</sub> throughout its sediments in the same concentrations and further research will need to be done to estimate their abundance and potential worth.

The Rare Earth Oxides (REO) in the study area sediments are certainly enriched (see Appendix D). The fact that they exist in ore minerals such as apatite, zircon and monazite (already identified), that are durable and have high specific gravity makes them perfect candidates for dredge mining. Separately, the REO are quite valuable (Hedrick, 1995), worth between \$9 to \$2500 per pound, but even as a heavy mineral sand, they are

worth \$0.45-\$0.55 per pound (Table 4.6). Although REO are separated from one another in high purity by ion exchange methods, some of them, like cerium, are separated by taking advantage of its quadrivalent character. Today, many REO applications use a very substantial amount of the mixed REO just as they are obtained from their minerals ores or with extra cerium added or their lanthanum and cerium removed.

**Table 4.6**

**Rare-Earth Oxide (REO) Prices—by Various Corporations in 2000**

Oxide	Percent Purity*	Price per pound
Cerium	96-99.50	\$8.45-16.80
Dysprosium	95-96	\$38-60
Erbium	96-98	\$60-86
Europium	99.99	\$320-450
Gadolinium	99.99	\$55-60
Holmium	99.9	\$230-480
Lanthanum	99.99	\$8.75-10.50
Neodymium	95-99.9	\$11-40
Lutetium	99.99	\$2,000-2,500
Praseodymium	96.0	\$14.50-16.80
Samarium	96.0	\$30-34
Terbium	99.9	\$311-375
Thulium	99.9	\$1,600-1,650
Ytterbium	99.0	\$100-105
Yttrium	99.99	\$40-50

\* Purity expressed as percent of total REO (U.S. Department of the Interior, Hedrick, 1995).

Each of the REO has distinct individual properties and they form organic chelates, which are taken advantage of in modern ion-exchange methods of separation (Parker, 1984). Domestic mine production of REO increases every year because of their increasing demand in industrial applications and production is expected to increase substantially in the years to come (Hedrick, 1995). Apatite, zircon and monazite heavy mineral sands account for over 270 million pounds at 90% washplant efficiencies in this study area. Until the pre-feasibility study is completed, however, I will not assign any value to apatite and zircon as potential REO ore-grade minerals. My results indicate that 31 million pounds of monazite heavy mineral sands should be available in the study area at 90% washplant efficiencies with a value of approximately \$13 million dollars at \$0.45 per pound. Production costs are listed in Chapter 5.

It should be noted here again that the nearby communities of Gloversville and Johnstown in Fulton County could benefit tremendously from the mineral processing of these concentrates. Because of the tremendous capacity of the Gloversville-Johnstown Wastewater Treatment Plant, which was designed to accommodate a much larger leather industry (the leather industry is essentially a metals industry using metals and metal salts to denature rawskins), mineral's processing there should be a natural fit.

The last mineral abundance of economic significance in the study area sediments evaluated in this report are the magnetic fractions of iron, magnetite and titano-magnetite. ODM and QIT estimated between 7.1 – 7.6% (average 7.35%) magnetite in the THM population. This means that 0.77% magnetite exists in the whole sediment overburden of the study area or approximately 2.1 billion pounds at 90% washplant efficiencies.

Even though magnetite does not have significant ore-grade value, Platt's Metals weekly quotes are approximately \$35 per metric ton or \$0.015 per pound. Since magnetite has a specific gravity of 4.9-5.2 and is magnetic, it is easily obtainable and separable in washplant operations. In the study area, magnetite mineral ore worth exceeds \$30 million dollars. Production costs are listed in Chapter 5.

## CHAPTER 5

### SUMMARY

#### *Island rebuilding and shoreline reconfiguration*

The primary goal of this study was to identify potential mineral wealth in the sediments of the Great Sacandaga Lake study area and to evaluate the economic rationale for permanently deepening the lakebed as a remedy to water-level fluctuations that many find intolerable. If we chose to deepen with earth-moving heavy equipment, the lake would have to be drained and the costs would have to be covered by the new rebuilt Island property values and the shoreline reconfiguration improvements. A modest beginning project like the one proposed in the study area, with an ideal island configuration to maximize the shoreline worth, would create over 10 miles of shoreline and have a shorefront value of almost \$11 million dollars. The Hudson River-Black River Regulating District Board (HHBRRD) would be unable to hire the earth moving, roadwork and access bridges needed to access the property for many times that amount. The Board would need to issue bonds to pay for the costs of earth-moving equipment and apportion the costs along to beneficiaries, but would have to claim the improvements as assets because it couldn't sell the property to a developer at a loss. The downstream beneficiaries would gain a modest benefit due to the increase of more than 18 billion more gallons of water in the reservoir (just in this phase-20' deep x 4.5  $mi^2$ ), but would not be happy with the costs associated with the altruism of a deepening cause. The new lands permit (any newly created shoreline would be leased, like the current shoreline properties) beneficiaries would love the property, but both their permits and the permit

tenants on the other side of the lake would not be happy with the tremendous permit fees associated with such a project. Water-level drawdown would have to begin early in the season and the work would have to be done and completed in the coldest part of the year, which would make everyone displeased. And finally, if mineral wealth were available, an earth-moving heavy equipment method of recovering the mineral wealth could not be feasibly completed in a single season in the study area.

The water levels of the Great Sacandaga Lake are fixed historically, to regulate the flow of waters during times of high water and times of drought. The spillway at Conklingville is the point of regulation and it is unlikely that building a newer dam to raise water levels would be wise and in the interests of all involved. The remedy of choice would be to deepen the Lake and find a way to do it that benefits everyone including the HRBRRD. The only prudent method of deepening is to remove some of the lakebed so that the reservoir holds more water. The only plausible way to do this is to find some mechanism that maintains the continued operation of the reservoir and pays for itself at the same time. Therefore, the only logical way to proceed is with *dredge mining and washplant recoveries of the valuable heavies in the sediments of the lakebed.*

The primary investigations of the placer deposits (valuable heavies) have been done, at least in the study area and points upstream in the Sacandaga River. Considerable enrichments in heavy mineral sands that are worth recovering include almandine, ilmenite, magnetite, and the REO (residing in monazite, apatite and zircon heavy mineral sands). There are probably smaller but recoverable quantities of other multiple oxides, but the work to verify them has not been done yet. The heavy mineral concentrate mineralogy has been done and without any further mineral sand beneficiation, the

remunerations derived from the project prove that it can be accomplished without costs to anyone, satisfy all of the beneficiaries, and pay-off handsomely (see Table 5.1).

**Table 5.1**

**Heavy mineral sands. Estimated dredge mining worth of study area heavy mineral sands and projected worth after dredging the whole Reservoir.**

<b>Mineral</b>	<b>Study Area</b>	<b>Whole Reservoir</b>
Almandine	\$782,000,000	\$7,042,000,000
Ilmenite	\$279,000,000	\$2,511,000,000
Magnetite*	\$30,640,000	\$827,280,000
REE	\$13,000,000	\$351,000,000
<b>Total Potential Worth</b>	<b>\$1,104,640,000 Φ</b>	<b>\$10,731,280,000</b>

Φ Dredge depth 6.6 yd (twenty feet deep x 4.5 square miles) in three seasons (~seven months each).

The costs to operate a successful dredge-mining operation are well known from various sites around the world. In one titanium sands operation in Australia, using 500 tph (tons per hour) cutter-head suction dredges, 12 mo/yr, 7 days/wk, 24 hr/day in unconsolidated sands like Great Sacandaga Lake sediments, it costs \$0.32/m<sup>3</sup> without amortization, depletion, depreciation, interest or overhead. Running time is said to be ~95%. Recovery is >92%, determined by tailings, sand sampling and recovery of ore reserves, determined by drilling to be 98%. Floating washplants are used to recover, by specific gravity, rutile, zircon and ilmenite, using MDL spirals and Reichert cones (Hartman, 1996). Inflated estimated costs (in today's dollars) for recovering heavies from the sediments in the study area are listed in Table 5.2.

**Table 5.2**

**Heavy mineral sand dredging costs. Estimated costs to dredge mine the study area mineral sands and the projected costs to dredge the whole Reservoir.**

<b>Operation</b>	<b>Study Area</b>	<b>Whole Reservoir</b>
Employees	\$11,000,000	\$990,000,000
Dredging	\$45,000,000	\$405,000,000
Dredges	\$6,000,000	\$12,000,000
Wash Plants	\$15,300,000	\$30,600,000
Transportation	\$40,000,000	\$360,000,000
Executive Operations	\$1,500,000	\$13,500,000
Outside Contracting	\$500,000	\$4,500,000
Amortization & Interest	\$1,550,000	\$13,950,000
Insurances	\$3,000,000	\$27,000,000
Overhead	\$4,750,000	\$47,250,000
Tax Planning	\$5,800,000	\$52,200,000
Bridges (two in study area)	\$6,000,000	\$9,000,000
<b>Total Potential Costs</b>	<b>\$140,400,000Φ</b>	<b>\$1,965,000,000</b>

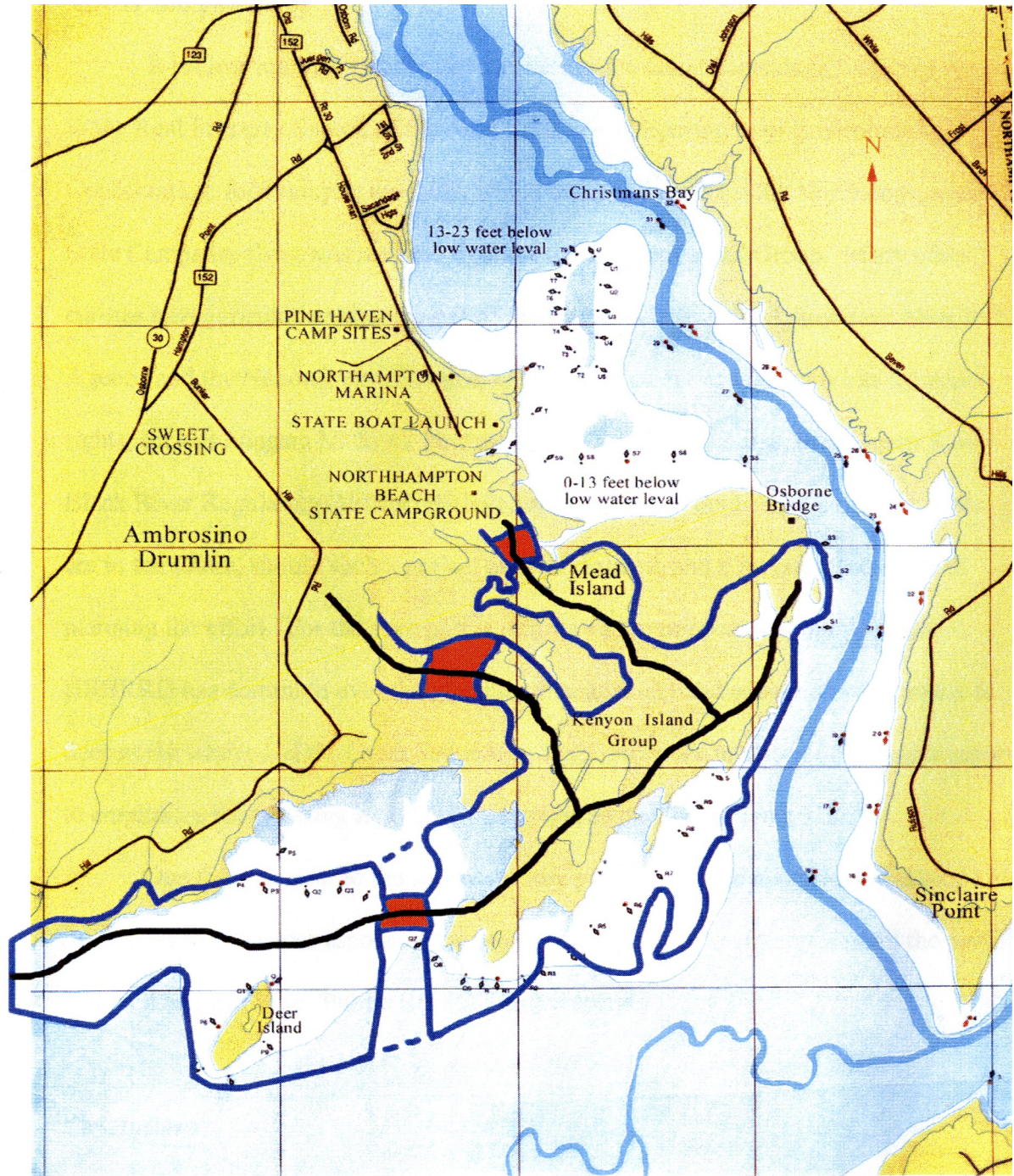
Φ Study area based on three seasons (7 month seasons x 24 hours per day).

The base of the rebuilt island would be shaped in an arcuate band across the islands of Deer and the Kenyon's, which is the skeletal end moraine to the east and south in the study area (see Figure 3.5). The remnants of the north and west skeletal end moraine extending from the Northampton Marina and the New York State Boat Launch would be dredged and removed to the greatest extent possible. End moraines are loaded with large rock debris and a late season heavy earth-moving deepening may be necessary to remove the shallow water threat of the north and west skeletal end moraine. Everything in between the two skeletal end moraines is dredgeable and worth a recovery consideration.

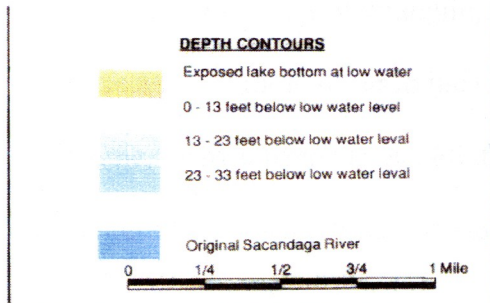


### ***Ideal Island***

The ideal island would have more than 10 miles of shoreline and be about 3.5  $mi^2$  in area. This arcuate ideal island would have as many appendages as practicable to provide as much shoreline, bays and inlets needed to give the new real property charm and character (Figure 5.0). Everything to the north and west would be dredged, the heavies removed and the tailings used to rebuild the island. Three scenarios for bridges could be imagined, no bridges, one bridge and/or two bridges to connect the roads to the building sites. In Table 5.2, the cost of building two bridges in the study area under the ideal island scheme, are included. Anything practical could be built on the ideal island, including a golf course, but it might also include a marina, a water filtration plant, wastewater treatment plant and homesites. A suitable name would be Osborne Crescent. There are 2240 acres in 3.5  $mi^2$ . The shoreline properties, over 530 of them, at 100' of beachfront and selling at over \$200 per beachfront foot would be worth near \$11,000,000. If the lots are 100' x 200' in length (~1/2 acre), that would use-up about 265 acres of real estate. Figuring about 10 miles of road right-of-ways and driveways (60' wide), uses-up only another 75 acres of real estate. The balance of the interior lands, some 1900 acres, can be used for anything, including restaurants and shops, parks and trees or even some kind of amusement park or resort. My preference would be to keep as much open space as possible, but I would expect the local people to have ideas of their own about how to successfully manage this newly created real property. The size of the ideal island could be somewhat different than proposed in the study area.



**Figure 5.1**  
**Approximate 3.5 square mile "Ideal Island,"**  
**Jimapco, 1991.**



- Proposed roads minus driveways
- Ideal Island "Osborne Crescent"
- Possible Bridges

### ***Access and ownership***

It is clear multiple entities have a stake in the Great Sacandaga Lake. According to the Real Property Bureau of the New York State Department of Environmental Conservation, for example, the DEC, which owns and operates the Northampton Beach State Campsite, also owns the land rights to the Kenyon Island Group. Many other entities have jurisdiction over the use of the lands, including the Adirondack State Park Agency and the National Park Service, while many others have interests in the water rights, such as Niagara Mohawk Power Corporation and of course, the Hudson River-Black River Regulating District (HRBRRD). There is no doubt that enormous obstacles are in the future, should such a project ever be realized, and a great deal of work in planning the effort. For the sake of this writing and in my personal opinion, the HRBRRD has dominion over the planning, management and execution of a project to deepen the lakebed of the Great Sacandaga Lake. No work has been done in this report to emphasize the planning aspects of this proposed deepening project.

One thing is clear to the success of any project with the magnitude and scope of work such as this paper espouses; the project has to satisfy the interests of all the parties involved and be in the “Public Benefit” for everyone.

### ***Conclusions***

So many years of wrangling and threats of litigation have passed since the vocal opponents of shallow water in the Great Sacandaga Lake began being heard. During all this time, many citizens from both the upstream and downstream interests have tried to come up with a palatable remedy to the navigation and recreational problems at the lake.

In this report, I have tried to show that a deepening project could actually be practical, environmentally sound and profitable at the same time.

All dredging operations are based on 5600 yd<sup>3</sup> per hour, with like washplant operation capacity. That translates to nearly 17,000,000 pounds of lakebed removed, heavies separated and tailings delivered to rebuild the ideal island in the study area per hour. Over 125,000 pounds of heavies would be separated per hour and over 3,000,000 pounds per day. This means over 60 truckloads of heavies will need to be transported to Amsterdam for transport along railways to a smelting plant, unless provisions are set-up to do the smelting in the Gloversville-Johnstown area. No provisions are made in this paper to build a smelting plant in the Gloversville-Johnstown communities, but the potential worth of smelted TiO<sub>2</sub> heavies increases to between \$1.8-2.4 billion dollars, just with the first year dredged TiO<sub>2</sub> heavies of the study area. The cost of transporting the heavies and the equipment to do it is estimated in Table 5.2. By the economy of scale, the cost of transportation increases only by the cost of fuel over the life of the project, if the whole reservoir is dredged for heavies. The total potential worth of smelted heavies if the whole reservoir is dredged, is \$20 billion dollars (a very conservative estimate).

The dredge pace is set to accomplish the 6.6 yd deep x 1.5 mi<sup>2</sup> task in the study area within a normal high water season, approximately seven months. The whole of the reservoir would take between 25-30 years to dredge and recover heavies. The dredging and washplant systems have been designed for proven efficiency, safety and visibility, operator comfort and particularly for quiet operations. No credit will be given the corporations who build and design this equipment because it can be built and supplied by several such companies.

The reservoir releases about 150 billion gallons of water each season. At >6 billion gallons of increased water capacity each dredging season, the removal of the lakebed will create over 180 billion gallons of drinking water capacity if the whole reservoir is dredged. This extra drinking water capacity, will be a priceless commodity in the new frontier. No attempt will be made in this report to quantify it as an asset. Certainly, a deeper Lake will have profound effects on the well-being of its fish and other fauna. If the deepening continues throughout the Lake, the disruption to the lakebed organisms (estimate 27 like-areas to be dredged) will be completely reversed in each area within a three-year period (based on discussions with local biology professors). The answer to this dilemma is to understand what fauna makes a living in soils, 6 months of the year and in water the remaining 6 months of the year?

The study area sediments proved to be extremely consistent throughout the sediment overburden of the study area. The heavy minerals identified are only slightly enriched from typical crustal sediments but as discussed they do have considerable value within the morainal ramparts investigated. It can be expected that at least another area of comparable size with sediments exactly like those of our study area can also be dredged for deepening and recovering heavy mineral sands, up the Sacandaga River. These two areas alone would take 6 seasons to dredge completely and be worth \$2.3 billion dollars. If the future studies of the GSL sediments indicate what I believe to be a terminal end moraine to the south and east of the Kenyon Island end moraine, at least another \$1 billion dollars worth of heavy mineral sands could potentially be recoverable.

This study has proved that a deepening project begun like the proposed study area project and expanded to include more and more of the lakebed sediments as appropriate, would be a perfect remedy to the dilemma of shallow water in the Great Sacandaga Lake. Furthermore, the public benefit and the economic advantages and spin-off businesses this project would deliver to the upstate economy are the obligation of the HRBRRD and the other regulating entities to allow to be brought to fruition. Much work needs to be done, but this project could develop through the planning phases, begun and start solving all the beneficiaries' problems within the next 5 years. It is my contention, that for future generations and the well-being of the industry of New York State, that an extra 180 billion gallons of drinking water is reason enough to undertake this project!

The next step I intend to take, beginning in the Fall of 2001 is to undertake a pre-feasibility study of the whole of the Great Sacandaga Lake. I intend to investigate the resource and take a more careful look at the methods of extraction and methods of processing the mineral sands. I want to investigate more thoroughly, the potential markets, capacities and costs and look carefully at potential environmental problems and remedies. There is a lot of work ahead to understand reasonably well about the quality and quantity of the mineral resources in the Great Sacandaga Lake. The objectives of this report have been successfully completed, though there is a long road ahead before investors might want to participate in a program to design, seek permits, start operations and begin marketing of these heavy mineral sands.

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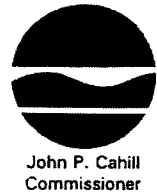
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# **Appendix A**

## **PERMITS AND HISTORY**

**New York State Department of Environmental Conservation**  
**Office of Natural Resources, Region 5**  
Route 86 - P.O. Box 296, Ray Brook, New York 12977  
Phone: (518) 897-1276 FAX: (518) 897-1370



January 27, 1999

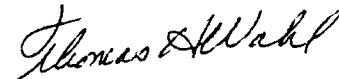
University of Albany  
c/o Arthur M. Ambrosino  
Dept. of Earth & Atmospheric Sciences  
1400 Washington Avenue, Room 351  
Albany, NY 12222

Dear Mr. Ambrosino:

Enclosed is a Temporary Revocable Permit which will allow you to take one inch soil cores with a maximum depth of 20', on a portion of the Northampton Campground. This permit is subject to the Terms and Conditions listed. **The permit will expire on April 1, 1999.**

Regional Forester Tom Martin, located in the Ray Brook Office, will act on behalf of the Department of Environmental Conservation as Regional Land Manager. Should you have any questions or concerns, please contact Mr. Martin at 518/897-1276.

Sincerely,



Thomas H. Wahl  
Regional Supervisor of Natural Resources

THW:mb

Enclosure

cc: T. Wolfe  
J. Grossman  
T. Kapelewski  
FR: D. Brooks  
ECO: M. Trottier  
File: TRP#2 File

New York State Department of Environmental Conservation

Reg 5 TRP 1996

TEMPORARY REVOCABLE PERMIT FOR THE  
USE OF STATE LANDS

(see page 2 for standard terms and conditions)

Name: University of Albany

Address: c/o Arthur M. Ambrosino  
Dept. of Earth & Atmospheric Sciences  
1400 Washington Avenue, Room 351  
Albany, NY 12222

Telephone Number: 518/442-4466

State Land Location: County: Fulton Town: Mayfield

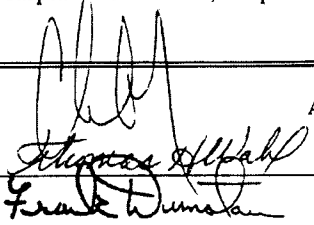
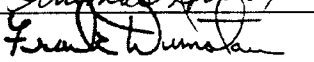
Land Designation: Intensive Use

Patent, Tract, Lot, Etc. Northampton Beach Campground

The Department of Environmental Conservation hereby grants permission to the above named permittee to use these State lands described above and shown on the attached map for the purposes stated on the attached application and as amended below subject to the standard terms and conditions listed on page 2 and to the following special terms and conditions. (Use additional sheet if necessary).

This TRP will allow you to take one inch soil cores with a maximum depth of 20', on a portion of the Northampton Campground.

Unless otherwise suspended or revoked, this permit shall expire on April 1, 1999.

STATE Museum  Regional Director  Central Office  (If required)	 	APPROVAL <i>1410-c 98</i>  Date <u>1/25/99</u>  Date <u>JAN 15 1999</u>
---	--	---

NOTICE OF EXPIRATION OF TEMPORARY REVOCABLE PERMIT

This is to inform you that this permit expired on  
and that all conditions of such permit have been satisfactorily met.

\_\_\_\_\_  
Date Regional Director

## **STANDARD TERMS AND CONDITIONS**

1. This permit shall at all times be subject to the approval of the Regional Land Manager and may be suspended or revoked at any time with due cause.
2. The permittee shall notify the Regional Land Manager at least 48 hours prior to commencing use and upon completion of use.
3. The activities authorized under this permit shall not interfere with normal administration of the area by the Department.
4. No damage will be done to State land, State facilities or boundary or survey markers.
5. The permittee is responsible for any inadvertent or deliberate damage caused by the exercise of this permit and will be held responsible for restoration, rehabilitation or repair at the permittee's expense.
6. No trees or other vegetation shall be cut, disturbed or removed unless specifically authorized by the Regional Land Manager.
7. The State land covered by this permit shall be kept free of litter and debris and be left in a condition satisfactory to the Regional Land Manager at the expiration of the permit or on completion of activities authorized by the permit.

**THE GREAT SACANDAGA LAKE ISLAND REBUILDING PROJECT:  
DREDGING AND REBUILDING THE KENYON ISLAND GROUP**

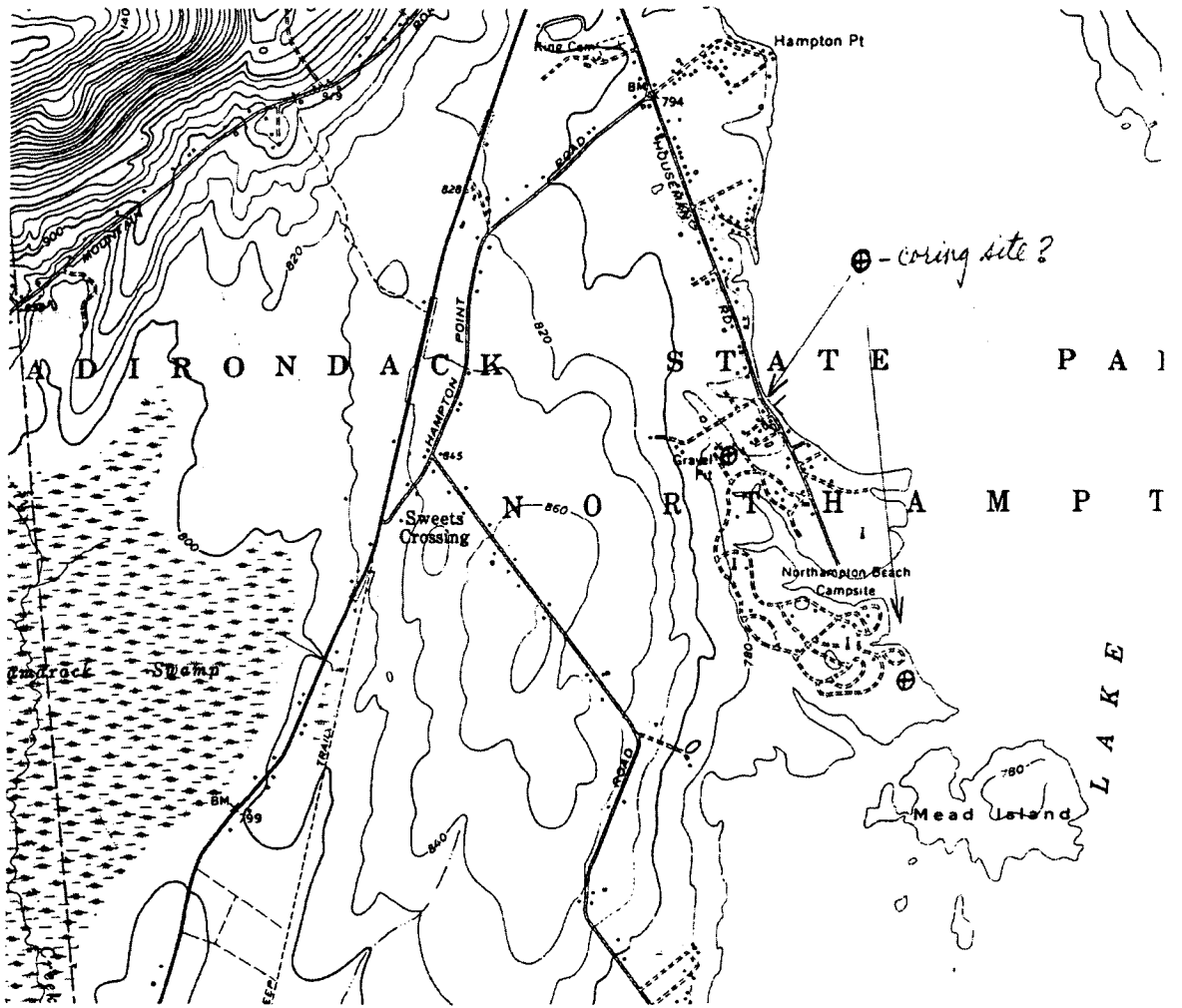
**AMBROSINO, Arthur M., Department of Earth and Atmospheric  
Sciences, University at Albany, 1400 Washington Avenue,  
Albany, N.Y. 12222, aa7148@cnsvox.albany.edu**

During the last decade, there has been increasing dispute between the downstream interests and the recreational users of the waters of the Great Sacandaga Lake. The downstream interests make up eleven power generation dams, Hudson River water/Ocean water salt front concerns at Poughkeepsie, and flood control at Albany, which was the principal reason the Great Sacandaga Lake was created. The recreational users are the landowners/boat operators, around the Lake.

Large water-level variations to accomodate the downstream interests cause large areas of very shallow water in the Great Sacandaga, especially in midsummer, which is a situation the boating public finds unacceptable.

Dredging is the only remedy to deepen the Lake significantly for boating; thus, it is of interest to know the types of sediments at various points in the Lake and what the environmental concerns would be. How many meters of sediments from the erosion of the islands have accumulated since the dam created the Lake in 1930, and how has the Basin filled with sediments since the last glacial epoch?

The Kenyon Island Group, perhaps the largest and most unnavigable shallow water shoal in the Great Sacandaga Lake, is the ideal location for such a possible dredging project. The island group is uninhabited, accessible and appears to be the ideal soil for dredging. The dredged tailings will be used to rebuild the islands. Core sampling should identify the underlying sediment stratigraphy and establish the feasibility of such a project and its worthiness. In addition, sediment cores will be analyzed and interpreted with respect to Climate Change since the last glacial epoch, approximately 15,000 years B.P..





STATE OF NEW YORK



Board of Hudson River-Black River Regulating District

Sacandaga Field Office

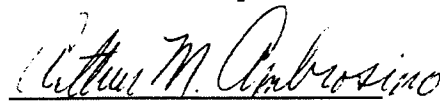
737 Bunker Hill Rd., Mayfield, New York 12117

Phone (518) 661-5535

FAX (518) 661-5720

August 20, 1998

On this date Mr. Arthur Ambrosino was allowed to borrow the original Power Commission maps to have them scanned and will then return them immediately.

  
Arthur M. Ambrosino  
Arthur Ambrosino

STATE OF NEW YORK



Board of Hudson River-Black River Regulating District

Sacandaga Field Office


737 Bunker Hill Rd., Mayfield, New York 12117

Phone (518) 661-5535

FAX (518) 661-5720

September 2, 1998

On this date Mr. Arthur Ambrosino returned the Original Power Commission maps to the Sacandaga Field Office.

  
Marlene Junquera



Board of Hudson River-Black River Regulating District  
350 Northern Boulevard, Albany, New York 12204 Phone (518) 465-3491  
FAX (518) 432-2485

**FAX COVER SHEET**

TO: ARTHUR AMBROSINO  
FROM: TOM BREWER  
DATE: 11/12/98  
SUBJECT: CORE SAMPLES

NO. OF PAGES (including cover sheet): 1

MESSAGE: THIS WILL CONFIRM THAT YOU ARE  
AUTHORIZED TO COLLECT CORE SEDIMENT  
SAMPLES IN THE AREA OF THE KENNON ISLANDS  
AT GREAT SACANDAGA LAKE ON 11/13/98.

*Thomas E. Brewer*  
CHIEF ENGINEER

IF YOU HAVE ANY PROBLEMS OR QUESTIONS RECEIVING,  
PLEASE CALL OUR OFFICE IMMEDIATELY.



Maps to swear by...not at!

LICENSING AGREEMENT

DATE: 8/25/98

CUSTOMER: Arthur M. Ambrosino - Sony Albany  
442. 4466

LICENSE NUMBER: C 982508 fax 492-5825

JIMAPCO herein authorizes the above mentioned customer to reproduce  
1 slide copies of Sacandaga Lake

JIMAPCO requires that the most recent edition be used and that the following copy appear on each reproduction:

" Reproduced with permission of JIMAPCO, Inc. Copyright 1998 " <sup>see map</sup>

This is as one-time license and expires with the one slide

SPECIAL INSTRUCTIONS:

Please call if you have any questions.

C Fisk  
Christina T. Fisk, Vice President

## History of the study area

The Sacandaga Basin has long been an important geopolitical region in the east-central part of New York State, no less important to the Native Americans who named the lowlands “Sacandog,” meaning “Land of Waving Grass.” As early as the 1860’s, the New York State Legislature considered the idea of constructing a dam and building a reservoir in the basin. The Sacandaga Basin contains approximately one third of the waters of the Upper Hudson Watershed. For many years, disastrous floods regularly wreaked havoc in the communities along the Hudson River, from Glens Falls to Albany. In the early 1900’s, the Legislature gave its blessing to the New York State Water Supply Commission to survey the basin and identify where a suitable dam should be constructed. In 1908 the Water Supply Commission’s task was completed and the proposed “Flow Line” and the site of the dam at Conklingville were identified. The State Legislature established the Hudson River Regulating District in 1922 as a public benefit corporation. The District was given a broad spectrum of legal powers to accomplish this mission, including the authority to build and operate the reservoir, issue bonds and apportion costs on its beneficiaries to finance construction, maintenance and for operations. The District’s specific responsibilities are to reduce floods by excess run-off and to augment Hudson River water flow at times of drought or other periods when river flows are low. Work began on the dam in 1927, was completed in 1929 and the valley was flooded by 1930.

The project cost \$12,000,000, and the 1100-foot Conklingville Dam retains close to 300 billion gallons of water in the reservoir at the high water mark. 125 miles of shoreline defines the reservoir’s 42 square mile surface area. Construction of the

reservoir took more than 500 men and several years to complete. It submerged 3872 graves in 22 cemeteries, which required transburial. It submerged 5 miles of the Fonda, Johnstown and Gloversville Railroad, which was never rebuilt. It submerged 67 miles of roads connecting several communities beneath the waters and required more than 50 miles of new highway construction. Five existing bridges were destroyed and 10 new bridges had to be erected. Finally, over 1250 structures were moved, burned or pulled down to make room for the water.

During the last several decades, there has been increasing dispute between the downstream interests and the recreational users of the water of the Great Sacandaga Lake. The downstream interests include 11 power generation dams, Hudson River water/Ocean water salt front concerns at Poughkeepsie and flood control at Albany. Large water-level variations to accommodate the downstream interests cause large areas of very shallow water in the Great Sacandaga, especially in midsummer, which is a situation the landowners and boating public find unacceptable.

The spillway at the Conklingville Dam is 771' above sea level. At high water the study area (Kenyon Island Group including Mead and Deer Islands) have shorelines at about 768'. The lake levels begin to drop long before the low water season approaches. Water is let out at an average rate of just less than 1 1/2" per day and by the low water season over 150 billion gallons of water have been drained from the reservoir. It takes approximately 6 months for the transformation from high water to low water level status. As the water is let out at Conklingville at the rate of a little less than 4' per month, the shallowest parts of the shallow water shoal that surrounds the Kenyon Island Group becomes more and more dangerous to recreational boaters of the lake. The situation

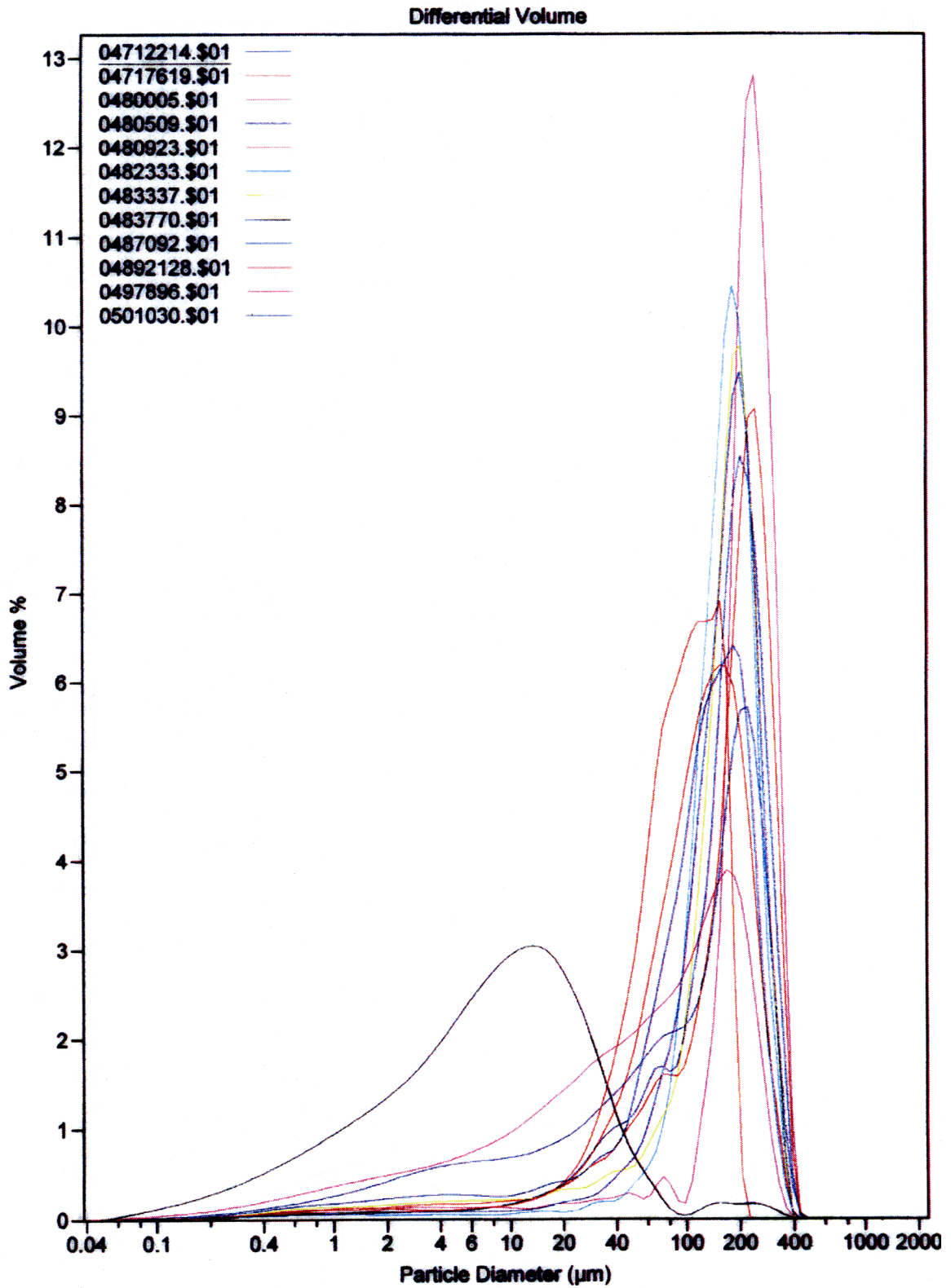
becomes so serious that almost nine miles of shallow warning buoys are strung around the greater shoal, warning boaters not to enter. Boaters are relegated to the marine buoyed Sacandaga River Channel for much of the boating season to get around the shallows.

In an effort to reach accommodations to all, an Upper Hudson-Sacandaga Comprehensive Settlement Negotiation Team is presently studying water level remedies (see Table below). The proposed remedy involves raising the high water mark to 770' over the next twenty years. Any remedy they agree upon will be a compromise to both upstream and downstream interests.

ADIRONDACK BOARDSAILING CLUB  
ADIRONDACK COUNCIL  
ADIRONDACK HYDRO DEVELOPMENT CORPORATION  
ADIRONDACK MOUNTAIN CLUB  
ADIRONDACK PARK AGENCY  
ADIRONDACK RIVER OUTFITTERS  
AMERICAN WHITEWATER ASSOCIATION  
FEDERAL ENERGY REGULATING COMMISSION  
FEEDER CANAL ALLIANCE  
FINCH, PRUYN AND COMPANY, INCORPORATED  
FULTON COUNTY  
GOMEZ AND SULLIVAN  
GREAT SACANDAGA LAKE ASSOCIATION  
GREAT SACANDAGA LAKE FISHERIES FEDERATION  
GREAT SACANDAGA LAKE MARINAS  
HUDSON RIVER RAFTING  
HUDSON RIVER-BLACK RIVER REGULATING DISTRICT  
INDIVIDUALS  
INTERNATIONAL PAPER  
MERCER COS  
NATIONAL PARK SERVICE  
NEW YORK RIVERS UNITED  
NEW YORK STATE CONSERVATION COUNCIL  
NYS DEPARTMENT OF ENVIRONMENTAL CONSERVATION  
NEW YORK STATE ENERGY GRID  
NIAGARA MOHAWK  
HUDSON RIVER MANAGER  
SARATOGA COUNTY  
TOWN OF HADLEY  
TROUT UNLIMITED  
UNITED STATES FISH AND WILDLIFE SERVICE  
WILDWATERS, INCORPORATED

**Appendix B**  
**PARTICLE SIZE VOLUME STATISTICS**  
**AND**  
**CORING INTERPRETATIONS**





**Figure 3.1a** Coulter LS Particle Size Analyzer, Sediment/Core Lab,  
Geology Department, Union College.

COULTER®

LS Particle Size Analyzer

6 Jul 1999

————— SEDIMENT/CORE LAB, GEOLOGY DEPT., UNION COLLEGE —————

File name: 04712214.\$01                      Group ID: 047122140  
Sample ID:  
Operator:                                      Run number: 2  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 9:32 30 Jun 1999                      Run length: 120 Seconds  
Pump speed: 100  
Obscuration: 2%  
PIDS Obscur: 20%  
Fluid: Water  
Software: 2.09                                      Firmware: 2.02 2.02

File name: 04717619.\$01                      Group ID: 047176192  
Sample ID:  
Operator:                                      Run number: 12  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 11:34 30 Jun 1999                      Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 3%  
PIDS Obscur: 26%  
Fluid: Water  
Software: 2.09                                      Firmware: 2.02 2.02

File name: 0480005.\$01                      Group ID: 0480005  
Sample ID:  
Operator:                                      Run number: 14  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 11:51 30 Jun 1999                      Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 2%  
PIDS Obscur: 19%  
Fluid: Water  
Software: 2.09                                      Firmware: 2.02 2.02

## SEDIMENT/CORE LAB, GEOLOGY DEPT., UNION COLLEGE

File name: 0480509.\$01                      Group ID: 0480509  
Sample ID:  
Operator:                                      Run number: 4  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                      Fluid Module  
Start time: 10:27 30 Jun 1999              Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 6%  
PIDS Obscur: 54%  
Fluid: Water  
Software: 2.09                              Firmware: 2.02 2.02

File name: 0480923.\$01                      Group ID: 0480923  
Sample ID:  
Operator:                                      Run number: 10  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                      Fluid Module  
Start time: 11:18 30 Jun 1999              Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 2%  
PIDS Obscur: 25%  
Fluid: Water  
Software: 2.09                              Firmware: 2.02 2.02

File name: 0482333.\$01                      Group ID: 0482333  
Sample ID:  
Operator:                                      Run number: 3  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                      Fluid Module  
Start time: 10:15 30 Jun 1999              Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 1%  
PIDS Obscur: 12%  
Fluid: Water  
Software: 2.09                              Firmware: 2.02 2.02

## SEDIMENT/CORE LAB, GEOLOGY DEPT., UNION COLLEGE

File name: 0483337.\$01 Group ID: 0483337  
Sample ID:  
Operator: Run number: 5  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230 Fluid Module  
Start time: 10:35 30 Jun 1999 Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 3%  
PIDS Obscur: 30%  
Fluid: Water  
Software: 2.09 Firmware: 2.02 2.02

File name: 0483770.\$01 Group ID: 0483770  
Sample ID:  
Operator: Run number: 15  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230 Fluid Module  
Start time: 12:00 30 Jun 1999 Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 22%  
PIDS Obscur: 91%  
Fluid: Water  
Software: 2.09 Firmware: 2.02 2.02

File name: 0487092.\$01 Group ID: 0487092  
Sample ID:  
Operator: Run number: 1  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230 Fluid Module  
Start time: 9:53 30 Jun 1999 Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 2%  
PIDS Obscur: 16%  
Fluid: Water  
Software: 2.09 Firmware: 2.02 2.02

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File name: 04892128.\$01                      Group ID: 04892128  
Sample ID:  
Operator:                                      Run number: 1  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 9:23 30 Jun 1999                      Run length: 120 Seconds  
Pump speed: 100  
Obscuration: 3%  
PIDS Obscur: 23%  
Fluid: Water  
Software: 2.09                                      Firmware: 2.02 2.02

File name: 0497896.\$01                      Group ID: 0497896  
Sample ID:  
Operator:                                      Run number: 16  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 12:10 30 Jun 1999                      Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 8%  
PIDS Obscur: 60%  
Fluid: Water  
Software: 2.09                                      Firmware: 2.02 2.02

File name: 0501030.\$01                      Group ID: 0501030  
Sample ID:  
Operator:                                      Run number: 1  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 8:41 30 Jun 1999                      Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 5%  
PIDS Obscur: 43%  
Fluid: Water  
Software: 2.09                                      Firmware: 2.02 2.02

## Volume Statistics (Arithmetic) 04712214.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%	S.D.:	89 $\mu\text{m}$
Mean:	168.6 $\mu\text{m}$	C.V.:	52.8%
Median:	174.2 $\mu\text{m}$	Skewness:	0.00299 Right skewed
D(3,2):	28.77 $\mu\text{m}$	Kurtosis:	-0.668 Platykurtic
Mode:	203.5 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	39.67	101.6	174.2	231.4	283.5

## Volume Statistics (Arithmetic) 04717619.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%	S.D.:	74.1 $\mu\text{m}$
Mean:	128.8 $\mu\text{m}$	C.V.:	57.5%
Median:	120.6 $\mu\text{m}$	Skewness:	0.526 Right skewed
D(3,2):	26.19 $\mu\text{m}$	Kurtosis:	-0.108 Platykurtic
Mode:	153.8 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	38.32	72.66	120.6	176.9	230.8

## Volume Statistics (Arithmetic) 0480005.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%	S.D.:	83.3 $\mu\text{m}$
Mean:	218.8 $\mu\text{m}$	C.V.:	38.1%
Median:	226.6 $\mu\text{m}$	Skewness:	-0.758 Left skewed
D(3,2):	25.56 $\mu\text{m}$	Kurtosis:	0.676 Leptokurtic
Mode:	245.2 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	116.3	181.1	226.6	273.5	316.5

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## Volume Statistics (Arithmetic)

0480509.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%		
Mean:	124.3 $\mu\text{m}$	S.D.:	99.8 $\mu\text{m}$
Median:	111.6 $\mu\text{m}$	C.V.:	80.3%
D(3,2):	7.889 $\mu\text{m}$	Skewness:	0.433 Right skewed
Mode:	223.4 $\mu\text{m}$	Kurtosis:	-0.905 Platykurtic

% <	10	25	50	75	90
Size $\mu\text{m}$	4.735	29.25	111.6	204.2	264.5

## Volume Statistics (Arithmetic)

0480923.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%		
Mean:	179.1 $\mu\text{m}$	S.D.:	96.6 $\mu\text{m}$
Median:	190.8 $\mu\text{m}$	C.V.:	53.9%
D(3,2):	22.32 $\mu\text{m}$	Skewness:	-0.157 Left skewed
Mode:	245.2 $\mu\text{m}$	Kurtosis:	-0.837 Platykurtic

% <	10	25	50	75	90
Size $\mu\text{m}$	35.12	103.4	190.8	251.0	301.3

## Volume Statistics (Arithmetic)

0482333.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%		
Mean:	169.5 $\mu\text{m}$	S.D.:	67 $\mu\text{m}$
Median:	168.9 $\mu\text{m}$	C.V.:	39.5%
D(3,2):	24.61 $\mu\text{m}$	Skewness:	0.0157 Right skewed
Mode:	185.3 $\mu\text{m}$	Kurtosis:	0.183 Leptokurtic

% <	10	25	50	75	90
Size $\mu\text{m}$	88.99	126.3	168.9	212.2	254.5

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## Volume Statistics (Arithmetic) 0483337.\$01

 Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%	S.D.:	80.5 $\mu\text{m}$
Mean:	162.7 $\mu\text{m}$	C.V.:	49.5%
Median:	169.9 $\mu\text{m}$	Skewness:	-0.228 Left skewed
D(3,2):	18.45 $\mu\text{m}$	Kurtosis:	-0.369 Platykurtic
Mode:	203.5 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	35.35	114.3	169.9	217.4	262.4

## Volume Statistics (Arithmetic) 0483770.\$01

 Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%	S.D.:	30.2 $\mu\text{m}$
Mean:	15.55 $\mu\text{m}$	C.V.:	194%
Median:	8.051 $\mu\text{m}$	Skewness:	6.37 Right skewed
D(3,2):	1.887 $\mu\text{m}$	Kurtosis:	50.4 Leptokurtic
Mode:	13.61 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	0.831	2.719	8.051	17.69	31.42

## Volume Statistics (Arithmetic) 0487092.\$01

 Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume	100.0%	S.D.:	75.7 $\mu\text{m}$
Mean:	166.6 $\mu\text{m}$	C.V.:	45.4%
Median:	169.0 $\mu\text{m}$	Skewness:	-0.0467 Left skewed
D(3,2):	22.49 $\mu\text{m}$	Kurtosis:	-0.308 Platykurtic
Mode:	203.5 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	65.51	115.6	169.0	218.5	263.8



Volume Statistics (Arithmetic) 04892128.\$01

Calculations from 0.040 µm to 2000 µm

Volume	100.0%	S.D.:	47.4 µm
Mean:	93.80 µm	C.V.:	50.6%
Median:	91.08 µm	Skewness:	0.0746 Right skewed
D(3,2):	16.01 µm	Kurtosis:	-0.762 Platykurtic
Mode:	153.8 µm		

% <	10	25	50	75	90
Size µm	32.64	58.70	91.08	130.3	160.2

Volume Statistics (Arithmetic) 0497896.\$01

Calculations from 0.040 µm to 2000 µm

Volume	100.0%	S.D.:	79.1 µm
Mean:	85.18 µm	C.V.:	92.8%
Median:	60.84 µm	Skewness:	0.851 Right skewed
D(3,2):	5.095 µm	Kurtosis:	-0.2 Platykurtic
Mode:	168.8 µm		

% <	10	25	50	75	90
Size µm	3.078	16.14	60.84	141.2	204.7

Volume Statistics (Arithmetic) 0501030.\$01

Calculations from 0.040 µm to 2000 µm

Volume	100.0%	S.D.:	78.7 µm
Mean:	129.9 µm	C.V.:	60.6%
Median:	126.3 µm	Skewness:	0.281 Right skewed
D(3,2):	13.02 µm	Kurtosis:	-0.445 Platykurtic
Mode:	185.3 µm		

% <	10	25	50	75	90
Size µm	18.55	71.51	126.3	185.5	234.5

COULTER®

LS Particle Size Analyzer

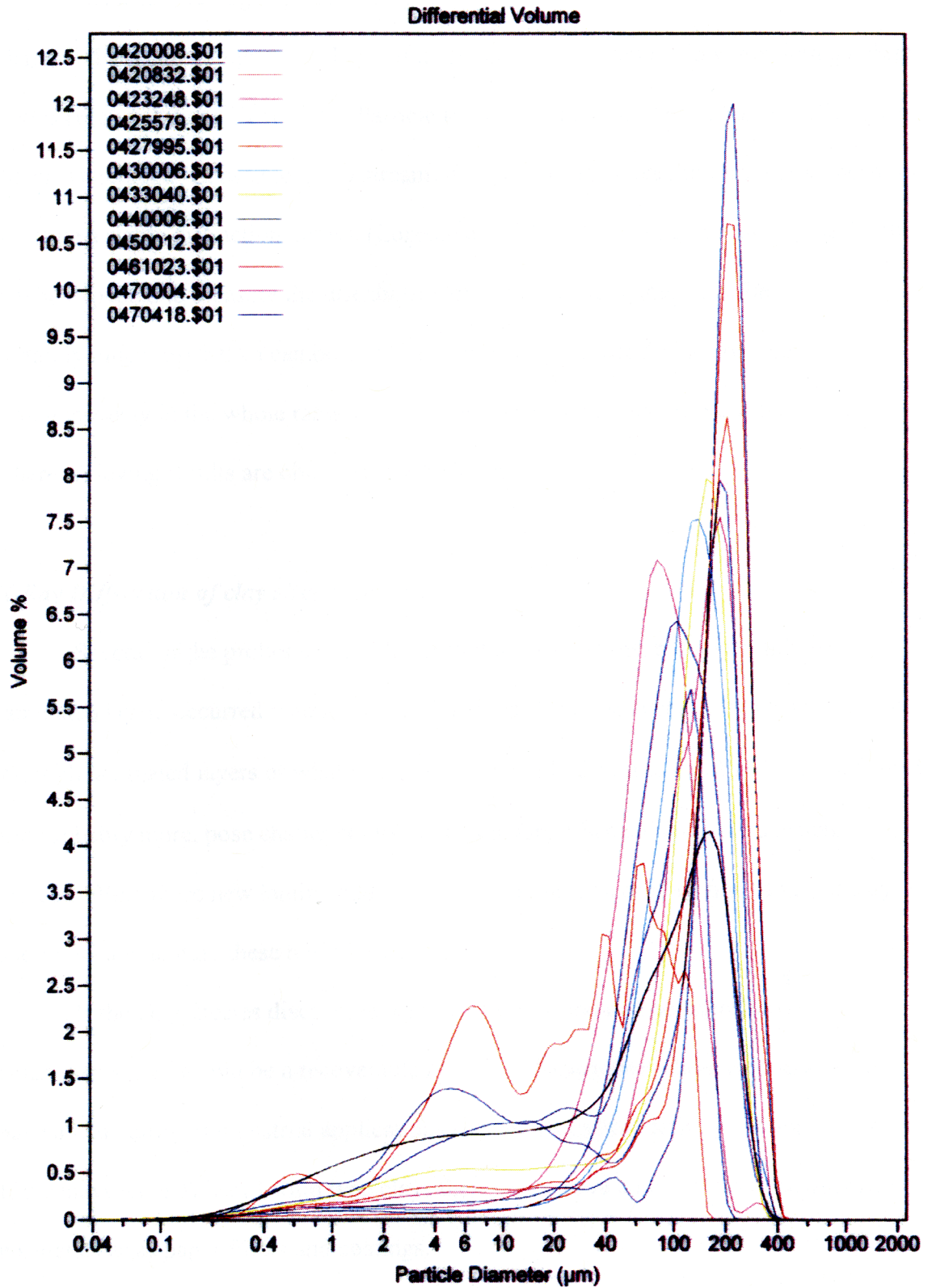
6 Jul 1999

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Particle	04712214.\$01	04717619.\$01	0480005.\$01	0480509.\$01	0480923.\$01	0482333.\$
Diameter	Volume	Volume	Volume	Volume	Volume	Volume
µm	%	%	%	%	%	%
1.000	0.455	0.543	0.644	2.46	0.811	0.350
2.000	0.779	0.872	1.23	5.20	1.52	0.457
5.000	2.83	2.75	1.93	10.52	3.27	0.885
20.00	7.90	9.31	2.09	12.04	6.58	1.72
50.00	18.77	37.95	3.95	20.12	16.34	20.17
125.0	49.95	40.92	52.68	34.08	44.98	64.40
250.0	18.60	6.81	36.41	12.88	25.31	11.22
500.0	0	0	0	0	0	0
1,000	0	0	0	0	0	0
2,000	0	0	0	0	0	0



**Figure 3.1b** Coulter LS Particle Size Analyzer, Sediment/Core Lab,  
Geology Department, Union College.

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File name: 0420008.\$01                      Group ID: 0420008  
Sample ID:  
Operator:                                      Run number: 2  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 10:06 30 Jun 1999              Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 2%  
PIDS Obscur: 24%  
Fluid: Water  
Software: 2.09                                Firmware: 2.02 2.02

File name: 0420832.\$01                      Group ID:  
Sample ID: 0420832                            Run number: 2  
Operator:  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 15:04 29 Jun 1999              Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 14%  
PIDS Obscur: 49%  
Fluid: Water  
Software: 2.09                                Firmware: 2.02 2.02

File name: 0423248.\$01                      Group ID: 0423248  
Sample ID:  
Operator:                                      Run number: 7  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                      Fluid Module  
Start time: 10:50 30 Jun 1999              Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 4%  
PIDS Obscur: 25%  
Fluid: Water  
Software: 2.09                                Firmware: 2.02 2.02

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File name: 0433040.\$01                      Group ID: 0433040  
 Sample ID:  
 Operator:                                      Run number: 13  
 Comments:  
 Optical model: Fraunhofer PIDS included  
 LS 230                                      Fluid Module  
 Start time: 11:41 30 Jun 1999              Run length: 120 Seconds  
 Pump speed: 81  
 Obscuration: 5%  
 PIDS Obscur: 49%  
 Fluid: Water  
 Software: 2.09                                  Firmware: 2.02 2.02

File name: 0440006.\$01                      Group ID: 0440006  
 Sample ID:  
 Operator:                                      Run number: 9  
 Comments:  
 Optical model: Fraunhofer PIDS included  
 LS 230                                      Fluid Module  
 Start time: 11:09 30 Jun 1999              Run length: 120 Seconds  
 Pump speed: 81  
 Obscuration: 9%  
 PIDS Obscur: 73%  
 Fluid: Water  
 Software: 2.09                                  Firmware: 2.02 2.02

File name: 0450012.\$01                      Group ID: Sacandaga  
 Sample ID: a0450012  
 Operator: Jaime                                Run number: 3  
 Comments:  
 Optical model: Fraunhofer PIDS included  
 LS 230                                      Fluid Module  
 Start time: 15:24 29 Jun 1999              Run length: 120 Seconds  
 Pump speed: 81  
 Obscuration: 19%  
 PIDS Obscur: 54%  
 Fluid: Water  
 Software: 2.09                                  Firmware: 2.02 2.02

File name: 0425579.\$01                      Group ID: 0425579  
Sample ID:  
Operator: jaimo                                Run number: 2  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                        Fluid Module  
Start time: 8:15 30 Jun 1999                Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 2%  
PIDS Obscur: 18%  
Fluid: Water  
Software: 2.09                                Firmware: 2.02 2.02

File name: 0427995.\$01                      Group ID: 0427995  
Sample ID:  
Operator:                                      Run number: 11  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                        Fluid Module  
Start time: 11:28 30 Jun 1999                Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 1%  
PIDS Obscur: 14%  
Fluid: Water  
Software: 2.09                                Firmware: 2.02 2.02

File name: 0430006.\$01                      Group ID: 0430006  
Sample ID:  
Operator:                                      Run number: 6  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                        Fluid Module  
Start time: 10:43 30 Jun 1999                Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 3%  
PIDS Obscur: 30%  
Fluid: Water  
Software: 2.09                                Firmware: 2.02 2.02

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File name: 0461023.\$01                      Group ID: 0461023  
Sample ID:  
Operator:                                      Run number: 8  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                      Fluid Module  
Start time: 10:59 30 Jun 1999              Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 4%  
PIDS Obscur: 37%  
Fluid: Water  
Software: 2.09                              Firmware: 2.02 2.02

File name: 0470004.\$01                      Group ID: 0470004  
Sample ID:  
Operator: jaime                              Run number: 1  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                      Fluid Module  
Start time: 8:05 30 Jun 1999              Run length: 120 Seconds  
Pump speed: 81  
Obscuration: 4%  
PIDS Obscur: 34%  
Fluid: Water  
Software: 2.09                              Firmware: 2.02 2.02

File name: 0470418.\$01                      Group ID:  
Sample ID: 0470418  
Operator:                                      Run number: 1  
Comments:  
Optical model: Fraunhofer PIDS included  
LS 230                                      Fluid Module  
Start time: 14:44 29 Jun 1999              Run length: 119 Seconds  
Pump speed: 81  
Obscuration: 13%  
PIDS Obscur: 50%  
Fluid: Water  
Software: 2.09                              Firmware: 2.02 2.02

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## Volume Statistics (Arithmetic) 0461023.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume:	100.0%		
Mean:	152.0 $\mu\text{m}$	S.D.:	88.5 $\mu\text{m}$
Median:	159.7 $\mu\text{m}$	C.V.:	58.2%
D(3,2):	14.42 $\mu\text{m}$	Skewness:	-0.0137 Left skewed
Mode:	203.5 $\mu\text{m}$	Kurtosis:	-0.635 Platykurtic

% <	10	25	50	75	90
Size $\mu\text{m}$	13.77	87.32	159.7	214.3	262.0

## Volume Statistics (Arithmetic) 0470004.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume:	100.0%		
Mean:	140.0 $\mu\text{m}$	S.D.:	78.8 $\mu\text{m}$
Median:	140.5 $\mu\text{m}$	C.V.:	56.3%
D(3,2):	14.82 $\mu\text{m}$	Skewness:	0.0732 Right skewed
Mode:	185.3 $\mu\text{m}$	Kurtosis:	-0.562 Platykurtic

% <	10	25	50	75	90
Size $\mu\text{m}$	20.38	85.75	140.5	195.9	243.7

## Volume Statistics (Arithmetic) 0470418.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume:	100.0%		
Mean:	117.2 $\mu\text{m}$	S.D.:	88.7 $\mu\text{m}$
Median:	129.7 $\mu\text{m}$	C.V.:	75.7%
D(3,2):	8.060 $\mu\text{m}$	Skewness:	0.0691 Right skewed
Mode:	185.3 $\mu\text{m}$	Kurtosis:	-1.3 Platykurtic

% <	10	25	50	75	90
Size $\mu\text{m}$	3.974	17.02	129.7	190.8	230.9



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Volume Statistics (Arithmetic) 0420008.\$01

Calculations from 0.040 µm to 2000 µm

Volume	100.0%		
Mean:	186.9 µm	S.D.:	82.9 µm
Median:	197.0 µm	C.V.:	44.4%
D(3,2):	19.83 µm	Skewness:	-0.536 Left skewed
Mode:	223.4 µm	Kurtosis:	0.0887 Leptokurtic

% <	10	25	50	75	90
Size µm	42.58	147.8	197.0	240.6	282.1

Volume Statistics (Arithmetic) 0420832.\$01

Calculations from 0.040 µm to 2000 µm

Volume-	100.0%		
Mean:	38.38 µm	S.D.:	37.1 µm
Median:	25.28 µm	C.V.:	96.7%
D(3,2):	5.880 µm	Skewness:	0.949 Right skewed
Mode:	66.44 µm	Kurtosis:	-0.136 Platykurtic

% <	10	25	50	75	90
Size µm	2.857	6.678	25.28	63.12	96.87

Volume Statistics (Arithmetic) 0423248.\$01

Calculations from 0.040 µm to 2000 µm

Volume	100.0%		
Mean:	77.98 µm	S.D.:	44.6 µm
Median:	73.86 µm	C.V.:	57.2%
D(3,2):	12.10 µm	Skewness:	1.02 Right skewed
Mode:	80.08 µm	Kurtosis:	3.24 Leptokurtic

% <	10	25	50	75	90
Size µm	25.08	48.25	73.86	103.5	132.9

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Volume Statistics (Arithmetic) 0425579.\$01

Calculations from 0.040 µm to 2000 µm

Volume:	100.0%		
Mean:	109.0 µm	S.D.:	60.2 µm
Median:	100.1 µm	C.V.:	55.2%
D(3,2):	15.94 µm	Skewness:	0.686 Right skewed
Mode:	105.9 µm	Kurtosis:	0.46 Leptokurtic

% <	10	25	50	75	90
Size µm	40.90	65.67	100.1	145.6	191.9

Volume Statistics (Arithmetic) 0427995.\$01

Calculations from 0.040 µm to 2000 µm

Volume:	100.0%		
Mean:	185.3 µm	S.D.:	80.3 µm
Median:	191.0 µm	C.V.:	43.4%
D(3,2):	29.66 µm	Skewness:	-0.218 Left skewed
Mode:	203.5 µm	Kurtosis:	-0.179 Platykurtic

% <	10	25	50	75	90
Size µm	67.34	137.7	191.0	238.6	284.5

Volume Statistics (Arithmetic) 0430006.\$01

Calculations from 0.040 µm to 2000 µm

Volume:	100.0%		
Mean:	125.4 µm	S.D.:	64.9 µm
Median:	121.5 µm	C.V.:	51.7%
D(3,2):	18.62 µm	Skewness:	0.42 Right skewed
Mode:	140.1 µm	Kurtosis:	0.343 Leptokurtic

% <	10	25	50	75	90
Size µm	45.89	80.49	121.5	166.6	209.4

## Volume Statistics (Arithmetic) 0433040.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume:	100.0%	S.D.:	74.4 $\mu\text{m}$
Mean:	119.1 $\mu\text{m}$	C.V.:	62.5%
Median:	127.2 $\mu\text{m}$	Skewness:	-0.0549 Left skewed
D(3,2):	8.611 $\mu\text{m}$	Kurtosis:	-0.814 Platykurtic
Mode:	153.8 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	5.488	59.49	127.2	172.9	212.3

## Volume Statistics (Arithmetic) 0440006.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume:	100.0%	S.D.:	78.2 $\mu\text{m}$
Mean:	83.70 $\mu\text{m}$	C.V.:	93.5%
Median:	65.34 $\mu\text{m}$	Skewness:	0.769 Right skewed
D(3,2):	5.064 $\mu\text{m}$	Kurtosis:	-0.31 Platykurtic
Mode:	168.8 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	1.948	9.899	65.34	140.5	198.8

## Volume Statistics (Arithmetic) 0450012.\$01

Calculations from 0.040  $\mu\text{m}$  to 2000  $\mu\text{m}$ 

Volume:	100.0%	S.D.:	54 $\mu\text{m}$
Mean:	61.74 $\mu\text{m}$	C.V.:	87.4%
Median:	54.09 $\mu\text{m}$	Skewness:	0.445 Right skewed
D(3,2):	5.640 $\mu\text{m}$	Kurtosis:	-1.1 Platykurtic
Mode:	127.6 $\mu\text{m}$		

% <	10	25	50	75	90
Size $\mu\text{m}$	2.509	7.543	54.09	108.1	139.9

COULTER®

LS Particle Size Analyzer

6 Jul 1999

————— SEDIMENT/CORE LAB, GEOLOGY DEPT., UNION COLLEGE —————

Particle	0420008.\$01	0420832.\$01	0423248.\$01	0425579.\$01	0427895.\$01	0430006.\$
Diameter	Volume	Volume	Volume	Volume	Volume	Volume
µm	%	%	%	%	%	%
1.000	0.979	2.44	1.01	0.654	0.352	0.848
2.000	1.58	11.28	1.21	0.707	0.559	1.01
5.000	3.24	26.94	4.00	1.80	1.65	1.77
20.00	3.51	22.23	18.59	10.11	4.12	6.00
50.00	7.34	29.66	60.40	50.61	13.65	41.10
125.0	61.32	2.85	12.50	32.62	58.74	44.23
250.0	20.61	0	0.542	2.21	20.23	3.47
500.0	0	0	0	0	0	0
1,000	0	0	0	0	0	0
2,000	0	0	0	0	0	0

## Core descriptions

In cores #1 and #4 we encountered small rocks in gravel horizons, which were about 2" in diameter and these rocks blocked the recovery of complete cores until the cutting shoe chewed them up enough to swallow them (Figure 2.1). This Figure should be referred to whenever detailed descriptions of the core horizons are being considered. None of the cores encountered any rocks of enough consequence to force us to relocate the truck for a second push.

**Core #1** was located just *SSW* of the four corners of old downtown Osborne Bridge. Its file number is 049 and its GPS (Global Positioning Satellite) coordinates are *N* (43°-10'-801") and *W* (74°-08'-998"). The elevation was approximately 766' above sea level. The site was also just north or on the edge of an end moraine, which was revealed further into the investigation of the study area. Every time we encountered any coarse soils/sediments, the cutting shoe plugged to a certain degree and the core recoveries were never complete, such was the case with Core #1. The probe moved down through 6" of gravel which could have been the remnants of an old road or at least the edge of one. This road could have been the road from Osborne Bridge to Northampton. The old New York State Water Commission Map #20 had not been cleaned-up at that time and no visible sign of an old road was detected at the site. We felt we were on the former lands of either J. Paul or E. Hammond (see Figure 1.5). The probe moved down through another 8" of developed soil horizon that was also gravelly. Both of the above horizons were sampled together for further testing. Then we hit some very coarse gravelly clay with stones as large as the cutting shoe, about 5" into it. Beyond this, the probe was unable to collect any more sediment for the next 18". The second section continued to

push aside the same sediments until it rounded and swallowed the obstructing cobble, more than 16" further down. It is believed that this horizon was a remnant braided or lateral stream flowing parallel to the adjacent end moraine. The next 13" were the same coarse gravely clay with smaller cobbles, 1-1 ½" at the largest. The last 18-19" of the first 8' push probe comprised a coated gravely clay. No attempt will be made here to predict the parent environment of this coated horizon but it did stand out from the background with further analysis and was completely saturated and assumed to be the water table. This was not hard to imagine since the site was the closest to the Sacandaga River, flowing just to the Northeast.

**Core #2** was located just *NW* of Osborne Bridge on the former lands of Foster Noyes (see Figure 1.5). It was located on the banks of the Sacandaga River Floodplain, which is pervasive north to the Northville Bridge, even at low water status. Its file number is 048 and its GPS coordinates are *N* (43°-10'-865") and *W* (74°-09'-809"). The elevation was approximately 765' above sea level. The top 5" was a very heavy sand indicative of a sandbar, but may also have been involved with ice rafted clast mechanics during spring melt ice jams due to the presence of interspersed pebbles. Next was a small 3" layer of a developed soil horizon, often referred to as the A-horizon. The probe then pushed through 14" of light gravel with black sands and another 10" of heavy gravel with black sands. All of the gravel horizons above could be considered glaciofluvial in nature, related to the annual melts on the front side of a retreating glacier. The bottom of the first section of probe core was approximately 1½ -2" of fine sand. The top of the next 4' probe had 2" of fine gravel with black sands. The next horizon down was 33" thick and contained alternating layers of light and dark sands with black sands. We first

determined them to be varved sands <50 $\mu$ m, deposited in quiet waters within one year, adjacent to a glacier. Upon later investigation, after running a few Ostracode Studies in both light and dark layers, questions were raised as to the real interpretation. Below this very fine sand horizon lay 7" of fine sand with black sands, 4" of fine sand without black sands, 7" more of medium sand with black sands and 4" of clay and fine sand with black sands. The bottom 36" of this 12" probed core was fine sand with alternating bands of interspersed black sands. The water table (saturation zone) was encountered at 6'. The last two 4' sections are indicative of beach or shoreline deposits alternating with quiet water deposition in a lake basin or floodplain influenced by a persistent river. It implies that all of these variations in sediment deposition may have already existed before glaciation or that they arrived after a glacial retreat. Another possibility exists, which is that they could be deformed glaciofluvial sediments of a former outwash plain. This poses more questions, which are talked about throughout the study. Core site #2 revealed a wide variety of core horizons and subsequently, several horizons were examined more thoroughly in this core than in any other. Nearly the whole of the upper 3 feet of the core horizons are probably the result of postglacial, fluvial deposition. Everything below 3' was consistent with a subglacial environment, typical of a kettle lake deposition.

**Core #3** was located just *E* of Mead Island on the former lands of L. D. Fritcher (see Figure 1.5). It was at an elevation of about 765' above sea level. Its file number is 050 and its GPS coordinates are *N* (43°-10'-821") and *W* (74°-09'-791"). The site was formerly farmed and a few small stone piles existed adjacent to Mead Island. The top 11" were a combination of a developing soil horizon and heavy beach sands at the uppermost few inches. This may be due to the nearness of the shoreline of Mead Island,

but it is clear, Mr. Fritcher farmed right up to the shore/tree line in his day. As the probe continued down the next 9" it encountered more heavy sand with black sands and small amounts of clay in the matrix. The next 3" contained gravel and black sands followed by another 3" of clayey gravel with black sands. The bottom 10" of the first section was again heavy sand and gravel with black sands. These soils were coarse enough to again clog the cutting shoe of the probe and another incomplete core was acquired. More gravel, perhaps a little lighter, met the advance of the second probe length and until the cutting shoe chugged its way through this 18" horizon, more than 15" was lost. The bottom 15" were again clayey gravel with black sands and was in the saturated zone. It's never clear where the probe core-cutting shoe gets hung up but the depth of the core bottom is still established. Core #3 is interpreted as stoss and lee topography where lodgement till slopes around an obstacle such as the Mead Island drumlin. This is likely subglacial in nature and could be another clue that everywhere within the study area the sediments could be simply made up of ice-contact deformations.

**Core #4** was located at the *NW* corner of Mead Island adjacent to a backwater inlet that faces the Sacandaga River floodplain to the east. At first glance, the inlet appears to be a streambed, but no upstream streambed or water flowing into it exists. Another such structure exists less than ½ mile north at the Northampton Beach State Campsite, which is used as a boat launch. Both of these structures are, in fact, relict streambed outlets and probably connected to the West. There the structures drained a kettle, which today is the Tamarack Swamp, due west about 1 mile. Kettles are subglacial happenings and further evidence that the study area is probably made up of ice-contact deformations.



Core #4's file number is 047 and its GPS coordinates are  $N(43^{\circ}-10'-715'')$  and  $W(74^{\circ}-10'-252'')$ . Its elevation was precisely at the high water mark of 768' above sea level. The site was on the former lands of Charles and Arietta Mead (see Figure 1.5). The first 4' of the first core section was medium beach sand with black sands, representative of the high water mark sites. Beneath this was approximately 14" of developed soil horizon. The next horizon down was very heavy gravel with black sands and cobbles large enough to plug the cutting sole, creating a 12" blank in the profile. It finally chewed the cobble up enough to swallow it and continued down through another 10" of light gravel with black sands. The gravel continued to lighten and contained less black sands as the second 4' section captured the top 8'. Somewhere along the push the remaining 36" of light gravel was heavy enough to compress the profile so that the sample was 6" short a complete core section. The same can be said for the next pushed section as it went on to capture 11" of clayey light gravel with black sands, 8" of light gravel with black sands and 2" of red clay light gravel with black sands before hitting the water table at 10'. Somewhere along this section push, probably while capturing the next 18" of medium gravel with black sand the collected sediments were less than complete. The remaining 4' in this third section was heavy sand with heavy black sands.

We had prepared for at least one deeply pushed core section and considered the water table at such a site. After we hit the water table at site #4 we installed a pointed cutting sole on the last 4' before lowering it down the core hole. It was designed to be ridged until the 12' mark was hit and then released to capture the sediments by the core liner in the same way that the sediments are collected above them. It worked well and we

collected the next 14" of heavy sand with black sands. The remaining 16" down to the 16" mark were heavy gravel with up to 1" stones, which gave us an 18" blank in this core capture. This horizon was in a matrix of fine sands and black sands. This core site had horizons that are consistent with streambed sediments and reflect a wandering high water flow subglacial drainage system of the kind mentioned earlier, perhaps such as in draining a kettle.

**Core #5** was located *WSW* of Mead Island on the edge of the largest eroded micro-island of a hummock field in the local greater Mead Islands group. The micro-island is usually submerged until late summer. The site was picked just on the eastern edge of the old Osborne Bridge Road and was 765' above sea level. Its file number is 046 and its GPS coordinates are *N* (43°-10'-527") and *W* (74°-10'-230"). It is unclear by the 1908 New York State Water Commission Maps who the former owner was but it could have been Horace Kenyon. In a straight line, the site was about 900' from core site #4 (see Figure 1.5). The first section behaved like the previous pushes, with gravel hang-ups at the cutting sole but we collected 80% of the sediments leaving a 10" blank at the top. The top 13" recovered was typical soil developed light gravelly loam with black sands. We pushed down through 18" of coated (oxidized) medium gravel with black sands and then another 8" of coated light gravel with black sands. The second section saw the same gravel compression and recovery was absent 8" at the top. The top 7" of this section was the same-coated light gravels with black sands that preceded it. Then we ran into 18" of heavy sand with black sands and some banded fine sand/clay matrix segments. The bottom 18" continued to be heavy sand with small bands of clay matrix

and heavier black sand layers. The hummocky topography of this site is consistent with kame and kettle microenvironments associated with subglacial ice-contact deformations.

**Core #6** was located about 800' further *SW* along the old Osborne Bridge gravel road from site #5 (see Figure 1.5). This site is approximately 760' above sea level and it is unclear who the former owner was, but was likely to be Horace Kenyon. The site was on the east side of the road and appears to be a former dirt road heading northwest to an anonymous unimproved road connected to the Bunker Hill road. Its file number is 045 and its GPS coordinates are *N* (43°-10'-431") and *W* (74°-10'-421"). Both this site and site #7 are similar regarding their sediment profiles. The top 12" of the first pushed section was developed soil horizon made up of loamy clay. Beneath this was 36" of densely packed fine sand and clay with what appears to be varved banding. The top 18" of the second section was more of the same densely packed fine sand/clay with banding. The next deeper horizon was 24" of fine sand and clay with black sand banding and was completely saturated (water table). The final 6" was fine sand and clay with black sands (the emphasis on the first component in abundance). This sediment profile is consistent with kettle lake environments, which are subglacial and represent ice-soil contact seasonal melting depositions. In retrospective, we should have marked a direction onto the side of the liners at both this site and core #7. That would have given us a better slump profile of these two sites, which were the only sites whose core horizons could be correlated in the study area.

**Core #7** was located at a bend in the old Osborne Bridge Road, *SW* of core #6 by some 1200' (see Figure 1.5). It's not perfectly clear from the maps but this property could have been formerly the same lands of Ed Gifford. This site was also at the 760'

elevation above sea level and on the west side of the road. Its file number was 044 and its GPS coordinates were  $N(43^{\circ}-10'-332'')$  and  $W(74^{\circ}-10'-487'')$ . The top 6" of the probe were probably the gravels of the old roadbed. The next 6" was a typical clayey loam developed soil horizon. The next 36" of the first section was densely packed fine sand and clay with varved banding. The top 12" of the second section was the same densely packed fine clay and varved banding. The remaining 36" of this 8' deep core was fine sand and clay with black sand banding and was completely saturated (water table). Again, this sediment profile is consistent with a subglacial kettle lake depositional environment.

**Core #8** was sited approximately 800' further *SW* along the old Osborne Bridge Road. It is on the east side of the road at 764' above sea level and was probably the former lands of Ammon Blowers (see Figure 1.5). Its file number is 043 and its GPS coordinates are  $N(43^{\circ}-10'-262'')$  and  $W(74^{\circ}-10'-603'')$ . The site was chosen because it was an apparent sandbar, which was of interest. The top 16" was deposited in the last 70 years and consists of heavy coated sand with black sands and a little fine sand and clay in the upper few inches. The next 24" in the first section consisted of a well-defined soil horizon of sand and light gravel with black sands. The last 8" was heavy sand and clay with varved banding. The next section of probe compressed somewhere along the push perhaps because of the first 12" of heavy sand and clay, or more likely because of the dark heavy coated sand and alternating clay bands and black sand banding which made up the last 31" of sediments. The last 31" of heavy coated sand horizon was completely saturated (water table). The sediment profile in this 8' core below the top newly formed sandbar represent alternating quiet water and fast water episodes of deposition. The

horizons are interpreted as stoss and lee topography where lodgement till slopes around large obstacles, such as the drumlin just to the *NW* of the site. This drumlin has never been identified, so it is named the *Ambrosino Drumlin*. It has an elevation of 880' above sea level, just southeast of Sweets Crossing. It is not clear as of this writing, but this drumlin appears to be of continental glaciation, rather than of the outlet or valley glacier type.

**Core #9** is located at the high water mark (768') on the shoreline of the old Green's Beach Campsites. It would be just south of the old Osborne Bridge Road and on the former lands of Ammon Blowers (see Figure 1.5). Its file number is 042 and its GPS coordinates are *N* (43°-10'-182") and *W* (74°-10'-798"). The top 8" of the upper 4' section was coarse heavy beach sand with black sand. The next 6" down was coated gravel with black sands. Below that was 18" of loamy heavy coated sand with black sands all showing typical soil development. The last 16" of the upper section were fine sand and clay mixtures, which lightened downward and contained black sands. This section slipped while in the liner or during splitting in half lengthwise. The bottom 4' section compressed at the lowest coarse gravel horizon and resulted in about a 7" blank at the top of the core. In (Figure 2.1) the last core in the rack, to the right of the circled #9, a 4" plug of dark gravel separated and at the top of the core section is evidenced. This is because during handling at the core site, the whole of the contents of the core slipped toward the bottom and protruded by 4". To avoid destroying the integrity of the protruded sediment, it was cut-off and slid down into the blank space at the top of the liner. The upper 6" of this core section consisted of fine sand and clay mixtures and black sands. The next 18" down was made up of medium sand with varved bands of

heavy black sands, especially in the upper half portion of the horizon. The bottom 16" was progressively coarser gravel going downward with black sands. The last 4" was darker due to the zone of saturation (water table). Core #9 was approximately 1200' further *SW* along the old Osborne Bridge Road and although it has the same profile of quiet water alternating fast water depositional character, the sediments seem to have undergone greater than usual secondary weathering. This is evident in the shades of color in the horizons, indicating groundwater's of totally different charge and suspended solids loads. The horizons are interpreted as stoss and lee topography where lodgement till slopes around large obstacles, perhaps such as an Ambrosino drumlin type.

## **Appendix C**

### **USGS WATER WELL RECORDS**

Date: Mon, 21 Jun 1999 09:38:28 -0400  
From: "Ronald V Allen, Hydrologic Technician, Troy, NY " <rallen@usgs.gov>  
To: aa7148@cnsvax.albany.edu  
Cc: "Ronald V Allen, Hydrologic Technician, Troy, NY " <rallen@usgs.gov>  
Subject: GWSI database retrieval, well records.

Arthur Ambrosino:

re: well records within 2-miles of: 431000, 0741000.

The 2-mile retrieval located one record ( local # FU 137). I re-searched using a 5-mile radius and picked-up 26 records. They are sorted by ascending latitude.

footnotes:

county, 035 = Fulton  
lat-long accuracy, F = within 5-seconds  
type analysis, B = common ions  
Primary use of water, H = domestic  
                                  S = stock  
                                  N = industrial  
                                  C = commercial  
aquifer code, 112TILL = pleistocene till  
                  112SAND =            sand  
                  112SDGV =            gravel  
                  371LLFL = Little Falls Dolomite  
                  371PSDM = Potsdam Sandstone  
                  364CNJR = Canajoharie Shale  
                  400PCMB = Precambrian Erathem

lith code, TILL = till  
              SHLE = shale  
              DLMT = dolomite  
              SAND = sand  
              ROCK = bedrock, undifferentiated  
              GRVL + gravel

Method constructed, D = dug  
                                  C = cable tool

type of finish, W = walled

Please contact me if you have questions.

Ron Allen

U.S. Geological Survey, Water Resources Division

U S	Ron Allen, Information Specialist	Voice: (518)285-5602
-+-	U.S. Geological Survey, WRD	Fax: (518)285-5601
G S	425 Jordan Road	USGS, NY web site,
	Troy, NY 12180-8349	<a href="http://ny.usgs.gov">http://ny.usgs.gov</a>



1DATE: 06/16/99  
 miles of, 43 10 00; 074 10 00.

GWSI retrieval, well records within 2  
 PAGE 1a

LAT/LONG		LAT	LONG	OF LAND	ALTITUDE	TOPO-	LATITUDE
LONGITUDE	DATUM	ACCURACY	SURFACE	DATUM	GRAPHIC		
LOCAL WELL NUMBER	SITE-ID	COUNTY	(DEGREES)	SETTING			
(DEGREES)	(CODE)	CODE	(FEET)	(CODE)	(DEGREES)	SETTING	
FU 135		430550074105901	035		430550		
0741100	NAD27	F 820	NGVD29		-		
FU 148		430604074095801	035		430605		
0741000	NAD27	F 850.00	NGVD29		-		
FU 136		430612074102801	035		430610		
0741030	NAD27	F 780.00	NGVD29		-		
FU 147		430626074094401	035		430625		
0740945	NAD27	F 790.00	NGVD29		-		
FU 146		430641074092201	035		430640		
0740920	NAD27	F 800.00	NGVD29		-		
FU 118		430728074140501	035		430730		
0741405	NAD27	F 780.00	NGVD29		-		
FU 117		430740074140901	035		430740		
0741410	NAD27	F 840.00	NGVD29		-		
FU 115		430815074142001	035		430815		
0741420	NAD27	F 790.00	NGVD29		-		
FU 116		430817074140901	035		430815		
0741410	NAD27	F 790.00	NGVD29		-		
FU 114		430824074142101	035		430825		
0741420	NAD27	F 820.00	NGVD29		-		
FU 145		430843074075601	035		430845		
0740755	NAD27	F 770.00	NGVD29		-		
FU 144		430848074072601	035		430850		
0740725	NAD27	F 790.00	NGVD29		-		
FU 113		430910074132501	035		430910		
0741325	NAD27	F 790.00	NGVD29		-		
FU 138		431111074115901	035		431110		
0741200	NAD27	F 820.00	NGVD29		-		
FU 108		431133074132301	035		431135		
0741325	NAD27	F 840.00	NGVD29		-		
FU 107		431141074131501	035		431140		
0741315	NAD27	F 860.00	NGVD29		-		
FU 143		431203074080601	035		431205		
0740805	NAD27	F 890.00	NGVD29		-		
FU 142		431229074083401	035		431230		
0740835	NAD27	F 930.00	NGVD29		-		
FU 141		431257074090201	035		431255		
0740900	NAD27	F 950.00	NGVD29		-		
FU 139		431354074112201	035		431355		
0741120	NAD27	F 750.00	NGVD29		-		
FU 112		430925074131001	035		430925		
0741310	NAD27	F 785.00	NGVD29		-		

FU 111			431026074134601	035	431025
0741345	NAD27	F	900.00	NGVD29	-
FU 110			431035074134001	035	431035
0741340	NAD27	F	890.00	NGVD29	-
FU 109			431048074133901	035	431050
0741340	NAD27	F	880.00	NGVD29	-
FU 137			431053074115901	035	431055
0741200	NAD27	F	820.00	NGVD29	-

FU 140			431355074111101	035	431355
0741110	NAD27	F	750.00	NGVD29	-

1DATE: 06/16/99 GWSI retrieval, well records within 2 miles of, 43 10 00; 074 10 00. PAGE 1b

DEPTH		DEPTH		PRIMARY		DIAMETER		LITH-	
TO	TO	METHOD	DEPTH	BOTTOM	USE	TYPE			
TOP	BOTTOM	CONST-	OF WELL	OF	OF	OF	AQUIFER		
(FEET)	(FEET)	RUCTED	(FEET)	CASING	WATER	CASING	CODE		LOGY
	LOCAL WELL NUMBER		FINISH	(FEET)		ANALYSES			CODE
					(FEET)	(IN)			
FU 135			26	H	--	--	112TILL		TILL
0	26	D	W	54	--	--			
FU 148			67.0	H	--	--	364CNJR		SHLE
44.0	--	C	--	--	6.00	--			
FU 136			65.0	H	--	--	--		--
--	--	-	--	--	--	--	364CNJR		SHLE
18.0	--	C	--	--	6.00	--			
FU 147			53.0	H	--	--	--		--
--	--	C	--	--	6.00	--			
FU 146			117	H	--	--	--		--
--	--	-	--	--	--	--			
--	--	C	--	--	6.00	B	--		--
FU 118			134	H	--	--	--		--
--	--	-	--	--	--	B	371LLFL		DLMT
114	--	C	--	--	6.00	--			
FU 117			130	S	--	--	371LLFL		DLMT
13.0	--	C	--	--	6.00	--			
FU 115			77.0	H	--	--	--		--
--	--	-	--	--	--	--	112SAND		SAND
--	--	C	--	--	6.00	--			
FU 116			99.0	H	--	--	112SAND		SAND
--	--	C	--	--	6.00	--			
FU 114			109	H	--	--	--		--
--	--	-	--	--	--	B	400PCMB		ROCK
82.0	--	C	--	--	6.00	--			
FU 145			10.0	H	--	--	112SDGV		GRVL
--	--	D	--	--	24.0	--			
FU 144			42.0	H	--	--	--		--
--	--	-	--	--	--	--	371LLFL		DLMT
--	--	C	--	--	6.00	--			
FU 113			73.0	H	--	--	--		--
--	--	-	--	--	--	--	112SAND		SAND
--	--	C	--	--	6.00	--			

FU 112	--	C	41.0	H	--	112TILL	TILL
--	--	--	--	--	6.00	--	--
FU 111	--	D	40.0	H	--	112TILL	TILL
--	--	--	--	--	48.0	--	--
FU 110	--	C	65.0	H	--	371LLFL	DLMT
--	--	--	--	--	6.00	--	--
FU 109	--	D	19.0	H	--	112TILL	TILL
--	--	--	--	--	20.0	--	--
FU 137	--	-	186	N	--	--	--
--	--	--	--	--	--	--	--
--	--	C	-	--	8.00	--	--
FU 138	--	C	140	C	--	371LLFL	DLMT
15.0	--	--	--	--	6.00	--	--
FU 108	--	D	38.0	H	--	112TILL	TILL
--	--	--	--	--	--	--	--
FU 107	--	D	12.0	H	--	112TILL	TILL
--	--	--	--	--	36.0	--	--
FU 143	--	-	14.0	H	--	--	--
--	--	--	--	--	--	--	--
--	--	D	-	--	--	112SAND	SAND
FU 142	--	D	18.0	H	--	112SAND	SAND
--	--	--	--	--	24.0	--	--
FU 141	--	-	62.0	S	--	--	--
--	--	--	--	--	--	--	--
3.00	--	C	-	--	3.00	371PSDM	SNDS
FU 139	--	-	128	C	--	--	--
--	--	--	--	--	--	--	--
--	--	C	-	--	6.00	112SDGV	GRVL
FU 140	--	C	125	H	--	400PCMB	ROCK
4.00	--	--	--	--	6.00	--	--

1DATE: 06/16/99  
miles of, 43 10 00; 074 10 00.

GWSI retrieval, well records within 2  
PAGE 1c

LOCAL WELL NUMBER OWNER	DISCHARGE REMARK (GPM)
FU 135 NYSZCZY, NICK	100 GPH, HARD WATER, TEMP 51
FU 148 HUGHES JOHN	FROM BULL.GW-24. ; 30= AVG.ANN.WL. 12.0
FU 136	PUMPED AT 6 GPM
MINKLER HARRY	FROM BULL.GW-24. ; 30= AVG.ANN.WL. 6.00
FU 147 PELCHER JOE	FROM BULL.GW-24. ; 30= AVG.ANN.WL. --
FU 146	Formerly site 430643074092201. --
FYFE JAMES	FROM BULL.GW-24. ; 30= AVG.ANN.WL. --
FU 118	Formerly site 430723074140601. --
SART LAUREN	FROM BULL.GW-24. ; 30= AVG.ANN.WL. 5.00

FU 117 WARNER V. J.	FROM BULL.GW-24. ; 30= AVG.ANN.WL. 6.00
FU 115	Formerly site 430816074142001.
VAN NOSTRAND DONALD	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
FU 116	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
WOJSKI A.	4.00
FU 114	Formerly site 430826074141001.
ZEA EARL	FROM BULL.GW-24. ; 30= AVG.ANN.WL. 2.50
FU 145	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
BOICE KATHERN	3.00
FU 144	Formerly site 430849074073701.
HAYDEN RALPH	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
FU 113	Formerly site 430912074131901.
FERENCY ESTHER	FROM BULL.GW-24.
FU 112	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
WARNER BURDETTE	7.50
FU 111	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
WARNER CYRUS	3.00
FU 109	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
STEWART EDITH	NOT USED AT PRESENT
FU 137	
U. S. FELDSPAR CORP.	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
FU 138	40.0
FERGUSON AND MITTLSTAEDT	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
FU 108	5.00
SHAW ED	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
FU 107	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
EDWARDS PALMER	Formerly site 431203074081001.
FU 143	
DARLING M.	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
FU 142	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
MADISON ANDY	Formerly site 431255074090301.
FU 141	
LINDSEY M. H.	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
FU 139	3.00 Formerly site 431353074112501.
HOBBS WILLIAM E.	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
FU 140	13.0
STANDHARDT FRED	FROM BULL.GW-24. ; 30= AVG.ANN.WL.
	1.00

## **Appendix D**

### **ALS CHEMEX ANALYSES AND MINERALOGY**



# Chemex Labs, Inc.

Analytical Chemists • Geochemists • Registered Assayers  
 994 Glendale Ave., Unit 3, Sparks, Nevada, U.S.A. 89431  
 PHONE: 775-356-5395 FAX: 775-355-0179

To: EARTH & ATMOSPHERIC SCIENCES  
 SUNY ALBANY  
 1400 WASHINGTON AVE.  
 ALBANY, NEW YORK  
 12222

A9919903

Comments: ATTN. ARTHUR M. AMBROSINO

## CERTIFICATE A9919903

TO: EARTH & ATMOSPHERIC SCIENCES

PROJECT: G.S.L.  
 ORDER #: 45855

SAMPLES submitted to our lab in Mississauga, ON.  
 This report was printed on 22-JUN-1999.

### SAMPLE PREPARATION

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION
299	1	Pulps prepped on other workorder

### ANALYTICAL PROCEDURES

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION	METHOD	DETECTION LIMIT	UPPER LIMIT
9301	1	Al %; ICP + ICP-MS package	ICP	0.01	25.0
9341	1	Sb ppm; ICP + ICP-MS package	ICP-MS	0.1	1000
9302	1	Ba ppm; ICP + ICP-MS package	ICP	10	10000
9303	1	Be ppm; ICP + ICP-MS package	ICP-MS/ICP	0.05	1000
9304	1	Bi ppm; ICP + ICP-MS package	ICP-MS/ICP	0.01	10000
9305	1	Cd ppm; ICP + ICP-MS package	ICP-MS/ICP	0.02	500
9306	1	Ca %; ICP + ICP-MS package	ICP	0.01	25.0
9307	1	Ce ppm; ICP + ICP-MS package	ICP-MS	0.01	500
9308	1	Cs ppm; ICP + ICP-MS package	ICP-MS	0.05	500
9309	1	Cr ppm; ICP + ICP-MS package	ICP	1	10000
9310	1	Co ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9311	1	Cu ppm; ICP + ICP-MS package	ICP	1	10000
9312	1	Ga ppm; ICP + ICP-MS package	ICP-MS	0.1	500
9313	1	Ge ppm; ICP + ICP-MS package	ICP-MS	0.1	500
9315	1	Fe %; ICP + ICP-MS package	ICP	0.01	25.0
9316	1	La ppm; ICP + ICP-MS package	ICP-MS	0.5	500
9317	1	Pb ppm; ICP + ICP-MS package	ICP-MS/ICP	0.5	10000
9318	1	Li ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9319	1	Mg %; ICP + ICP-MS package	ICP	0.01	15.00
9320	1	Mn ppm; ICP + ICP-MS package	ICP	5	10000
9321	1	Mo ppm; ICP + ICP-MS package	ICP	0.2	10000
9322	1	Ni ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9323	1	Nb ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9324	1	P ppm; ICP + ICP-MS package	ICP	10	10000
9325	1	K %; ICP + ICP-MS package	ICP	0.01	10.00
9326	1	Rb ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9327	1	Ag ppm; ICP + ICP-MS package	ICP-MS/ICP	0.05	100.0
9328	1	Sn %; ICP + ICP-MS package	ICP	0.01	10.00
9329	1	Sr ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9330	1	Ta ppm; ICP + ICP-MS package	ICP-MS	0.05	500
9331	1	Tb ppm; ICP + ICP-MS package	ICP-MS	0.05	500
9332	1	Tl ppm; ICP + ICP-MS package	ICP-MS	0.02	500
9333	1	Th ppm; ICP + ICP-MS package	ICP	0.2	500
9334	1	Ti %; ICP + ICP-MS package	ICP	0.01	10.00
9335	1	U ppm; ICP + ICP-MS package	ICP-MS/ICP	0.1	10000
9336	1	V ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9337	1	Y ppm; ICP + ICP-MS package	ICP	1	10000
9338	1	Zn ppm; ICP + ICP-MS package	ICP-MS	0.1	5000
9339	1	Zn ppm; ICP + ICP-MS package	ICP	2	10000

The results of this assay were based solely upon the content of the sample submitted. Any decision to invest should be made only after the potential investment value of the assets or deposit has been determined based on the results of assays of multiple samples of geologic materials collected by the prospective investor or by a qualified person selected by him/her and based on an evaluation of all engineering data which is available concerning any proposed project. Statement required by Nevada State Law NRS 519



# Chemex Labs, Inc.

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To: EARTH & ATMOSPHERIC SCIENCES  
SUNY ALBANY  
1400 WASHINGTON AVE.  
ALBANY, NEW YORK  
12222

Page Number : 1-A  
Total Pages : 1  
Certificate Date : 22-JUN-1993  
Invoice No. : 18919903  
P.O. Number : 45855  
Account : RCC

Project: G.S.L.  
Comments: ATTN: ARTHUR M. AMBROSINO

## CERTIFICATE OF ANALYSIS A9919903

SAMPLE	PREP CODE	Al % (ICP)	Sb ppm (ICP)	Ba ppm (ICP)	Be ppm (ICP)	Bi ppm (ICP)	Cd ppm (ICP)	Ca % (ICP)	Ce ppm (ICP)	Cs ppm (ICP)	Cr ppm (ICP)	Co ppm (ICP)	Cu ppm (ICP)	Ga ppm (ICP)	Ge ppm (ICP)
0101500	299 --	7.43	0.1	870	2.95	0.02	0.20	1.36	187.0	0.55	186	7.4	6	21.0	1.4

CERTIFICATION



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 Nevada, U.S.A. 89431  
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 ALBANY, NEW YORK  
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Page Number 1-B  
 Total Pages 1  
 Certificate Date 22-JUN-1999  
 Invoice No. 18919903  
 P O Number 45855  
 Account : RCO

Project: G.S.L.  
 Comments: ATTN: ARTHUR M. AMBROSINO

## CERTIFICATE OF ANALYSIS A9919903

SAMPLE	PREP CODE	Fe % (ICP)	La ppm (ICP)	Pb ppm (ICP)	Li ppm (ICP)	Hg % (ICP)	Mn ppm (ICP)	Mo ppm (ICP)	NI ppm (ICP)	Nb ppm (ICP)	P ppm (ICP)	K % (ICP)	Rb ppm (ICP)	Sr ppm (ICP)	Na % (ICP)
0101500	299 --	2.39	81.0	18.0	24.2	0.53	290	1.2	5.8	8.6	950	3.49	88.6	0.25	2.07

*Arthur M. Ambrosino*

CERTIFICATION:





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SUNY ALBANY  
1400 WASHINGTON AVE.  
ALBANY, NEW YORK  
12222

Project: G.S.I.  
Comments: ATTN: ARTHUR M. AMBROSINO

Page Number : 1-C  
Total Pages : 1  
Certificate Date: 22-JUN 1999  
Invoice No. : 19919903  
P.O Number : 45855  
Account : RCO

## CERTIFICATE OF ANALYSIS A9919903

SAMPLE	PREP CODE	Sr ppm (ICP)	Ta ppm (ICP)	Te ppm (ICP)	Tl ppm (ICP)	Tb ppm (ICP)	Ti % (ICP)	H ppm (ICP)	D ppm (ICP)	V ppm (ICP)	Y ppm (ICP)	Zn ppm (ICP)
0101500	299 --	266	0.50	< 0.05	0.50	12.0	0.31	0.7	1.0	37	30.7	66

*Arthur M. Ambrosino*

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To: EARTH & ATMOSPHERIC SCIENCES  
 SUNY ALBANY  
 1400 WASHINGTON AVE.  
 ALBANY, NEW YORK  
 12222

A9919902

Comments: ATTN: ARTHUR M. AMBROSINO

CERTIFICATE		A9919902	
(CO) - EARTH & ATMOSPHERIC SCIENCES Object: G.S.L O #: 45855 samples submitted to our lab in Mississauga, ON. this report was printed on 24-JUN-1999.			
SAMPLE PREPARATION			
HEMEX CODE	NUMBER SAMPLES	DESCRIPTION	
217 9300	1 1	Geochem ring entire sample T127 special code	

ANALYTICAL PROCEDURES					
CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION	METHOD	DETECTION LIMIT	UPPER LIMIT
975	1	Au ppb: ICP-fluorescence package	FA-ICP-AFS	2	10000
976	1	Pt ppb: ICP-fluorescence package	FA-ICP-AFS	5	10000
977	1	Pd ppb: ICP-fluorescence package	FA-ICP-AFS	2	10000
13	1	As ppm: HNO3-aqua regia digest	AAS-HYDRIDE/EDL	1	10000
20	1	Hg ppb: HNO3-HCl digestion	AAS-FLAMELESS	10	100000

The results of this assay were based solely upon the content of the sample submitted. Any decision to invest should be made only after the potential investment value of the claim or deposit has been determined based on the results of assays of multiple samples of geologic materials collected by the prospective investor or by a qualified person selected by him/her and based on an evaluation of all engineering data which is available concerning any proposed project. Statement required by Nevada State Law NRS 519



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1400 WASHINGTON AVE.  
ALBANY, NEW YORK  
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Project: G.S.L.  
Comments: ATTN: ARTHUR M AMBROSINO

Page Number 1  
Total Pages 1  
Certificate Date: 24 JUN 1989  
Invoice No. 19919902  
P.O. Number 45855  
Account RCO

CERTIFICATE OF ANALYSIS		A9919902						
SAMPLE	PREP CODE	Au ppb AFS	Pt ppb AFS	Pd ppb AFS	As ppm	Hg ppb		
0101500	2179300	< 2	< 5	< 2	4	< 10		

*Arthur M. Ambrosino*

CERTIFICATION:



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To: EARTH & ATMOSPHERIC SCIENCES  
 SUNY ALBANY  
 1400 WASHINGTON AVE.  
 ALBANY, NEW YORK  
 12222

A8919911

Comments: ATTN: ARTHUR M. AMBROSINO

## CERTIFICATE A9919911

(CO) - EARTH & ATMOSPHERIC SCIENCES

Object: G.S.L  
 O #: 45856

amples submitted to our lab in Mississauga, ON.  
 his report was printed on 22-JUN-1999.

### SAMPLE PREPARATION

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION
299	8	Pulp; prepped on other workorder

### ANALYTICAL PROCEDURES

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION	METHOD	DETECTION LIMIT	UPPER LIMIT
9301	8	Al %; ICP + ICP-MS package	ICP	0.01	25.0
9341	8	Sb ppm; ICP + ICP-MS package	ICP-MS	0.1	1000
9302	8	Ba ppm; ICP + ICP-MS package	ICP	10	10000
9303	8	Be ppm; ICP + ICP-MS package	ICP-MS/ICP	0.05	1000
9304	8	B1 ppm; ICP + ICP-MS package	ICP-MS/ICP	0.01	10000
9305	8	Cd ppm; ICP + ICP-MS package	ICP-MS/ICP	0.02	500
9306	8	Ca %; ICP + ICP-MS package	ICP	0.01	25.0
9307	8	Ce ppm; ICP + ICP-MS package	ICP-MS	0.01	500
9308	8	Ce ppm; ICP + ICP-MS package	ICP-MS	0.05	500
9309	8	Cr ppm; ICP + ICP-MS package	ICP	1	10000
9310	8	Co ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9311	8	Cu ppm; ICP + ICP-MS package	ICP	1	10000
9312	8	Ca ppm; ICP + ICP-MS package	ICP-MS	0.1	500
9313	8	Ce ppm; ICP + ICP-MS package	ICP-MS	0.1	500
9315	8	Fe %; ICP + ICP-MS package	ICP	0.01	25.0
9316	8	La ppm; ICP + ICP-MS package	ICP-MS	0.5	500
9317	8	Pb ppm; ICP + ICP-MS package	ICP-MS/ICP	0.5	10000
9318	8	Li ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9319	8	Mg %; ICP + ICP-MS package	ICP	0.01	15.00
9320	8	Mn ppm; ICP + ICP-MS package	ICP	5	10000
9321	8	Mo ppm; ICP + ICP-MS package	ICP	0.2	10000
9322	8	Ni ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9323	8	Nb ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9324	8	P ppm; ICP + ICP-MS package	ICP	10	10000
9325	8	K %; ICP + ICP-MS package	ICP	0.01	10.00
9326	8	Rb ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9327	8	Ag ppm; ICP + ICP-MS package	ICP-MS/ICP	0.05	100.0
9328	8	Na %; ICP + ICP-MS package	ICP	0.01	10.00
9329	8	Sr ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9330	8	Ta ppm; ICP + ICP-MS package	ICP-MS	0.05	100.0
9331	8	Te ppm; ICP + ICP-MS package	ICP-MS	0.05	500
9332	8	Tl ppm; ICP + ICP-MS package	ICP-MS	0.02	500
9333	8	Tl %; ICP + ICP-MS package	ICP	0.01	500
9334	8	W ppm; ICP + ICP-MS package	ICP-MS/ICP	0.1	10000
9335	8	V ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9336	8	Y ppm; ICP + ICP-MS package	ICP	1	10000
9337	8	X ppm; ICP + ICP-MS package	ICP-MS	0.1	500
9338	8	Zn ppm; ICP + ICP-MS package	ICP	2	10000

The results of this assay were based solely upon the content of the sample submitted. Any decision to invest should be made only after the potential investment value of the claim or deposit has been determined based on the results of assays of multiple samples of geologic materials collected by the prospective investor or by a qualified person selected by him/her and based on an evaluation of all engineering data which is available concerning any proposed project. Statement required by Nevada State Law NRS 519



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..

Page Number : 1 A  
 Total Pages : 1  
 Certificate Date: 22-JUN 1999  
 Invoice No. : 19919911  
 P.O. Number : 45856  
 Account : RCO

Project: G.S.L  
 Comments: ATTN: ARTHUR M. AMBROSINO

**CERTIFICATE OF ANALYSIS A9919911**

SAMPLE	PREP CODE	Al % (ICP)	Sb ppm (ICP)	Ba ppm (ICP)	Be ppm (ICP)	Bi ppm (ICP)	cd ppm (ICP)	Ca % (ICP)	Ce ppm (ICP)	Cs ppm (ICP)	Cr ppm (ICP)	Co ppm (ICP)	Cu ppm (ICP)	Ga ppm (ICP)	Ge ppm (ICP)
0491832	299 --	6.50	0.3	490	1.55	0.06	0.50	2.12	52.7	0.90	294	12.8	< 1	18.9	1.5
0427995	299 --	5.99	< 0.1	400	0.60	0.01	0.14	1.79	36.2	0.50	273	6.8	4	9.8	0.9
0425579	299 --	6.82	0.1	510	1.40	0.02	0.22	2.40	45.5	0.60	273	8.8	4	17.3	1.2
0482333	299 --	6.61	0.1	490	1.15	0.03	0.36	2.63	56.8	0.55	260	11.6	1	16.9	1.2
0423268	299 --	7.50	0.1	590	1.55	0.03	0.38	2.47	62.0	0.85	166	9.6	1	21.7	1.3
0487092	299 --	5.95	< 0.1	510	0.65	0.01	0.22	1.98	36.4	0.65	183	6.2	3	13.9	1.0
0480509	299 --	6.67	0.1	480	1.30	0.08	0.52	2.18	47.2	0.80	267	12.2	1	18.8	1.4
0480005	299 --	6.28	0.1	460	1.55	0.03	0.38	2.46	44.4	0.70	259	13.2	3	18.5	1.4

*Arthur M. Ambrosino*

CERTIFICATION:



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Page Number : 1 B  
 Total Pages : 1  
 Certificate Date : 22-JUN 1999  
 Invoice No. : 19919911  
 P. O. Number : 45856  
 Account : RCO

Project : G.S.L  
 Comments : ATTN ARTHUR M. AMBROSINO

## CERTIFICATE OF ANALYSIS A9919911

SAMPLE	PREP CODE	Fe % (ICP)	La ppm (ICP)	Pb ppm (ICP)	Li ppm (ICP)	Mg % (ICP)	Mn ppm (ICP)	Mo ppm (ICP)	Ni ppm (ICP)	Nb ppm (ICP)	P ppm (ICP)	K % (ICP)	Rb ppm (ICP)	Ag ppm (ICP)	Na % (ICP)
0491832	299 --	4.86	21.0	22.0	14.0	0.91	810	2.6	13.0	16.8	670	1.98	68.6	0.50	1.49
0427995	299 --	4.36	18.5	9.5	8.8	0.89	895	1.4	7.8	6.6	420	2.14	45.6	0.10	1.52
0425579	299 --	3.27	20.0	14.0	8.2	0.69	535	2.0	15.6	9.6	600	2.28	71.0	0.25	1.99
0482333	299 --	4.32	23.5	15.5	10.6	0.97	715	2.0	11.4	14.2	740	2.12	65.8	0.50	1.78
0423248	299 --	3.17	25.5	17.0	7.2	0.64	485	1.4	10.6	10.4	880	2.51	78.0	0.50	2.29
0487092	299 --	2.82	16.5	15.0	3.4	0.52	460	1.6	62.7	7.6	460	2.49	66.8	0.25	1.86
0480509	299 --	4.11	19.5	58.0	14.4	0.83	515	2.2	13.4	12.6	420	2.06	70.8	0.30	1.58
0480005	299 --	4.78	18.5	70.5	14.2	1.03	765	2.2	14.6	15.0	300	2.07	72.6	0.30	1.60

*Arthur M. Ambrosino*

CERTIFICATION



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 Total Pages : 1  
 Certificate Date: 22-JUN-1999  
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 P. O. Number : 45856  
 Account : RCO

Project: G.S.L  
 Comments: ATTN: ARTHUR M. AMBROSINO

**CERTIFICATE OF ANALYSIS A9919911**

SAMPLE	PREP CODE	Sr ppm (ICP)	Ta ppm (ICP)	Te ppm (ICP)	Tl ppm (ICP)	Tb ppm (ICP)	Ti % (ICP)	W ppm (ICP)	U ppm (ICP)	V ppm (ICP)	Y ppm (ICP)	Zn ppm (ICP)
0491832	299 --	216	1.05	< 0.05	0.44	7.4	1.02	1.2	1.2	105	32.2	96
0427995	299 --	175.0	0.50	< 0.05	0.28	4.8	0.69	0.7	0.6	84	26.6	66
0425579	299 --	279	0.70	< 0.05	0.42	4.4	0.56	1.0	0.8	62	28.9	52
0482333	299 --	272	1.25	< 0.05	0.40	5.4	0.86	1.0	1.2	92	33.6	62
0423248	299 --	301	0.75	< 0.05	0.58	5.4	0.50	0.8	1.2	62	32.7	54
0487092	299 --	228	0.50	< 0.05	0.42	4.0	0.46	0.7	0.8	51	22.8	46
0480509	299 --	244	0.70	< 0.05	0.44	5.6	0.79	1.1	1.0	84	27.0	88
0480005	299 --	247	0.75	< 0.05	0.44	4.8	0.96	0.7	0.8	95	35.4	74

*Arthur M. Ambrosino*

CERTIFICATION:



# Chemex Labs, Inc.

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 994 Glendale Ave., Unit 3,  
 Sparks, Nevada, U.S.A. 89431  
 PHONE: 775-356-5395 FAX: 775-355-0179

To EARTH & ATMOSPHERIC SCIENCES  
 SUNY ALBANY  
 1400 WASHINGTON AVE  
 ALBANY, NEW YORK  
 12222

A9919909

Comments: ATTN: ARTHUR M AMBROSINO

## CERTIFICATE A9919909

CO) - EARTH & ATMOSPHERIC SCIENCES

Project G.S.L.  
 J.#: 45856

Samples submitted to our lab in Mississauga, ON.  
 is report was printed on 24-JUN-1999.

## SAMPLE PREPARATION

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION
217 9300	8 8	Geochem ring entire sample T127 special code

ANALYTICAL PROCEDURES					
CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION	METHOD	DETECTION LIMIT	UPPER LIMIT
975	8	Au Ppb: ICP-fluorescence package	FA-ICP-AFS	2	10000
976	8	Pt Ppb: ICP-fluorescence package	FA-ICP-AFS	5	10000
977	8	Pd Ppb: ICP-fluorescence package	FA-ICP-AFS	2	10000
13	8	As Ppb: HNO3-aqua regia digest	AAS-HYDRIDE/EDL	1	10000
20	8	Hg Ppb: HNO3-HCl digestion	AAS-FLAMELESS	10	100000

The results of this assay were based solely upon the content of the sample submitted. Any decision to invest should be made only after the potential investment value of the claim or deposit has been determined based on the results of assays of multiple samples of geologic materials collected by the prospective investor or by a qualified person selected by him/her and based on an evaluation of all engineering data which is available concerning any proposed project. Statement required by Nevada State Law NRS 519





# Chemex Labs, Inc.

Analytical Chemists \* Geochemists \* Registered Assayers  
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Page Number : 1  
 Total Pages : 1  
 Certificate Date : 24 JUN 1999  
 Invoice No. : 19919909  
 P.O. Number : 45856  
 Account : RCC

Project : G.S.L.  
 Comments : ATTN: ARTHUR M AMBROSINO

## CERTIFICATE OF ANALYSIS A9919909

SAMPLE	PREP CODE	Au ppb AFS	Pt ppb AFS	Pd ppb AFS	As ppm	Hg ppb			
0491832	2179300	< 2	< 5	< 2	< 1	30			
0427995	2179300	< 2	< 5	< 2	< 4	< 10			
0425579	2179300	< 2	< 5	< 2	1	< 10			
0482333	2179300	< 2	< 5	< 2	2	< 10			
0423248	2179300	< 2	< 5	< 2	1	< 10			
0487092	2179300	< 2	< 5	< 2	5	< 10			
0480509	2179300	< 2	< 5	< 2	1	30			
0480005	2179300	< 2	< 5	< 2	< 1	< 10			

*Arthur M. Ambrosino*

CERTIFICATE



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 ALBANY, NEW YORK  
 12222

A9919907

Comments: ATTN: ARTHUR M. AMBROSINO

## CERTIFICATE A9919907

TO: EARTH & ATMOSPHERIC SCIENCES

Project: G.S.L.  
 # : 45855

samples submitted to our lab in Mississauga, ON.  
 This report was printed on 22-JUN-1999.

### SAMPLE PREPARATION

EMEX CODE	NUMBER SAMPLES	DESCRIPTION
299	18	Pulp; prepped on other workorder

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION	METHOD	DETECTION LIMIT	UPPER LIMIT
9301	18	Al %; ICP + ICP-MS package	ICP	0.01	25.0
9341	18	Sb ppm; ICP + ICP-MS package	ICP-MS	0.1	1000
9302	18	Ba ppm; ICP + ICP-MS package	ICP-MS	10	10000
9303	18	Be ppm; ICP + ICP-MS package	ICP-MS/ICP	0.05	1000
9304	18	Bi ppm; ICP + ICP-MS package	ICP-MS/ICP	0.01	10000
9305	18	Cd ppm; ICP + ICP-MS package	ICP-MS/ICP	0.02	500
9306	18	Ca %; ICP + ICP-MS package	ICP	0.01	25.0
9307	18	Ce ppm; ICP + ICP-MS package	ICP-MS	0.01	500
9308	18	Cr ppm; ICP + ICP-MS package	ICP-MS	0.05	500
9309	18	Cu ppm; ICP + ICP-MS package	ICP-MS/ICP	1	10000
9310	18	Co ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9311	18	Cu ppm; ICP + ICP-MS package	ICP-MS	1	10000
9312	18	Ga ppm; ICP + ICP-MS package	ICP-MS	0.1	500
9313	18	Ge ppm; ICP + ICP-MS package	ICP-MS	0.1	500
9315	18	Fe %; ICP + ICP-MS package	ICP	0.01	25.0
9316	18	La ppm; ICP + ICP-MS package	ICP-MS	0.5	500
9317	18	Pb ppm; ICP + ICP-MS package	ICP-MS/ICP	0.5	10000
9318	18	Li ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9319	18	Mg %; ICP + ICP-MS package	ICP	0.01	15.00
9320	18	Mn ppm; ICP + ICP-MS package	ICP	5	10000
9321	18	Mo ppm; ICP + ICP-MS package	ICP	0.2	10000
9322	18	Ni ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9323	18	Nb ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9324	18	P ppm; ICP + ICP-MS package	ICP	10	10000
9325	18	K %; ICP + ICP-MS package	ICP	0.01	10.00
9326	18	Rb ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9327	18	Ag ppm; ICP + ICP-MS package	ICP-MS/ICP	0.05	100.0
9328	18	Na %; ICP + ICP-MS package	ICP	0.01	10.00
9329	18	Sr ppm; ICP + ICP-MS package	ICP-MS/ICP	0.2	10000
9330	18	Ta ppm; ICP + ICP-MS package	ICP-MS/ICP	0.05	100.0
9331	18	Te ppm; ICP + ICP-MS package	ICP-MS	0.05	500
9332	18	Tl ppm; ICP + ICP-MS package	ICP-MS	0.02	500
9333	18	Th ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9334	18	Ti %; ICP + ICP-MS package	ICP	0.01	10.00
9335	18	W ppm; ICP + ICP-MS package	ICP-MS/ICP	0.1	10000
9336	18	U ppm; ICP + ICP-MS package	ICP-MS	0.2	500
9337	18	V ppm; ICP + ICP-MS package	ICP	1	10000
9338	18	Y ppm; ICP + ICP-MS package	ICP-MS	0.1	500
9339	18	Zn ppm; ICP + ICP-MS package	ICP	2	10000

### ANALYTICAL PROCEDURES

The results of this assay were based solely upon the content of the sample submitted. Any decision to invest should be made only after the potential investment value of the claim or deposit has been determined based on the results of assays of multiple samples of geologic materials collected by the prospective investor or by a qualified person selected by him/her and based on an evaluation of all engineering data which is available concerning any proposed project. Statement required by Nevada State Law NRS 619



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..

Page Number : 1.A  
 Total Pages : 1  
 Certificate Date: 22 JUN 1999  
 Invoice No. : 19919907  
 P.O. Number : 45855  
 Account : RCC

Project: G.S.I.  
 Comments: ATTN: ARTHUR M. AMBROSINO

**CERTIFICATE OF ANALYSIS A9919907**

SAMPLE	PREP CODE	Al % (ICP)	Sb ppm (ICP)	Ba ppm (ICP)	Be ppm (ICP)	Bi ppm (ICP)	Cd ppm (ICP)	Ca % (ICP)	Ce ppm (ICP)	Cs ppm (ICP)	Cr ppm (ICP)	Co ppm (ICP)	Cu ppm (ICP)	Ga ppm (ICP)	Ge ppm (ICP)
0433040	299 --	7.37	0.2	570	2.65	0.05	0.32	1.61	44.6	1.25	208	9.2	1	20.5	1.4
0467896	299 --	6.57	0.2	500	1.75	0.03	0.22	2.08	45.2	0.65	295	10.0	5	17.5	1.4
0461023	299 --	5.90	0.4	490	2.10	0.04	0.24	1.26	42.5	1.00	324	8.2	6	16.9	1.4
047122140	299 --	6.18	0.1	500	1.65	0.03	0.14	1.56	49.4	0.90	274	7.6	7	17.2	1.3
0497896	299 --	6.26	0.2	580	2.70	0.04	0.34	1.44	65.1	1.30	227	10.0	4	19.6	1.5
0420832	299 --	7.14	0.2	540	1.90	0.04	0.40	1.69	54.8	1.05	214	10.0	1	21.0	1.5
0480923	299 --	7.00	0.1	510	1.30	0.02	0.20	2.42	42.2	0.75	271	11.2	3	18.7	1.4
0420008	299 --	5.75	0.1	340	1.40	0.04	0.56	2.05	69.4	0.55	355	17.2	< 1	17.9	2.0
047176192	299 --	6.84	0.1	570	2.25	0.03	0.26	1.78	66.7	1.00	223	10.8	6	19.5	1.6
0501030	299 --	7.34	0.3	550	2.00	0.03	0.30	2.39	50.1	0.80	275	11.6	6	19.3	1.3
0470418	299 --	6.00	0.1	630	1.75	0.06	0.30	1.92	70.7	0.70	247	9.2	1	19.8	1.5
0483770	299 --	8.57	0.4	670	3.70	0.13	0.26	1.67	132.0	3.65	56	19.4	15	28.3	1.9
0470004	299 --	5.75	0.2	500	1.45	0.04	0.30	1.35	35.4	0.85	366	7.4	7	15.7	1.3
0440006	299 --	7.12	0.3	570	2.00	0.08	0.40	1.55	53.6	1.50	219	11.8	4	20.3	1.4
04892128	299 --	7.27	0.1	570	2.50	0.03	0.36	2.37	60.4	0.90	174	8.6	4	20.0	1.4
0481317	299 --	6.81	0.2	540	1.85	0.06	0.22	2.08	57.6	1.10	244	11.2	8	18.4	1.4
0450012	299 --	6.81	0.1	560	1.90	0.06	0.38	1.63	51.7	1.10	225	8.2	< 1	20.0	1.4
0430006	299 --	7.00	0.2	550	1.55	0.04	0.32	1.94	56.4	1.10	218	11.2	4	19.4	1.4

*Arthur M. Ambrosino*

CERTIFICATION:



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Project: G.S.I.  
 Comments: ATTN: ARTHUR M. AMBROSINO

Page Number : 1-B  
 Total Pages : 1  
 Certificate Date : 22-JUN-1999  
 Invoice No. : 19919907  
 P.O. Number : 45855  
 Account : RCC

SAMPLE	PREP CODE	Fe % (ICP)	La ppm (ICP)	Pb ppm (ICP)	Li ppm (ICP)	Mg % (ICP)	Mn ppm (ICP)	Mo ppm (ICP)	Ni ppm (ICP)	Nb ppm (ICP)	P ppm (ICP)	K % (ICP)	Rb ppm (ICP)	Ag ppm (ICP)	Na % (ICP)	CERTIFICATE OF ANALYSIS	
																A9919907	
0433040	299 --	3.23	20.5	17.0	18.0	0.56	430	1.6	11.0	10.6	360	2.49	89.6	0.40	1.75		
0467896	299 --	3.43	21.0	14.5	16.6	0.74	565	2.0	11.0	18.4	400	2.27	79.2	0.45	1.74		
0461023	299 --	2.93	18.0	17.0	17.8	0.52	455	2.0	10.8	6.4	380	2.85	88.6	0.30	1.38		
047122140	299 --	2.54	26.0	15.0	20.8	0.59	330	1.6	11.0	6.0	390	2.56	91.2	0.15	1.71		
0497896	299 --	3.54	32.5	17.5	16.6	0.55	565	2.0	10.6	13.6	630	2.71	108.0	0.50	1.59		
0420832	299 --	3.17	22.5	17.5	18.0	0.59	545	1.6	11.0	10.8	530	2.35	90.0	0.55	1.78		
0480923	299 --	3.23	18.0	42.5	21.0	0.84	500	2.0	12.2	8.2	330	2.27	83.2	0.15	1.76		
0420008	299 --	8.52	30.0	13.0	15.0	1.18	1495	3.4	13.6	30.0	380	1.65	57.4	0.75	1.14		
047176192	299 --	4.08	34.0	17.0	18.2	0.76	645	3.2	12.0	12.8	750	2.67	92.6	0.45	1.85		
0501030	299 --	3.60	19.0	16.0	16.8	0.87	585	2.0	19.4	11.0	580	2.27	79.4	0.40	1.80		
0470418	299 --	5.03	29.5	17.0	12.6	0.71	810	2.6	6.8	15.0	1530	2.27	71.4	0.35	1.49		
0483770	299 --	4.57	57.5	22.5	38.0	0.94	680	1.0	25.8	14.0	920	3.01	145.5	0.35	1.68		
0470004	299 --	2.79	16.0	22.5	13.4	0.51	415	2.2	9.8	8.6	210	2.40	83.2	0.40	1.46		
0440006	299 --	3.85	24.0	26.0	22.8	0.71	575	1.8	13.8	10.8	580	2.38	88.2	2.95	1.55		
0489218	299 --	3.03	27.0	18.5	11.0	0.60	490	1.6	14.2	10.8	840	2.63	90.0	0.60	2.20		
0483337	299 --	3.43	24.5	15.0	19.2	0.79	525	2.8	89.5	9.6	460	2.35	81.6	0.25	1.66		
0450012	299 --	3.53	22.5	19.0	15.6	0.54	525	1.8	9.2	12.4	510	2.37	83.4	0.50	1.63		
0430006	299 --	3.37	23.5	17.0	13.4	0.68	490	1.8	11.4	11.0	600	2.44	85.6	0.55	1.82		

*Arthur M. Ambrosino*

CERTIFICATION



**Chemex Labs, Inc.**  
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Project: G.S.I.  
 Comments: ATTN: ARTHUR M. AMBROSINO

Page Number : 1 C  
 Total Pages : 1  
 Certificate Date: 22-JUN-1999  
 Invoice No : 19919907  
 P.O. Number : 45955  
 Account : RCC

**CERTIFICATE OF ANALYSIS A9919907**

SAMPLE	PRSP CODE	Sr ppm (ICP)	Ta ppm (ICP)	To ppm (ICP)	Tl ppm (ICP)	Tb ppm (ICP)	Ti % (ICP)	W ppm (ICP)	U ppm (ICP)	V ppm (ICP)	Y ppm (ICP)	Zn ppm (ICP)
0433040	299 --	226	0.40	< 0.05	0.52	5.2	0.52	0.8	1.0	61	26.8	58
0457896	299 --	249	0.50	< 0.05	0.44	5.0	0.60	0.9	0.8	65	31.4	54
0461023	299 --	167.0	0.55	< 0.05	0.52	5.8	0.41	0.9	0.8	49	24.8	62
047122140	299 --	203	0.45	< 0.05	0.52	7.0	0.30	0.8	0.8	44	24.2	48
0497896	299 --	175.0	0.75	< 0.05	0.64	7.0	0.54	1.1	1.4	61	46.5	66
0430832	299 --	230	0.85	< 0.05	0.52	6.2	0.51	0.8	1.2	61	29.6	60
0480923	299 --	290	0.50	< 0.05	0.48	4.6	0.56	1.0	0.8	66	23.2	56
0420008	299 --	153.5	1.75	< 0.05	0.34	11.4	1.98	1.5	1.8	162	76.4	92
047176192	299 --	226	0.60	< 0.05	0.56	9.4	0.67	0.9	1.4	76	40.9	80
0501030	299 --	286	1.05	< 0.05	0.46	4.8	0.64	0.9	1.0	74	27.6	70
0470418	299 --	206	0.85	< 0.05	0.50	3.8	0.84	1.0	0.8	55	40.9	112
0483770	299 --	228	0.90	< 0.05	0.90	13.4	0.52	0.8	2.0	97	54.7	108
0470004	299 --	183.5	0.40	< 0.05	0.54	4.6	0.44	1.7	1.0	48	23.2	62
0440006	299 --	230	0.45	< 0.05	0.60	6.6	0.57	0.9	1.2	75	35.7	88
04892128	299 --	277	0.55	< 0.05	0.56	6.0	0.51	0.8	1.4	58	35.8	54
0483377	299 --	251	0.40	< 0.05	0.54	6.0	0.54	1.0	1.0	70	27.2	62
0450012	299 --	216	0.50	< 0.05	0.54	5.8	0.60	1.0	1.4	65	30.1	58
0430006	299 --	243	0.50	< 0.05	0.54	6.6	0.55	1.0	1.4	67	29.5	56

*Arthur M. Ambrosino*

CERTIFICATION:



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A9919904

Comments: ATTN: ARTHUR M. AMBROSINO

## CERTIFICATE A9919904

CO) EARTH & ATMOSPHERIC SCIENCES  
 Project: G.S.L. 45855  
 Samples submitted to our lab in Missisquoi, ON.  
 is report was printed on 24-JUN-1999.

## SAMPLE PREPARATION

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION
217	18	Geochem ring entire sample
9300	18	T127 special code

## ANALYTICAL PROCEDURES

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION	METHOD	DETECTION LIMIT	UPPER LIMIT
975	18	Au Ppb: ICP-fluorescence package	FA-ICP-AFS	2	10000
976	18	Pt Ppb: ICP-fluorescence package	FA-ICP-AFS	5	10000
977	18	Pd Ppb: ICP-fluorescence package	FA-ICP-AFS	2	10000
13	18	As Ppm: HNO3-aqua regia digest	AAS-HYDRIDE/EDL	1	10000
20	18	Hg Ppb: HNO3-RC1 digestion	AAS-FLAMELESS	10	100000

The results of this assay were based solely upon the content of the sample submitted. Any decision to invest should be made only after the potential investment value of the claim or deposit has been determined based on the results of assays of multiple samples of geologic materials collected by the prospective investor or by a qualified person selected by him/her and based on an evaluation of all engineering data which is available concerning any proposed project. Statement required by Nevada State Law NRS 518



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 SUNY ALBANY  
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 ALBANY, NEW YORK  
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 Certificate Date: 24 JUN 1999  
 Invoice No. 19919904  
 P. O. Number : 45855  
 Account RCO

Project: G.S.L.  
 Comments: ATTN: ARTHUR M. AMBROSINO

## CERTIFICATE OF ANALYSIS A9919904

SAMPLE	PREP CODE	Au ppb AFS	Pt ppb AFS	Pd ppb AFS	As ppm	Hg ppb				
0433040	2179300	< 2	< 5	< 2	< 4	30				
0467896	2179300	< 2	< 5	12	< 1	< 10				
0461023	2179300	< 2	< 5	< 2	< 2	< 10				
047122140	2179300	< 2	< 5	< 2	< 2	< 10				
0497896	2179300	< 2	< 5	< 2	< 1	< 10				
0420832	2179300	< 2	< 5	< 2	3	30				
0480923	2179300	< 2	< 5	< 2	2	< 10				
0420008	2179300	< 2	< 5	< 2	4	< 10				
047176192	2179300	< 2	< 5	< 2	4	< 10				
0501030	2179300	< 2	< 5	< 2	< 1	< 10				
0470418	2179300	< 2	< 5	< 2	< 1	40				
0483770	2179300	< 2	< 5	< 2	< 1	< 10				
0470004	2179300	< 2	< 5	< 2	< 1	< 10				
0440006	2179300	< 2	< 5	< 2	< 1	40				
04892128	2179300	< 2	< 5	< 2	< 1	< 10				
0483337	2179300	< 2	< 5	< 2	1	< 10				
0450012	2179300	< 2	< 5	< 2	4	30				
0430006	2179300	< 2	< 5	< 2	< 1	< 10				

*Arthur M. Ambrosino*

CERTIFICATION:



**OVERBURDEN DRILLING MANAGEMENT LIMITED**

Mr. Arthur M. Ambrosino  
Trackside I, Apt. 3C2  
Johnstown, NY  
U.S.A. 12095

September 07, 2000

Fax: (514) 442-5825

Dear Mr. Ambrosino:

Re: Heavy Mineral Suite of Estuary Sand Sample AA-01, New York

We have completed our heavy mineral processing of the above composited drill core sample and are pleased to report the results.

We extracted the -2.0 mm heavy minerals using a shaking table followed by a heavy liquid separation at S.G. 3.2 (Table 1), performed a ferromagnetic separation to remove magnetite, and sieved off two nonferromagnetic size fractions — medium (0.25-0.5 mm) and fine (0.125-0.25 mm) sand — for mineral logging. To ensure that we would be able to log all of the minerals accurately, we checked their paramagnetic properties beforehand and analyzed representative grains by energy dispersive x-ray spectrometry (EDS; Table 2) using our scanning electron microscope (SEM). These analyses were done on unpolished grains and, although reliable, were not sufficiently precise to be recorded for publication.

The medium and fine sand sized heavy minerals were found to contain 7.1 and 7.6 weight percent magnetite respectively (Table 1). The percentages of nonferromagnetic minerals were determined by counting grains and are therefore expressed in grain percent of the nonferromagnetic fraction, not weight percent of the total concentrate. The minerals were scattered thinly over a lined plate and 200 contiguous grains were counted along the lines. The minimum count of one grain therefore equals 0.5 percent (Table 3).

The major minerals present are almandine, hornblende, two clinopyroxenes (brown to dark green augite and pale green to colourless diopside), orthopyroxene (dark to pale brown hypersthene, bronzite and enstatite) and ilmenite. Minor minerals occurring at the 0.5-2 percent level are titanite, leucoxene, apatite, zircon and goethite. Only a low trace of epidote was observed and hematite, rutile, kyanite and staurolite are conspicuously absent. The general abundance of pyroxene and ilmenite suggests major inputs from a mafic plutonic terrane. If this is the case, the unexamined 1-2 mm fraction of our heavy mineral concentrate may contain gabbro grains.



Page 2

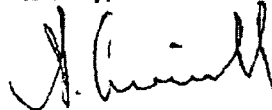
September 07, 2000

Mr. Arthur M. Ambrosino  
Heavy Mineral Suite of Estuary Sand Sample AA-01

The fine sand fraction contains significantly more orthopyroxene (16.5 versus 7 percent) and correspondingly less clinopyroxene (17.5 versus 24.5 percent) than the medium sand fraction (Table 3), possibly because the orthopyroxene occurs in less equant, splintery grains, but the proportions of the other major and minor nonferromagnetic minerals and of magnetite (Table 1) are similar. Therefore it may be assumed that the minerals of the -0.125 mm suite, which are too fine-grained to log, are also similar.

I hope these observations are helpful. Please call me if you have any questions.

Sincerely,

A handwritten signature in black ink, appearing to read "Stuart Averill". The signature is fluid and cursive, with a large initial "S" and a long, sweeping tail.

Stuart Averill,  
President

Bulk Sample Tabling Weights and Descriptions

Sample Number	Weight (kg)		Sample Description										CLASS				
	Bulk Rec'd	Table <2 mm Class	Clasts >2.0 mm Percentage					Matrix <2.0 mm Distribution						Colour			
			Size	V/S	OR	LS	OT	SU	SD	ST	CY	SAND			CLAY		
AA-01	8.4	8.2	0.2	8.0	P	10	90	0	0	0	U	+	Y	-	B	B	SILTY SAND

HMC Processing Weights

Sample Number	<2.0 mm Table Concentrate Heavy Liquid Separation S.G. 3.20																
	Fenestragic Fraction (g)					Nonfenestragic Fraction (g)											
	Total Mag	Heavy Liquid	Lights	Total	Wt% of	Total Mag	Heavy Liquid	Lights	Total	Wt% of							
AA-01	1276.4	696.6	44.4	17.7	15.2	7.5%	8.3	7.1%	3.2	535.4	38.0	168.1	187.8	108.6	28.3	4.6	

Table 1 - Sample weights and descriptions.

<u>No. of Grains</u>	<u>Description</u>	<u>Binocular Microscope I.D.</u>	<u>SEM I.D.</u>
1	Dark green	Augite or diopside	Augite
1	Brown	Augite or orthopyroxene	Augite
1	Beige-white	Clinopyroxene	Augite
3	Colourless, elongate	Kyanite or diopside	Diopside
1	Very pale green	Clinopyroxene	Diopside
5	Brown-orange to brown	Orthopyroxene	Hypersthene-bronzite
2	Pale brown to colourless	Enstatite or epidote	Bronzite-enstatite
1	Orange-brown	Leucoxene or goethite	Goethite
1	Colourless to white	Apatite	Apatite

Table 2 - SEM analyses of representative medium sand grains by EDS.

Mineral	0.25-0.5 mm		0.125-0.25 mm	
	No. of Grains	% of Grains	No. of Grains	% of Grains
Almandine	69	34.5	65	32.5
Hornblende	50	25.0	42	21.0
Augite	28	14.0	18	9.0
Diopside	21	10.5	17	8.5
Orthopyroxene	14	7.0	33	16.5
Ilmenite	13	6.5	21	10.5
Titanite	1	0.5	0	0.0
Leucoxene	0	0.0	1	0.5
Apatite	2	1.0	2	1.0
Zircon	1	0.5	0	0.0
Goethite	1	0.5	1	0.5
	200	100	200	100

Table 3 - 200-grain counts of medium and fine sand size fractions.



# ALS Chemex

Chemex Labs, Inc.  
 Analytical Chemistry & Geochronology & Regenerant Assays  
 994 Glendale Ave., Unit 3 Sparks  
 Nevada, U.S.A. 89431  
 PHONE 775-356-5395 FAX 775-355-0179

To: EARTH & ATMOSPHERIC SCIENCES  
 SUNY ALBANY  
 1400 WASHINGTON AVE  
 ALBANY, NEW YORK  
 12222 USA

A0025777

Comments ATTN: ARTHUR M. AMBROSINO CC: ARTHUR M. AMBROSINO

## CERTIFICATE A0025777

Client: EARTH & ATMOSPHERIC SCIENCES

Project: G.S.L.

Samples submitted to our lab in Sparks, NV.  
 This report was printed on 29-SEP-2000.

## SAMPLE PREPARATION

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION
235	1	Pen con ring to approx 150 mesh
299	5	Pulp: prepped on other workorder

## ANALYTICAL PROCEDURES

CHEMEX CODE	NUMBER SAMPLES	DESCRIPTION	METHOD	DETECTION LIMIT	UPPER LIMIT
382	1	TiO2 %; Peroxide fush-Fe precip	AAS	0.01	100.0
332	1	TbO2 %; Assay	NAA	0.001	100.0
873	1			0.001	100.00
359	1	Ce %; Assay	NAA	0.01	100.0
372	1	La %; Assay	NAA	0.001	100.0
451	1	FeO %; Acid decomposition	TITRATION	0.01	100.0
327	1	Fe %; HClO4-HNO3-HP digestion	AAS	0.01	100.0

The results of this assay were based solely upon the content of the sample submitted. Any decision to invest should be made only after the potential investment value of the claim or deposit has been determined based on the results of assays of multiple samples of geologic materials collected by the prospective investor or by a qualified person selected by him/her and based on an evaluation of all engineering data which is available concerning any proposed project. Statement required by Nevada State Law NRS 519



**ALS Chemex**

2000 Lakes Blvd.  
 An American Enterprise • One Technology Center • Redwood City, California  
 994 Glendale Ave. Unit 3, Sparks, Nevada 89431  
 PHONE 775-356-5395 FAX 775-355-0179

To: EARTH & ATMOSPHERIC SCIENCES  
 SUNY ALBANY  
 1400 WASHINGTON AVE  
 ALBANY, NEW YORK  
 12222 USA

Page Number 1  
 Total Pages 1  
 Certificate Date 29 SEP 2000  
 Invoice No. 10025777  
 P.O. Number  
 Account RCO

Project G.S.L.  
 Comments ATTN: ARTHUR M. AMBROSINO CC: ARTHUR M. AMBROSINO

SAMPLE		PREP CODE	TiO2 %	TbO2 NA %	Y2O3 %	Ce NAA %	La NAA %	FeO %	Fe % tot.dig	CERTIFICATE OF ANALYSIS	
										A0025777	
TIGSL01		235	6.56								
THGSL02		299		0.003							
YGSL03		299			0.026						
CEGSL04		299				0.01					
LAGSL05		299					0.007				
PEGSL06		299						18.55	25.1		

*Arthur M. Ambrosino*

CERTIFICATION

**Test T871 - Sacandaga Lake, New York**

<b>SAMPLE:</b>	<b>T871A 20101</b>	<b>T871B 20102</b>	<b>T871C 20103</b>
----------------	------------------------	------------------------	------------------------

Slimes(%)	18.5	7.7	11.1
THM (%)	8.5	4.1	4.3
THM d <sub>50</sub> (µm)	193	190	301

**THM MINERALOGY**

	Calc.	SEM	Calc	SEM	Calc.	SEM
Titanomag	7.2	1.2	5.9	1.6	5.7	-
Altered Ilmenite		1.2		-		0.8
Ilmenite	16.8	20.8	13.7	14.8	9.4	9.2
Ilm./Silicate/Fe oxide		2.8		1.2		2.0
Rutile	0.3	-	0.4	-	0.3	0.4
Other Ti Minerals	4.7		4.4		3.7	
Zircon	0.6	0.8	0.6	1.2	0.3	0.4
Monazite	0.2	-	0.3	-	0.3	-
Other Heavy Minerals	70.1	73.2	74.7	81.2	80.2	87.2

**ILMENITE PRODUCT CHEMISTRY**

	Best Mag Fractions	MicroProbe	BMF	MP	BMF	MP
TiO <sub>2</sub>	44.11	47.53	42.00	48.00	37.22	47.28
FeT	37.12	39.04	34.94	38.50	33.90	39.26
Al <sub>2</sub> O <sub>3</sub>	0.68		1.41		2.16	
CaO	0.54	0.03	1.26	0.01	1.65	0.09
MgO	0.74	0.51	0.93	0.44	1.58	0.42
P <sub>2</sub> O <sub>5</sub>	0.02		0.03		0.08	
MnO	0.79	0.47	0.94	0.58	0.81	0.56
SiO <sub>2</sub>	2.81	0.19	6.17	0.28	9.80	0.23
Cr <sub>2</sub> O <sub>3</sub>	0.04	0.03	0.03	0.03	0.05	0.03
V <sub>2</sub> O <sub>5</sub>	0.26		0.21		0.24	
ZrO <sub>2</sub>	0.02		0.01		0.01	

**ILMENITE PRODUCT (Conductor)**

As % of conductor fraction	54.8	35.7	31.4
As % of THM	14.6	10.5	7.5
As % of sample	1.2	0.4	0.3

Note: MicroProbe average oxide values for 83, 88 and 66 grains respectively.

## **Appendix E**

### **RADIOLOGICAL SAFETY DATA**



U.C.L.A. Radiation Safety

Spectrum name: AN1E202.AN1

Sample description

Monsanite Sand Samples; 03/06/2000

Spectrum Filename: An1E202.An1

Acquisition information

Start time 06-Mar-00 10:34:46  
Live time 3600  
Real time 3608  
Dead time .23%  
Detector/Geometry IDs 0 & 0

Detector system

CHS-GAMMA SPEC1 MCB 9

Calibration

Filename: An1E202.An1  
Created: 10-Feb-00 15:58:22 & 15-Feb-00 16:20:38  
I-129 Calibration; 10/20/99; charcoal cartridge- 2" from the det  
e  
Zero offset .797 keV; Gain .301 keV/channel

Library Files

Main analysis library: C:\USER\WYCHCTER.LIB

Analysis parameters

Start channel 10 for an energy of 3.81keV  
Stop channel 8144 for an energy of 2455.19keV  
Peak rejection level 50.000%  
Activity scaling factor  $1.0000E+00 / ( 3.7000E+04 * 1.0000E+00 )$   
= 2.7027

E-05

Detection limit method:  
MDA - EG&G ORTEC method

Additional random error: 1.0000000E+00  
Additional systematic error: 1.0000000E+00  
Background width: average of five points.

Corrections

Decay correct to date

Status

YES

Comments

15-Feb-00 12:00:00

```

Decay during acquisition      NO
Peaked background correction  YES      C:\USER\BKG1.Pbc
                                  14-Jul-99 14:02:21
Absorption (Internal)        NO
Geometry correction          NO
Random summing                NO

```

```

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2
U.C.L.A. Radiation Safety      Spectrum name: AN1E202.AN1

```

Energy calibration normalized difference: .1313

\*\*\*\*\* UNIDENTIFIED PEAK SUMMARY \*\*\*\*\*

CTED	CHANNEL	ENERGY	COUNTS	BACKGROUND	NET AREA	EFFICIENCY	UNCERT	FWHM	SUSPECTED
IDE					* AREA	1 SIGMA %	keV	NUCL	
	53.02	16.76	1810.	2035.	6.944E+04	3.70	1.461	-	
	D								
	64.34	20.17	1702.	1799.	7.238E+04	4.01	1.467	-	
	D								
	2451.09	739.02	184.	1985.	0.000E+00	2.44	2.394	-	
	D								
	2578.47	777.39	210.	899.	0.000E+00	4.04	2.419	-	
	D								

s Peak fails shape tests.  
D Peak area deconvoluted.

-----  
This section based on library: C:\USER\WYCHCTER.LIB

\*\*\*\*\* IDENTIFIED PEAK SUMMARY \*\*\*\*\*

NUCLIDE	PEAK CENTROID	BACKGROUND	NET AREA	INTENSITY	UNCERT
FWHM	CHANNEL	ENERGY	COUNTS	COUNTS	CTS/SEC 1 SIGMA %
keV					

ZN-65	24.36	8.13	0.	0.	.000	.00	
.000D							
BI-207	32.56	10.60	1.	0.	.000	.00	
.000D							
PB-214	33.22	10.80	1025.	0.	.000	.00	1
.449D							
PB-210	33.22	10.80	1.	0.	.000	123.29	
.000D							
TH-234	41.53	13.30	912.	2881.	.800	2.38	1
.454D							
PA-234	42.52	13.60	3713.	278.	.077	31.60	1
.455D							
Y-88	44.35	14.15	4157.	379.	.105	24.62	1
.456D							
CO-57	45.21	14.41	1343.	0.	.000	.00	
.288D							
Y-88	49.83	15.80	4392.	501.	.139	19.25	1
.459D							
SN-113	77.06	24.00	2544.	1073.	.298	7.32	1
.474D							
SN-113	77.76	24.21	3097.	514.	.143	15.95	1
.475D							
SN-113	88.02	27.30	2767.	799.	.222	9.95	1
.481D							
SN-113	90.01	27.90	2868.	672.	.187	11.92	1
.482D							
CS-137	103.03	31.82	2542.	845.	.235	9.11	1
.489D							
CS-137	104.26	32.19	3062.	308.	.086	26.05	1
.490D							
CS-137	118.24	36.40	2467.	836.	.232	9.09	1
.498D							
EU-152	128.61	39.52	2950.	359.	.100	22.05	1
.504D							
BI-212	129.73	39.86	3314.	0.	.000	.00	1
.504D							
EU-152	130.60	40.12	3065.	319.	.089	25.17	1
.505D							
EU-155	137.87	42.31	1726.	0.	.000	.00	
.778D							
EU-154	137.87	42.31	674.	0.	.000	162.12	
.417D							
EU-155	140.16	43.00	1105.	0.	.000	.00	
.513D							
EU-154	140.16	43.00	1105.	0.	.000	318.02	
.513D							
EU-155	147.80	45.30	1994.	0.	.000	.00	
.000D							
EU-152	148.13	45.40	3326.	283.	.079	29.38	1
.514D							
PB-210	151.85	46.52	3724.	0.	.000	.00	1
.517D							

EU-152 152.12 46.60 3469. 255. .071 33.21 1  
 .517D

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 3  
 U.C.L.A. Radiation Safety Spectrum name: AN1E202.AN1

Nuclide	Channel	Energy	Background	Net area	Cnts/sec	Uncert	FWH
M							
EU-155	159.09	48.70	3085.	0.	.000	.00	
.277D							
EU-154	159.09	48.70	2635.	0.	.000	372.03	
.000D							
EU-154	163.41	50.00	3695.	149.	.041	58.34	1
.523D							
PB-214	174.04	53.20	3270.	19.	.005	422.44	1
.529D							
EU-155	196.65	60.01	5078.	0.	.000	.00	1
.145D							
PA-234	207.25	63.20	4528.	0.	.000	.00	1
.547							
TH-234	208.35	63.53	2819.	115.	.032	62.53	
.692s							
TL-208	239.13	72.80	6538.	725.	.201	16.20	1
.565D							
BI-207	239.13	72.80	1854.	0.	.000	44.84	
.478D							
PB-214	245.81	74.81	8074.	148.	.041	86.30	1
.568D							
PB-212	245.81	74.81	5239.	0.	.000	10736.27	
.727D							
TL-208	246.34	74.97	6811.	1373.	.381	8.92	1
.569D							
BI-207	246.34	74.97	5199.	0.	.000	10790.98	
.447D							
PB-214	253.44	77.11	6128.	158.	.044	70.32	1
.572D							
PB-212	253.44	77.11	2558.	0.	.000	491.34	
.000D							
TL-208	278.98	84.80	5501.	573.	.159	18.78	1
.586D							
BI-207	278.98	84.80	1767.	0.	.000	68.70	
.314D							
EU-155	284.46	86.45	6921.	17.	.005	674.48	1
.589D							
PB-214	286.95	87.20	6148.	287.	.080	39.02	1
.591D							
PB-212	286.95	87.20	4190.	0.	.000	299.61	
.247D							

BI-207	287.28	87.30	4549.	0.	.000	.00	
.217D							
PB-214	295.59	89.80	6197.	129.	.036	86.67	1
.595D							
PB-212	295.59	89.80	1091.	0.	.000	49.66	
.156D							
U-235	296.12	89.96	2113.	0.	.000	.00	
.245D							
TH-234	304.16	92.38	6127.	106.	.029	104.74	1
.600D							
TH-234	305.55	92.80	6013.	231.	.064	47.91	1
.601D							
U-235	307.38	93.35	3993.	0.	.000	.00	
.497D							
PA-234	311.76	94.67	6158.	64.	.018	173.00	1
.604D							
PA-234	324.28	98.44	5718.	58.	.016	184.25	1
.611D							
AC-228	327.63	99.45	5720.	0.	.000	.00	1
.612D							
PA-234	328.46	99.70	5679.	0.	.000	.00	1
.613D							
U-235	346.07	105.00	1977.	0.	.000	.00	
.000D							
EU-155	347.10	105.31	5495.	122.	.034	86.63	1
.623D							
U-235	359.81	109.14	1255.	0.	.000	.00	
.000D							
PA-234	365.99	111.00	5600.	101.	.028	104.87	1
.633D							
PA-234	377.61	114.50	5198.	98.	.027	104.95	1
.639D							
EU-152	401.79	121.78	4001.	9.	.003	962.13	1
.651D							
CO-57	402.75	122.07	4010.	0.	.000	.00	1
.652D							
EU-154	406.17	123.10	5062.	39.	.011	256.91	1
.654D							
PA-234	413.48	125.30	4688.	0.	.000	.00	1
.657D							
AC-228	426.10	129.10	4801.	0.	.000	.00	1
.664D							
PA-234	433.34	131.28	4393.	60.	.017	156.81	1
.668D							
CO-57	450.44	136.43	4205.	0.	.000	.00	1
.676D							
FE-59	471.10	142.65	4212.	48.	.013	192.96	1
.687D							
U-235	474.78	143.76	3170.	0.	.000	.00	
.000D							
PA-234	496.00	150.15	2874.	98.	.027	92.28	
.000s							

U-235 .372s	545.11	164.94	3414.	231.	.064	51.01	
U-235 .758D	614.12	185.72	1269.	13.	.004	375.20	1

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U.C.L.A. Radiation Safety Spectrum name: AN1E202.AN1

Nuclide	Channel	Energy	Background	Net area	Cnts/sec	Uncert	FWH
M							
RA-226 .759D	615.02	185.99	1408.	0.	.000	.00	1
PA-234 .000D	615.05	186.00	1116.	0.	.000	223.58	
FE-59 .769D	636.11	192.34	2019.	106.	.029	60.68	1
PA-234 .782D	664.20	200.80	1342.	0.	.000	.00	1
U-235 .785D	668.58	202.12	1783.	24.	.007	250.65	1
PA-234 .786D	671.17	202.90	2108.	53.	.015	122.21	1
U-235 .790D	679.18	205.31	2249.	10.	.003	650.38	1
AC-228 .796D	692.76	209.40	2126.	123.	.034	53.82	1
PA-234 .290D	750.77	226.87	1306.	26.	.007	235.23	
PA-234 .000D	750.77	226.87	1320.	0.	.000	342.39	
PB-212 .841D	789.82	238.63	1718.	422.	.117	14.72	1
RA-224 .845D	797.69	241.00	2030.	30.	.008	210.26	1
PB-214 .846D	800.74	241.92	1728.	113.	.031	52.96	1
EU-152 .851D	809.88	244.67	1713.	6.	.002	928.67	1
EU-154 .856D	821.07	248.04	1632.	14.	.004	403.50	1
PA-234 .857D	823.92	248.90	1615.	40.	.011	143.11	1
SN-113 .866D	844.31	255.04	1210.	34.	.009	146.92	1
AC-228 .889D	894.98	270.30	1357.	50.	.014	104.80	1
PA-234 .892D	900.96	272.10	1355.	26.	.007	199.97	1

TL-208	918.42	277.36	1114.	43.	.012	111.40	1
.899D							
PA-234	972.68	293.70	759.	25.	.007	159.80	1
.923D							
PB-214	977.73	295.22	960.	86.	.024	52.07	1
.925D							
PB-212	993.90	300.09	878.	52.	.015	81.16	1
.932D							
CR-51	1055.46	318.63	1034.	93.	.026	73.27	1
.108s							
AC-228	1088.00	328.43	128.	2.	.001	403.70	
.000s							
AC-228	1121.10	338.40	774.	134.	.037	30.66	1
.985D							
EU-152	1140.69	344.30	907.	35.	.010	124.24	1
.993D							
PB-214	1165.03	351.63	952.	308.	.086	23.51	1
.613							
PA-234	1224.70	369.60	760.	61.	.017	65.46	2
.027D							
PA-234	1233.33	372.20	744.	25.	.007	156.39	2
.030D							
SN-113	1301.51	392.74	553.	70.	.019	63.51	
.652s							
AC-228	1356.84	409.40	718.	55.	.015	70.23	2
.078D							
EU-152	1362.45	411.09	659.	25.	.007	148.44	2
.080D							
EU-152	1468.55	443.05	272.	37.	.010	65.15	
.550s							
PA-234	1520.19	458.60	368.	20.	.005	139.96	2
.137D							
AC-228	1534.80	463.00	639.	51.	.014	70.91	2
.142D							
CS-134	1570.95	473.89	267.	0.	.000	2310.84	
.253s							
PA-234	1680.22	506.80	707.	51.	.014	75.30	2
.190D							
TL-208	1693.23	510.72	665.	212.	.059	18.52	2
.195D							
PA-234	1702.79	513.60	726.	50.	.014	77.83	2
.198D							
PA-234	1726.03	520.60	617.	0.	.000	.00	2
.205D							
PA-234	1727.36	521.00	524.	59.	.016	56.45	2
.205D							
CS-134	1867.65	563.26	427.	10.	.003	286.50	2
.248D							
PA-234	1878.41	566.50	501.	22.	.006	145.02	2
.252D							
PA-234	1885.71	568.70	518.	19.	.005	167.38	2
.254D							

PA-234 .254D	1887.57	569.26	549.	0.	.000	.00	2
PA-234 .000D	1887.57	569.26	64.	0.	.000	661.21	
CS-134 .254D	1887.67	569.29	549.	0.	.000	.00	2

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Nuclide M	Channel	Energy	Background	Net area	Cnts/sec	Uncert	FWH
BI-207 .255D	1888.93	569.67	580.	0.	.000	.00	2
PA-234 .259D	1903.31	574.00	323.	13.	.004	196.94	2
TL-208 .268D	1933.65	583.14	462.	188.	.052	17.76	2
EU-154 .276D	1962.07	591.70	273.	16.	.004	150.60	2
PA-234 .286D	1998.92	602.80	268.	7.	.002	329.98	2
CS-134 .287D	2005.09	604.66	407.	9.	.003	302.35	2
BI-214 .292D	2020.56	609.32	392.	198.	.055	15.79	2
PA-234 .348s	2065.25	622.78	196.	39.	.011	62.17	
PA-234 .330D	2166.23	653.20	380.	20.	.006	137.15	2
CS-137 .337D	2194.18	661.62	359.	19.	.005	140.61	2
PA-234 .339D	2204.07	664.60	395.	2.	.000	1596.98	2
BI-214 .340D	2206.89	665.45	380.	0.	.000	.00	2
PA-234 .341D	2211.04	666.70	380.	20.	.006	138.66	2
PA-234 .343D	2221.33	669.80	223.	16.	.005	130.34	2
PA-234 .361D	2296.68	692.50	305.	21.	.006	119.13	2
EU-154 .550D	2296.68	692.50	112.	0.	.000	127.81	
PA-234 .366D	2318.59	699.10	343.	40.	.011	67.68	2
NB-94 .368D	2329.88	702.50	319.	38.	.011	68.70	2



PA-234 .444D	2717.90	819.40	176.	13.	.004	151.59	2
PA-234 .446D	2735.49	824.70	182.	20.	.006	98.93	2
PA-234 .447D	2740.80	826.30	183.	0.	.000	.00	2
PA-234 .450D	2756.74	831.10	138.	10.	.003	168.08	2

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Nuclide M	Channel	Energy	Background	Net area	Cnts/sec	Uncert	FWH
MN-54 .452D	2769.05	834.81	123.	0.	.000	.00	2
AC-228 .452D	2771.67	835.60	93.	9.	.002	157.88	2
MN-56 .458	2808.18	846.60	105.	0.	.000	.00	2
CO-56 .000s	2805.00	845.64	27.	6.	.002	102.85	
TL-208 .465D	2854.22	860.47	126.	23.	.006	71.73	2
EU-152 .468D	2877.19	867.39	149.	15.	.004	117.11	2
NB-94 .470D	2889.50	871.10	144.	0.	.000	.00	2
EU-154 .471D	2896.47	873.20	145.	0.	.000	.00	2
PA-234 .472D	2908.09	876.70	140.	11.	.003	158.70	2
PA-234 .474D	2920.73	880.51	159.	0.	.000	.00	2
PA-234 .000D	2920.73	880.51	35.	0.	.000	357.07	
PA-234 .475D	2929.79	883.24	135.	21.	.006	81.00	2
SC-46 .407D	2949.77	889.26	56.	0.	.000	.00	
Y-88 .481D	2978.85	898.02	128.	13.	.004	123.37	2
PA-234 .481D	2980.77	898.60	125.	0.	.000	.00	2
AC-228 .051	3020.85	910.68	109.	165.	.046	16.42	2
PA-234 .492D	3067.06	924.60	94.	10.	.003	146.25	2

PA-234 .492D	3074.03	926.70	131.	7.	.002	240.84	2
BI-214 .495D	3098.43	934.05	124.	35.	.010	47.36	2
PA-234 .502s	3136.67	945.57	76.	29.	.008	61.47	
PA-234 .500	3148.04	949.00	90.	0.	.000	.00	2
EU-152 .505D	3197.83	964.00	113.	0.	.000	.00	2
AC-228 .505D	3199.82	964.60	127.	18.	.005	91.98	2
AC-228 .506D	3214.09	968.90	127.	92.	.025	20.28	2
CO-56 .368D	3242.50	977.46	86.	0.	.000	.00	
PA-234 .509D	3246.94	978.80	122.	7.	.002	238.01	2
PA-234 .509D	3252.58	980.50	129.	2.	.001	868.99	2
PA-234 .510D	3262.21	983.40	104.	12.	.003	125.87	2
EU-154 .514D	3305.02	996.30	62.	7.	.002	167.90	2
EU-154 .516D	3333.23	1004.80	119.	21.	.006	76.02	2
CO-56 .645D	3442.84	1037.83	29.	0.	.000	.00	
CS-134 .316s	3443.50	1038.03	8.	0.	.000	.00	
BI-207 .404s	3523.81	1062.23	41.	6.	.002	164.58	
RB-86 .528D	3571.60	1076.63	82.	14.	.004	98.29	2
EU-152 .529D	3602.04	1085.80	99.	24.	.007	62.72	2
EU-152 .529D	3615.38	1089.82	110.	26.	.007	60.23	2
FE-59 .530D	3646.57	1099.22	63.	11.	.003	110.19	2
EU-152 .531D	3689.21	1112.07	89.	20.	.005	71.36	2
ZN-65 .531D	3700.66	1115.52	133.	0.	.000	.00	2
BI-214 .531D	3716.45	1120.28	97.	77.	.021	21.44	2
SC-46 .531D	3717.25	1120.52	167.	0.	.000	.00	2
BI-214 .262s	3835.05	1156.02	47.	17.	.005	68.52	
CS-134 .514s	3871.65	1167.05	76.	45.	.013	44.03	

CO-60 .301D	3892.16	1173.23	86.	0.	.000	.00
CO-56 .306D	3898.46	1175.13	70.	0.	.000	.00
EU-152 .510s	4025.92	1213.54	76.	14.	.004	124.29
BI-214 .523D	4107.44	1238.11	121.	40.	.011	42.23 2
CO-56 .252D	4108.01	1238.28	76.	0.	.000	.00

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Nuclide	Channel	Energy	Background	Net area	Cnts/sec	Uncert	FWH
M NA-22 .516D	4228.32	1274.54	43.	0.	.000	.00	2
EU-154 .516D	4229.18	1274.80	41.	9.	.002	109.19	2
BI-214 .515D	4249.62	1280.96	74.	13.	.003	101.32	2
FE-59 .512D	4284.79	1291.56	58.	14.	.004	83.95	2
EU-152 .510D	4310.11	1299.19	53.	15.	.004	73.37	2
CO-60 .078s	4415.33	1330.90	54.	39.	.011	48.85	1
PA-234 .493D	4488.64	1353.00	45.	19.	.005	54.11	2
CO-56 .497D	4512.60	1360.22	6.	0.	.000	.00	
CS-134 .489D	4528.89	1365.13	33.	10.	.003	84.75	2
NA-24 .488D	4540.24	1368.55	45.	9.	.003	107.68	2
BI-214 .484D	4570.43	1377.65	34.	27.	.008	35.84	2
PA-234 .477D	4625.00	1394.10	25.	7.	.002	116.30	2
EU-154 .476D	4637.28	1397.80	38.	6.	.002	144.67	2
BI-214 .474D	4649.55	1401.50	32.	19.	.005	46.62	2
BI-214 .471D	4671.05	1407.98	22.	26.	.007	31.78	2
EU-152 .471D	4671.38	1408.08	47.	0.	.000	.00	2

PA-234	4819.09	1452.60	21.	3.	.001	264.09	2
.449D							
AC-228	4840.98	1459.20	1036.	127.	.035	37.01	2
.446D							
K-40	4846.12	1460.75	210.	973.	.270	3.83	2
.445D							
AC-228	4954.90	1493.54	7.	40.	.011	24.05	
.362s							
BI-214	5016.44	1512.09	0.	34.	.009	17.15	
.452s							
AC-228	5283.55	1592.61	21.	40.	.011	34.19	
.442s							
EU-154	5299.79	1597.50	3.	6.	.002	56.68	
.407s							
BI-212	5377.64	1620.97	4.	30.	.008	25.07	
.435s							
AC-228	5408.19	1630.18	4.	13.	.004	38.31	
.420s							
BI-214	5511.34	1661.28	4.	2.	.001	136.31	2
.303D							
PA-234	5535.29	1668.50	8.	9.	.003	53.56	2
.296D							
PA-234	5617.29	1693.22	0.	14.	.004	26.73	
.377s							
BI-214	5735.94	1728.99	2.	23.	.006	25.41	1
.271s							
BI-214	5852.31	1764.08	7.	66.	.018	14.71	1
.006s							
BI-207	5872.68	1770.22	3.	0.	.000	.00	
.362D							
CO-56	5876.90	1771.49	3.	0.	.000	.00	
.374D							
MN-56	6008.60	1811.20	1.	0.	.000	.00	
.302s							
Y-88	6091.00	1836.04	1.	0.	.000	87.18	
.241s							
BI-214	6126.93	1846.88	8.	9.	.002	65.50	
.480s							
CO-56	6684.00	2014.85	0.	1.	.000	95.92	
.129s							
CO-56	6748.00	2034.15	0.	2.	.001	70.71	
.302s							
MN-56	7008.13	2112.60	2.	0.	.000	.00	
.559D							
BI-214	7023.00	2117.08	1.	1.	.000	141.42	
.257s							
BI-214	7314.32	2204.94	10.	2.	.001	232.70	
.309s							
BI-214	8118.00	2447.35	12.	2.	.001	281.07	
.000s							

s Peak fails shape tests.

D Peak area deconvoluted.

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\*\*\*\*\* S U M M A R Y O F L I B R A R Y P E A K U S A G E \*\*  
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NUCLIDE	AVERAGE ACTIVITY microCi	ENERGY keV	PEAK ACTIVITY CODE microCi	MDA VALUE microCi	COMMENTS
NA-22 9.99E+01	0.00000E+00	1274.54	0.000E+00 ! (	0.0000E+00	9.50E+02
NA-24 1.00E+02 9.98E+01	0.00000E+00	1368.55 2754.10	0.000E+00 % 0.000E+00 =	0.0000E+00 0.0000E+00	6.26E-01
K-40 1.07E+01	0.00000E+00	1460.75	0.000E+00 (	0.0000E+00	4.68E+11
SC-46 1.00E+02 1.00E+02	0.00000E+00	1120.52 889.26	0.000E+00 ! ( 0.000E+00 & (	0.0000E+00 0.0000E+00	8.38E+01
CR-51 9.83E+00	0.00000E+00	320.07	0.000E+00 &	0.0000E+00	2.77E+01
MN-54 1.00E+02	0.00000E+00	834.81	0.000E+00 ! (	0.0000E+00	3.12E+02
MN-56 9.90E+01 2.72E+01 1.43E+01	0.00000E+00	846.60 1811.20 2112.60	0.000E+00 ( 0.000E+00 % 0.000E+00 ? (	0.0000E+00 0.0000E+00 0.0000E+00	1.07E-01
FE-59 5.65E+01 4.32E+01	0.00000E+00	1099.22 1291.56 192.34	0.000E+00 % 0.000E+00 % 0.000E+00 %	0.0000E+00 0.0000E+00 0.0000E+00	4.51E+01

3.11E+00						
1.03E+00		142.65	0.000E+00	%	0.0000E+00	
CO-56	0.00000E+00	846.75	0.000E+00	%	0.0000E+00	7.73E+01
1.00E+02		1238.28	0.000E+00	?	0.0000E+00	
6.76E+01		2598.58	0.000E+00	=	0.0000E+00	
1.69E+01		1771.49	0.000E+00	?	0.0000E+00	
1.57E+01		1037.83	0.000E+00	?	0.0000E+00	
1.40E+01		2034.92	0.000E+00	%	0.0000E+00	
7.89E+00		3253.64	0.000E+00	=	0.0000E+00	
7.41E+00		1360.22	0.000E+00	?	0.0000E+00	
4.33E+00		2015.36	0.000E+00	%	0.0000E+00	
3.08E+00		3202.30	0.000E+00	=	0.0000E+00	
3.04E+00		1175.13	0.000E+00	?	0.0000E+00	
2.28E+00		3273.19	0.000E+00	=	0.0000E+00	
1.75E+00		977.46	0.000E+00	?	0.0000E+00	
1.44E+00		3010.20	0.000E+00	=	0.0000E+00	
1.00E+00						
CO-57	0.00000E+00	122.07	0.000E+00	!(	0.0000E+00	2.70E+02
8.56E+01		136.43	0.000E+00	!(	0.0000E+00	
1.06E+01		14.41	0.000E+00	?(	2.5799E-04	
9.50E+00						

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Nuclide	Ave activity	Energy	Activity	Code	Peak MDA	Comments
CO-58	0.00000E+00	810.75	0.000E+00	%	0.0000E+00	7.08E+01
9.94E+01						

CO-60	0.00000E+00	1332.51	0.000E+00	&(	0.0000E+00	1.93E+03
1.00E+02		1173.23	0.000E+00	?(	0.0000E+00	
9.99E+01						
ZN-65	0.00000E+00	1115.52	0.000E+00	!(	0.0000E+00	2.44E+02
5.08E+01		8.13	0.000E+00	?(	1.9134E-06	
3.87E+01						
RB-86	0.00000E+00	1076.63	0.000E+00	%	0.0000E+00	1.86E+01
8.76E+00						
Y-88	0.00000E+00	1836.01	0.000E+00	%	0.0000E+00	1.07E+02
9.94E+01		898.02	0.000E+00	%	0.0000E+00	
9.40E+01		14.15	1.656E-04		8.4177E-05	
5.25E+01		15.80	1.414E-03		5.6452E-04	
8.70E+00						
NB-94	0.00000E+00	871.10	0.000E+00	!(	0.0000E+00	7.41E+06
1.00E+02		702.50	0.000E+00	%	0.0000E+00	
1.00E+02						
NB-95	0.00000E+00	765.82	0.000E+00	(	0.0000E+00	3.52E+01
9.90E+01						
SN-113	0.00000E+00	391.71	0.000E+00	%	0.0000E+00	1.15E+02
6.42E+01		24.21	5.050E-04		1.6907E-04	
3.90E+01		24.00	1.959E-03		3.1440E-04	
2.07E+01		27.30	3.508E-03		7.5013E-04	
1.10E+01		27.90	1.626E-02		4.1225E-03	
2.10E+00		255.04	0.000E+00	%	0.0000E+00	
2.07E+00						
CS-134	0.00000E+00	604.66	0.000E+00	%	0.0000E+00	7.53E+02
9.76E+01		795.76	0.000E+00	%	0.0000E+00	
8.54E+01		569.29	0.000E+00	!	0.0000E+00	
1.54E+01		801.84	0.000E+00	!	0.0000E+00	
8.73E+00						

8.38E+00		563.26	0.000E+00 %	0.0000E+00	
3.04E+00		1365.13	0.000E+00 %	0.0000E+00	
1.80E+00		1167.86	0.000E+00 ?	0.0000E+00	
1.46E+00		475.35	0.000E+00 &	0.0000E+00	
1.00E+00		1038.50	0.000E+00 %	0.0000E+00	
CS-137	0.00000E+00	661.62	0.000E+00 %	0.0000E+00	1.10E+04
8.46E+01		32.19	6.810E-03	3.6768E-03	
3.70E+00		31.82	3.423E-02	6.7480E-03	
1.92E+00		36.40	6.601E-02	1.2997E-02	
1.04E+00					
EU-152	7.38047E-04	40.12	7.380E-04 (	3.6687E-04	4.64E+03
3.00E+01		121.78	0.000E+00 %	0.0000E+00	
2.92E+01		344.30	0.000E+00 %	0.0000E+00	
2.70E+01		1408.08	0.000E+00 -	0.0000E+00	
2.12E+01		39.52	1.596E-03 +	7.3273E-04	
1.60E+01					

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Nuclide	Ave activity	Energy	Activity	Code	Peak MDA	Comments
1.46E+01		964.00	0.000E+00 -		0.0000E+00	
1.36E+01		1112.07	0.000E+00 %		0.0000E+00	
1.30E+01		778.90	0.000E+00 %		0.0000E+00	
1.03E+01		1085.80	0.000E+00 %		0.0000E+00	
9.00E+00		45.40	1.791E-03 +		1.0865E-03	
7.62E+00		244.67	0.000E+00 %		0.0000E+00	



4.18E+00		867.39	0.000E+00 %	0.0000E+00	
2.84E+00		444.00	0.000E+00 %	0.0000E+00	
2.26E+00		411.09	0.000E+00 %	0.0000E+00	
2.10E+00		46.60	6.644E-03 +	4.5427E-03	
1.72E+00		1089.82	0.000E+00 %	0.0000E+00	
1.71E+00		1299.19	0.000E+00 %	0.0000E+00	
1.45E+00		1212.89	0.000E+00 %	0.0000E+00	
EU-154	0.00000E+00	123.10	0.000E+00 %	0.0000E+00	3.10E+03
4.05E+01		1274.80	0.000E+00 %	0.0000E+00	
3.55E+01		723.30	0.000E+00 %	0.0000E+00	
1.97E+01		1004.80	0.000E+00 %	0.0000E+00	
1.76E+01		43.00	0.000E+00 %	4.5412E-04	Energy dup
lication		873.20	0.000E+00 !	0.0000E+00	
1.13E+01		996.30	0.000E+00 %	0.0000E+00	
1.07E+01		42.31	0.000E+00 %	6.5748E-04	Energy dup
lication		248.04	0.000E+00 %	0.0000E+00	
6.60E+00		591.70	0.000E+00 %	0.0000E+00	
4.60E+00		48.70	0.000E+00 %	1.7911E-03	Energy dup
lication		756.70	0.000E+00 %	0.0000E+00	
4.10E+00		1597.30	0.000E+00 %	0.0000E+00	
1.70E+00		692.50	0.000E+00 %	0.0000E+00	Energy dup
lication		1397.80	0.000E+00 %	0.0000E+00	
1.60E+00		50.00	0.000E+00 %	8.7175E-03	
1.00E+00					
EU-155	0.00000E+00	86.45	0.000E+00 %	1.7177E-04	1.81E+03
3.27E+01		105.31	0.000E+00 %	0.0000E+00	
2.18E+01					

lication		43.00	0.000E+00 ?	4.6009E-04	Energy dup
lication		42.31	0.000E+00 ?	1.1056E-03	Energy dup
lication		48.70	0.000E+00 ?	2.0873E-03	Energy dup
1.36E+00		45.30	0.000E+00 ?	5.4037E-03	
1.21E+00		60.01	0.000E+00 !	6.3765E-03	
TL-208	0.00000E+00	583.14	0.000E+00 (	0.0000E+00	2.15E-03
8.60E+01		510.72	0.000E+00 (	0.0000E+00	
2.25E+01		860.47	0.000E+00 %	0.0000E+00	
1.20E+01		277.36	0.000E+00 %	0.0000E+00	
6.50E+00		74.97	1.110E-02 +	2.0843E-03	Energy dup
lication		72.80	1.065E-02 +	3.5695E-03	Energy dup
lication		763.30	0.000E+00 !(	0.0000E+00	
1.70E+00		84.80	1.157E-02 +	4.4931E-03	Energy dup
lication					

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 U.C.L.A. Radiation Safety Spectrum name: AN1E202.AN1

Nuclide	Ave activity	Energy	Activity	Code	Peak MDA	Comments
PB-210	0.00000E+00	10.80	0.000E+00 %		4.7401E-06	7.45E+03
2.43E+01		46.52	0.000E+00 !		2.3912E-03	
4.00E+00						
PB-212	0.00000E+00	238.63	0.000E+00 (		0.0000E+00	4.43E-01
4.31E+01		77.11	0.000E+00 %		2.2626E-04	Energy dup
lication		74.81	0.000E+00 %		6.1086E-04	Energy dup
lication		87.20	0.000E+00 %		6.8807E-04	Energy dup
lication		300.09	0.000E+00 %		0.0000E+00	

3.27E+00		89.80	0.000E+00 ?(	1.2324E-03 Energy dup
lication				
PB-214	0.00000E+00	351.99	0.000E+00 ?(	0.0000E+00 1.86E-02
3.71E+01		295.22	0.000E+00 %	0.0000E+00
1.92E+01		10.80	0.000E+00 !(	1.3785E-04 Energy dup
lication		77.11	0.000E+00 %	5.7718E-04 Energy dup
lication		241.92	0.000E+00 %	0.0000E+00
7.47E+00		74.81	0.000E+00 %	1.1583E-03 Energy dup
lication		87.20	1.819E-03 +	1.4489E-03 Energy dup
lication		53.20	0.000E+00 %	6.7038E-03
1.10E+00		785.95	0.000E+00 !(	0.0000E+00
1.09E+00		89.80	0.000E+00 %	4.9786E-03 Energy dup
lication				
BI-207	0.00000E+00	569.67	0.000E+00 !(	0.0000E+00 1.39E+04
9.80E+01		1063.62	0.000E+00 &	0.0000E+00
7.70E+01		74.97	0.000E+00 %	1.5093E-04 Energy dup
lication		10.60	0.000E+00 ?(	3.1394E-06
3.50E+01		72.80	0.000E+00 ?(	1.5808E-04 Energy dup
lication		84.80	0.000E+00 %	2.0935E-04 Energy dup
lication		1770.22	0.000E+00 ?(	0.0000E+00
7.00E+00		87.30	0.000E+00 ?(	1.1271E-03
4.00E+00				
BI-212	0.00000E+00	727.17	0.000E+00 %	0.0000E+00 4.20E-02
1.18E+01		1620.56	0.000E+00 ?	0.0000E+00
2.75E+00		785.42	0.000E+00 !	0.0000E+00
2.00E+00		39.86	0.000E+00 !	1.0513E-02
1.10E+00				
BI-214	0.00000E+00	609.32	0.000E+00 (	0.0000E+00 1.38E-02

4.61E+01					
1.59E+01	1764.51	0.000E+00	(	0.0000E+00	
1.50E+01	1120.28	0.000E+00	(	0.0000E+00	
5.92E+00	1238.11	0.000E+00	(	0.0000E+00	
4.99E+00	2204.12	0.000E+00	&	0.0000E+00	
4.89E+00	768.36	0.000E+00	(	0.0000E+00	
4.02E+00	1377.65	0.000E+00	(	0.0000E+00	
3.16E+00	934.05	0.000E+00	(	0.0000E+00	
3.05E+00	1729.60	0.000E+00	?(	0.0000E+00	
2.48E+00	1407.98	0.000E+00	(	0.0000E+00	
2.19E+00	1509.19	0.000E+00	&(	0.0000E+00	
2.12E+00	1847.44	0.000E+00	%	0.0000E+00	

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Spectrum name: AN1E202.AN1

Nuclide	Ave activity	Energy	Activity	Code	Peak MDA	Comments
		1155.19	0.000E+00	%	0.0000E+00	
1.69E+00		665.45	0.000E+00	!(	0.0000E+00	
1.56E+00		2447.71	0.000E+00	%	0.0000E+00	
1.55E+00		1280.96	0.000E+00	%	0.0000E+00	
1.47E+00		1401.50	0.000E+00	(	0.0000E+00	
1.39E+00		806.17	0.000E+00	!(	0.0000E+00	
1.23E+00		2118.54	0.000E+00	&	0.0000E+00	
1.21E+00		1661.28	0.000E+00	%	0.0000E+00	
1.15E+00						
RA-224	0.00000E+00	241.00	0.000E+00	%	0.0000E+00	3.66E+00
3.90E+00						

RA-226	0.00000E+00	185.99	0.000E+00	!(	0.0000E+00	5.84E+05
3.28E+00						
AC-228	0.00000E+00	911.07	0.000E+00	(	0.0000E+00	2.55E-01
2.90E+01						
1.75E+01		968.90	0.000E+00	(	0.0000E+00	
1.20E+01		338.40	0.000E+00	(	0.0000E+00	
5.45E+00		964.60	0.000E+00	%	0.0000E+00	
4.84E+00		794.80	0.000E+00	!(	0.0000E+00	
4.64E+00		463.00	0.000E+00	%	0.0000E+00	
4.55E+00		209.40	0.000E+00	%	0.0000E+00	
3.77E+00		270.30	0.000E+00	%	0.0000E+00	
3.71E+00		1587.90	0.000E+00	&(	0.0000E+00	
3.36E+00		328.00	0.000E+00	%	0.0000E+00	
2.93E+00		129.10	0.000E+00	!(	0.0000E+00	
2.23E+00		409.40	0.000E+00	%	0.0000E+00	
1.95E+00		1630.40	0.000E+00	? (	0.0000E+00	
1.82E+00		835.60	0.000E+00	%	0.0000E+00	
1.62E+00		772.10	0.000E+00	%	0.0000E+00	
1.36E+00		99.45	0.000E+00	!(	3.1646E-03	
1.10E+00		755.20	0.000E+00	!(	0.0000E+00	
1.05E+00		1495.80	0.000E+00	&(	0.0000E+00	
1.04E+00		1459.20	0.000E+00	(	0.0000E+00	
TH-234	6.52686E-03	13.30	6.527E-03	(	1.9815E-04	2.41E+01
9.80E+00						
3.90E+00		63.29	0.000E+00	%	1.4047E-03	
3.00E+00		92.80	1.672E-03	-	1.6318E-03	
2.57E+00		92.38	0.000E+00	%	1.9137E-03	

PA-234	5.47395E-05	13.60	5.474E-05	(	3.4359E-05	2.79E-01
1.14E+02		98.44	0.000E+00	%	1.7480E-04	
2.51E+01		946.00	0.000E+00	%	0.0000E+00	
2.00E+01		131.28	0.000E+00	%	0.0000E+00	
2.00E+01		94.67	0.000E+00	%	3.0778E-04	
1.55E+01		883.24	0.000E+00	%	0.0000E+00	
1.20E+01		926.70	0.000E+00	%	0.0000E+00	
1.10E+01		569.26	0.000E+00	%	0.0000E+00	Energy dup
lication		111.00	0.000E+00	%	0.0000E+00	
8.55E+00		733.00	0.000E+00	%	0.0000E+00	
8.50E+00						

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U.C.L.A. Radiation Safety Spectrum name: AN1E202.AN1

Nuclide	Ave activity	Energy	Activity	Code	Peak MDA	Comments
		949.00	0.000E+00	-	0.0000E+00	
7.80E+00		152.70	0.000E+00	&	0.0000E+00	
7.20E+00		880.51	0.000E+00	%	0.0000E+00	Energy dup
lication		226.87	0.000E+00	&	0.0000E+00	Energy dup
lication		831.10	0.000E+00	%	0.0000E+00	
5.60E+00		808.10	0.000E+00	-	0.0000E+00	
4.90E+00		99.70	0.000E+00	-	9.1173E-04	
4.70E+00		699.10	0.000E+00	%	0.0000E+00	
4.60E+00		880.51	0.000E+00	-	0.0000E+00	Energy dup
lication		898.60	0.000E+00	-	0.0000E+00	
4.00E+00		1394.10	0.000E+00	%	0.0000E+00	
3.90E+00						

lication	226.87	0.000E+00	&	0.0000E+00	Energy dup
3.60E+00	824.70	0.000E+00	%	0.0000E+00	
3.40E+00	796.20	0.000E+00	-	0.0000E+00	
3.30E+00	805.50	0.000E+00	%	0.0000E+00	
3.20E+00	826.30	0.000E+00	-	0.0000E+00	
3.20E+00	706.00	0.000E+00	%	0.0000E+00	
3.20E+00	293.70	0.000E+00	%	0.0000E+00	
3.15E+00	114.50	0.000E+00	%	0.0000E+00	
lication	569.26	0.000E+00	-	0.0000E+00	Energy dup
3.10E+00	248.90	0.000E+00	%	0.0000E+00	
3.10E+00	63.20	0.000E+00	-	2.1941E-03	
3.00E+00	742.81	0.000E+00	%	0.0000E+00	
3.00E+00	568.70	0.000E+00	%	0.0000E+00	
2.80E+00	924.60	0.000E+00	%	0.0000E+00	
2.80E+00	369.60	0.000E+00	%	0.0000E+00	
2.70E+00	876.70	0.000E+00	%	0.0000E+00	
2.40E+00	983.40	0.000E+00	%	0.0000E+00	
2.20E+00	980.50	0.000E+00	%	0.0000E+00	
2.20E+00	819.40	0.000E+00	%	0.0000E+00	
2.00E+00	574.00	0.000E+00	%	0.0000E+00	
lication	186.00	0.000E+00	&	0.0000E+00	Energy dup
1.70E+00	1353.00	0.000E+00	%	0.0000E+00	
1.70E+00	566.50	0.000E+00	%	0.0000E+00	
1.60E+00	786.27	0.000E+00	%	0.0000E+00	
1.60E+00	780.80	0.000E+00	-	0.0000E+00	
1.50E+00	793.60	0.000E+00	%	0.0000E+00	

1.50E+00	669.80	0.000E+00 %	0.0000E+00
1.50E+00	666.70	0.000E+00 %	0.0000E+00
1.50E+00	664.60	0.000E+00 %	0.0000E+00
1.50E+00	458.60	0.000E+00 %	0.0000E+00
1.50E+00	372.20	0.000E+00 %	0.0000E+00
1.40E+00	978.80	0.000E+00 %	0.0000E+00
1.40E+00	754.80	0.000E+00 %	0.0000E+00
1.40E+00	506.80	0.000E+00 %	0.0000E+00
1.30E+00	1694.00	0.000E+00 -	0.0000E+00
1.30E+00	692.50	0.000E+00 %	0.0000E+00 Energy dup
1.30E+00	653.20	0.000E+00 %	0.0000E+00

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 U.C.L.A. Radiation Safety Spectrum name: AN1E202.AN1

Nuclide	Ave activity	Energy	Activity	Code	Peak MDA	Comments
1.30E+00		602.80	0.000E+00 %		0.0000E+00	
1.20E+00		1452.60	0.000E+00 %		0.0000E+00	
1.20E+00		521.00	0.000E+00 %		0.0000E+00	
1.20E+00		520.60	0.000E+00 -		0.0000E+00	
1.20E+00		513.60	0.000E+00 %		0.0000E+00	
1.20E+00		272.10	0.000E+00 %		0.0000E+00	
1.10E+00		202.90	0.000E+00 %		0.0000E+00	
1.10E+00		125.30	0.000E+00 -		0.0000E+00	
1.08E+00		200.80	0.000E+00 -		0.0000E+00	
1.06E+00		1668.50	0.000E+00 %		0.0000E+00	



1.00E+00		623.50	0.000E+00 %	0.0000E+00	
U-235	0.00000E+00	185.72	0.000E+00 %	0.0000E+00	1.39E+09
5.40E+01		143.76	0.000E+00 &	0.0000E+00	
1.05E+01		205.31	0.000E+00 %	0.0000E+00	
4.70E+00		163.35	0.000E+00 &	0.0000E+00	
4.70E+00		93.35	0.000E+00 ?	1.5583E-03	
2.50E+00		109.14	0.000E+00 ?	0.0000E+00	
1.50E+00		89.96	0.000E+00 ?	1.9844E-03	
1.50E+00		202.12	0.000E+00 %	0.0000E+00	
1.00E+00		105.00	0.000E+00 &	0.0000E+00	
1.00E+00					

( - This peak used in the nuclide activity average.

\* - Peak is too wide, but only one peak in library.

! - Peak is part of a multiplet and this area went negative during deconvolution.

? - Peak is too narrow.

@ - Peak is too wide at FW25M, but ok at FWHM.

% - Peak fails sensitivity test.

\$ - Peak identified, but first peak of this nuclide failed one or more qualification tests.

+ - Peak activity higher than counting uncertainty range.

- - Peak activity lower than counting uncertainty range.

= - Peak outside analysis energy range.

& - Calculated peak centroid is not close enough to the library energy centroid for positive identification.

P - Peakbackground subtraction

\*\*\*\*\* SUMMARY OF NUCLIDES IN SAMPLE \*\*

NUCLIDE	TIME OF COUNT ACTIVITY	TIME CORRECTED ACTIVITY	UNCERTAINTY COUNTING	1 SIGMA TOTAL
	uCi/grams	uCi/grams		

NA-22	<	0.00E+00	0.00E+00	
NA-24	<	0.00E+00	>12 HALFLIVES	
K-40	<	0.00E+00	0.00E+00	

SC-46 < 0.00E+00 0.00E+00  
 CR-51 < 0.00E+00 0.00E+00  
 MN-54 < 0.00E+00 0.00E+00

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U.C.L.A. Radiation Safety Spectrum name: AN1E202.AN1

MN-56	<	0.00E+00	>12 HALFLIVES		
FE-59	<	0.00E+00	0.00E+00		
CO-56	<	0.00E+00	0.00E+00		
CO-57	<	0.00E+00	0.00E+00		
CO-58	<	0.00E+00	0.00E+00		
CO-60	<	0.00E+00	0.00E+00		
ZN-65	<	0.00E+00	0.00E+00		
RB-86	<	0.00E+00	0.00E+00		
Y-88	<	0.00E+00	0.00E+00		
NB-94	<	0.00E+00	0.00E+00		
NB-95	<	0.00E+00	0.00E+00		
SN-113	<	0.00E+00	0.00E+00		
CS-134	<	0.00E+00	0.00E+00		
CS-137	<	0.00E+00	0.00E+00		
EU-152		1.9947E-08	2.0007E-08	25.17%	25.30%
EU-154	<	0.00E+00	0.00E+00		
EU-155	<	4.64E-09	4.68E-09		
TL-208	<	0.00E+00	>12 HALFLIVES		
PB-210	<	1.28E-10	1.28E-10		
PB-212	<	0.00E+00	>12 HALFLIVES		
PB-214	<	0.00E+00	>12 HALFLIVES		
BI-207	<	0.00E+00	0.00E+00		
BI-212	<	0.00E+00	>12 HALFLIVES		
BI-214	<	0.00E+00	>12 HALFLIVES		
RA-224	<	0.00E+00	0.00E+00		
RA-226	<	0.00E+00	0.00E+00		
AC-228	<	0.00E+00	>12 HALFLIVES		
TH-234		1.7640E-07	3.1303E-07	2.38%	3.48%
PA-234		1.4794E-09	>12 HALFLIVES	31.60%	31.70%
U-235	<	0.00E+00	0.00E+00		

----- S U M M A R Y -----

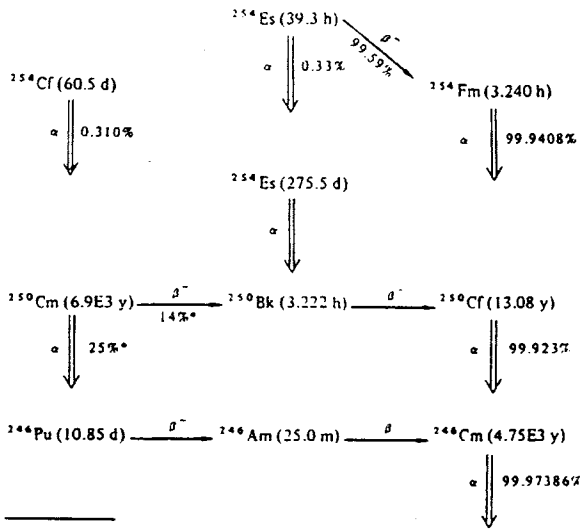
TOTAL ACTIVITY ( 3.8 to 2455.2 keV) 3.3303310E-07 uCi/grams

The library has energies which are not separable.

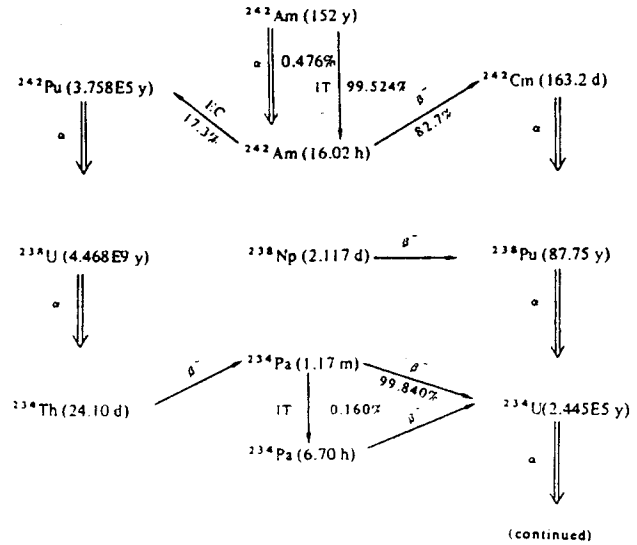
Analyzed by: \_\_\_\_\_

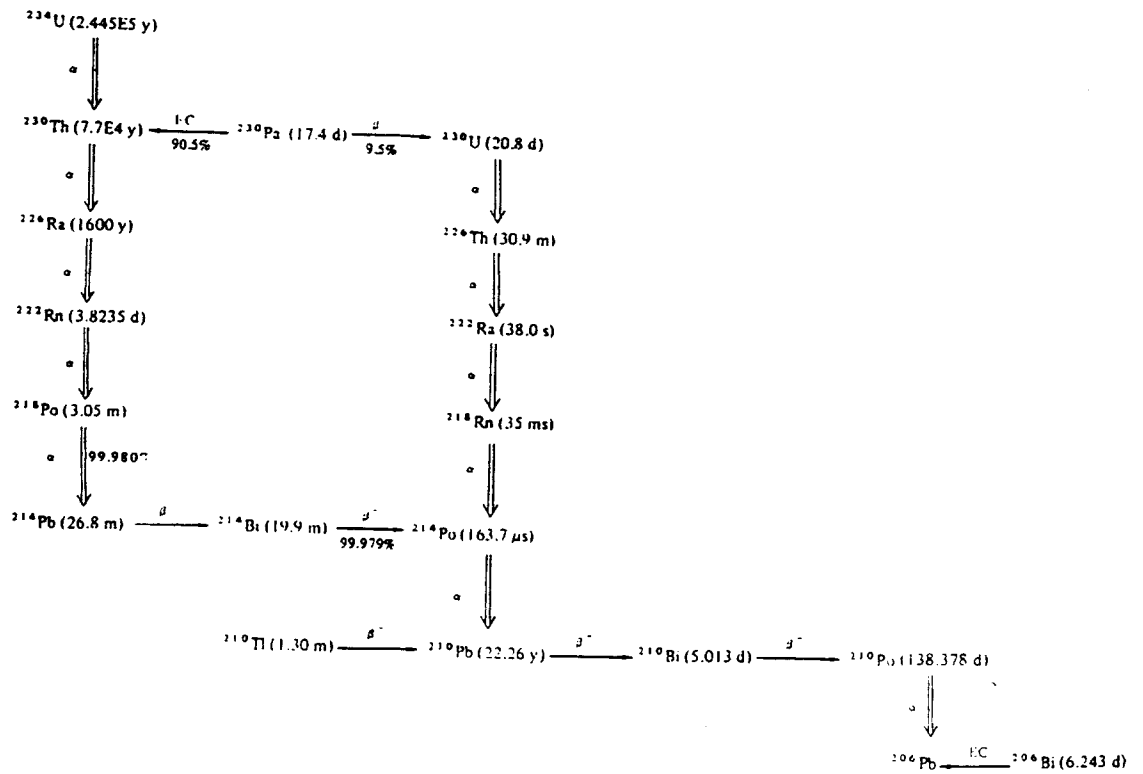
MFH

Uranium Series



\*Branching ratio based on systematics; decay has not been observed.





DIAGRAMS OF RADIOACTIVE DECAY CHAINS