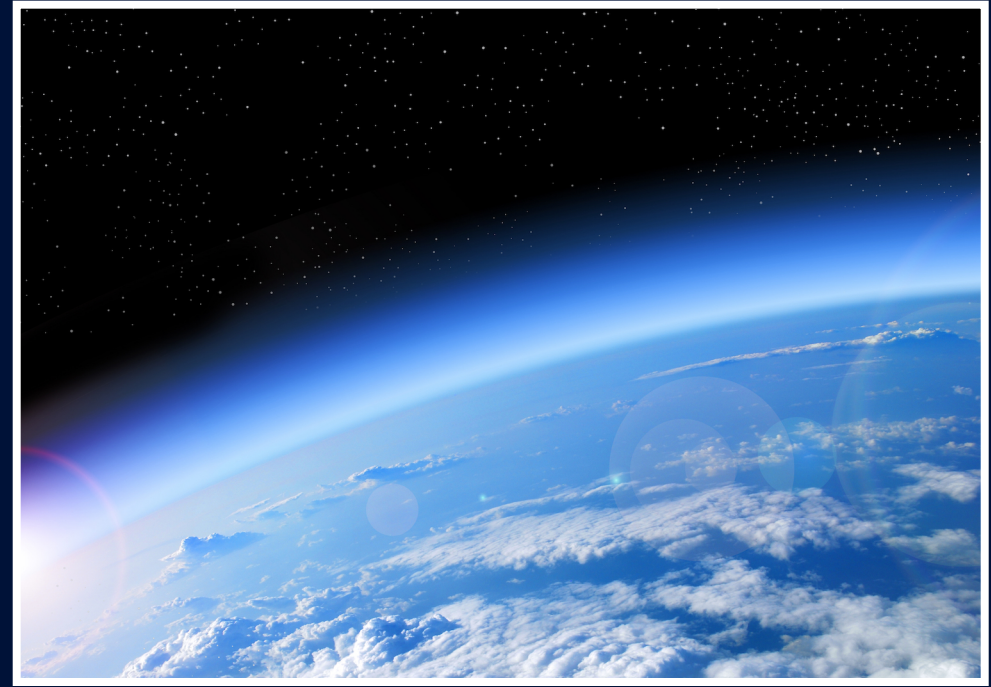


# Composition of the atmosphere

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~ The **composition** of the atmosphere is not constant, but there are **two gases** that make up **99%** of clean, dry air: **nitrogen** ( $\text{N}_2$ ; 78%) and **oxygen** ( $\text{O}_2$ ; 21%).

Chemical Composition of Air		
Name	Symbol	% by volume
Nitrogen	N <sub>2</sub>	78.084 %
Oxygen	O <sub>2</sub>	20.9476 %
Argon	Ar	0.934 %

~ Most of the remaining *1* % is **argon** (Ar; .93%), leaving < .07% for various other gases, water vapor, and tiny solid and liquid particles.

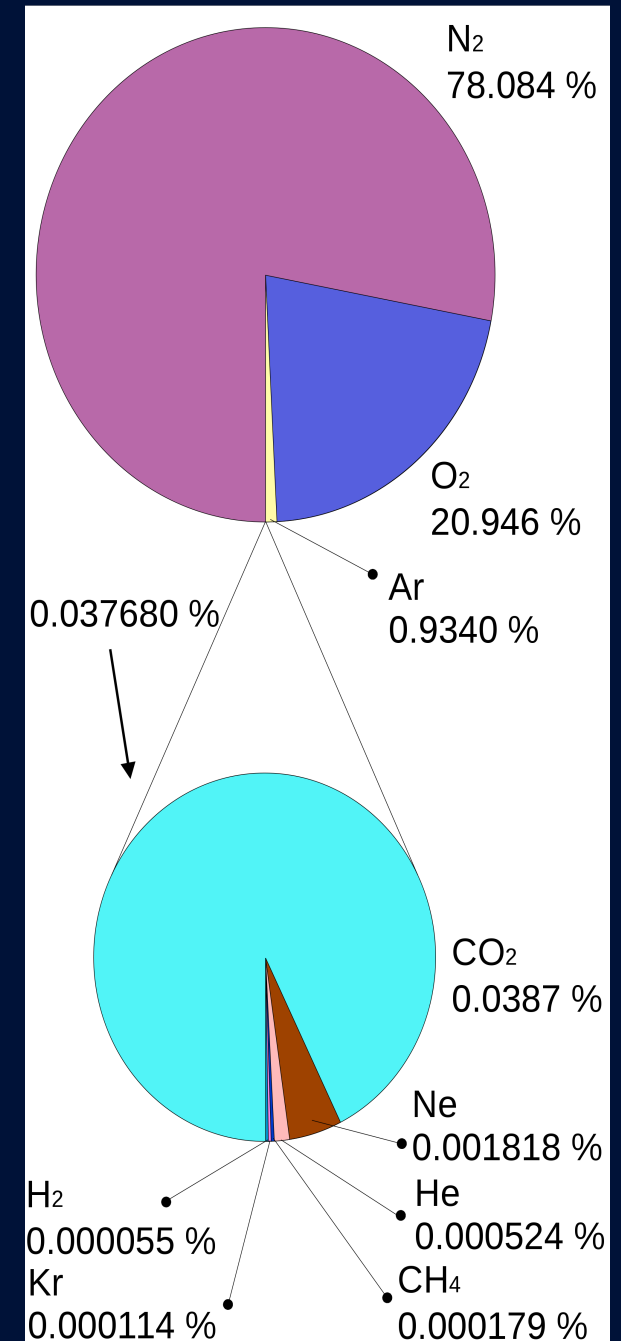


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Neon	Ne	0.001818 %
Methane	CH <sub>4</sub>	0.0002 %
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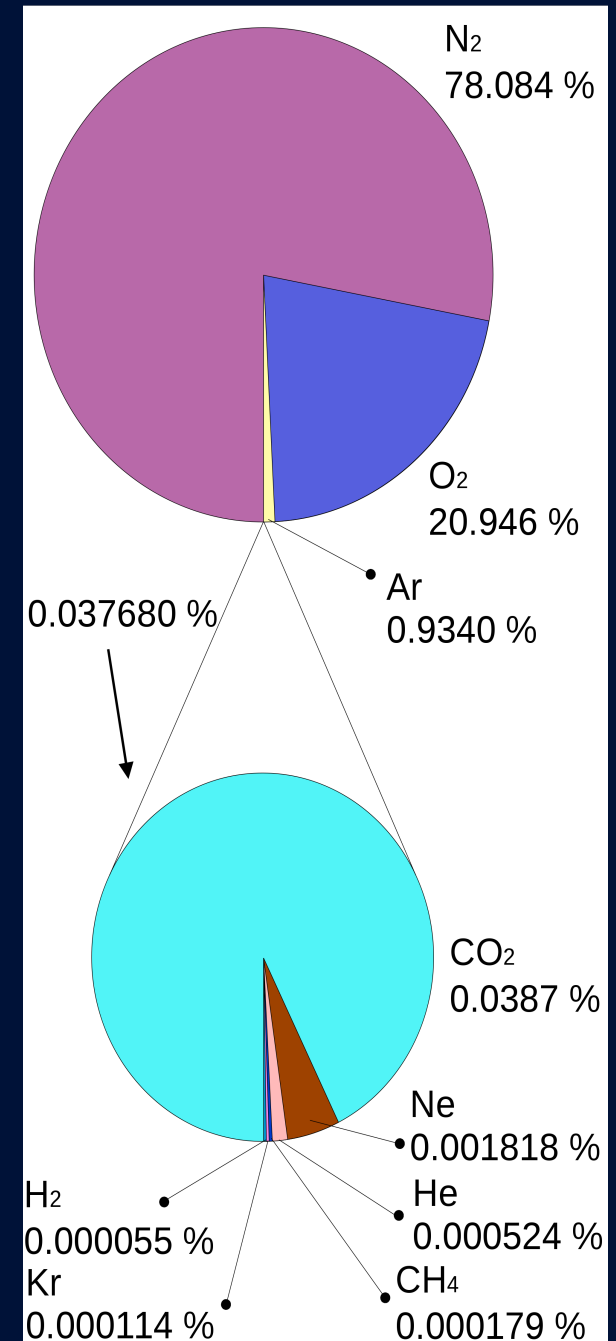




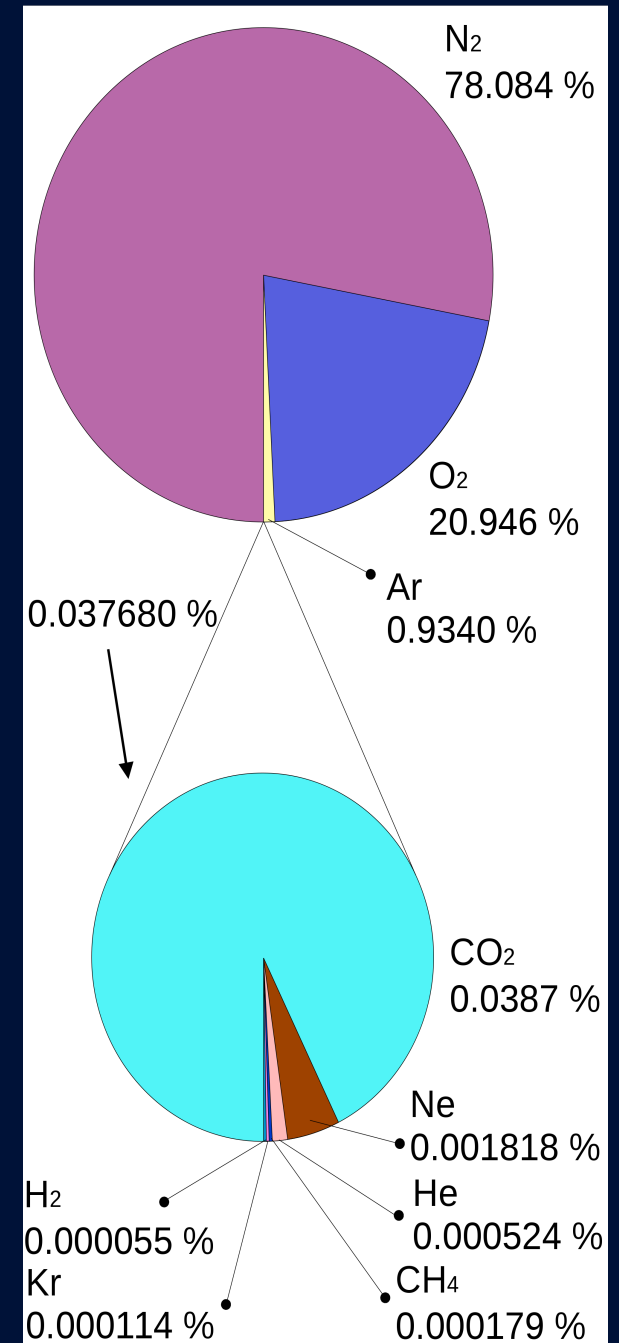
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~ Most of the remaining **1%** is **argon** (Ar; .93%), leaving < .07% for various other gases, water vapor, and tiny solid and liquid particles.

~ It is this < .07% that is most **important meteorologically** as N<sub>2</sub>, O<sub>2</sub>, and Ar are of great importance to life, but have little effect on weather.



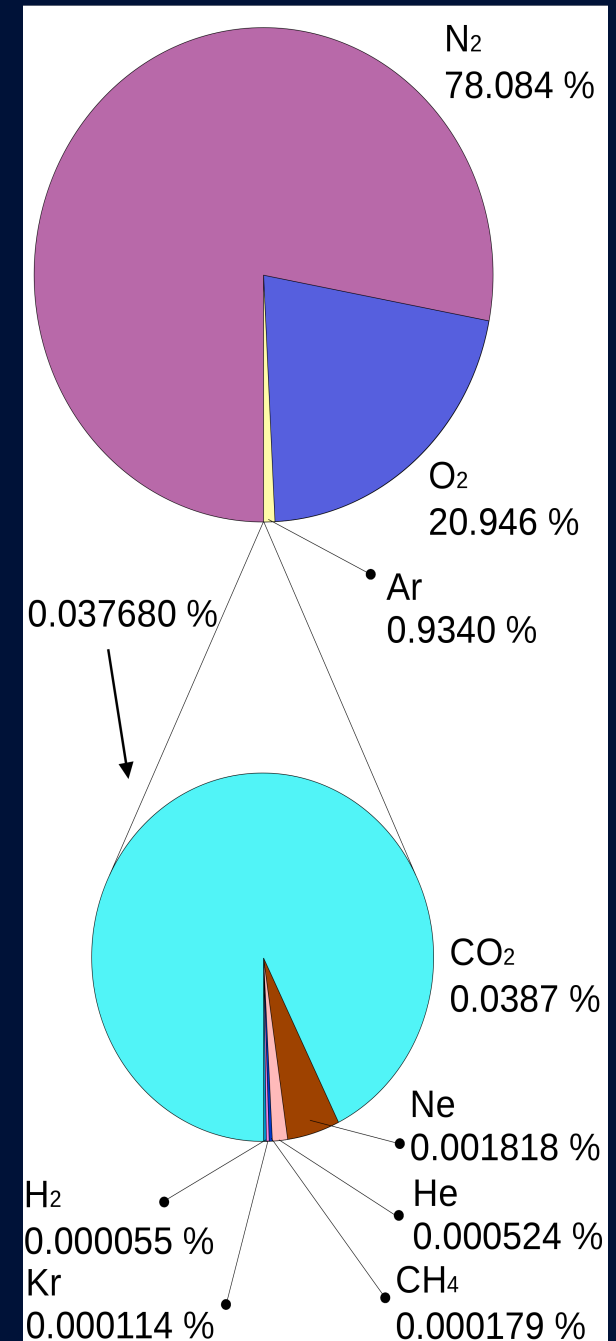
~ The most plentiful gas remaining is **carbon dioxide** (CO<sub>2</sub>) at **.0387%** or 387 parts per million (ppm).



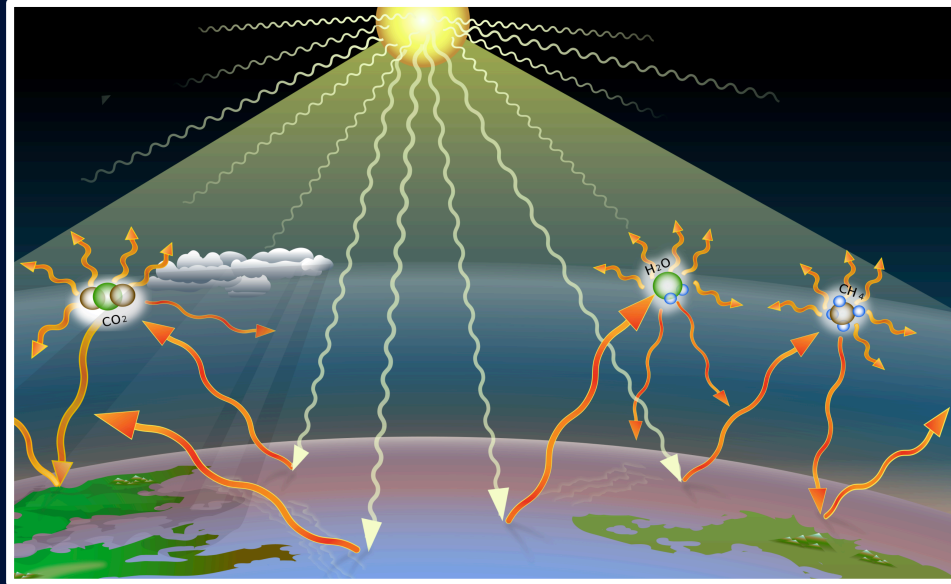


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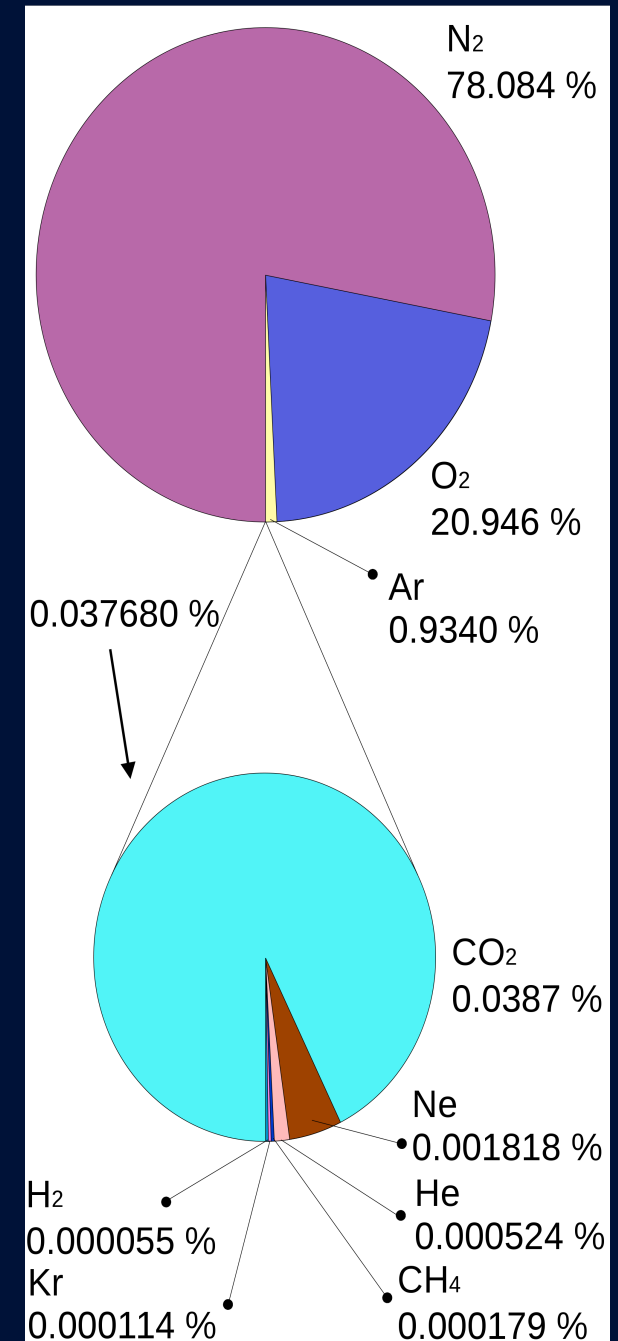
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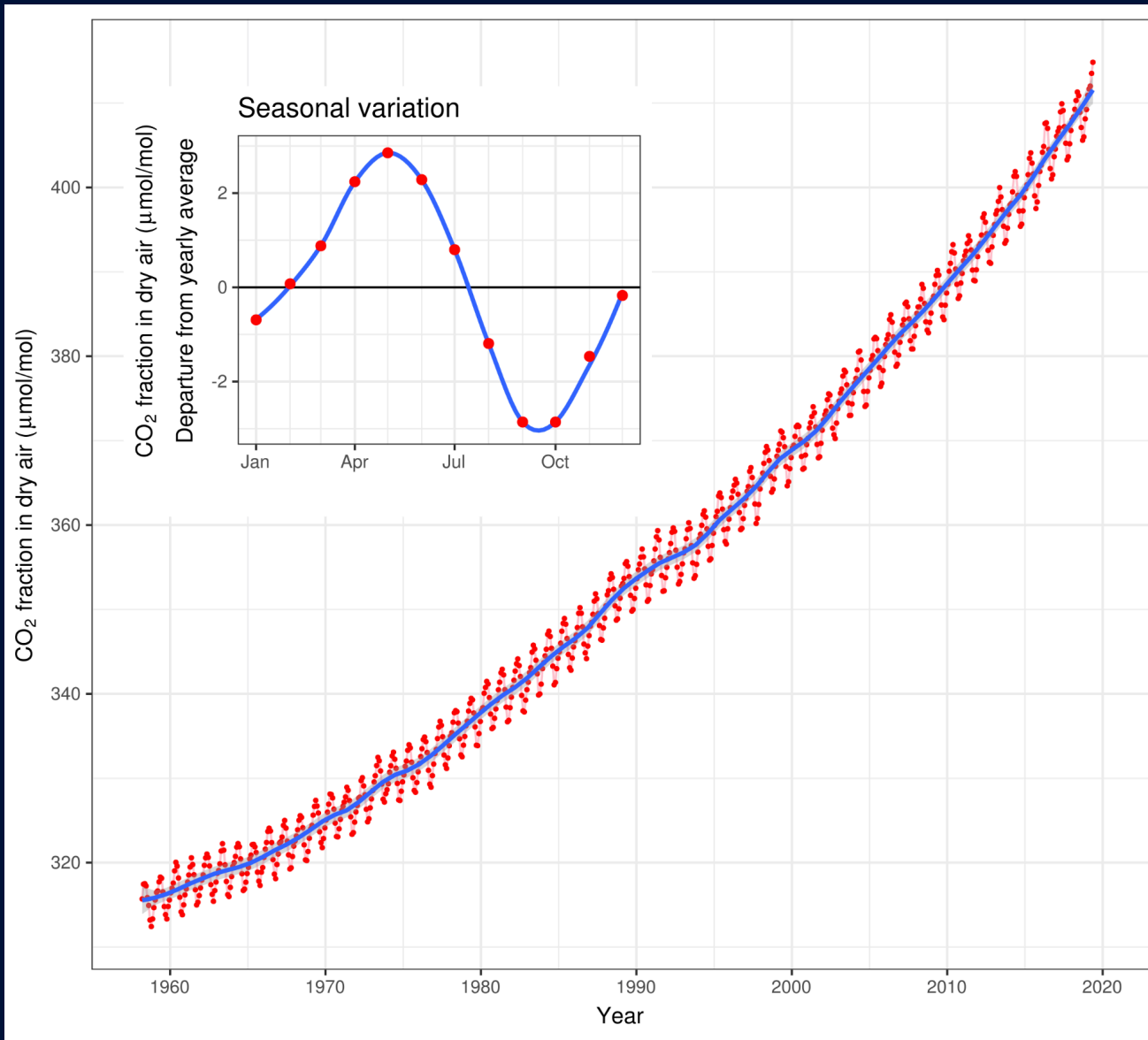
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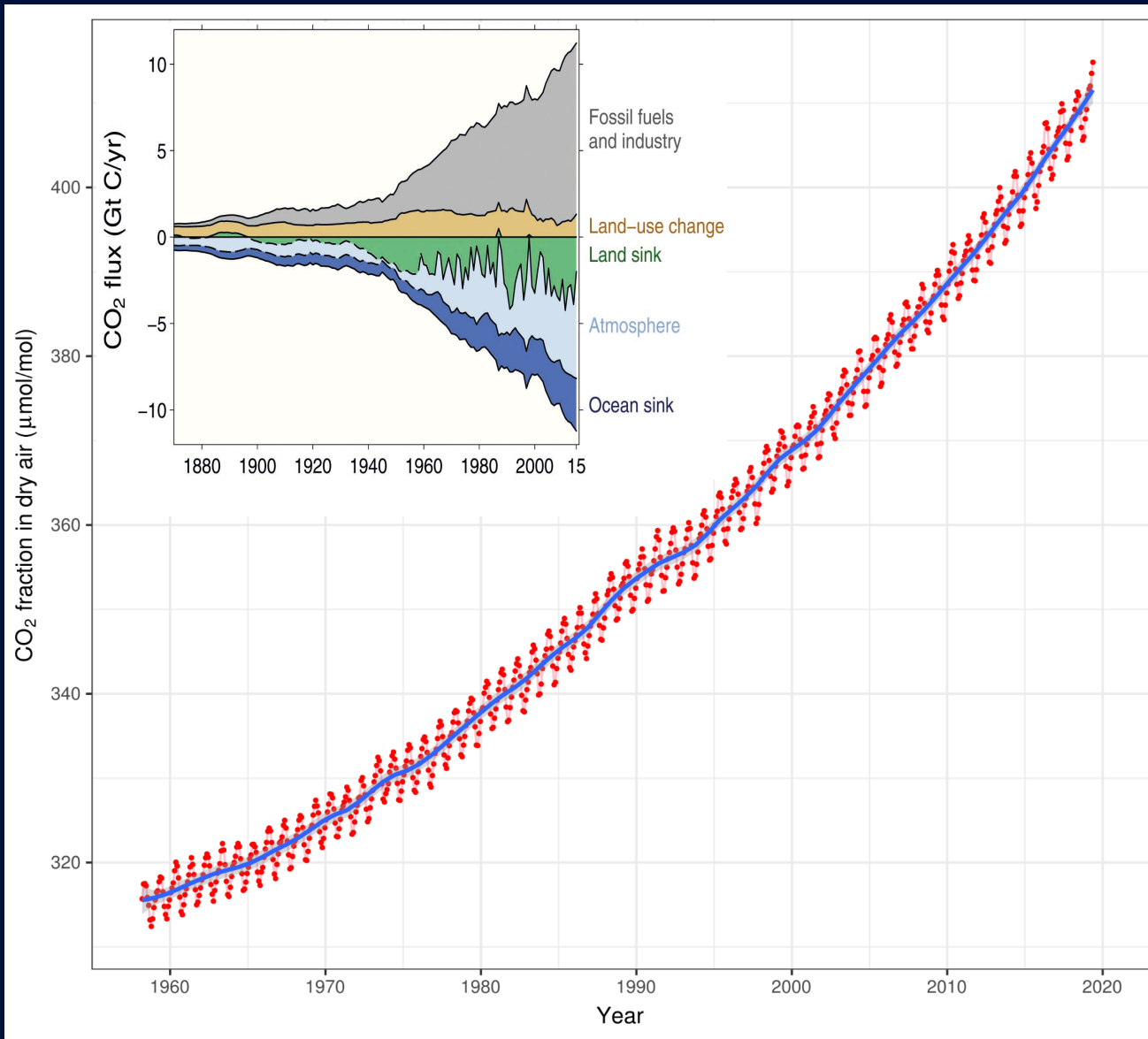
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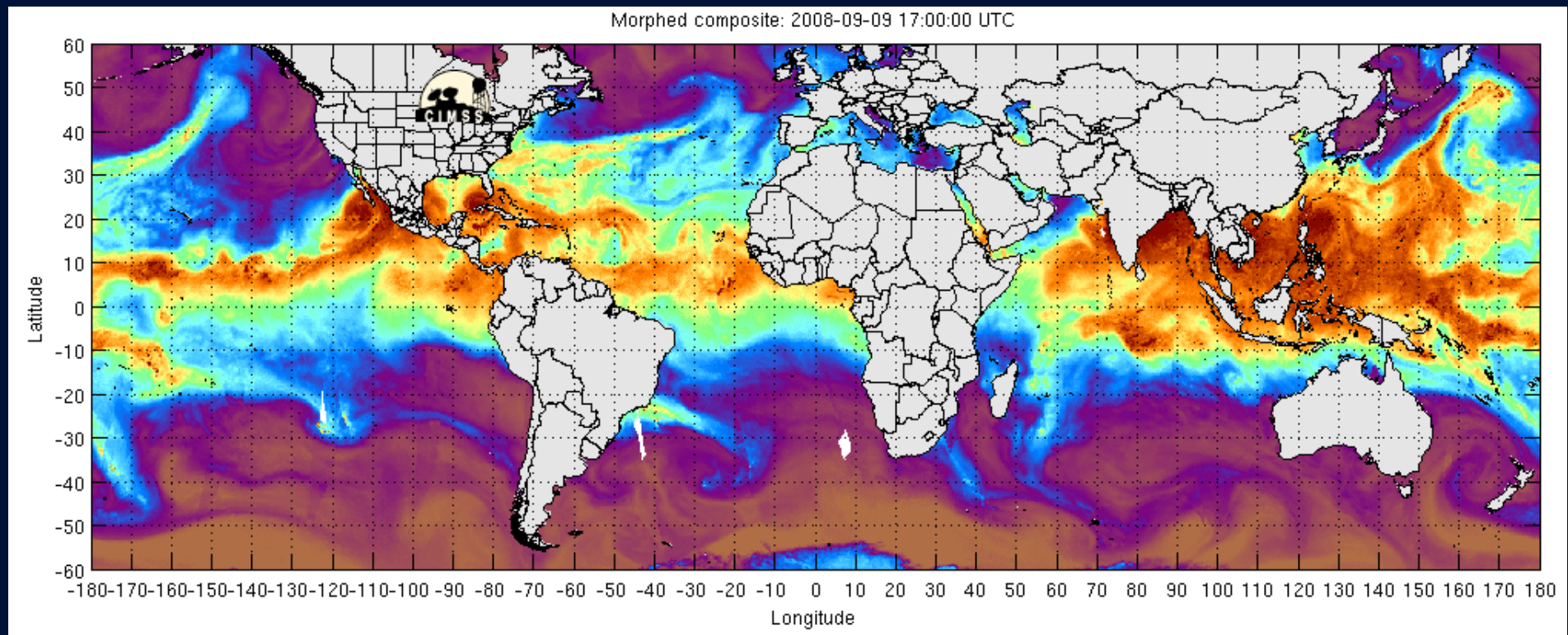
~ CO<sub>2</sub> has been *increasing* steadily over the last century due to the *burning* of fossil fuels.

~ Some CO<sub>2</sub> is taken up by *plants* and *oceans*, but a significant portion *remains* in the *atmosphere*, contributing to the *warming* of our *planet*.

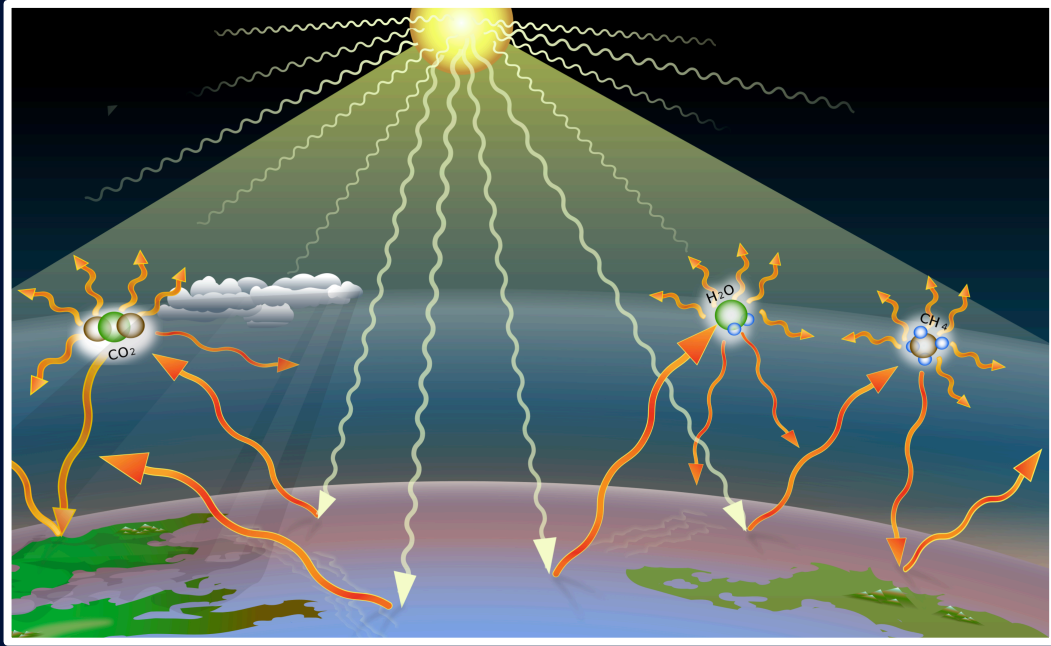


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- ~ The amount of **water vapor** in the atmosphere can account for ~0% (desert) to 4% (rainforest) of the air by volume.

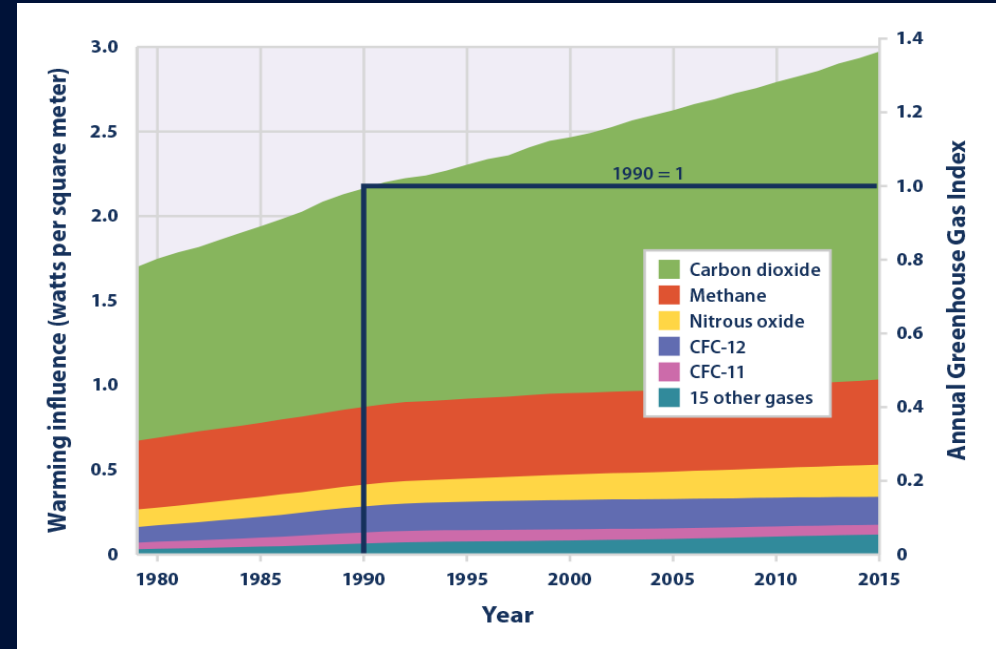
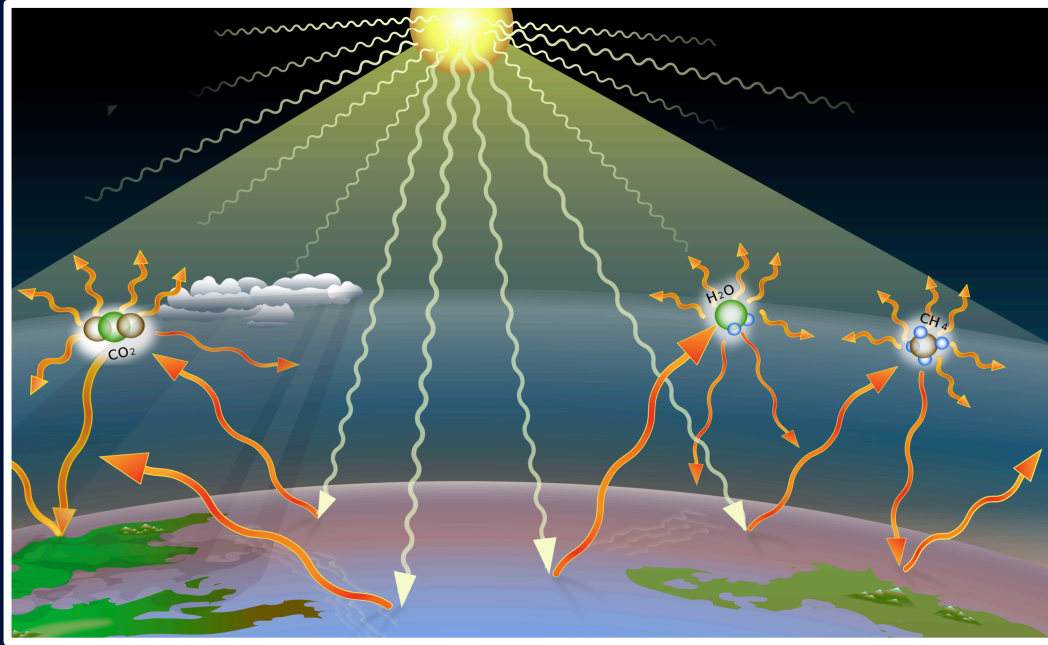


~ Like  $\text{CO}_2$  and *methane* ( $\text{CH}_4$ ), *water vapor* is an **absorber** of **Earth's heat energy**.





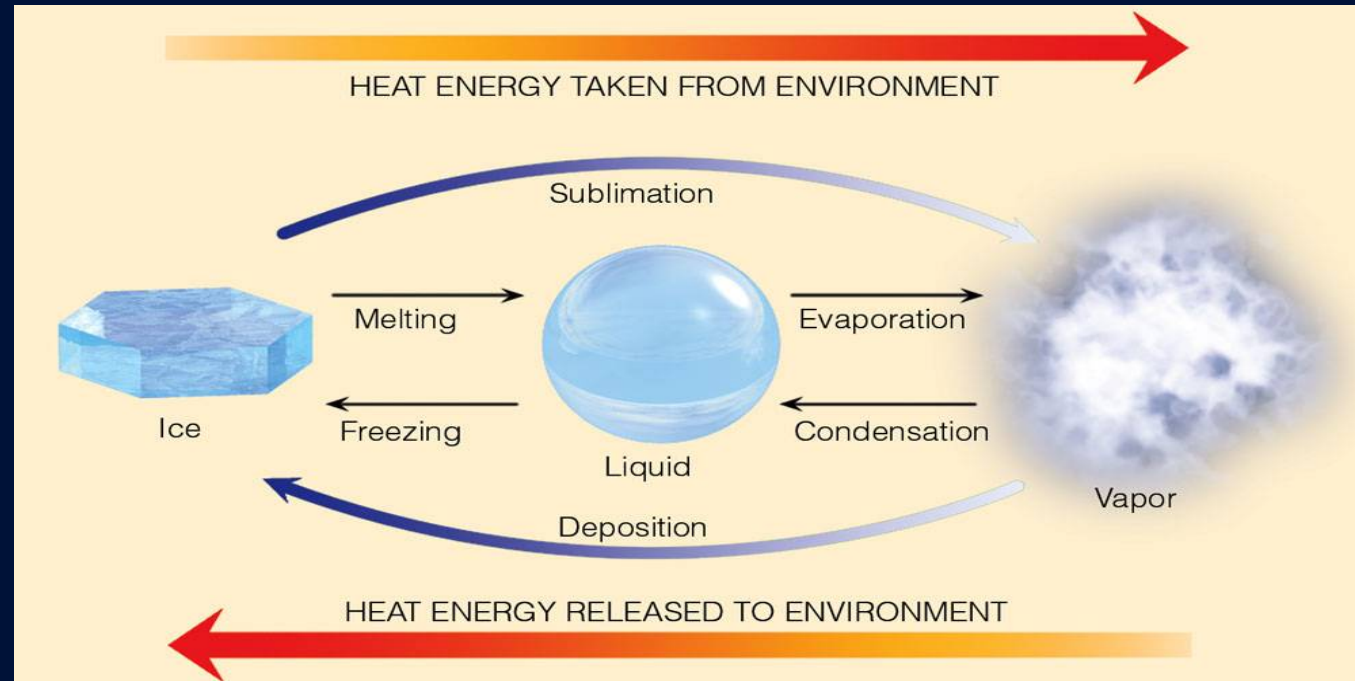
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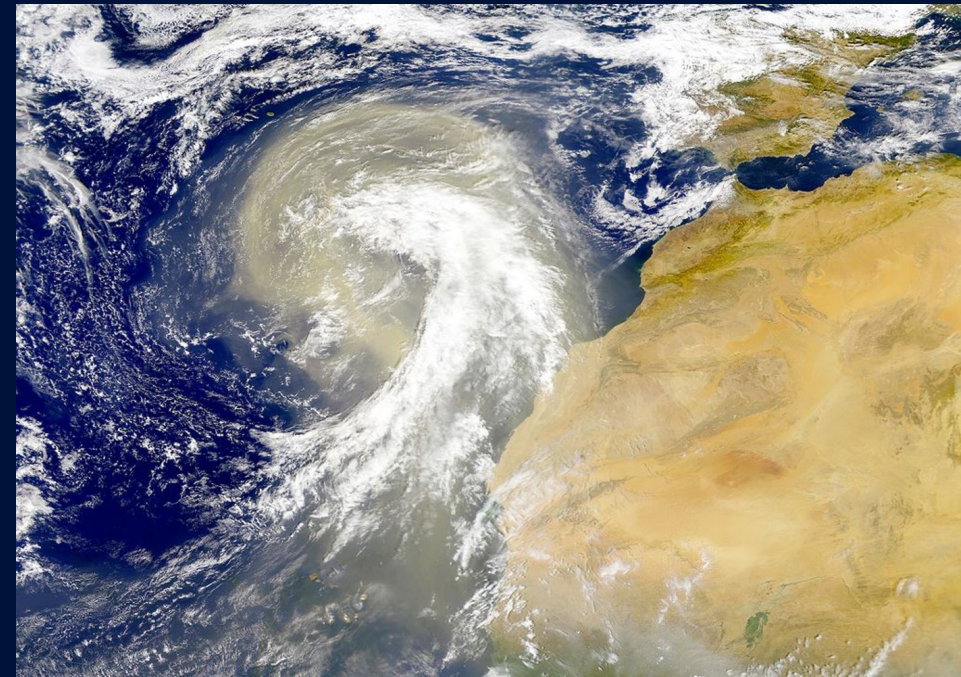
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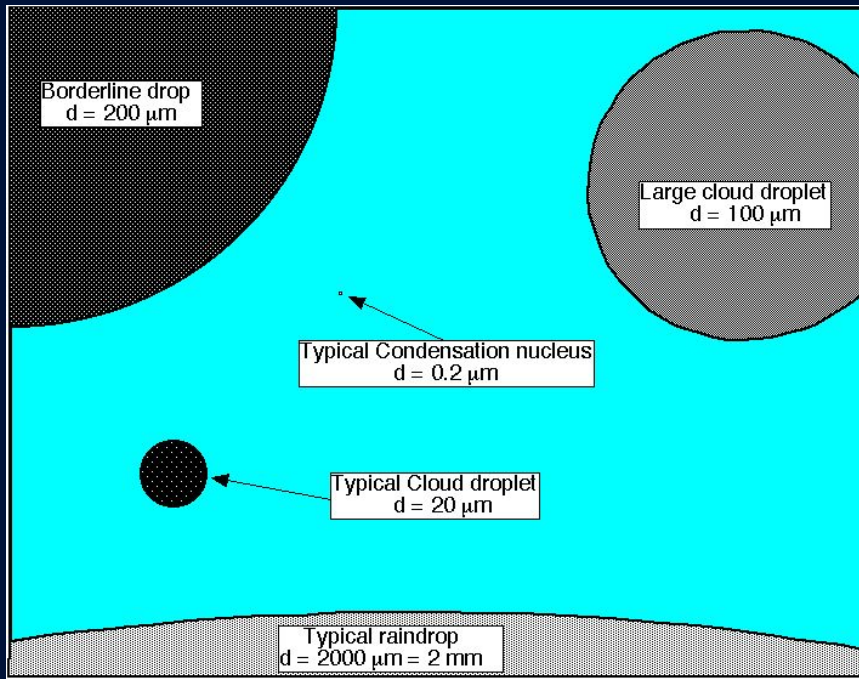


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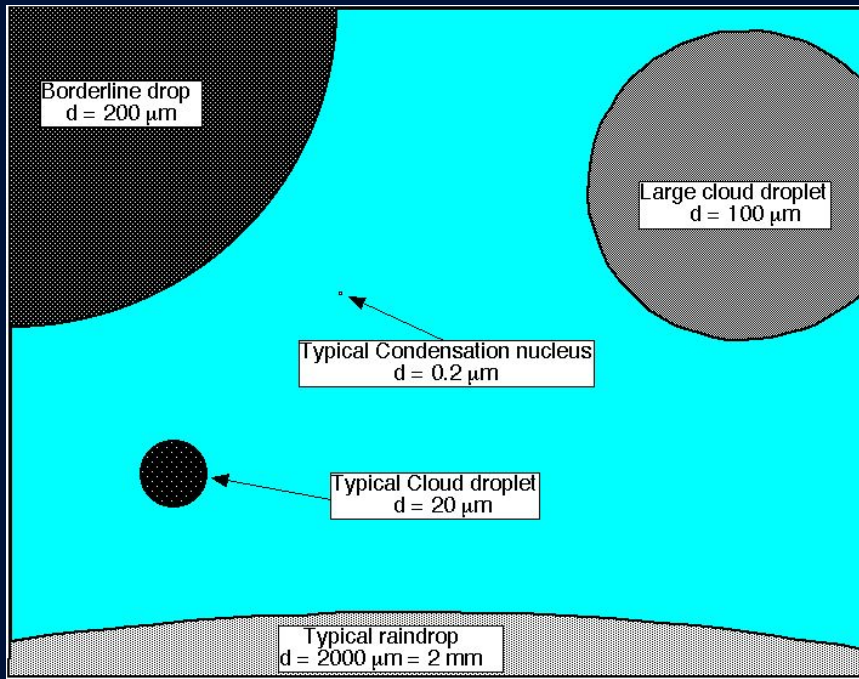


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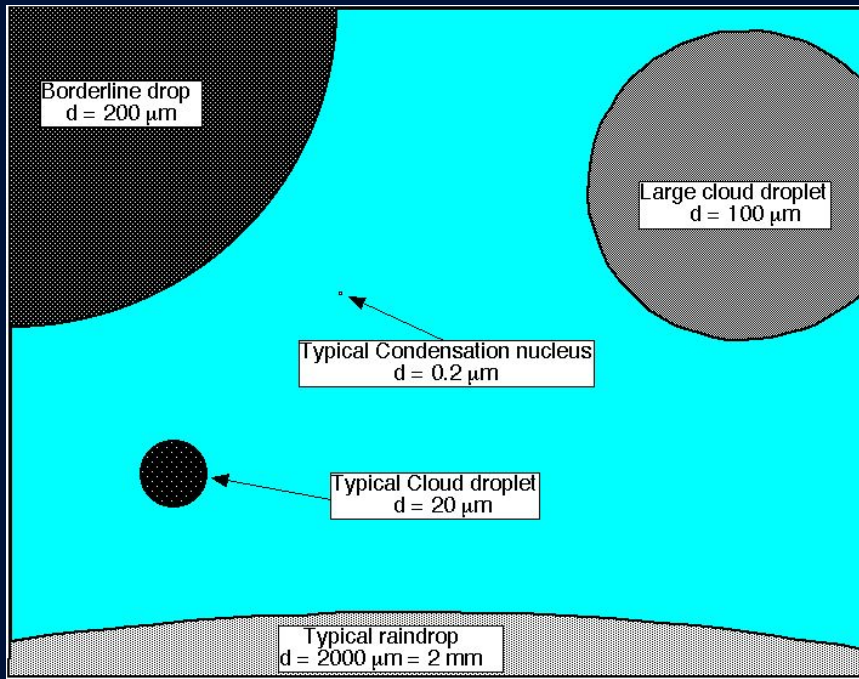




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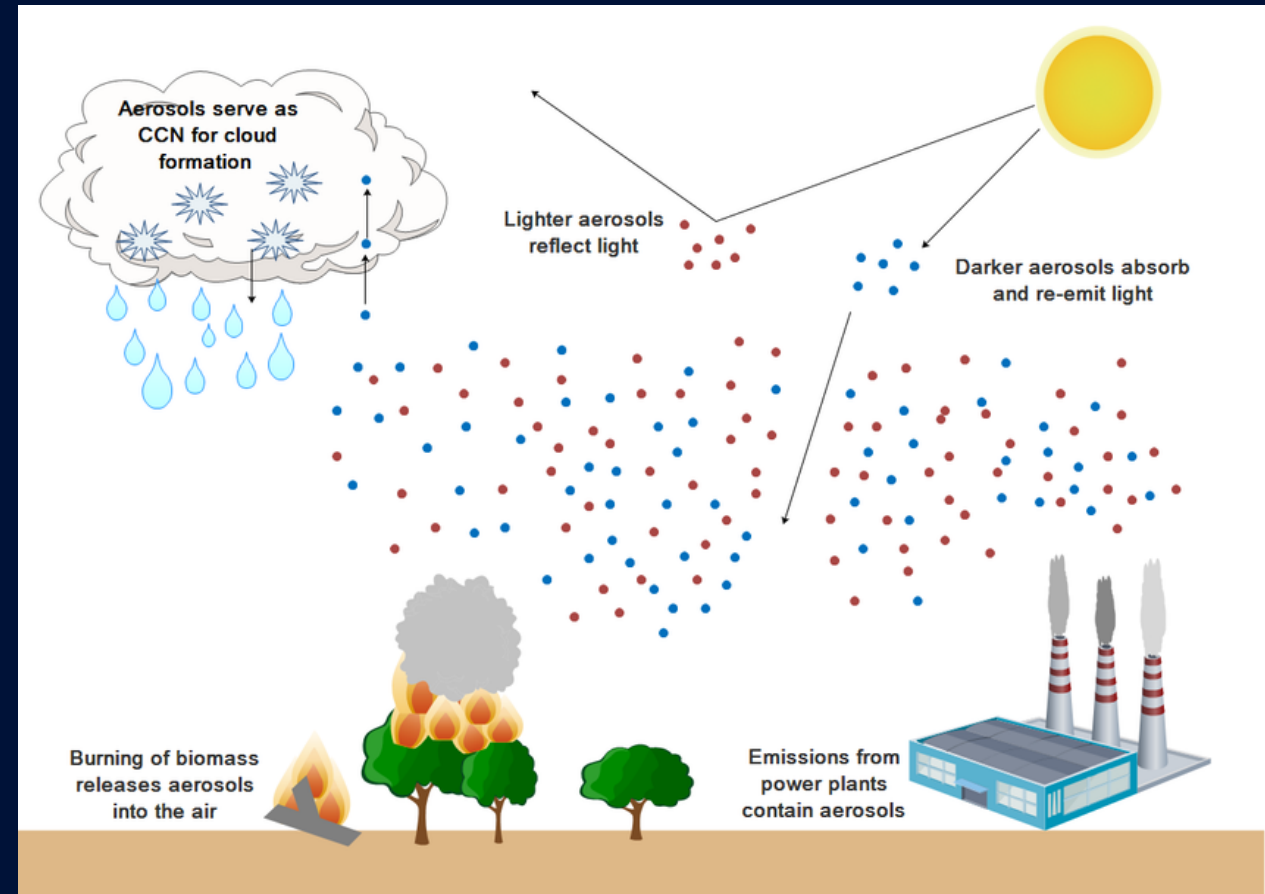
~ They also **absorb** and **reflect** incoming solar and Earth's heat **energy**, and must be considered in the **energy balance** of the planet.



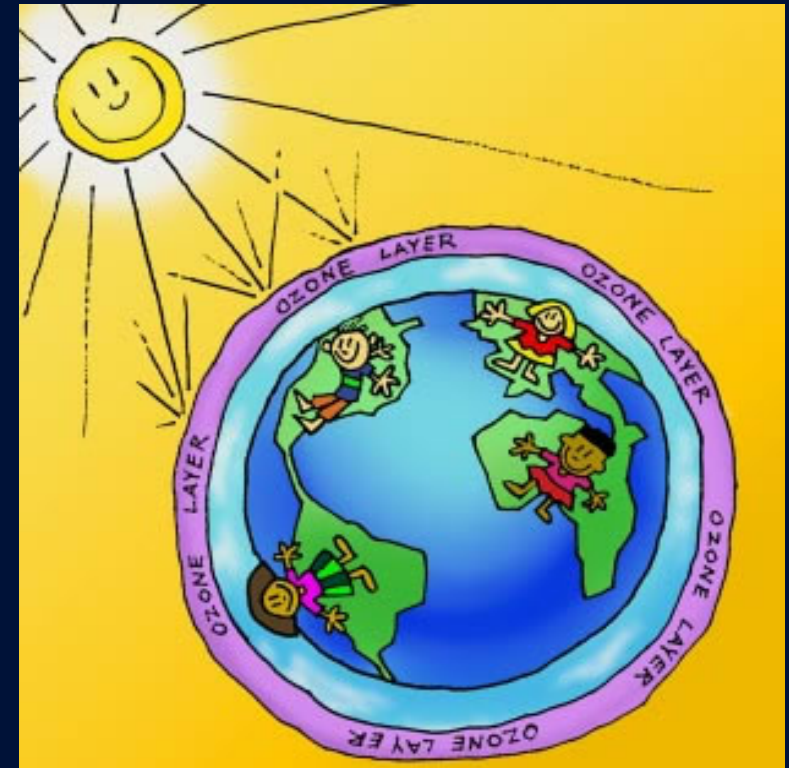


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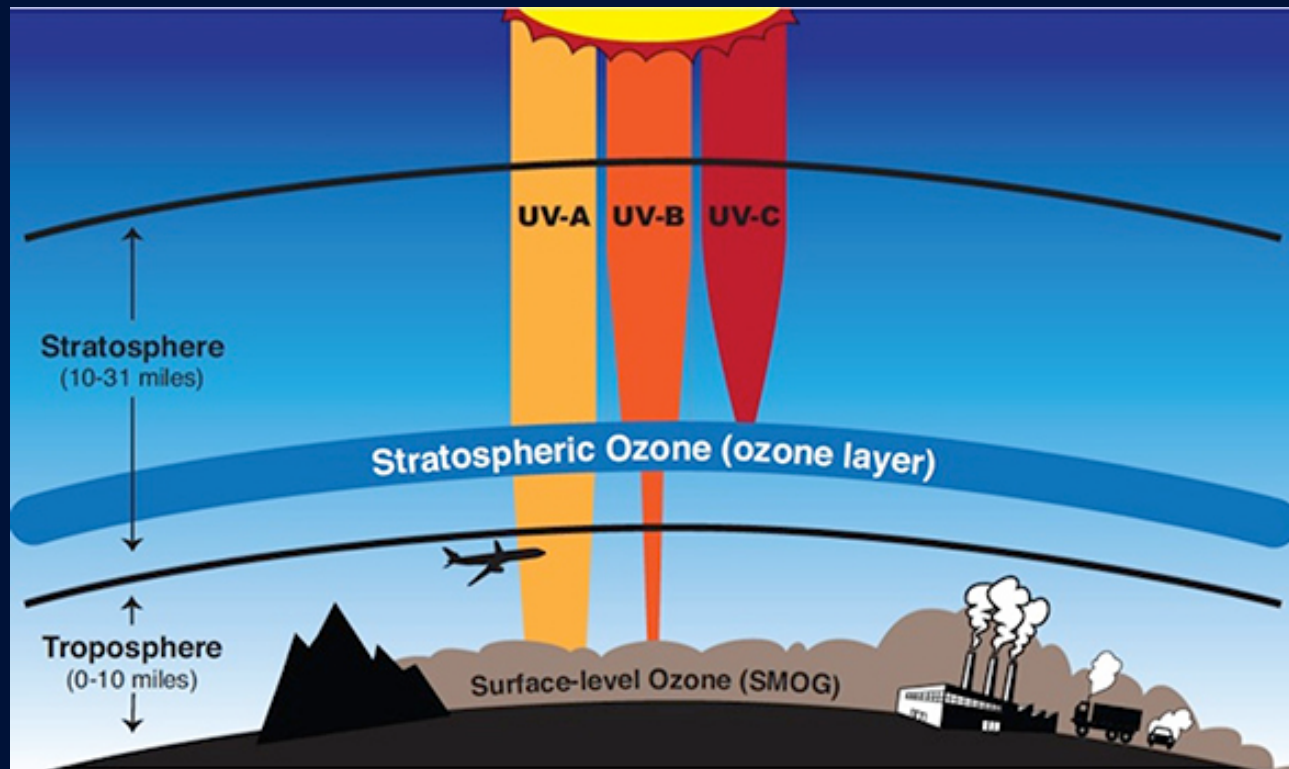
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~ Finally, an important gas for meteorology and climate is **ozone**.

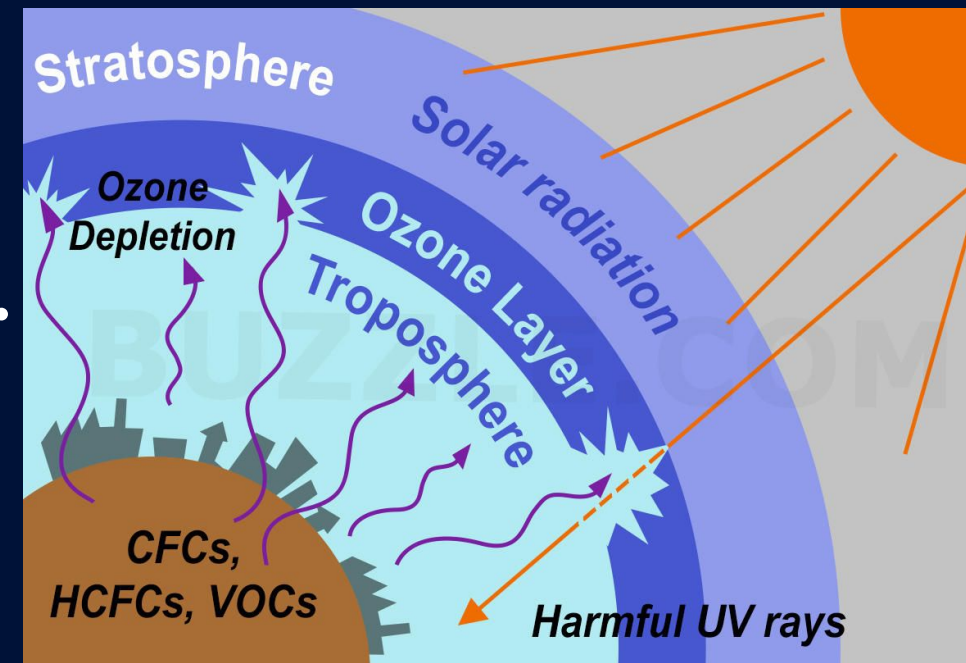


- ~ Finally, an important gas for meteorology and climate is **ozone**.
- ~ The **ozone ( $O_3$ ) layer** (20–30 km above the Earth) is **crucial** to life on the planet, as it **absorbs** harmful **ultraviolet (UV) radiation** from the Sun.



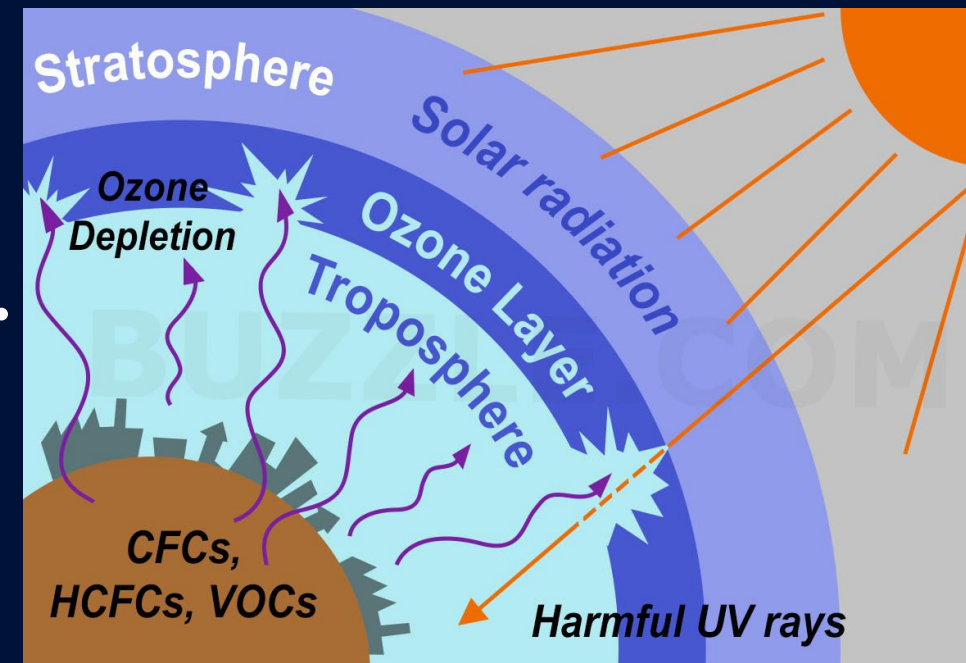
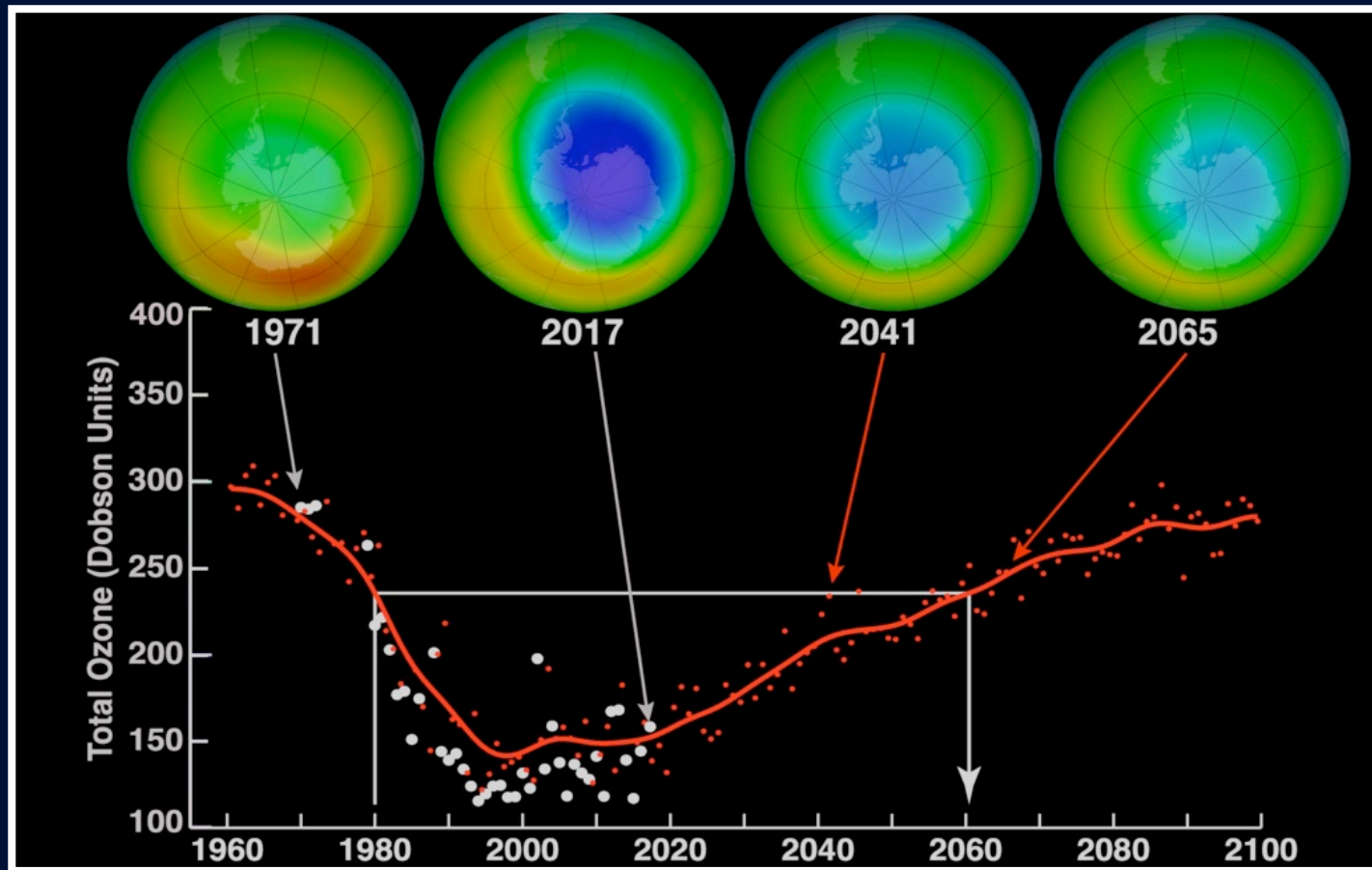
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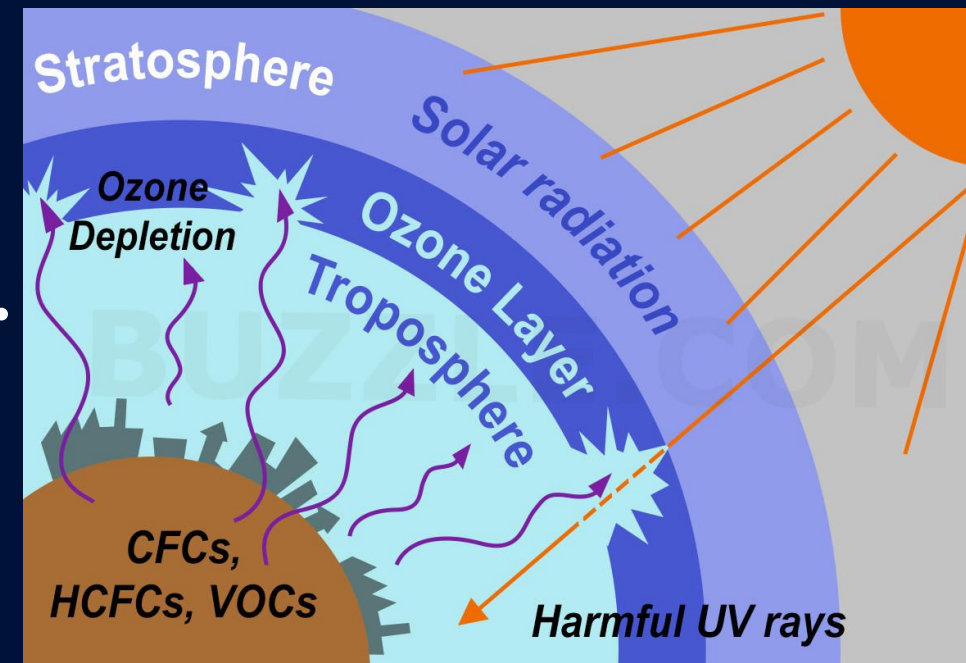
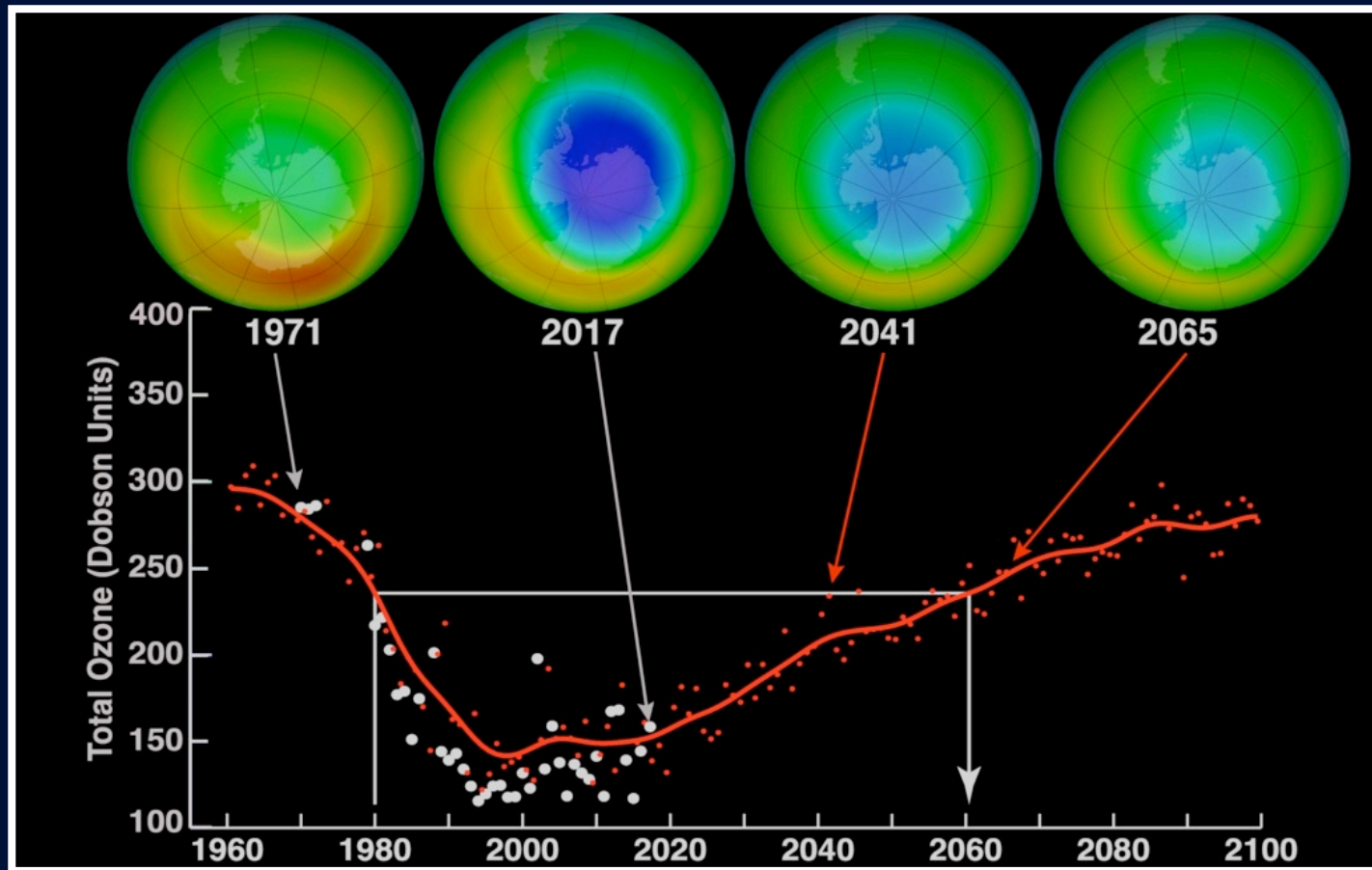




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~ Scientists discovered the problem and international **laws** were **established** to **control** the **pollutants** that destroy  $O_3$ .

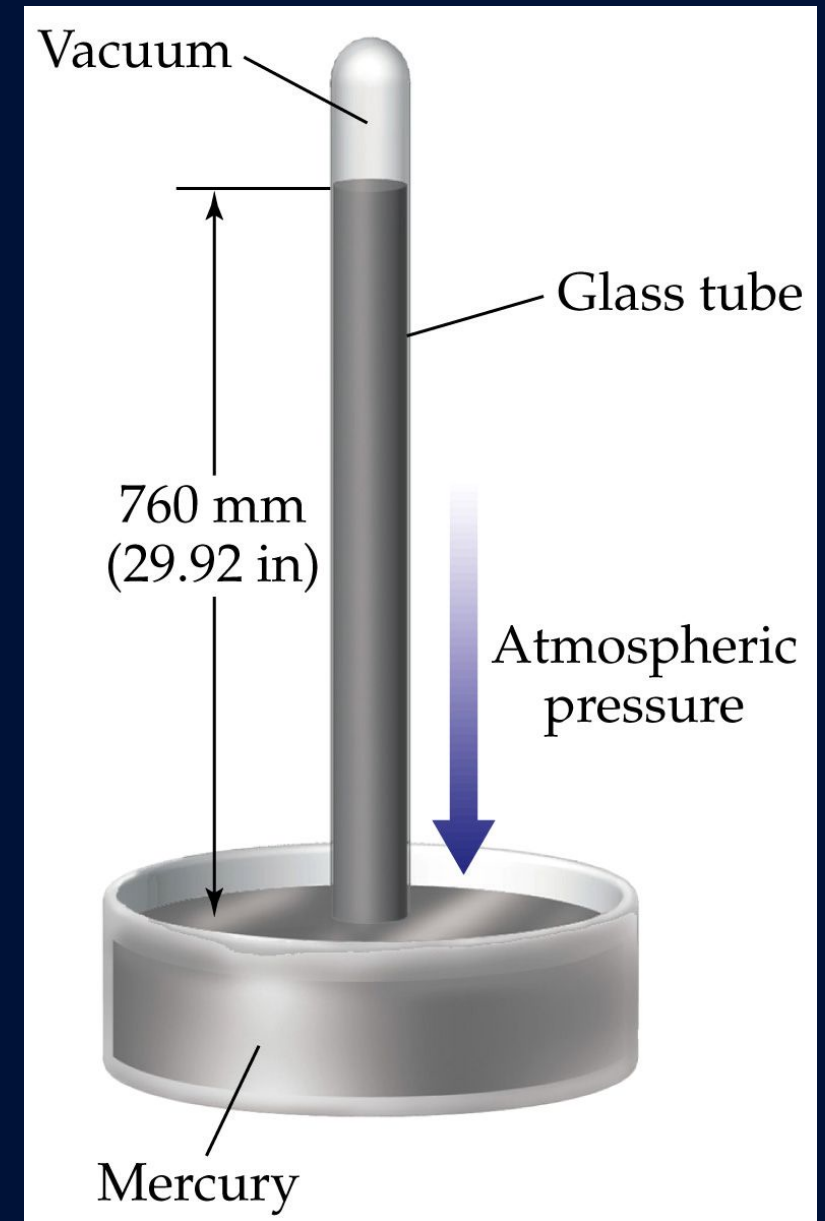
# Vertical extent and structure of the atmosphere

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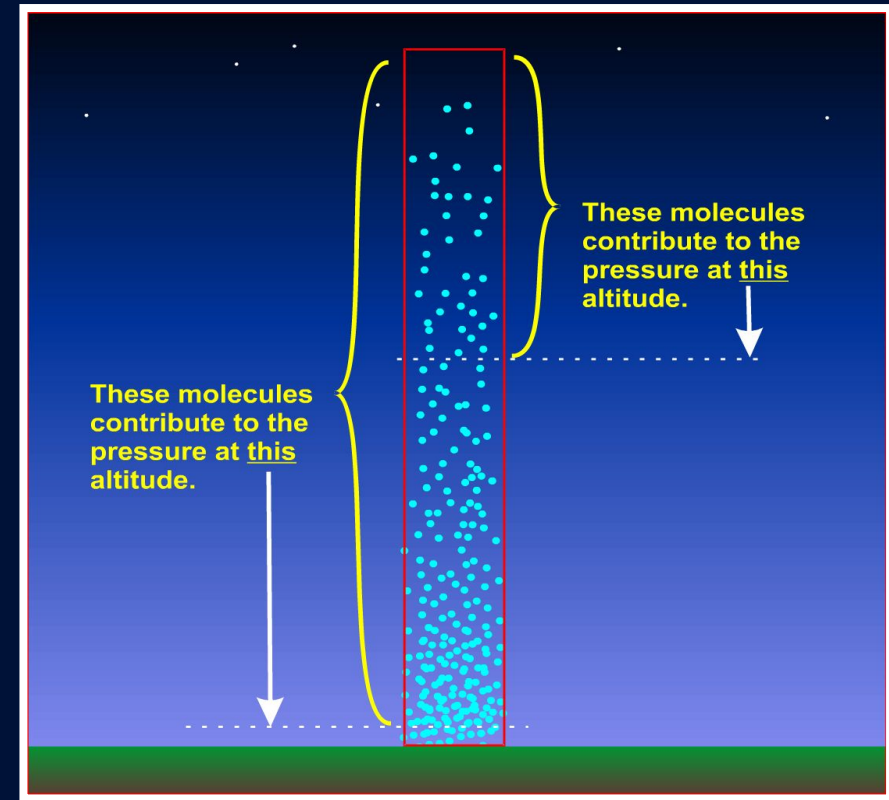
- ~ There is *no* distinct **demarcation** between our atmosphere and space, but a **rapid thinning** of the atmosphere where air **molecules** become **too few to detect**.
- ~ One way to explore the extent of the atmosphere is through examining **how** atmospheric **pressure changes with height**.



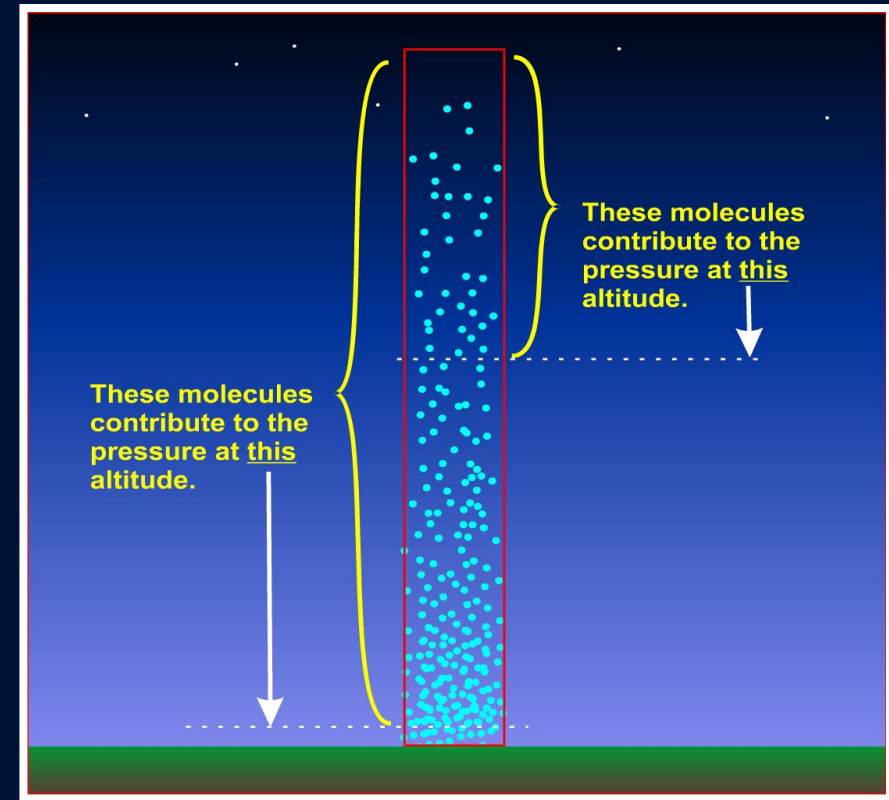
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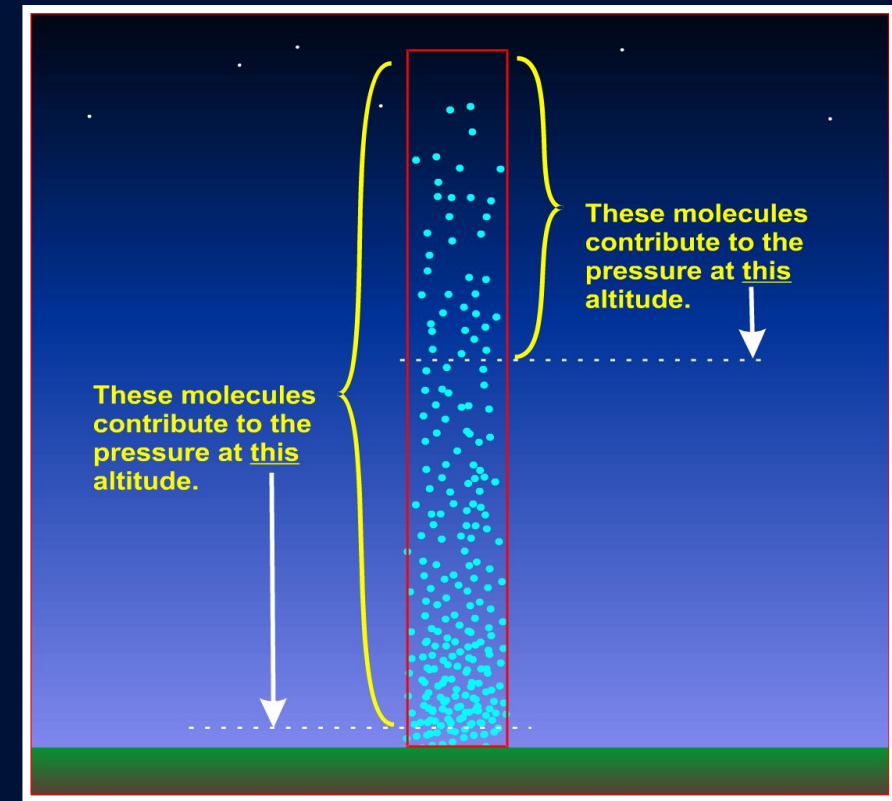
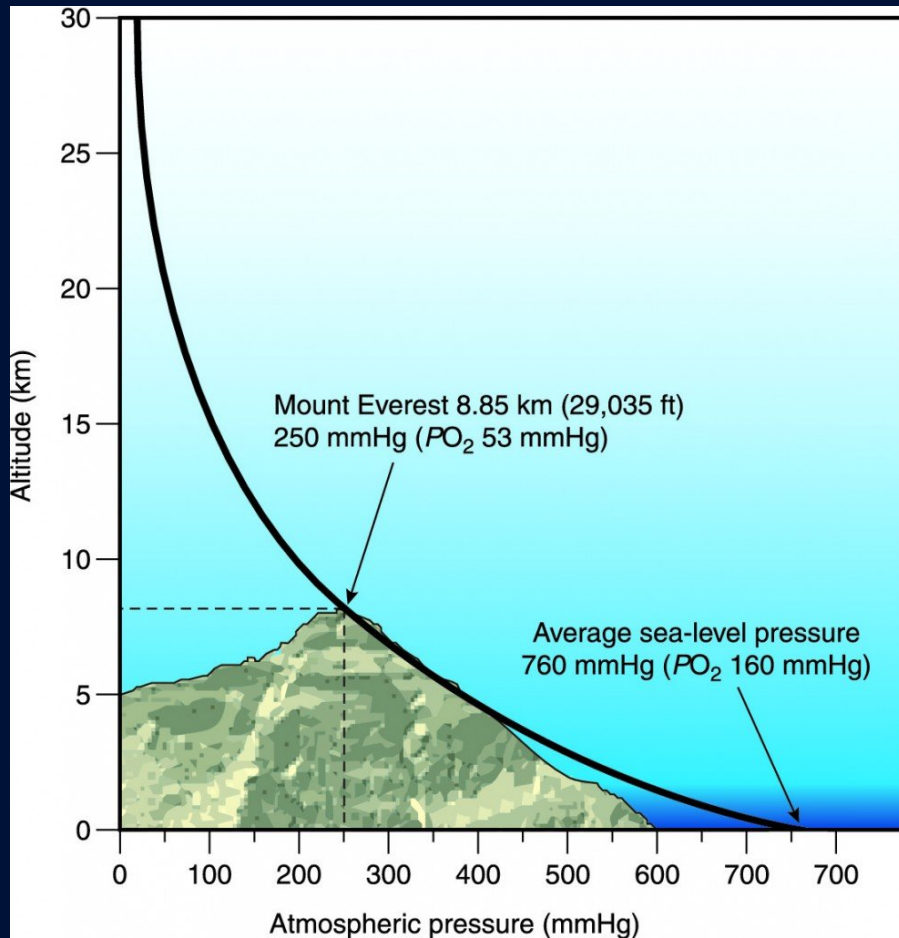


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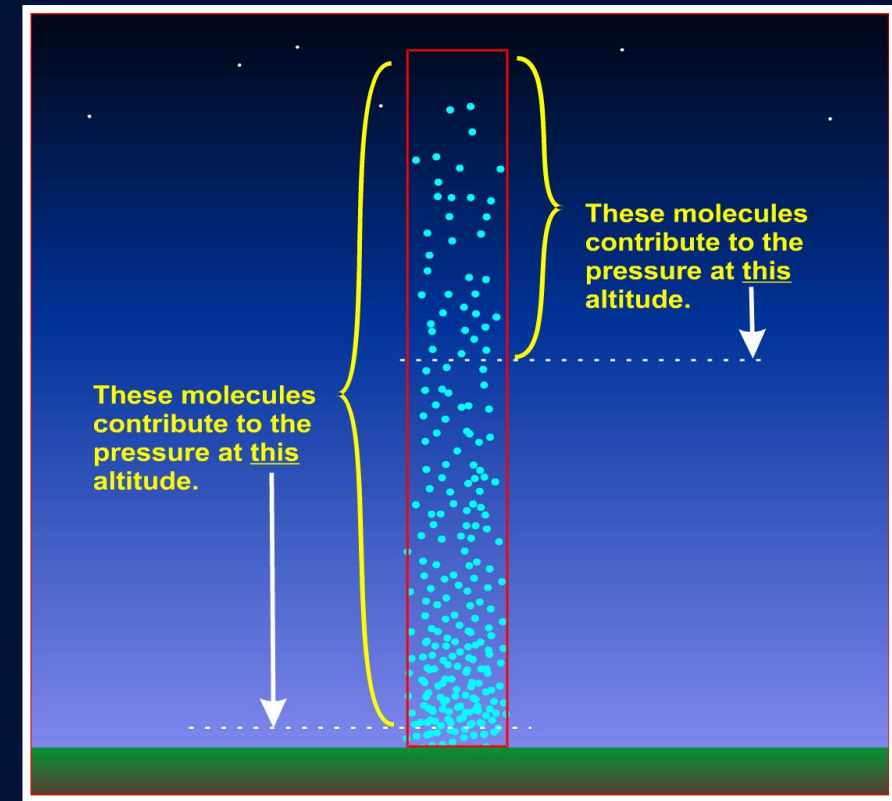
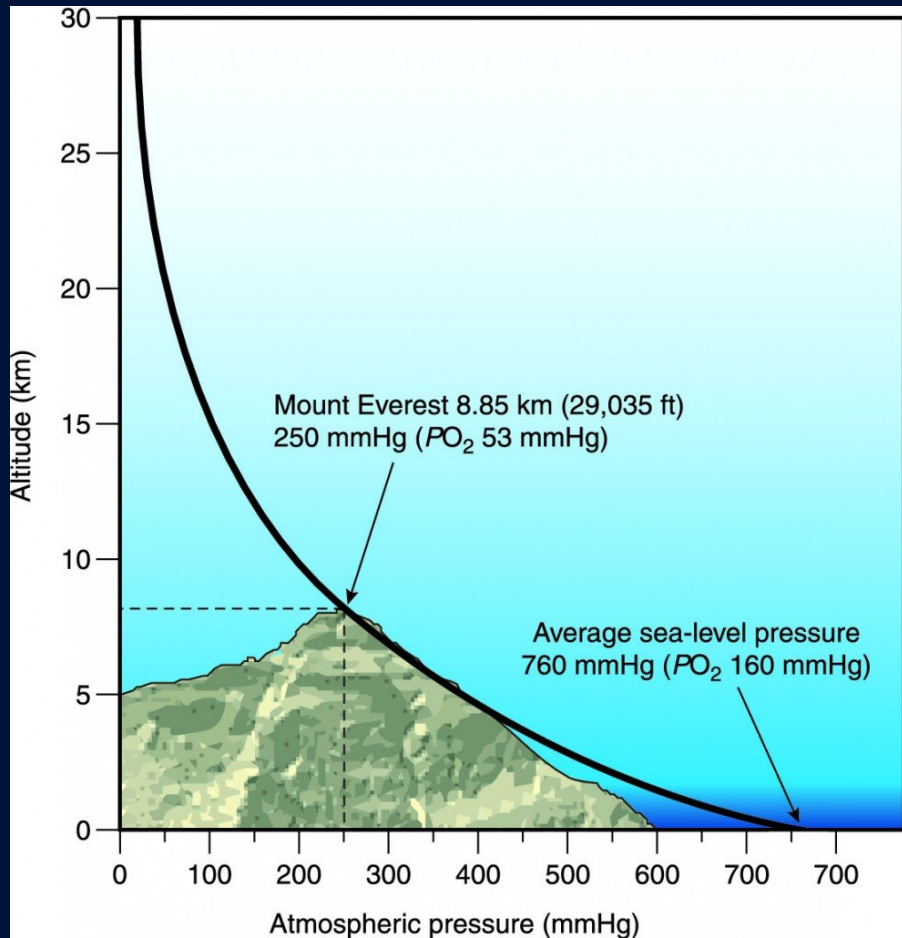


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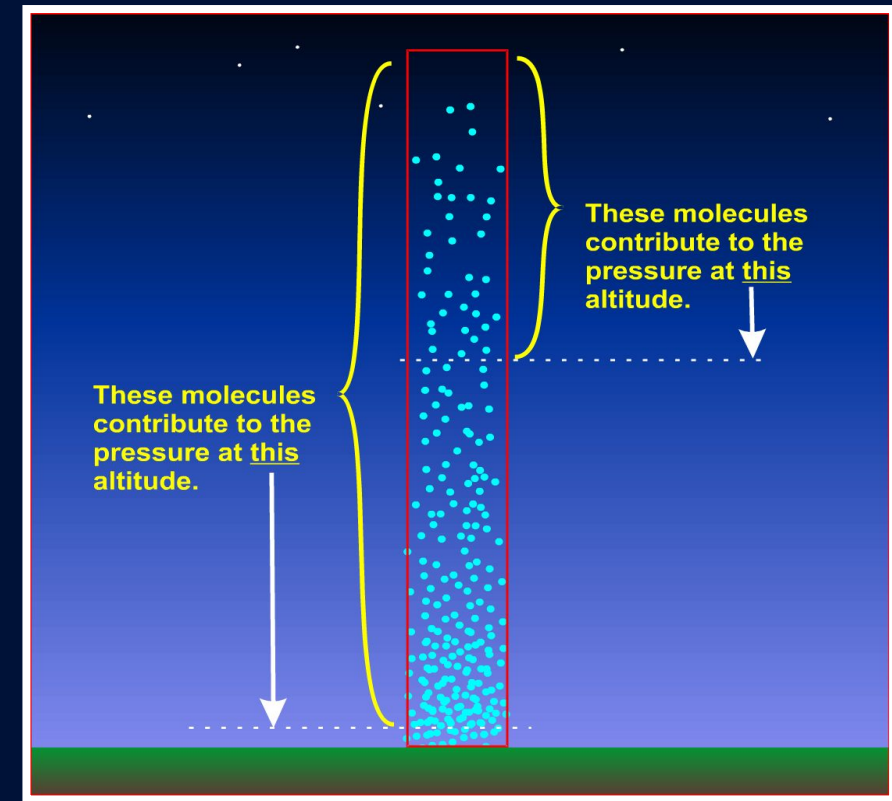
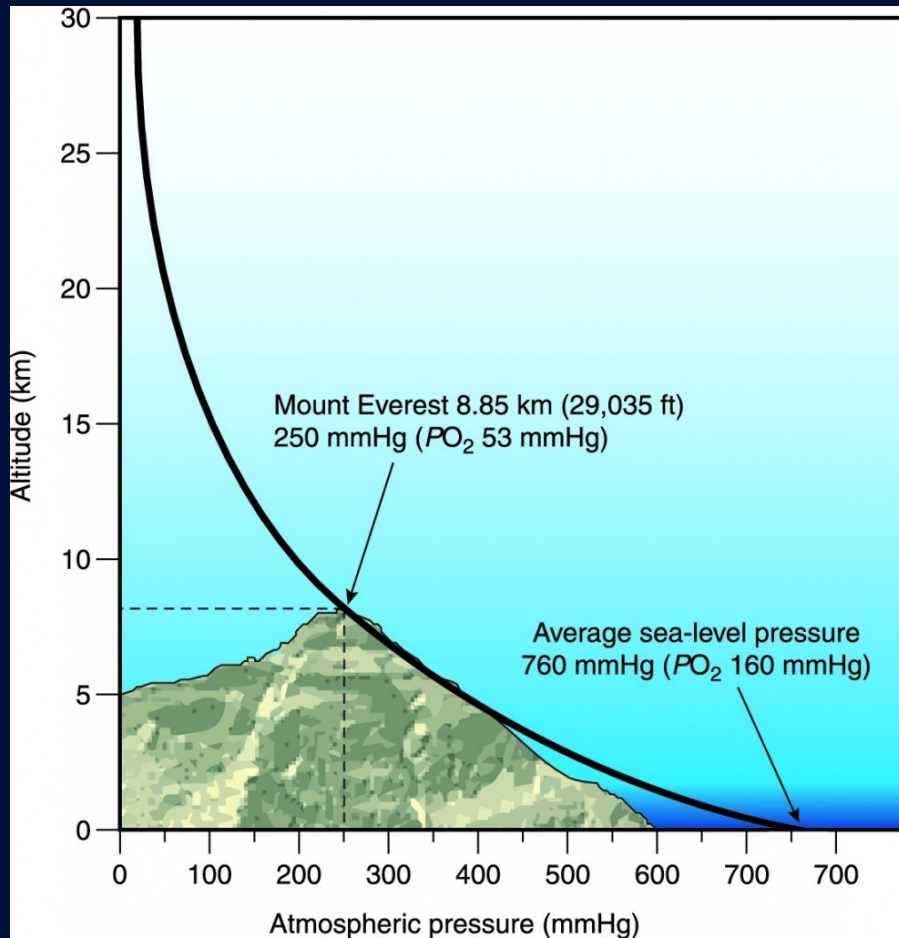


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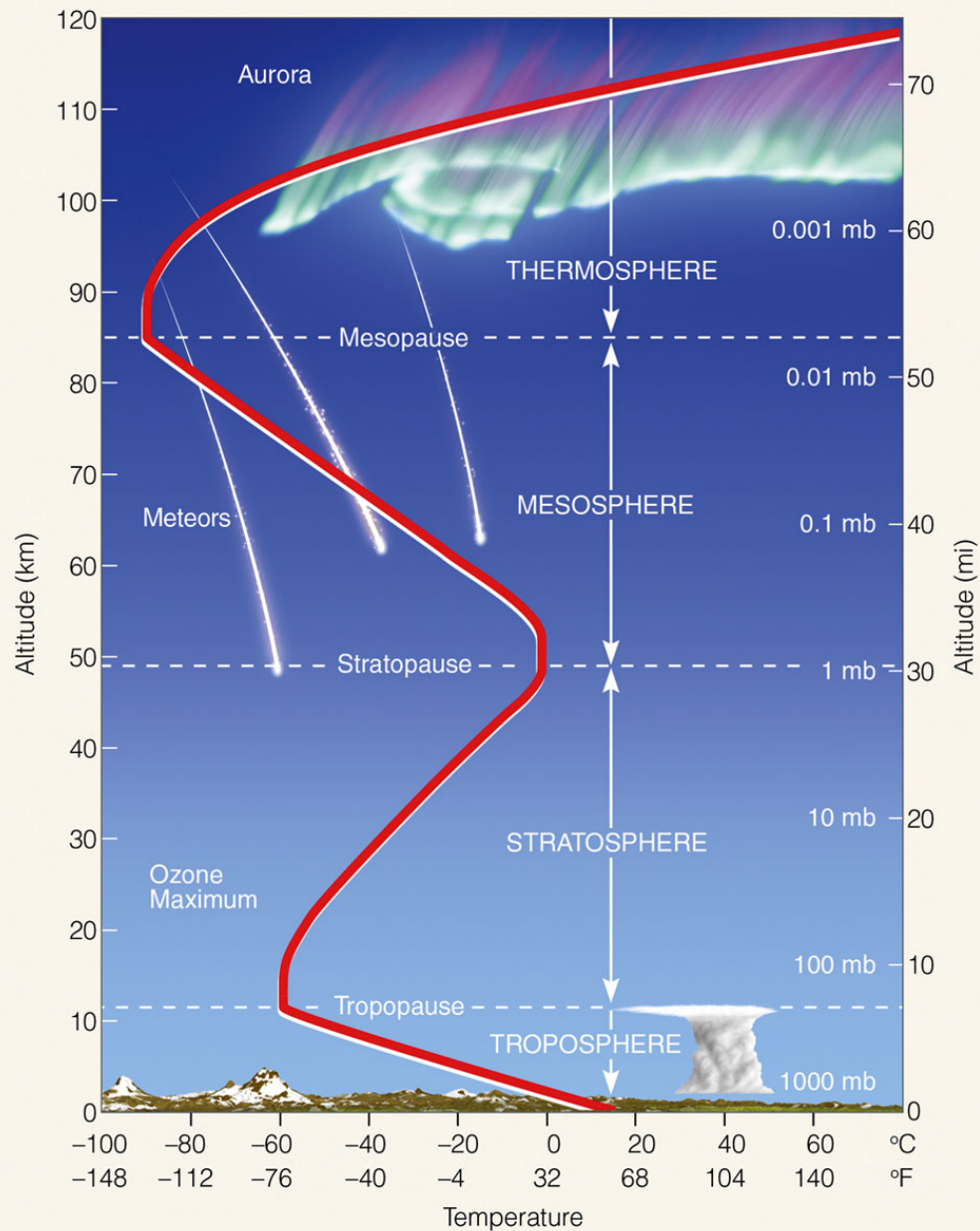
~ This compressibility means **50%** of the *air* molecules in the atmosphere lie **below 6 km** and **90% below 16 km**.



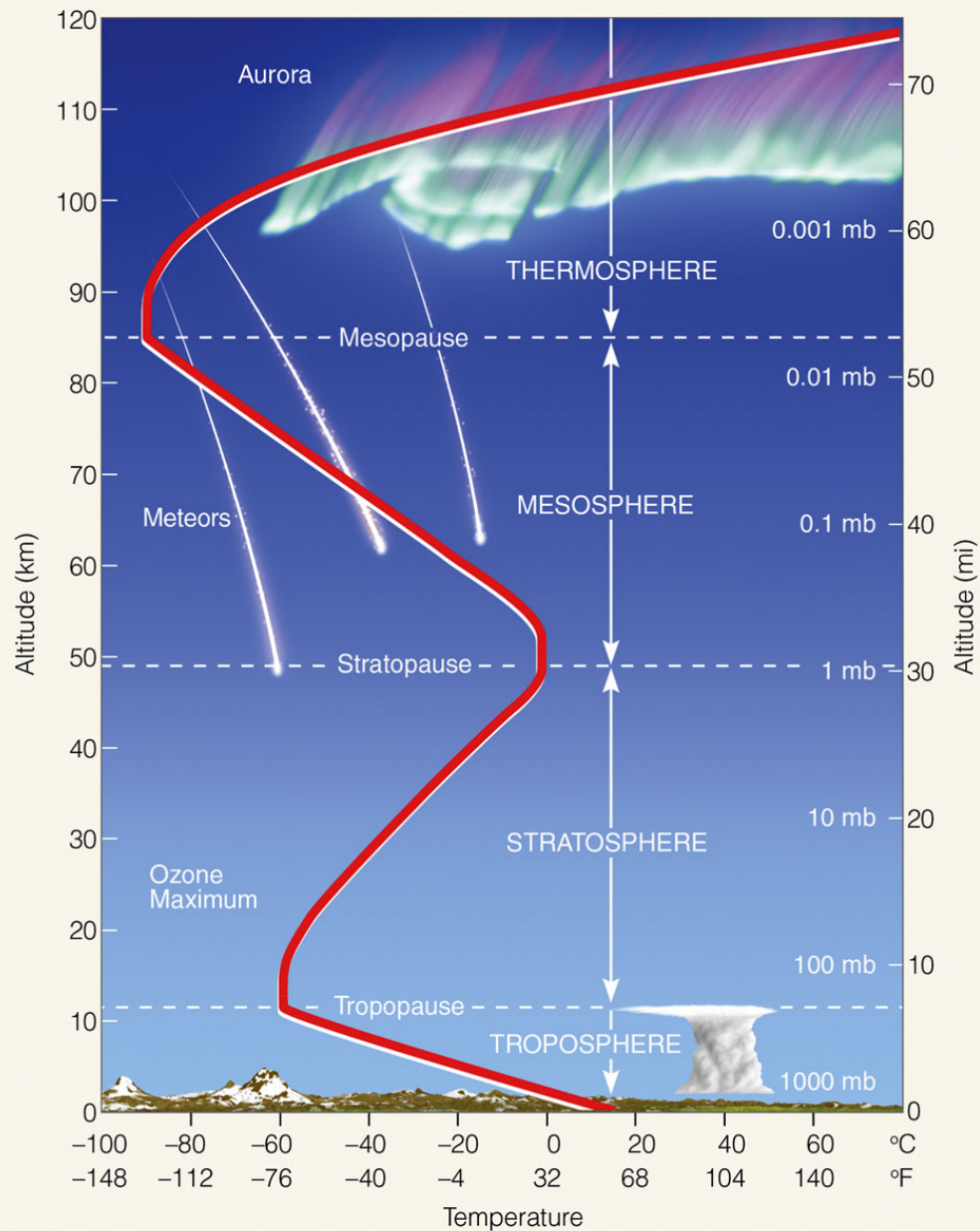
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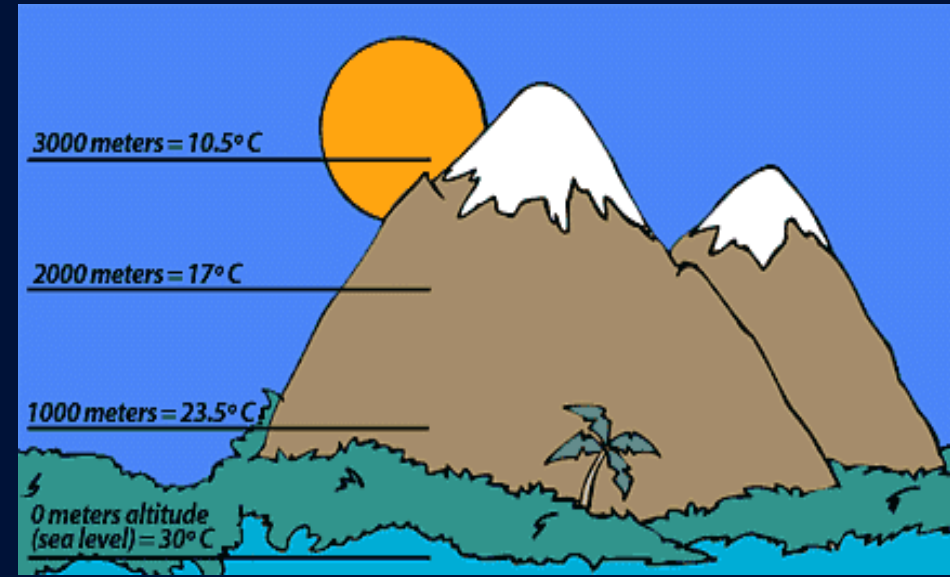
~ Studies reveal *four* distinct *thermal layers*, the bottom of which, where *temperature decreases* with height, is called the *troposphere*.

~ This ***decrease*** in ***temperature*** with ***height*** is called a ***lapse rate***.



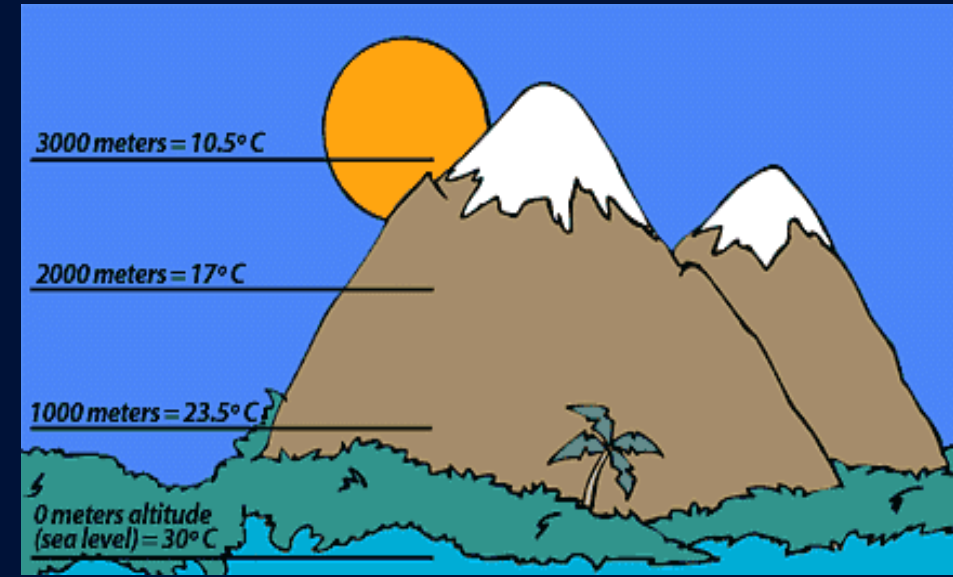
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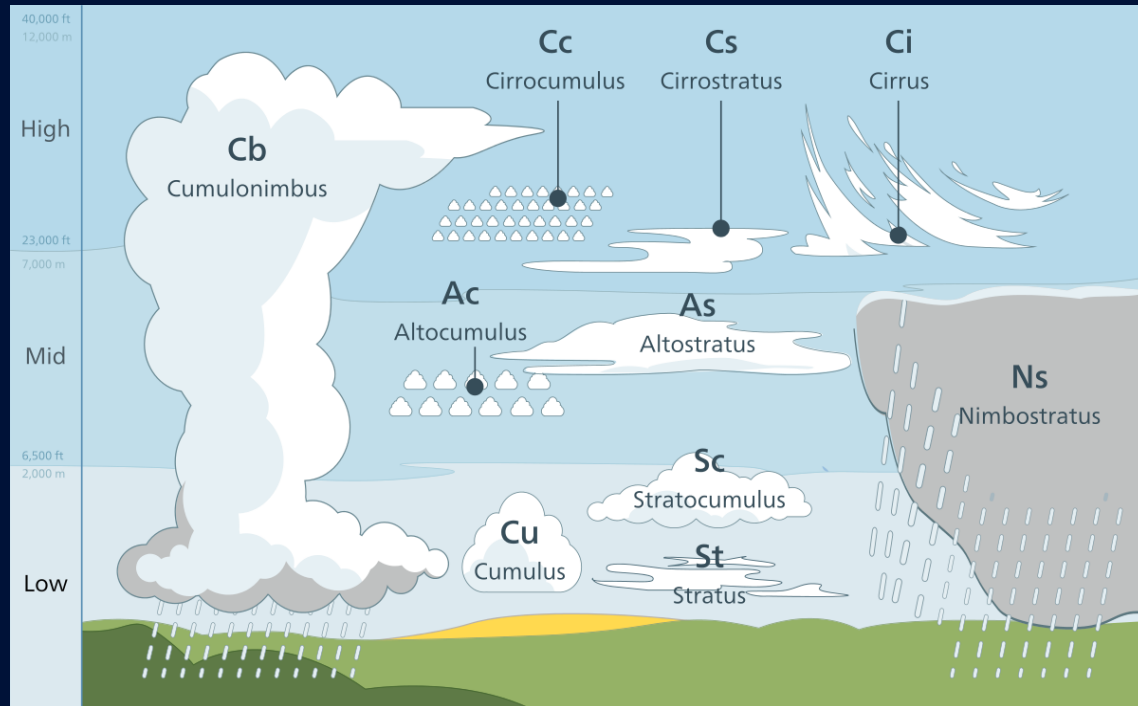
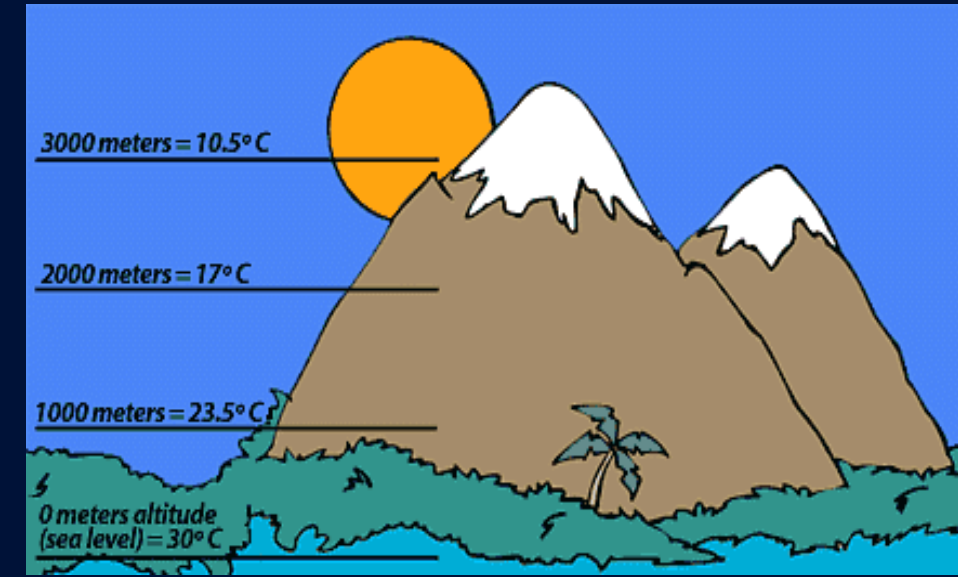
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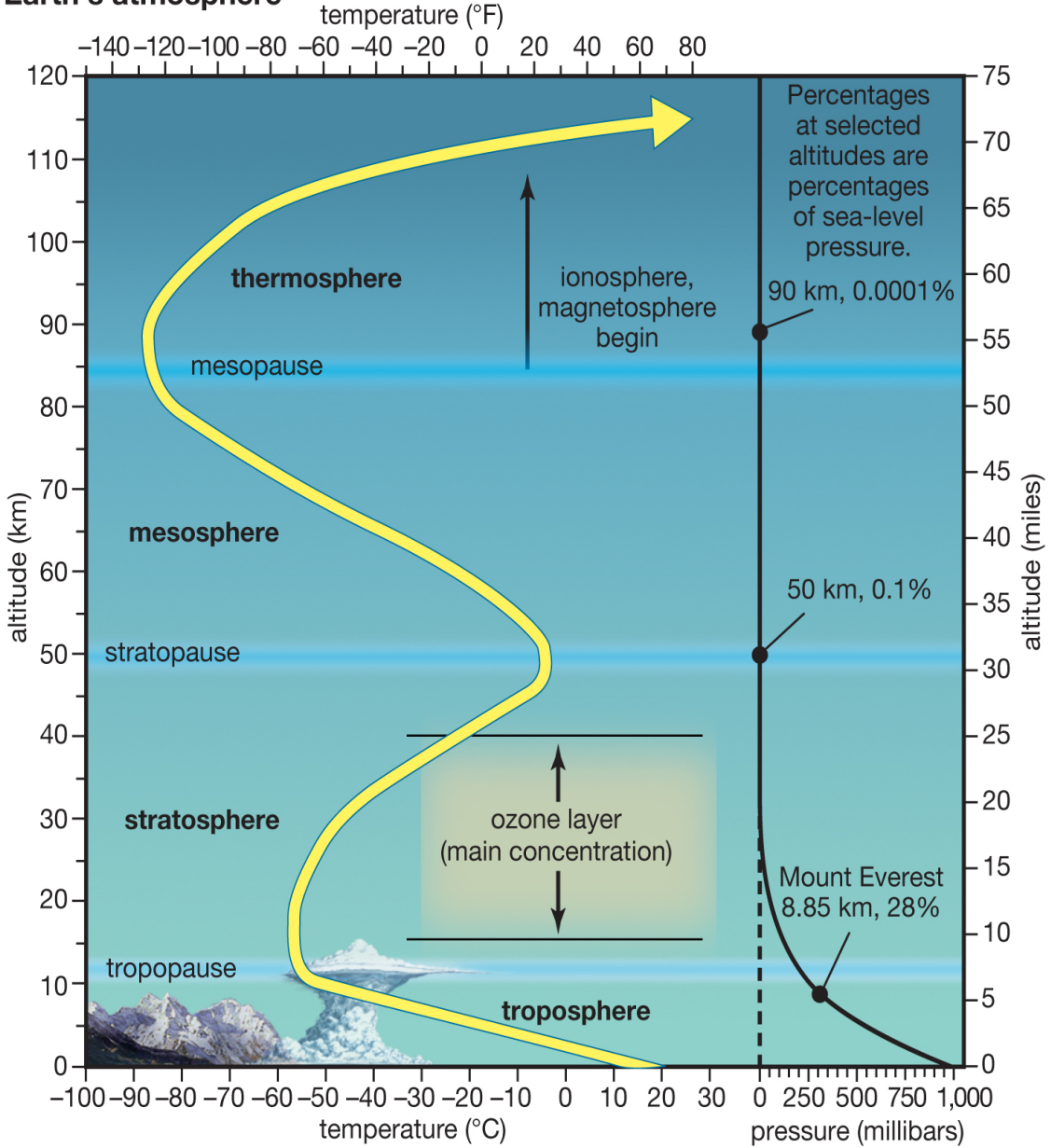
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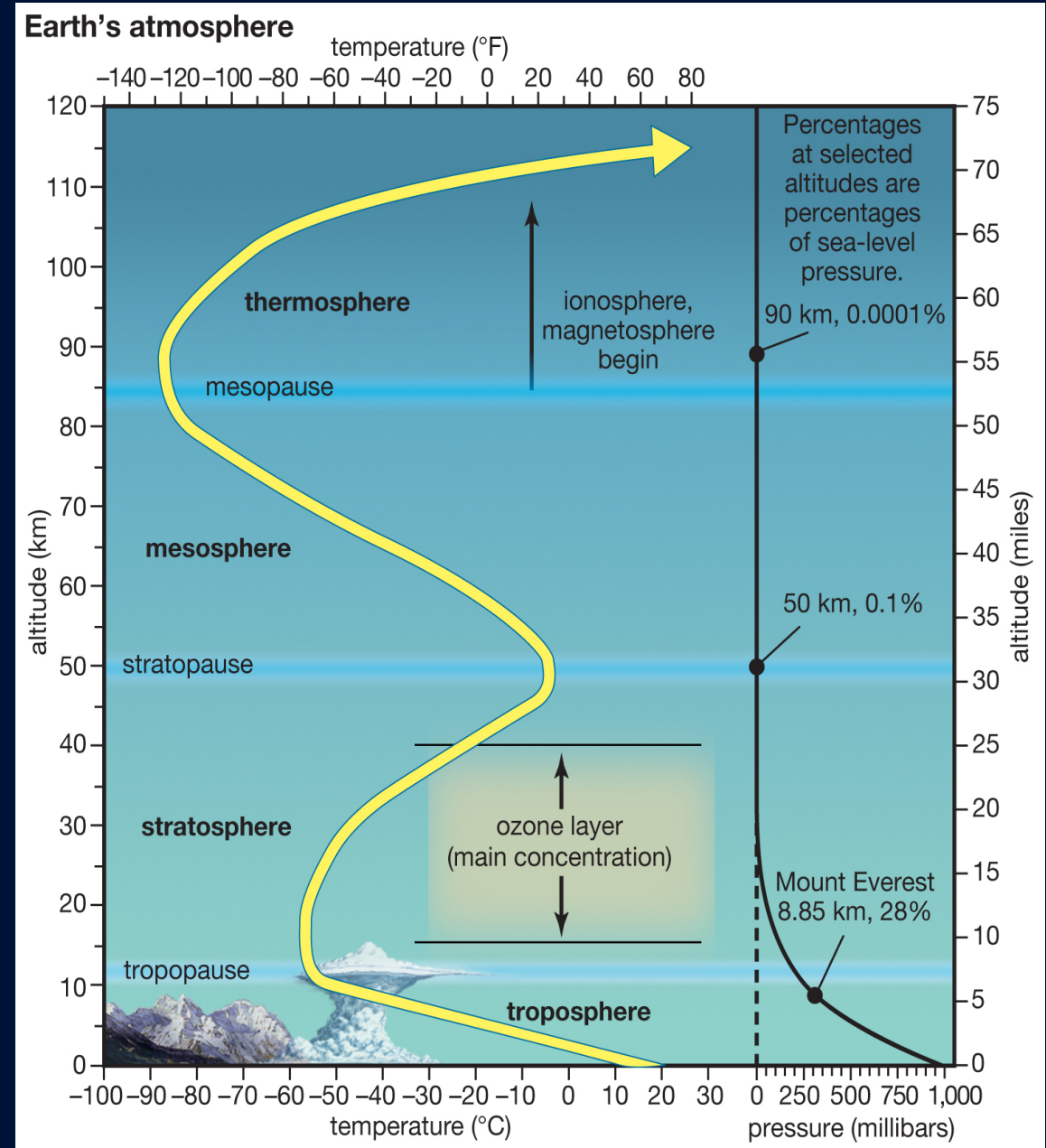
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Earth's atmosphere

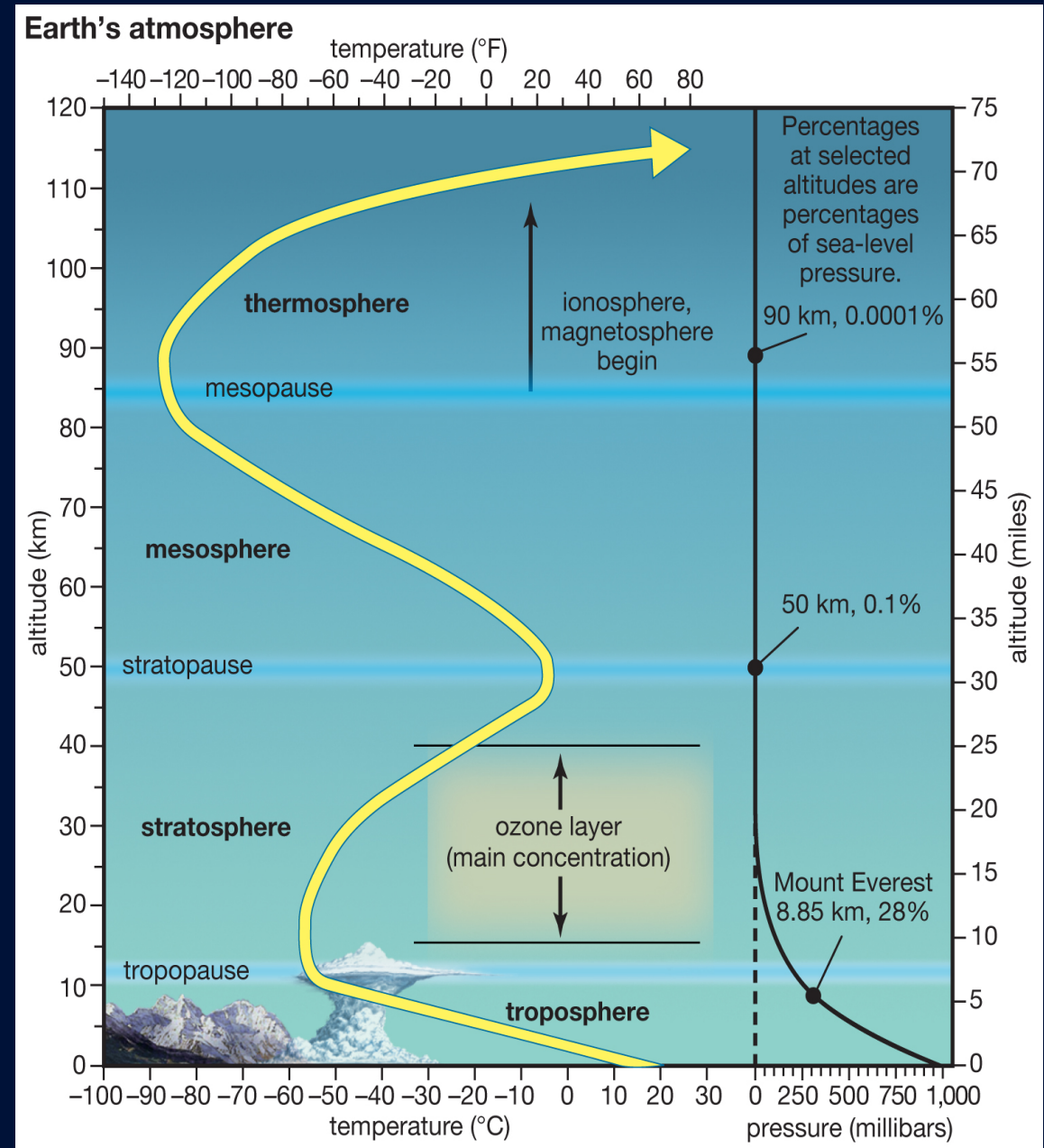


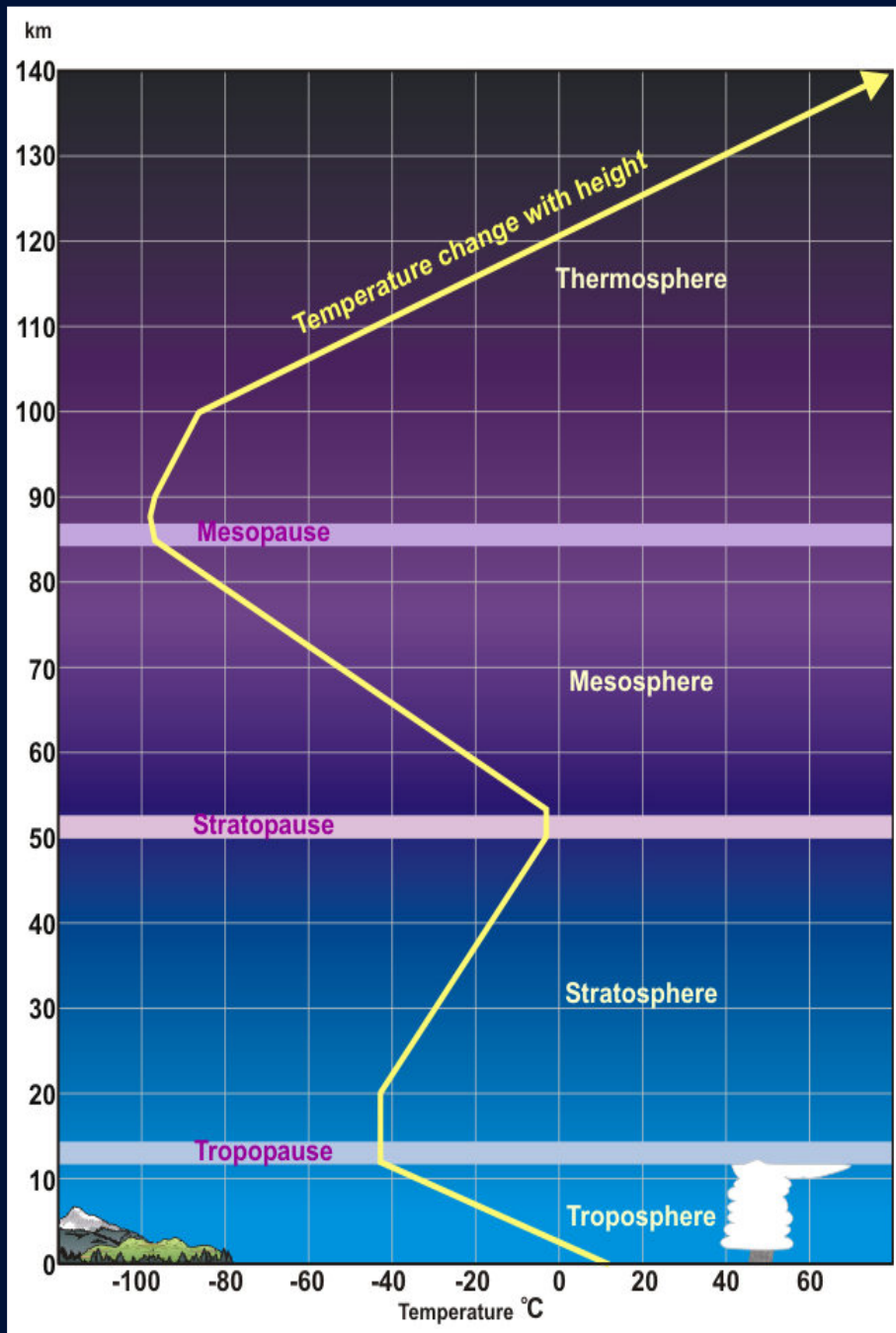
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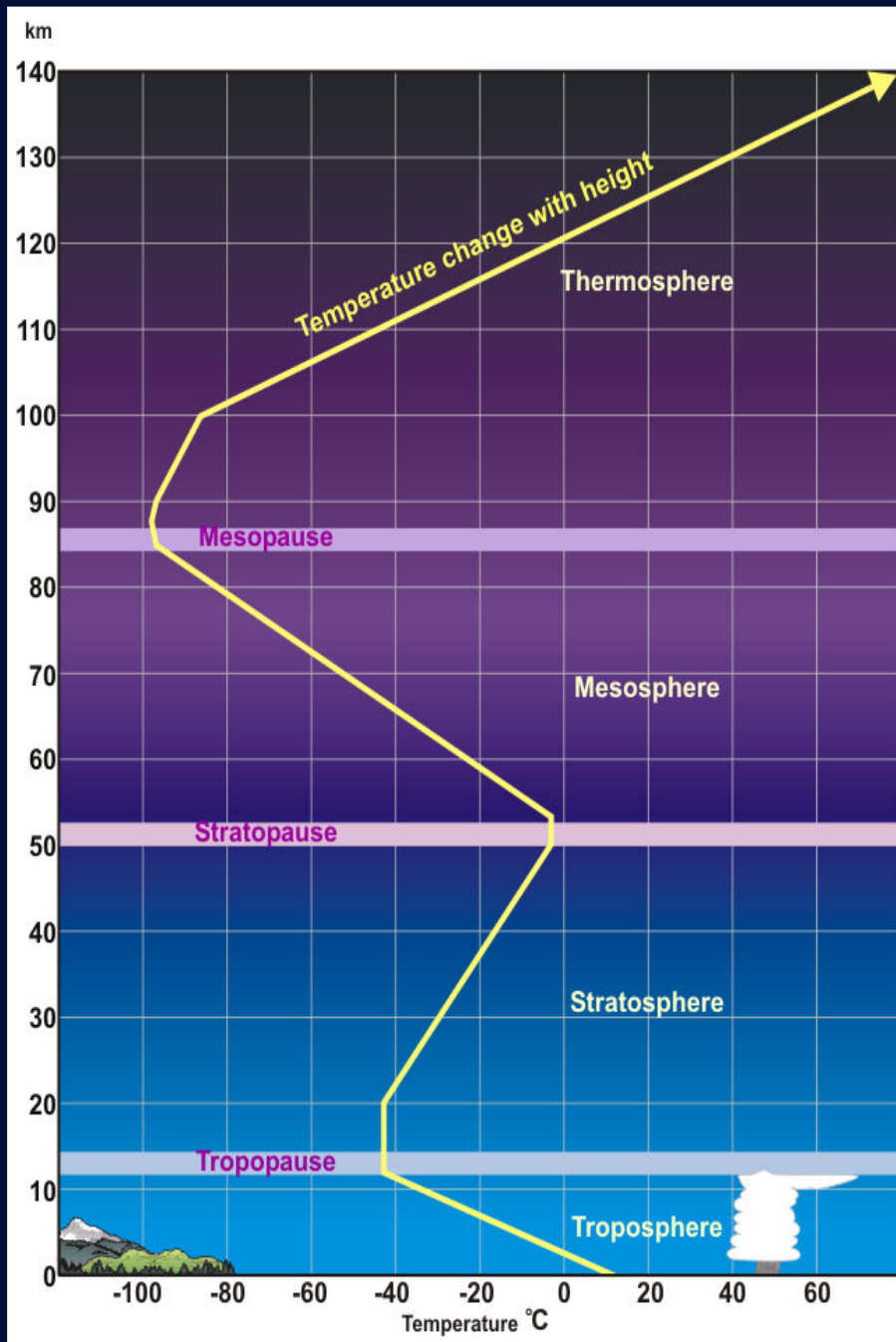
~ **Temperatures increase** with height in the **stratosphere**, located 12–50 km above the Earth, due to **ozone** which absorbs UV light from the Sun.

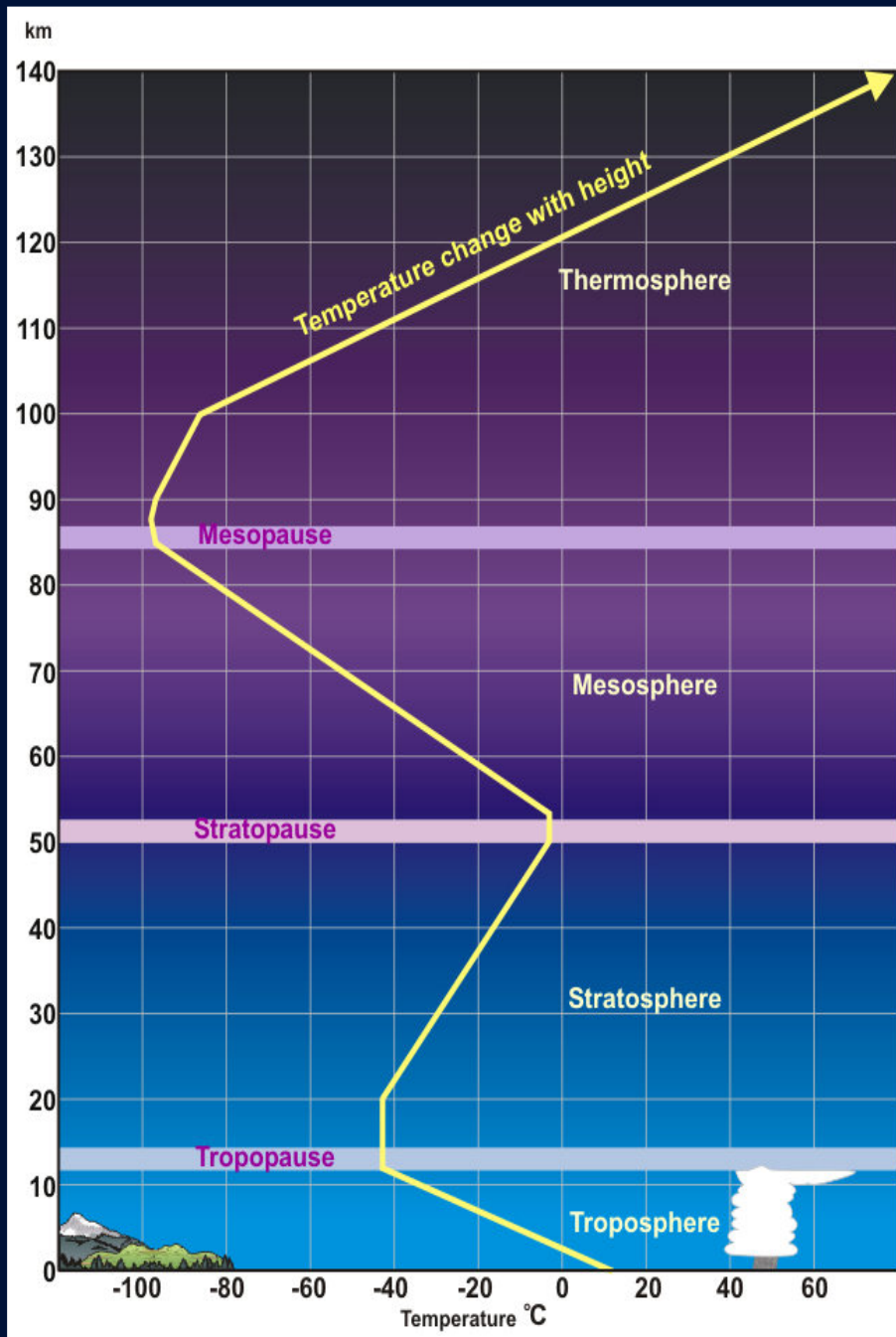






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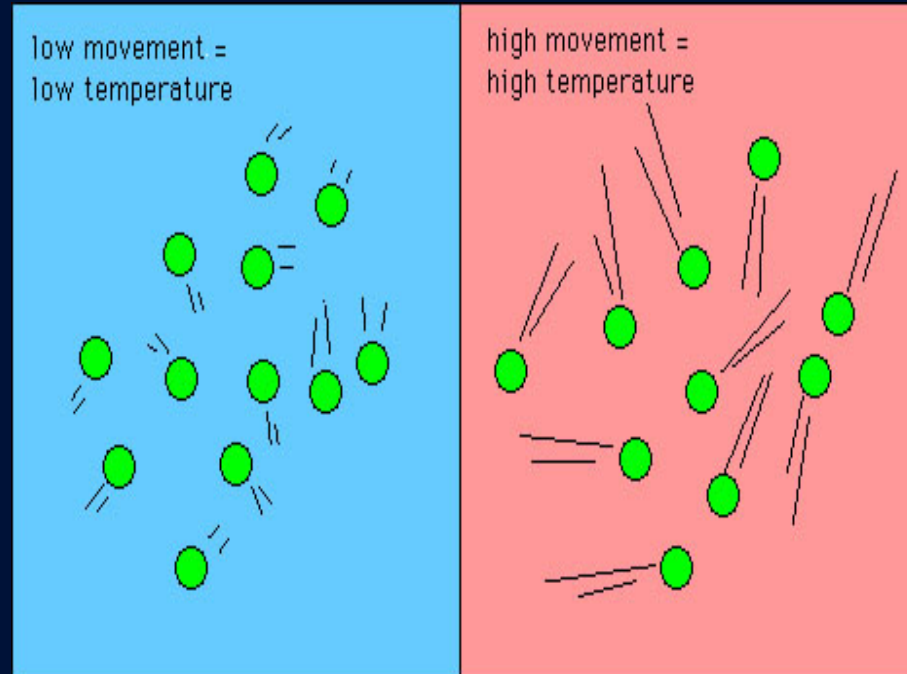
~ *At a height of ~85 km, **temperatures** stop decreasing at the **mesopause**, and begin **increasing** as we enter the **thermosphere**, which extends far **into space**.*

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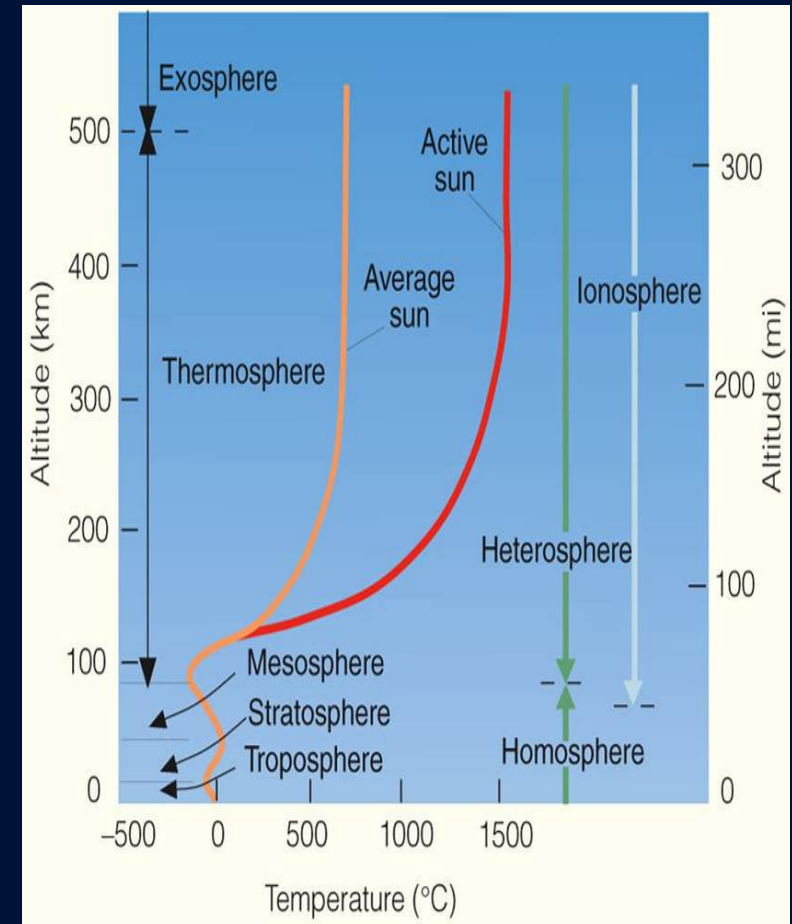


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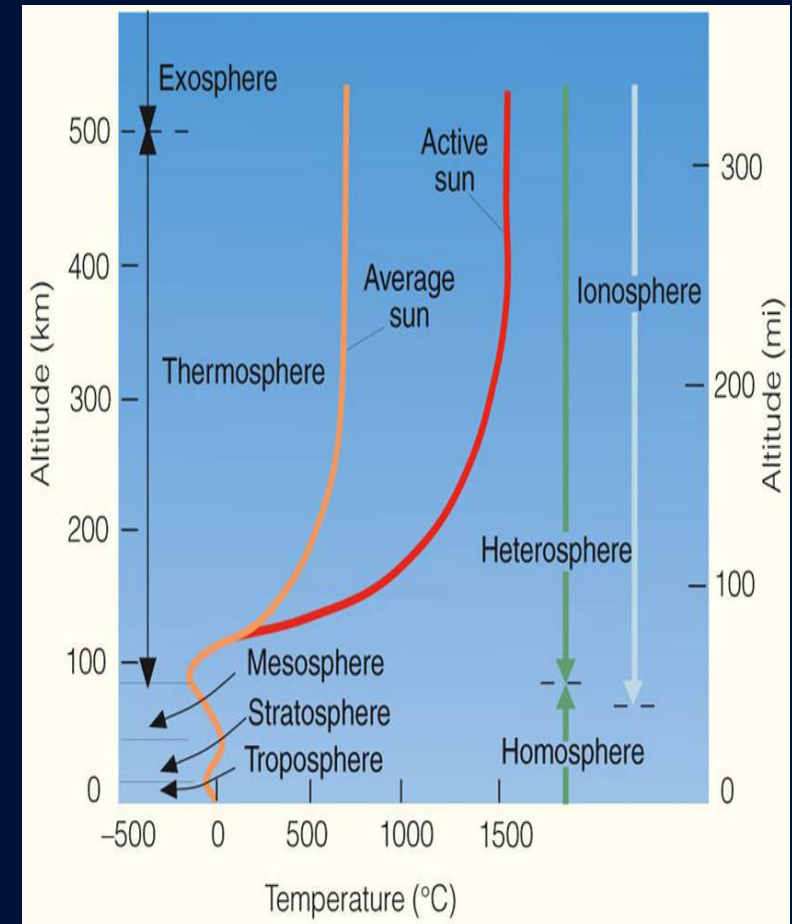
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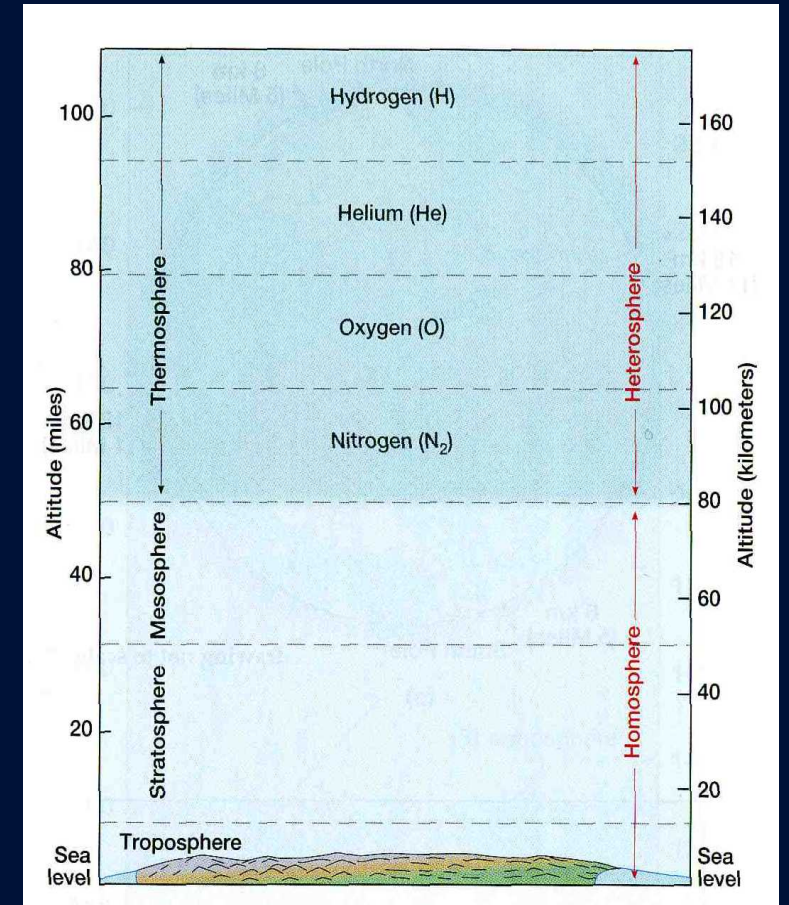
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~ In the *heterosphere*, the gases are stratified into *layers*, with *nitrogen* being the *lowest* level, then *oxygen*, *helium*, and *hydrogen*; i.e., from the *heaviest* to the *lightest*.

~ The *heterosphere* is *electrically-charged* (the *ionosphere*), where gases take on a *positive charge* as they lose electrons, *absorbing* high-energy *radiation*.



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